

A new method for constructing pyrido[3',2':4,5]imidazo[1,2-*b*]pyridazine system and preparation of its derivatives

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General Information

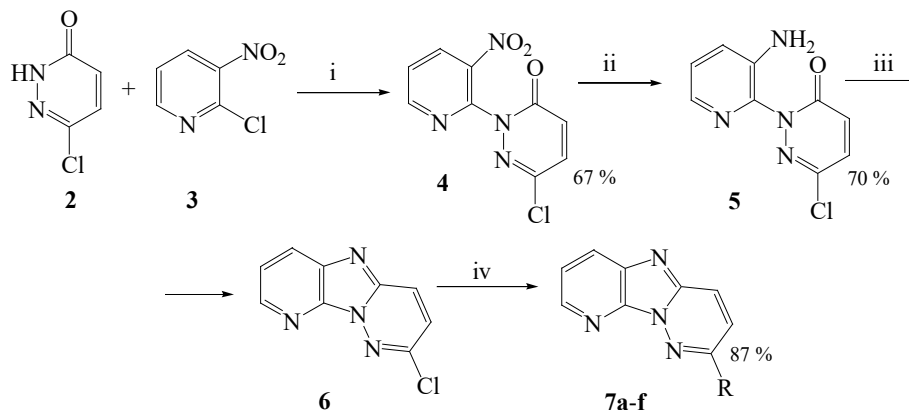
NMR spectra of new compounds **6**, **7c-f** were recorded on a spectrometer Bruker Avance 600 (600 MHz), of compounds **4**, **5**, **7a**, **7b** on a spectrometer Varian Unity-300 (300 MHz) in DMSO-*d*₆. Chemical shifts of nuclei ¹H and ¹³C were measured relatively the residual signals of deuterated solvent (see δ values in Ref. [S1]). Coupling constants (*J*) are reported in Hz. Melting points were determined by using Fisher-Johns Melting Point Apparatus (Fisher Scientific) and are uncorrected. Elemental analysis was performed using a EuroEA 3000 analyzer by burning a portion of the sample in an oxygen flow followed by chromatography of the combustion gases. ESI mass spectra for compounds **5,6** were registered on a Agilent 6470 device. The reaction and purity of the obtained compounds were monitored by TLC (plates with Al₂O₃ III activity grade, eluent CHCl₃, development of TLC plates by exposition to iodine vapors in iodine chamber).

Quantum-chemical calculations were performed using quantum chemical program packages Gaussian 09 [S2] and Firefly 8.0 [S3], which is partially based on the GAMESS (US) source code [S4] at the DFT level of theory (B3LYP/6-311++G^{**}).

Starting compounds **2** and **3** are commercially available from TCI Shanghai (China) and Chemieliva Pharmaceutical (China), respectively.

2. Synthesis and characterization of compounds 4-7.

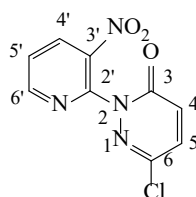
The general scheme:



7: R = MeO (**a**), PhS (**b**), H₂NNH (**c**), 4-Me-piperidinino (**d**), morpholino (**e**), 4-Ph-piperazino (**f**)

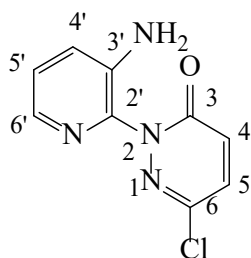
Reagents and conditions: i, K₂CO₃, 60°C, 3 h; ii, Fe, HCl, reflux, 2 h; iii, POCl₃, 100 °C, 40 min; iv, for **7a**: Na, MeOH, reflux, 1 h, for **7b**: Na, EtOH, PhSH, reflux, 10 min, for **7c**: N₂H₄·H₂O, EtOH, reflux, 1 h, for **7d**: 4-methylpiperidine, MeO(CH₂)₂OH, reflux, 4 h, for **7e**: morpholine, MeO(CH₂)₂OH, reflux, 1,5 h, for **7f**: 4-Ph-piperazine, tetramethylurea, reflux, 4 h

6-Chloro-2-(3-nitropyridin-2-yl)pyridazin-3(2H)-one (**4**).



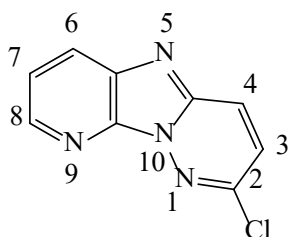
A mixture of 15.8 g (0.1 mol) of chloropyridazinone **2** and chloronitropyridine **3** with 15 g of K₂CO₃ in 50 ml of DMF was heated at 60°C for 1 h with stirring and kept at this temperature for 2 h. The mixture was poured into 250 ml of H₂O, and the precipitate of N-heteroaryl derivative **4** was filtered off and washed with water (3×20 ml). Yield of **4** 16.9 g (67%), mp 176-178°C (EtOH). IR spectrum (powder), ν , cm⁻¹: 1602 s (C=O), 1585 s (C=N), 1534 s (C=N). ¹H NMR (300 MHz), δ , ppm: 7.24 (d, *J* 9.8, 1H, H-5), 7.78 (d, *J* 9.9, 1H, H-4), 7.92 (dd, *J* 8.2, 4.7, 1H, H-5'), 8.70 (dd, *J* 8.1, 1.6, 1H, H-4'), 8.97 (dd, *J* 4.8, 1.6, 1H, H-6'). Found, %: C 42.71; H 1.92; Cl 14.18; N 22.12. C₉H₅ClN₄O₃. Calcd., %: C 42.79; H 2.00; Cl 14.03; N 22.18.

2-(3-Aminopyridin-2-yl)-6-chloropyridazin-3(2H)-one (5).



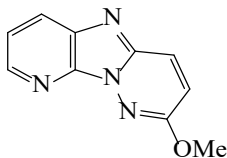
A suspension of 8 g of powdered Fe metal in 100 ml EtOH and 2 ml of conc. HCl was boiled with vigorous stirring for 1 h with portionwise addition of 2.53 g (10 mmol) of pyridazinone **4**, and then the mixture was boiled for another 2 h. The hot reaction mixture was filtered from excess of Fe, the filtrate was evaporated with a water bath to 20 ml volume, and 75 ml of water was added. The resulting crystals of amine **5** were filtered off and washed with water (3x5 ml). Yield 1.56 g (70%), mp 175-176°C (EtOAc : EtOH, 1:1). IR spectrum (powder), ν , cm^{-1} : 3408 m (NH_2), 3329 m (NH_2), 1660 s (C=O), 1577s (C=N). ^1H NMR (300 MHz), δ , ppm: 5.51 (s, 2H, NH_2), 7.14 (d, J 9.7, 1H, H-5), 7.16-7.24 (m, 2H, H-4', H-5'), 7.62 (d, J 9.8, 1H, H-4), 7.67 (dd, J 4.8, 1.8, 1H, H-6'). HRMS (ESI): m/z $[\text{M}+\text{H}]^+$ calcd for $\text{C}_9\text{H}_7\text{ClN}_4\text{O}$ 223.63; found: 223.07.

2-Chloropyrido[3',2':4,5]imidazo[1,2-*b*]pyridazine (6).



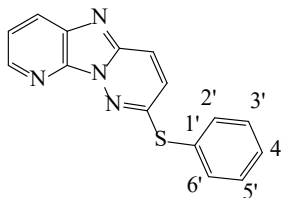
A mixture of 2.05 g (0.01 mmol) of pyridazinone **5** and 12 ml of POCl_3 was kept for 40 min at 100°C, then cooled, carefully poured into 120 ml of H_2O and after decomposition of excess of POCl_3 at room temperature, it was neutralized with a solution of NH_4OH to pH 7-8. The resulting precipitate of chloro derivative **6** was filtered off and washed with water (5x3 ml). Yield 1.78 g (87%), mp. 242-244°C (EtOAc : EtOH, 1:1). IR spectrum (powder), ν , cm^{-1} : 1577 w (C=N), 1523 m (C=N), 1468 m (C=N). ^1H NMR (600 MHz), δ , ppm: 7.66 (1 H, dd, J 8.3, 4.5, 7-H), 7.71 (1 H, d, J 9.7, 3-H), 8.36 (1 H, d, J 9.7, 4-H), 8.39 (1 H, dd, J 8.3, 1.4, 6-H), 8.65 (1 H, dd, J 4.5, 1.4, 8-H). ^{13}C NMR (600 MHz), δ , ppm: 122.8, 126.2, 129.3, 129.6, 135.8, 141.9, 142.2, 145.6, 146.3. Mass-spectrum (ESI): m/z $[\text{M}+\text{H}]^+$: calcd for $\text{C}_9\text{H}_5\text{ClN}_4$ 205.62; found: 205.08.

2-Methoxypyrido[3',2':4,5]imidazo[1,2-*b*]pyridazine (7a).



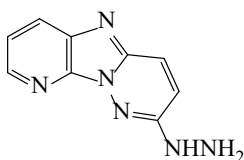
To a solution of 0.23 g (0.01 mol) Na in 10 ml MeOH, 2.05 g (0.01 mol) of chloropyridoimidazopyridazine **6** was added, and this was boiled for 1 h. The reaction mixture was filtered, and the filtrate was evaporated to dryness in vacuum. Yield 1.88 g (94%), m.p. 154-155°C (EtOAc). IR spectrum (powder), ν , cm^{-1} : 1576 m (C=N), 1496 m (C=N), 1496 w (C=N). ^1H NMR (300 MHz), δ , ppm: 4.05 (s, 3H, MeO), 7.30 (d, J 9.8, 1H, H-3), 7.54 (dd, J 8.3, 4.6, 1H, H-7), 8.19 (d, J 9.9, 1H, H-4), 8.29 (dd, J 8.2, 1.5, 1H, H-6), 8.55 (dd, J 4.5, 1.5, 1H, H-8). Found, %: C 59.88; H 3.92; N 27.91. $\text{C}_{10}\text{H}_8\text{N}_4\text{O}$. Calcd., %: C 59.99; H 4.03; N 27.99.

2-(Phenylthio)pyrido[3',2':4,5]imidazo[1,2-*b*]pyridazine (7b).



To a solution of 0.23 g (0.01 mol) Na in 150 ml EtOH, 1.10 g (0.01 mol) thiophenol was added followed by 2.05 g (0.01 mol) of chloropyridoimidazopyridazine **6**. The reaction mixture was boiled for 10 min, diluted with 50 ml of water, and phenylthio derivative **6b** was filtered off. Yield 2.11 g (76%), mp. 108-109°C (cyclohexane). IR spectrum (powder), ν , cm^{-1} : 1575 m (C=N), 1526 m (C=N), 1319 s (CS). ^1H NMR (300 MHz), δ , ppm: 7.30 (d, J 9.7, 1H, H-3), 7.51-7.55 (m, 3H, H-3' – H-5'), 7.60 (dd, J 8.3, 4.5, 1H, H-7), 7.68-7.71 (m, 2H, H-2',6'), 8.17 (d, J 9.8, 1H, H-4), 8.33 (dd, J 8.2, 1.4, 1H, H-6), 8.57 (dd, J 4.6, 1.4, 1H, H-8). Found, %: C 64.65; H 3.54; N 20.07; S 11.59. $\text{C}_{15}\text{H}_{10}\text{N}_4\text{S}$. Calcd., %: C 64.73; H 3.62; N 20.13; S 11.52

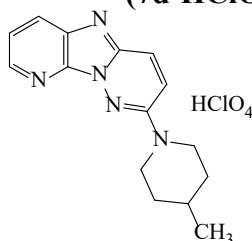
2-Hydrazinopyrido[3',2':4,5]imidazo[1,2-*b*]pyridazine (7c).



A solution of 0.41 g (0.002 mol) chloropyridoimidazopyridazine **6** and 1 ml of hydrazine hydrate in 10 ml of EtOH was boiled for 1 h, evaporated to dryness in a water bath, the residue is treated with 5 ml of water, the reaction product is filtered off and washed with water (3×3 ml). Yield 0.31 g (78%), mp. 276-277°C (MeCN). IR spectrum (powder), ν , cm^{-1} : 3298 m (NH_2), 3262 m (NH), 1587 s (C=N),

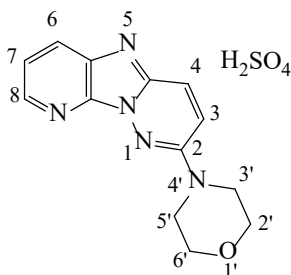
1503 s (C=N). ^1H NMR (600 MHz), δ , ppm: 4.37 (s, 2H, NH_2), 7.10 (d, J 9.8, 1H, H-3), 7.42 (dd, J 8.2, 4.5, 1H, H-7), 7.90 (d, J 9.9, 1H, H-4), 8.18 (dd, J 8.2, 1.4, 1H, H-6), 8.29 (s, 1H, NH), 8.45 (dd, J 4.5, 1.4, 1H, H-8). ^{13}C NMR (600 MHz), δ , ppm: 118.72, 119.87, 126.18, 127.63, 134.44, 141.92, 142.12, 143.30, 155.06. Found, %: C 53.92; H 4.00; N 41.91. $\text{C}_9\text{H}_8\text{N}_6$. Calcd., %: C 53.99; H 4.03; N 41.98.

2-(4-Methylpiperidin-1-yl)pyrido[3',2':4,5]imidazo[1,2-*b*]pyridazine hydroperchlorate (7d·HClO₄).



A mixture of 0.41 g (0.002 mol) chloropyridoimidazopyridazine **6**, 2 ml of 4-methylpiperidine and 4 ml of 2-methoxyethanol was boiled for 4 h, 15 ml of water was added, the solid was filtered off and washed with water (3×3 ml). Yield 0.38 g (71%). To convert it into hydroperchlorate, a solution of the base in EtOH was treated with equimolar amount of 60% HClO_4 , the salt was filtered off, washed with water, mp. 237-238°C (EtOH). IR spectrum (powder), ν , cm^{-1} : 1578 m (C=N), 1498 m (C=N). ^1H NMR (600 MHz), δ , ppm: 0.94 (d, J 6.5, 3H, Me), 1.23 (m, 2H, H-3',5'-ax), 1.70-1.73 (m, 1H, H-4'), 1.78 (d, J 12.9, 2H, H-3',5'-eq), 3.07 (t, J 12.1, 2H, H-2',6'-ax), 4.35 (d, J 13.2, 2H, H-2',6'-eq), 7.76 (dd, J 8.3, 4.6, 1H, H-7), 8.06 (d, J 10.2, 1H, H-3), 8.25 (d, J 10.2, 1H, H-4), 8.40 (d, J 8.2, 1H, H-6), 8.71 (d, J 4.6, 1H, H-8), 9.33 (s, 1H, NH). ^{13}C NMR (600 MHz), δ , ppm: 21.50, 30.05, 33.04, 45.58, 122.51, 122.69, 123.42, 124.58, 126.14, 137.85, 139.82, 145.98, 154.99. Found, %: C 48.90; H 4.86; Cl 9.76; N 18.62. $\text{C}_{15}\text{H}_{17}\text{N}_5\cdot\text{HClO}_4$. Calcd., %: C 48.98; H 4.93; Cl 9.64; N 19.04.

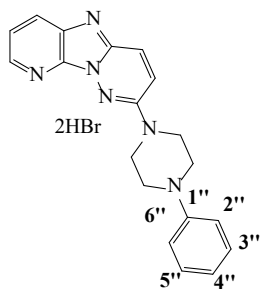
2-(Morpholin-4-yl)pyrido[3',2':4,5]imidazo[1,2-*b*]pyridazine dihydrosulfate (7e·H₂SO₄).



A mixture of 2.05 g (0.01 mol) of chloropyridoimidazopyridazine **6**, 1.74 g, 1.72 ml (0.02 mol) of morpholine in 8 ml of 2-methoxyethanol was boiled for 1.5 h and diluted 20 ml of water. Sodium bicarbonate (1 g) was added, the mixture was extracted with 15 ml of CHCl_3 , the extract was dried with Na_2SO_4 and evaporated to dryness to afford 1.81 g (71%) of free base **7e**. To convert the base to dihydrosulfate, its solution in EtOH was treated with sulfuric acid until a weakly acidic reaction, the salt

was filtered and washed with water, mp 234-236°C (EtOH). IR spectrum (powder), ν , cm^{-1} : 1608 m (C=N), 1578 m (C=N). ^1H NMR (600 MHz), δ , ppm: 3.59-3.68 (m, 4H, H-3',5'), 3.77(dd, 4H, H-2',6'), 7.70 (dd, 1H, J 8.3, 4.6, H-7), 7.97 (d, J 10.2, 1H, H-3), 8.16 (br s, 2H, $2\text{N}^+\text{H}$), 8.30 (d, J 10.2, 1H, H-4), 8.38 (dd, 1H, J 8.3, 1.4, H-6), 8.66 (dd, J 4.6, 1.4, 1H, H-8). Found, %: C 44.16; H 4.23; N 19.80; S 9.20. $\text{C}_{13}\text{H}_{13}\text{N}_5\text{O}\cdot\text{H}_2\text{SO}_4$. Calcd., %: C 44.19; H 4.28; N 19.82; S 9.07

2-(4-Phenylpiperazin-1-yl)pyrido[3',2':4,5]imidazo[1,2-*b*]pyridazine dihydrobromide (7f·2HBr).



A solution of 0.41 g (0.002 mol) of chloropyridoimidazopyridazine **6** and 0.65 g (0.004 mol) of 1-phenylpiperazine in 6 ml of tetramethylurea was boiled for 4 h. After cooling, the reaction mixture was poured into 15 ml of H_2O , the precipitate of free amine **7f** was filtered off, washed with water (3×3 ml) and dried. Yield 0.54 g (82%). To convert it into dihydrobromide, a solution of the base in EtOH was treated with conc. HBr and the precipitated salt was filtered off and washed with water, mp 225-226°C (EtOH). IR spectrum (powder), ν , cm^{-1} : 3478 w (^+NH), 3443 w (^+NH), 1573 s (C=N), 1485 s (C=N). ^1H NMR (600 MHz), δ , ppm: 3.56 (t, J 5.2, 4H, H-2',6'), 4.04 (t, J 5.2, 4H, H-3',5'), 7.05-7.13 (m, 1H, H-4''), 7.36-7.39 (m, 4H, H-2'', 3'', 5'', 6''), 7.50 (br s, 2H, 2^+NH), 7.81 (dd, J 8.3, 4.6, 1H, H-7), 8.24 (d, J 10.2, 1H, H-3), 8.45 (d, J 10.1, 1H, H-4), 8.48 (d, J 8.3, 1H, H-6), 8.75 (dd, J 4.6, 1.5, 1H, H-8). ^{13}C NMR (600 MHz), δ , ppm: 44.29, 44.76, 47.73, 49.90, 117.83, 117.95, 122.74, 122.94, 123.77, 124.62, 125.79, 129.37, 131.60, 137.93, 139.66, 146.29, 155.14. Found, %: C 46.31; H 4.06; Br 32.55; N 17.02. $\text{C}_{19}\text{H}_{18}\text{N}_6\cdot 2\text{HBr}$. Calcd., %: C 46.36; H 4.10; Br 32.47; N 17.07.

Quantum chemical data (B3LYP/6-311++G**)

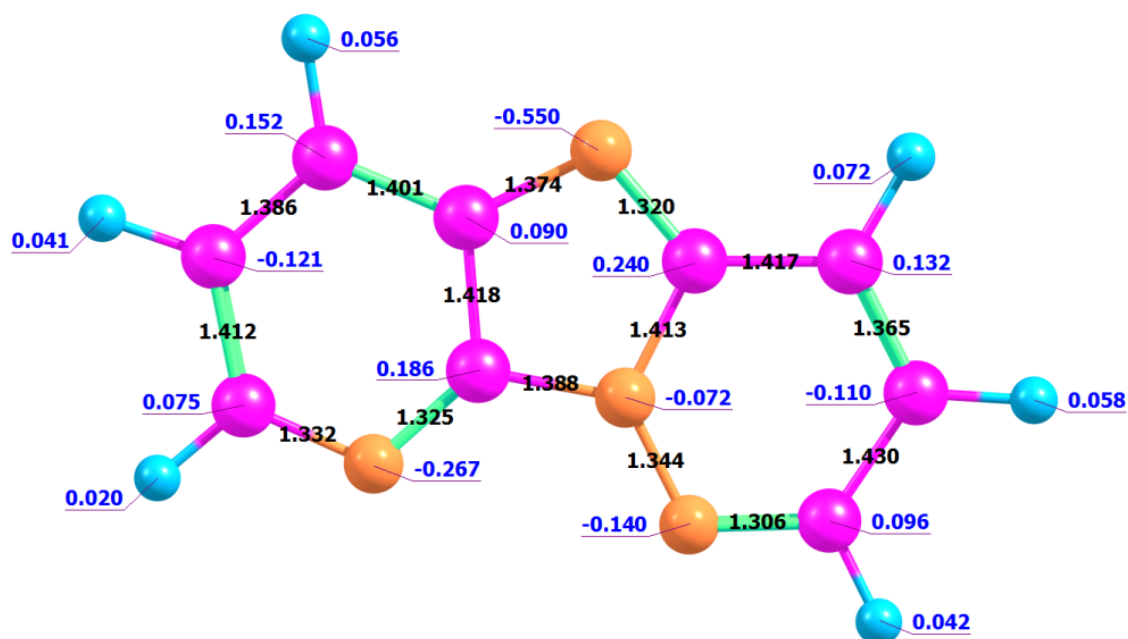


Figure S1. Molecule of tricycle 1. Here and below, the numbers indicate bond lengths (Å) and atomic charges (APT), respectively.

$E_{\text{tot}} = -565.691611637$			A.U.	C	2.773946000	-1.191553000	-0.000006000
symmetry c1				C	3.300159000	0.137621000	0.000104000
C	-2.809033000	-1.183437000	0.000072000	C	2.446371000	1.202541000	0.000114000
C	-3.312528000	0.135349000	0.000060000	H	-3.497710000	-2.022685000	0.000115000
C	-2.449032000	1.219619000	0.000014000	H	-4.385599000	0.285979000	0.000091000
N	-1.512919000	-1.491501000	0.000025000	H	-2.804277000	2.243275000	-0.000020000
C	-0.709372000	-0.438144000	-0.000048000	H	3.436702000	-2.049270000	-0.000048000
C	-1.078072000	0.931038000	-0.000040000	H	4.373748000	0.278708000	0.000175000
N	0.678072000	-0.422513000	-0.000063000	H	2.782630000	2.231365000	0.000125000
C	1.054054000	0.939605000	0.000026000				
N	0.018816000	1.758232000	-0.000161000				
N	1.501108000	-1.484675000	-0.000119000				

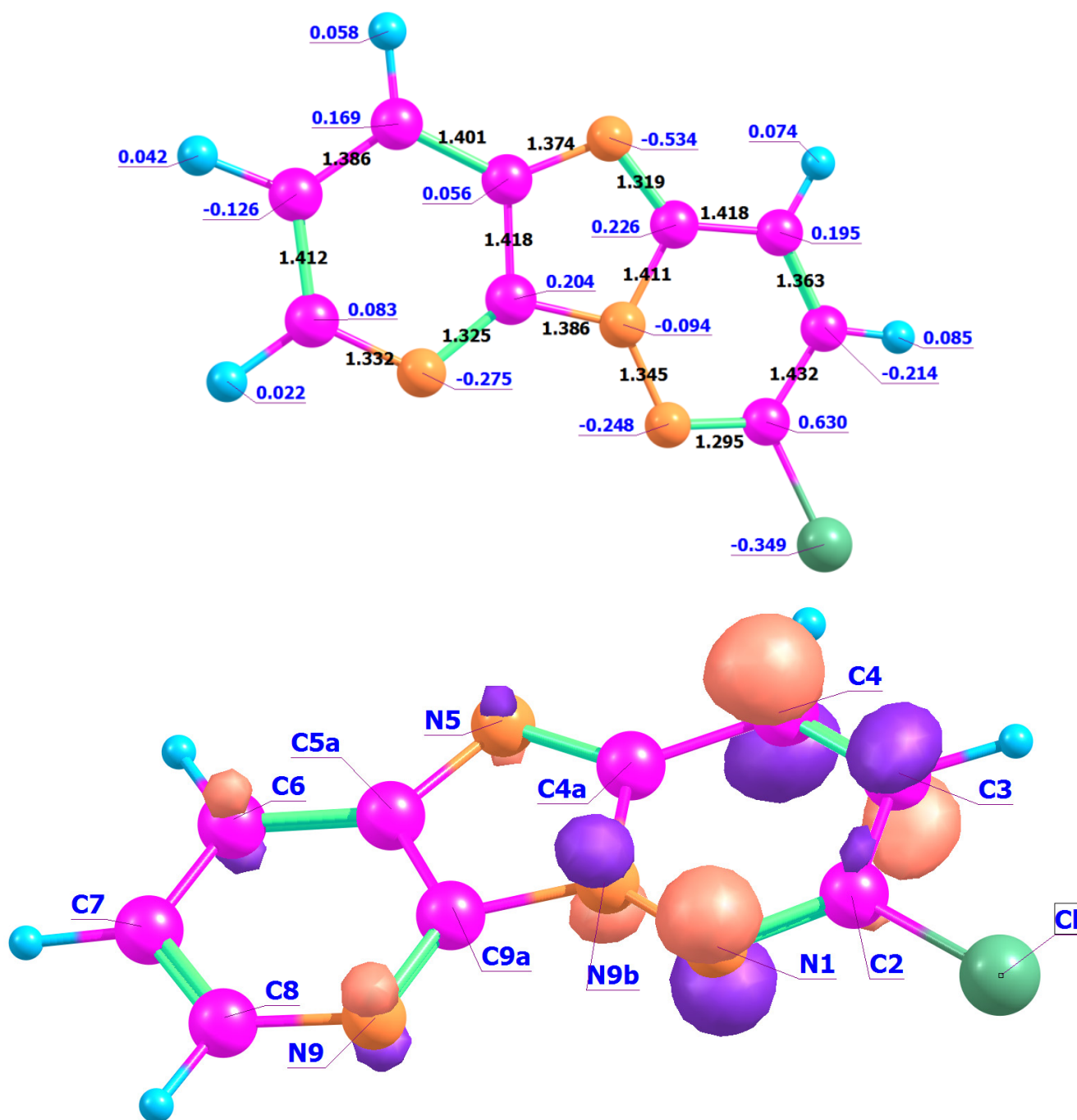


Figure S2. Chloroderivative **6** and its LUMO; light brown color corresponds to a positive sign of the wave function, and purple color corresponds to a negative sign

$E_{\text{tot}} = -1025.31148626$ A.U.							
symmetry c1							
C	-3.051245000	-1.711078000	0.000020000	N	1.170685000	-0.791777000	0.000067000
C	-3.903903000	-0.585447000	-0.000010000	C	2.297928000	-0.154415000	0.000124000
C	-3.378762000	0.696927000	-0.000040000	C	2.448680000	1.269990000	0.000063000
N	-1.721049000	-1.644718000	0.000029000	C	1.327717000	2.045626000	0.000030000
C	-1.245006000	-0.408609000	0.000005000	Cl	3.741266000	-1.140305000	-0.000078000
C	-1.981781000	0.803306000	-0.000031000	H	-3.477591000	-2.709403000	0.000046000
N	0.080776000	-0.003430000	0.000021000	H	-4.976118000	-0.741315000	-0.000008000
C	0.061946000	1.407305000	-0.000003000	H	-4.006170000	1.580262000	-0.000073000
N	-1.159677000	1.904177000	-0.000062000	H	3.441835000	1.697091000	0.000060000
				H	1.367928000	3.127138000	-0.000017000

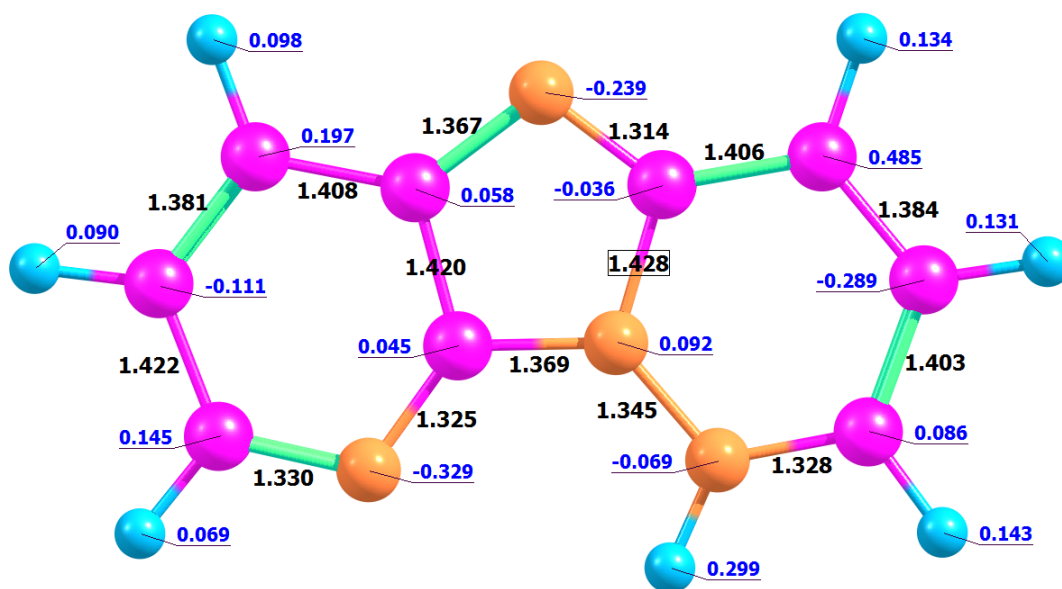


Figure S3. 1H-Protonated form of tricycle 1.

$E_{\text{tot}} = -566.038258110$ A.U.				N	-1.518562000	-1.393999000	0.000017000
symmetry c1				C	-2.822394000	-1.143485000	0.000030000
C	2.800787000	-1.204821000	0.000037000	C	-3.298134000	0.176089000	0.000019000
C	3.323608000	0.117628000	0.000016000	C	-2.412942000	1.239383000	-0.000003000
C	2.491898000	1.220509000	-0.000010000	H	3.478606000	-2.051226000	0.000051000
N	1.501519000	-1.490663000	0.000024000	H	4.398998000	0.244005000	0.000016000
C	0.733413000	-0.411583000	-0.000007000	H	2.871778000	2.234533000	-0.000035000
C	1.108200000	0.957877000	-0.000024000	H	-1.109863000	-2.328147000	0.000024000
N	-0.635076000	-0.380226000	-0.000007000	H	-3.467141000	-2.012460000	0.000048000
C	-1.028605000	0.992421000	-0.000018000	H	-4.366913000	0.338011000	0.000020000
N	0.019709000	1.785399000	-0.000082000	H	-2.753584000	2.267597000	-0.000028000

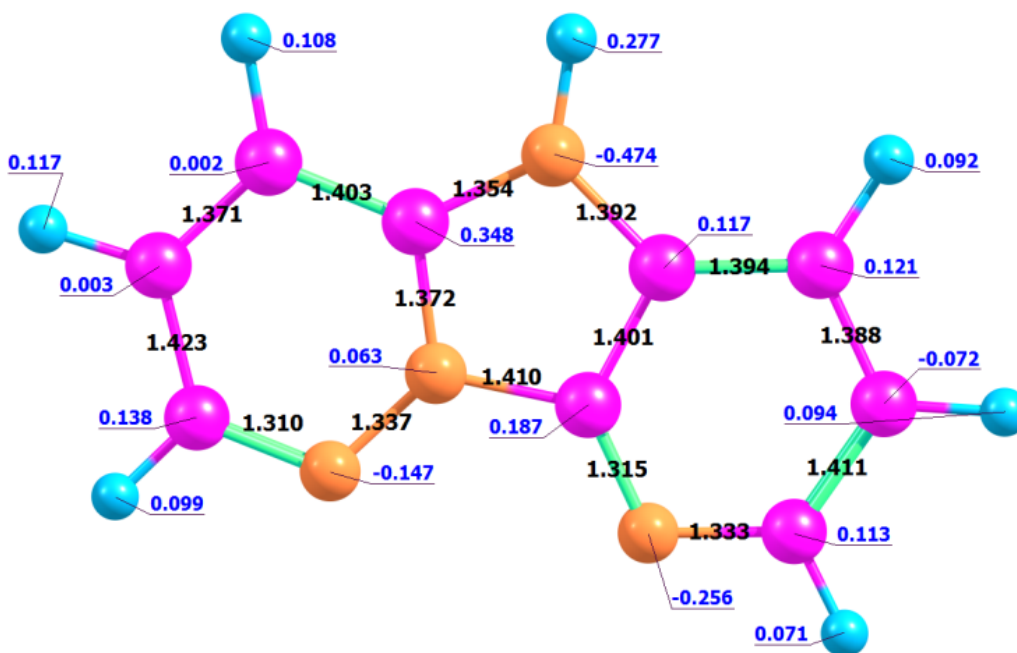


Figure S4. 5H-Protonated form of tricycle 1

$E_{\text{tot}} = -566.064423743$ A.U.				N	1.482380000	-1.486565000	-0.000043000
symmetry c1				C	2.767402000	-1.230244000	0.000006000
C	-2.799500000	-1.226838000	0.000033000	C	3.319124000	0.081062000	0.000048000
C	-3.330936000	0.080593000	0.000041000	C	2.484629000	1.169051000	0.000036000
C	-2.494981000	1.188481000	0.000011000	H	-3.464870000	-2.082866000	0.000058000
N	-1.494459000	-1.497598000	-0.000007000	H	-4.405610000	0.211932000	0.000082000
C	-0.721406000	-0.434114000	-0.000043000	H	-2.882432000	2.199698000	0.000016000
C	-1.130354000	0.905657000	-0.000045000	H	0.047134000	2.705425000	-0.000165000
N	0.688179000	-0.410554000	-0.000032000	H	3.404565000	-2.106425000	0.000013000
C	1.108740000	0.895680000	-0.000008000	H	4.394479000	0.205696000	0.000106000
N	0.015581000	1.695028000	-0.000009000	H	2.848668000	2.188397000	0.000047000

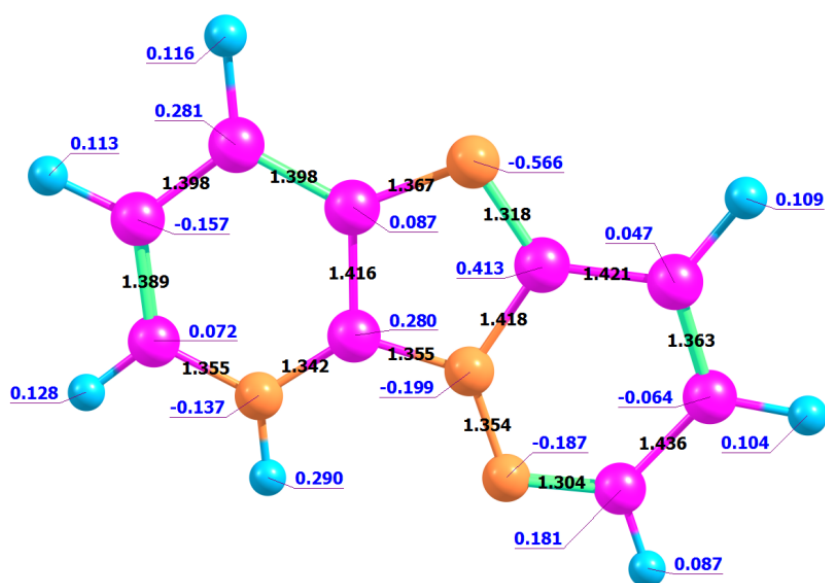
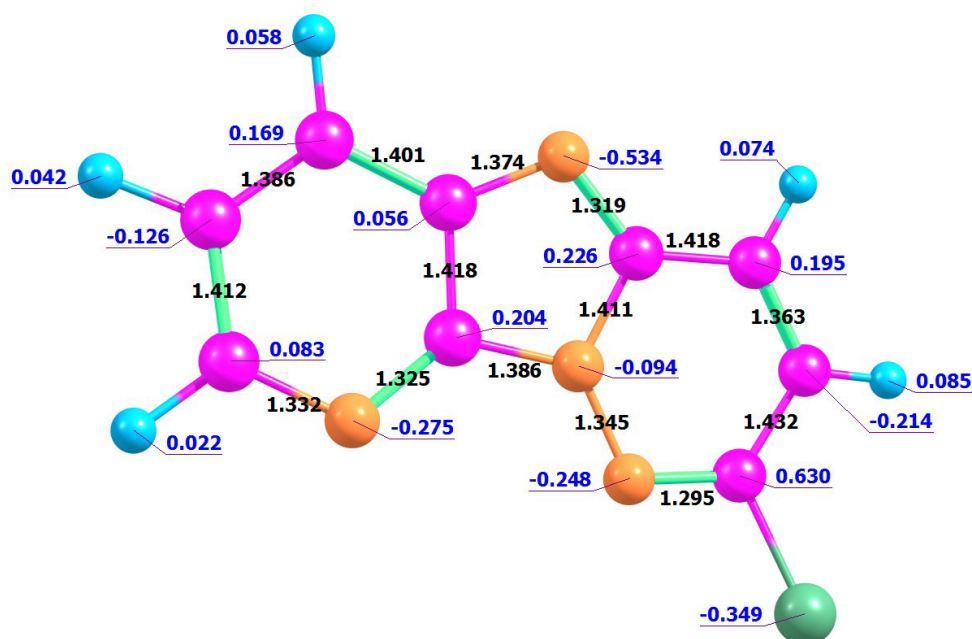
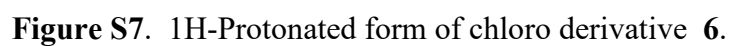


Figure S5. 9H-Protonated form of tricycle 1.

$E_{\text{tot}} = -566.062643792$ A.U.				$E_{\text{tot}} = -566.062643792$ A.U.			
symmetry c1				symmetry c1			
C	-2.858197000	-1.134656000	0.000071000	C	3.311148000	0.117870000	0.000088000
C	-3.308616000	0.179393000	0.000066000	C	2.487509000	1.204267000	0.000074000
C	-2.417089000	1.255620000	-0.000004000	H	-3.520192000	-1.989111000	0.000112000
N	-1.530034000	-1.404126000	0.000015000	H	-4.376482000	0.352017000	0.000124000
C	-0.655719000	-0.385702000	-0.000083000	H	-2.771272000	2.279620000	0.000005000
C	-1.047975000	0.975301000	-0.000069000	H	-1.190557000	-2.363420000	-0.000016000
N	0.699526000	-0.393153000	-0.000055000	H	3.422451000	-2.075462000	0.000012000
C	1.086278000	0.970971000	0.000006000	H	4.387224000	0.235660000	0.000160000
N	0.051294000	1.787767000	-0.000076000	H	2.851016000	2.223687000	0.000092000



$E_{\text{tot}} = -1025.31148626$ A.U.				$E_{\text{tot}} = -1025.31148626$ A.U.			
symmetry	c1			N	-1.159677000	1.904177000	-0.000062000
				N	1.170685000	-0.791777000	0.000067000
C	-3.051245000	-1.711078000	0.000020000	C	2.297928000	-0.154415000	0.000124000
C	-3.903903000	-0.585447000	-0.000010000	C	2.448680000	1.269990000	0.000063000
C	-3.378762000	0.696927000	-0.000040000	C	1.327717000	2.045626000	0.000030000
N	-1.721049000	-1.644718000	0.000029000	Cl	3.741266000	-1.140305000	-0.000078000
C	-1.245006000	-0.408609000	0.000005000	H	-3.477591000	-2.709403000	0.000046000
C	-1.981781000	0.803306000	-0.000031000	H	-4.976118000	-0.741315000	-0.000008000
N	0.080776000	-0.003430000	0.000021000	H	-4.006170000	1.580262000	-0.000073000
C	0.061946000	1.407305000	-0.000003000	H	3.441835000	1.697091000	0.000060000
				H	1.367928000	3.127138000	-0.000017000

S12

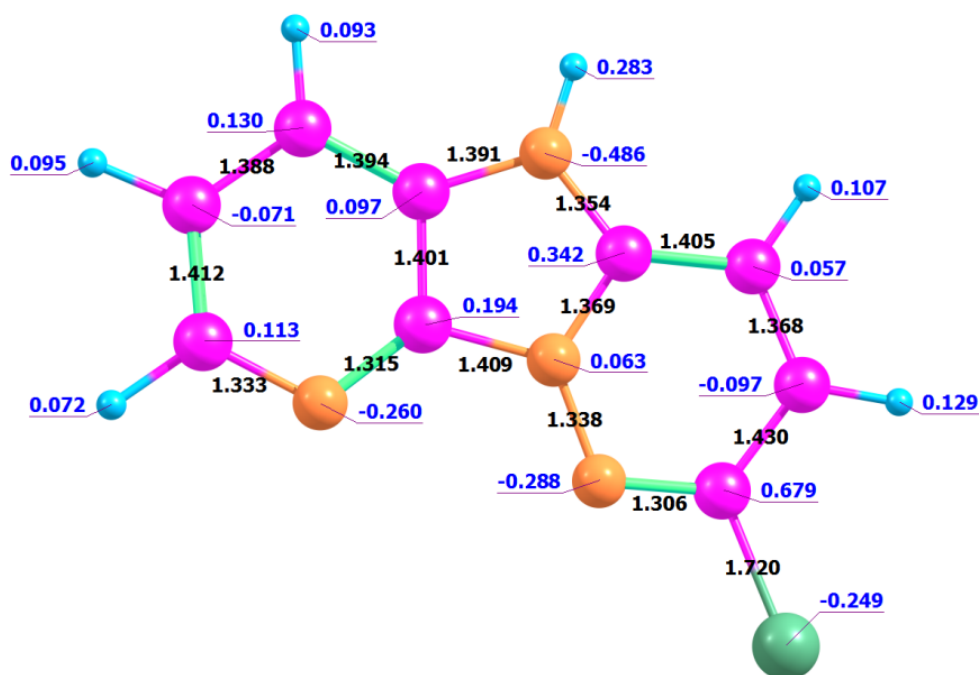


Figure S8. 5H-Protonated form of chloro derivative **6**.

$E_{\text{tot}} = -1025.67889553$ A.U.				N	1.169784000	-0.782115000	0.000061000
symmetry c1				C	2.324417000	-0.171781000	0.000018000
C	-3.001782000	-1.765454000	0.000127000	C	2.483269000	1.249231000	0.000277000
C	-3.887959000	-0.666171000	-0.000150000	C	1.371018000	2.045086000	0.000264000
C	-3.407394000	0.635534000	-0.000177000	Cl	3.723101000	-1.173543000	-0.000247000
N	-1.674398000	-1.648546000	0.000328000	H	-3.391890000	-2.777029000	0.000307000
C	-1.241027000	-0.407180000	0.000107000	H	-4.954859000	-0.850435000	-0.000399000
C	-2.018871000	0.758512000	-0.000048000	H	-4.070234000	1.491900000	-0.000366000
N	0.100444000	0.022758000	0.000155000	H	-1.409232000	2.820818000	-0.000542000
C	0.127229000	1.391399000	0.000005000	H	3.479093000	1.672145000	0.000495000
N	-1.148968000	1.843907000	-0.000286000	H	1.432968000	3.125745000	0.000372000

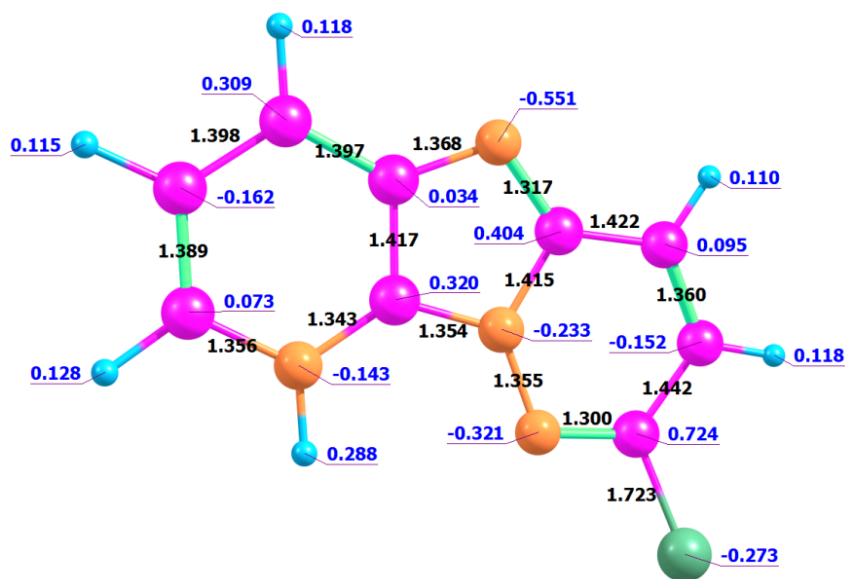


Figure S9. 9H-Protonated form of chloro derivative **6**.

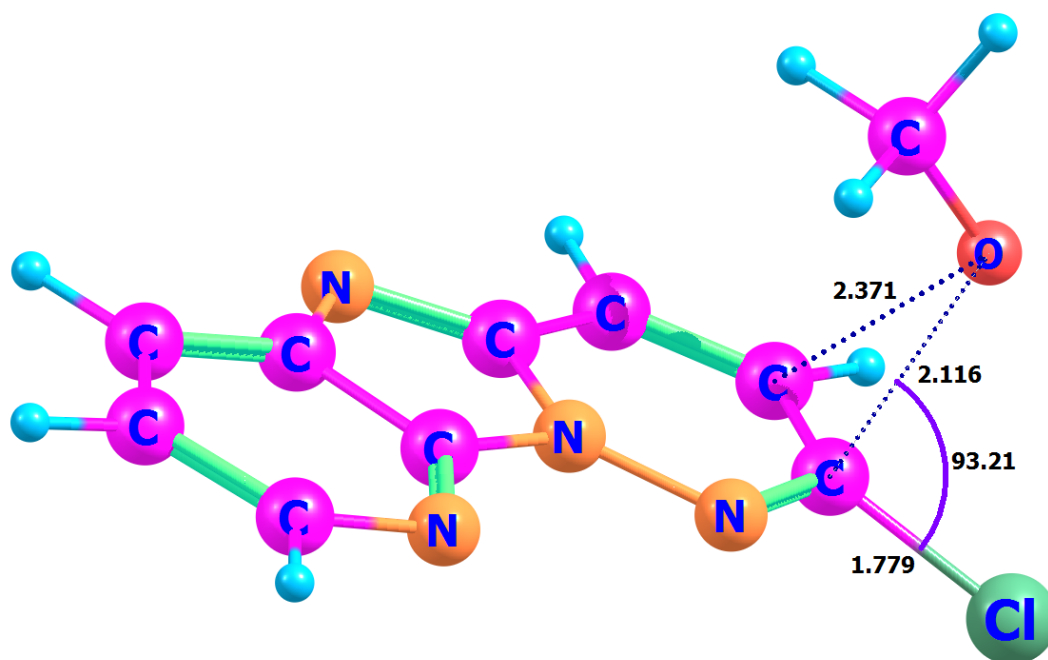


Figure S11. Transition state TS1 formed in the system **6** – MeO[−].

$E_{\text{tot}} = -1140.49744278$, $\lambda_i = -153.9 \text{ cm}^{-1}$				C	1.937297000	-0.308766000	-0.364823000
symmetry c1				C	2.091042000	1.085084000	-0.721106000
C	-3.443980000	-1.641865000	0.374040000	C	0.988173000	1.875897000	-0.797803000
C	-4.275022000	-0.522922000	0.202832000	H	-3.881310000	-2.606065000	0.620696000
C	-3.724278000	0.718384000	-0.115291000	H	-5.347245000	-0.637388000	0.320983000
N	-2.111117000	-1.619782000	0.254768000	H	-4.338123000	1.601659000	-0.255469000
C	-1.608985000	-0.430339000	-0.045472000	H	3.087784000	1.453665000	-0.907955000
C	-2.337658000	0.781689000	-0.249074000	H	1.036959000	2.932812000	-1.022696000
N	-0.289744000	-0.073333000	-0.232675000	Cl	3.285457000	-1.395309000	-0.774797000
C	-0.281995000	1.282063000	-0.537642000	O	2.873613000	0.379700000	1.403087000
N	-1.499076000	1.823025000	-0.553862000	C	1.904627000	0.684064000	2.327738000
N	0.796121000	-0.921593000	-0.200329000	H	2.337196000	0.801782000	3.341839000
				H	1.369383000	1.640213000	2.116357000
				H	1.125068000	-0.101987000	2.417390000

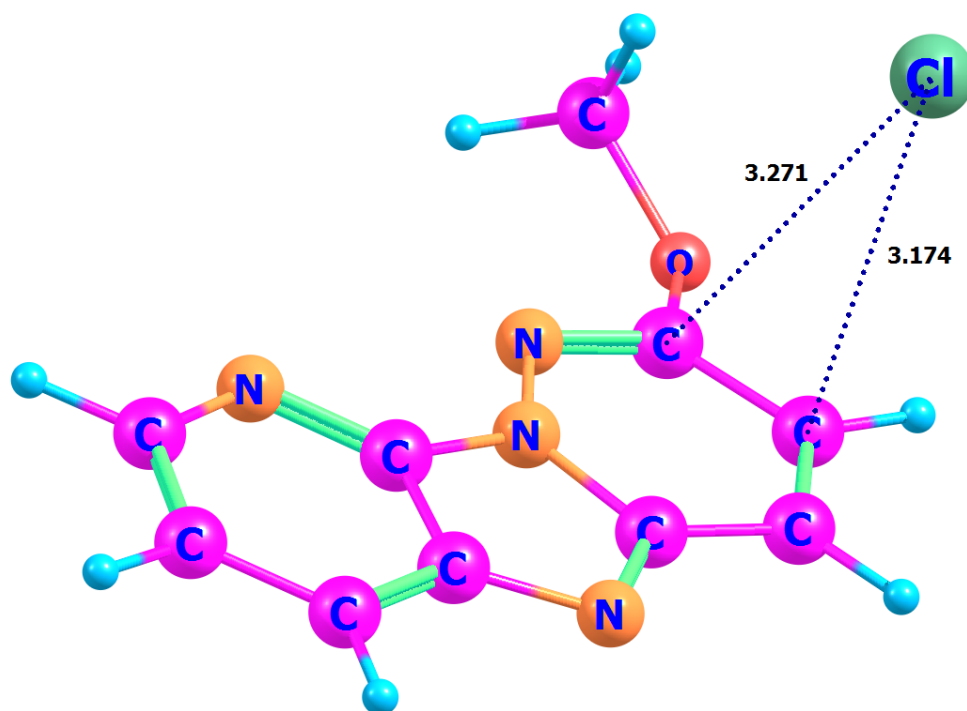


Figure S12. The weak complex between methoxy derivative **7a** and chloride ion formed from transition state **TS1**

$$E_{\text{tot}} = -1140.57802900 \quad \text{A.U.}$$

symmetry c1

C	-3.559312000	1.591040000	0.546649000	C	0.801282000	-1.912190000	-0.881386000
C	-4.365760000	0.438758000	0.584699000	H	-3.994293000	2.561831000	0.768196000
C	-3.822402000	-0.809930000	0.301030000	H	-5.415342000	0.538610000	0.838672000
N	-2.256407000	1.588454000	0.250897000	H	-4.417808000	-1.715930000	0.320795000
C	-1.759913000	0.387898000	-0.008588000	H	2.878122000	-1.453342000	-1.189391000
C	-2.460903000	-0.853009000	-0.012686000	H	0.870695000	-2.987091000	-0.978347000
N	-0.462092000	0.048837000	-0.346910000	O	2.782191000	1.084638000	-1.084461000
C	-0.448203000	-1.334541000	-0.534019000	C	2.936991000	2.276990000	-0.296904000
N	-1.638425000	-1.897600000	-0.340586000	H	3.874093000	2.720643000	-0.631263000
N	0.579130000	0.901840000	-0.513323000	H	2.106929000	2.967384000	-0.459763000
C	1.704418000	0.326226000	-0.831412000	H	3.011690000	1.982968000	0.753153000
C	1.878569000	-1.092859000	-0.999750000	Cl	3.520255000	-0.699599000	1.688076000

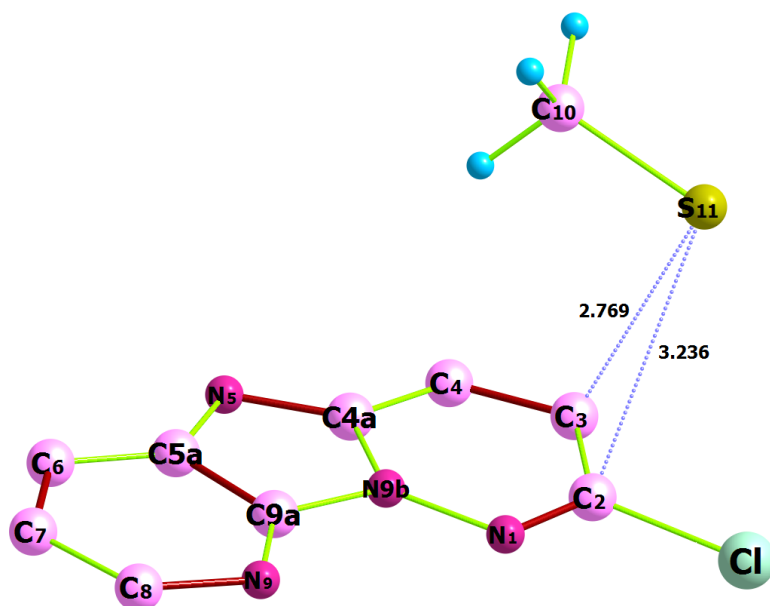


Figure S13. The starting reactant complex **6·MeS⁻** from which transition state **TS2** is formed.

$E_{\text{tot}} = -1463.50996812 \text{ A.U.}$							
symmetry c1							
Cl	2.635387000	2.387476000	0.158454000	C	-1.978910000	0.394094000	-0.152973000
C	0.883106000	-0.979336000	1.451346000	N	-2.637842000	1.298207000	-0.855613000
C	1.871836000	-0.072626000	1.100892000	C	-3.858078000	-1.052141000	0.232177000
C	1.454905000	1.105728000	0.393836000	C	-4.577298000	-0.113977000	-0.509591000
N	0.270732000	1.361865000	-0.050411000	C	-3.940938000	1.024767000	-1.023023000
N	-1.533822000	-1.500546000	1.103669000	H	1.943409000	-2.700619000	-2.017501000
C	-0.420876000	-0.781913000	0.955404000	H	1.597252000	-2.523589000	-0.290998000
N	-0.636989000	0.379659000	0.192258000	H	2.995144000	-3.477301000	-0.813431000
C	-2.499113000	-0.798989000	0.429913000	C	2.409087000	-2.579507000	-1.034834000
				S	3.430937000	-1.072082000	-0.957574000

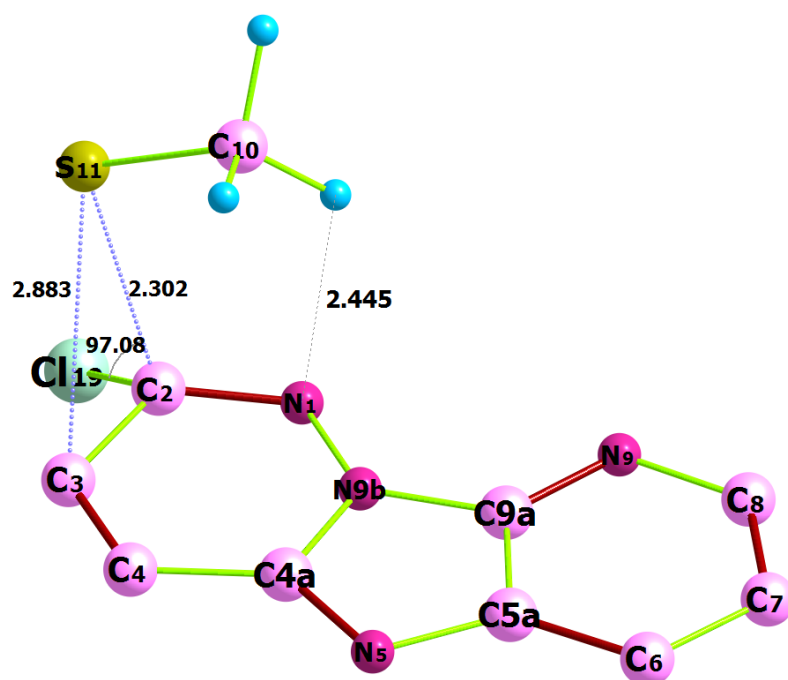


Figure S14. Transition state **TS2** formed in the system **6 – MeS⁻**.

symmetry c1, $E_{\text{tot}} = -1463.49683586$ a.u., $\lambda_i = -145$ cm ⁻¹				N	0.636498000	-0.486402000	-0.727657000
C	-3.543848000	-1.619197000	-0.608642000	C	1.784185000	0.135930000	-0.475395000
C	-4.424128000	-0.644139000	-0.105584000	C	1.836157000	1.520091000	0.000966000
C	-3.933332000	0.584690000	0.330449000	C	0.705813000	2.160125000	0.374892000
N	-2.220808000	-1.464165000	-0.710628000	Cl	3.033688000	-0.157466000	-1.818042000
C	-1.773448000	-0.286584000	-0.291335000	S	2.985035000	-0.814877000	1.242640000
C	-2.555582000	0.785807000	0.240452000	C	1.500598000	-1.348116000	2.156141000
N	-0.475464000	0.178399000	-0.257966000	H	1.742945000	-2.231294000	2.751995000
C	-0.534058000	1.459471000	0.263351000	H	1.113525000	-0.569672000	2.820325000
N	-1.768789000	1.855181000	0.576344000	H	0.729971000	-1.630880000	1.430858000

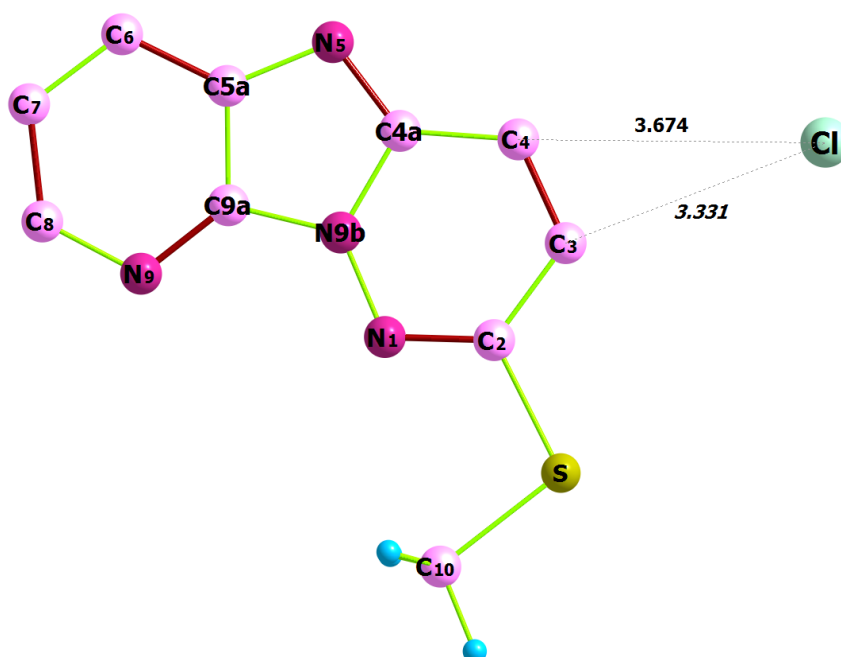
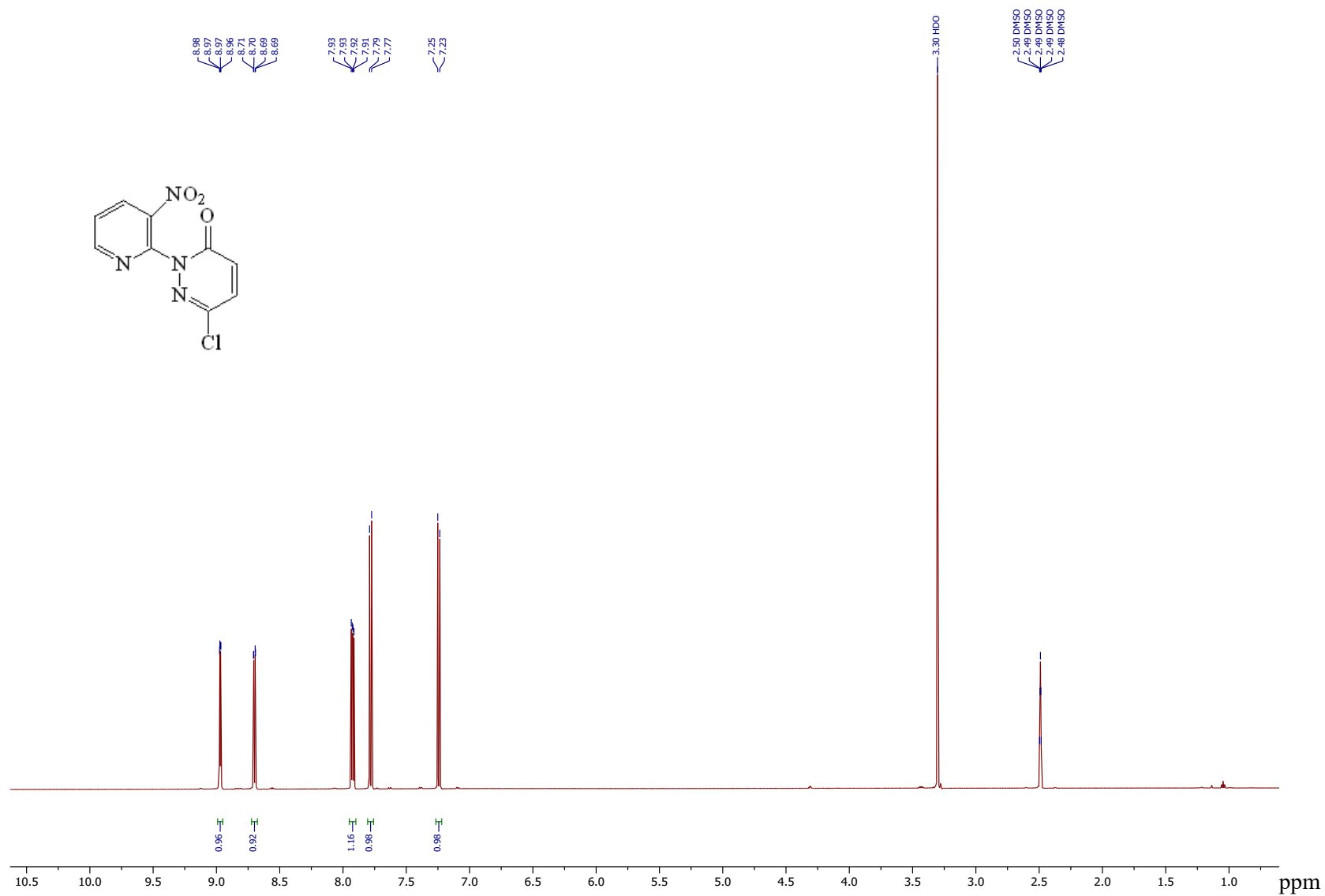


Figure S15. Complex of reaction products in the system **6** – MeS⁻

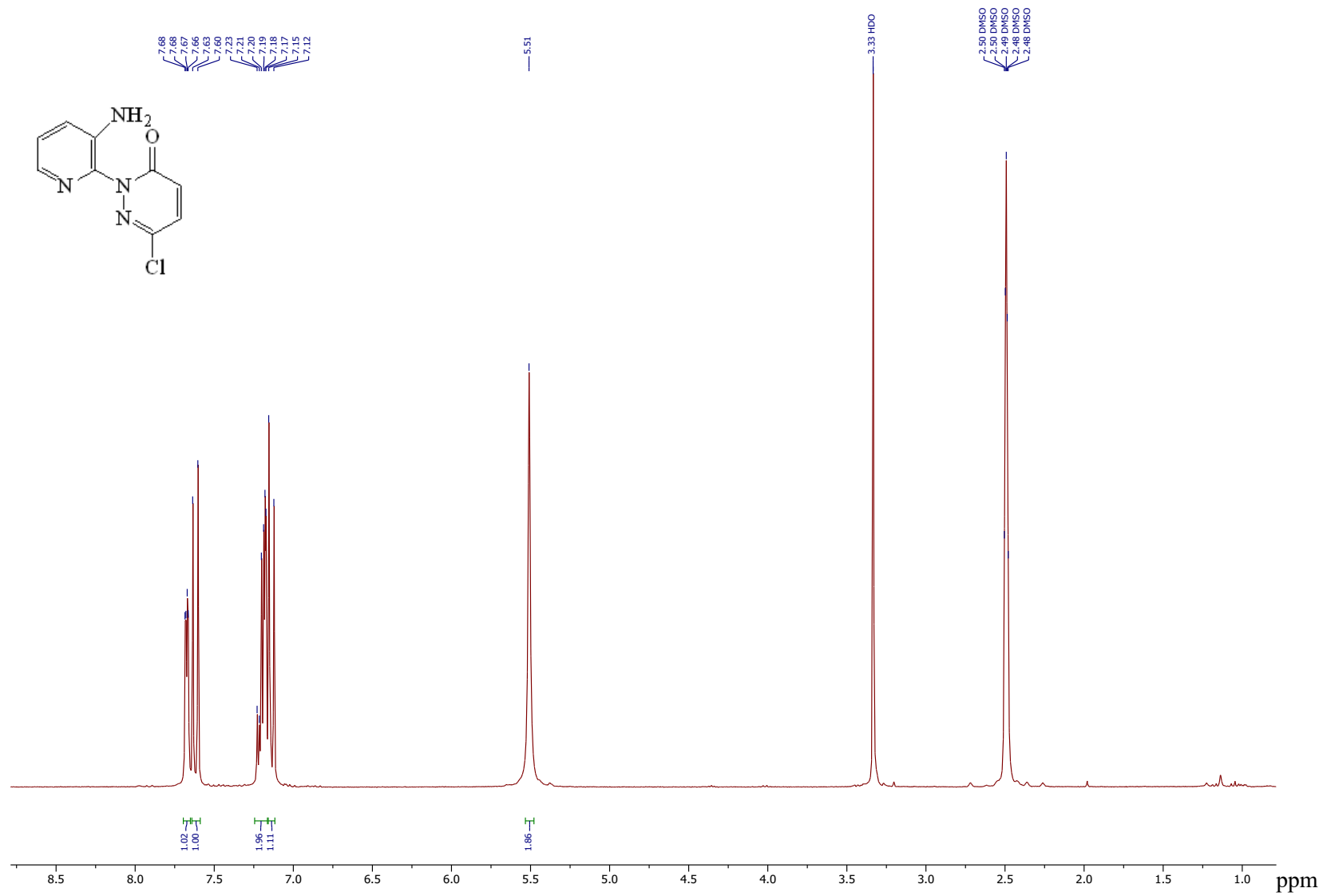
$E_{\text{tot}} = -1463.56126236$ A.U. symmetry Cs				C	1.349894686	-1.199835303	0.000000000
C	-0.283637314	4.341684697	0.000000000	S	-2.543467314	-2.167750303	0.000000000
C	1.087478686	4.660709697	0.000000000	C	-3.819108314	-0.866584303	0.000000000
C	2.045348686	3.653496697	0.000000000	H	-4.774924314	-1.392399303	0.000000000
N	-0.765452314	3.095850697	0.000000000	H	-3.734007314	-0.242088303	0.888300000
C	0.166935686	2.154143697	0.000000000	Cl	1.739675314	-4.853152697	0.000000000
C	1.578741686	2.334871697	0.000000000				
N	0.000000686	0.779952697	0.000000000				
C	1.286216686	0.219106697	0.000000000				
N	2.245004686	1.138573697	0.000000000				
N	-1.176558314	0.111063697	0.000000000				
C	-1.061442314	-1.191862303	0.000000000				
C	0.187001686	-1.906318303	0.000000000				

3. NMR Spectra of compounds 4-7.

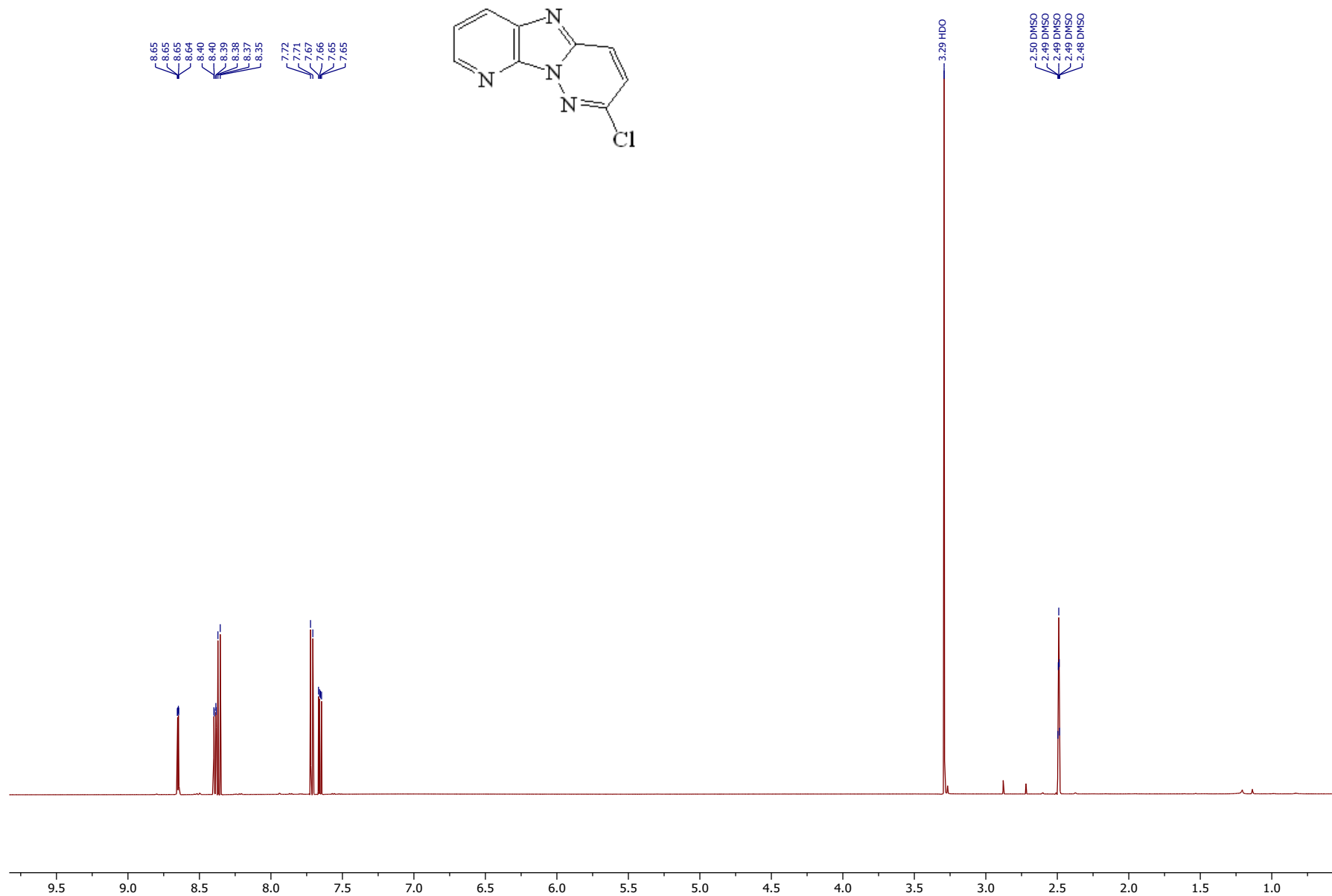
6-Chloro-2-(3-nitropyridin-2-yl)pyridazin-3(2H)-one (4).

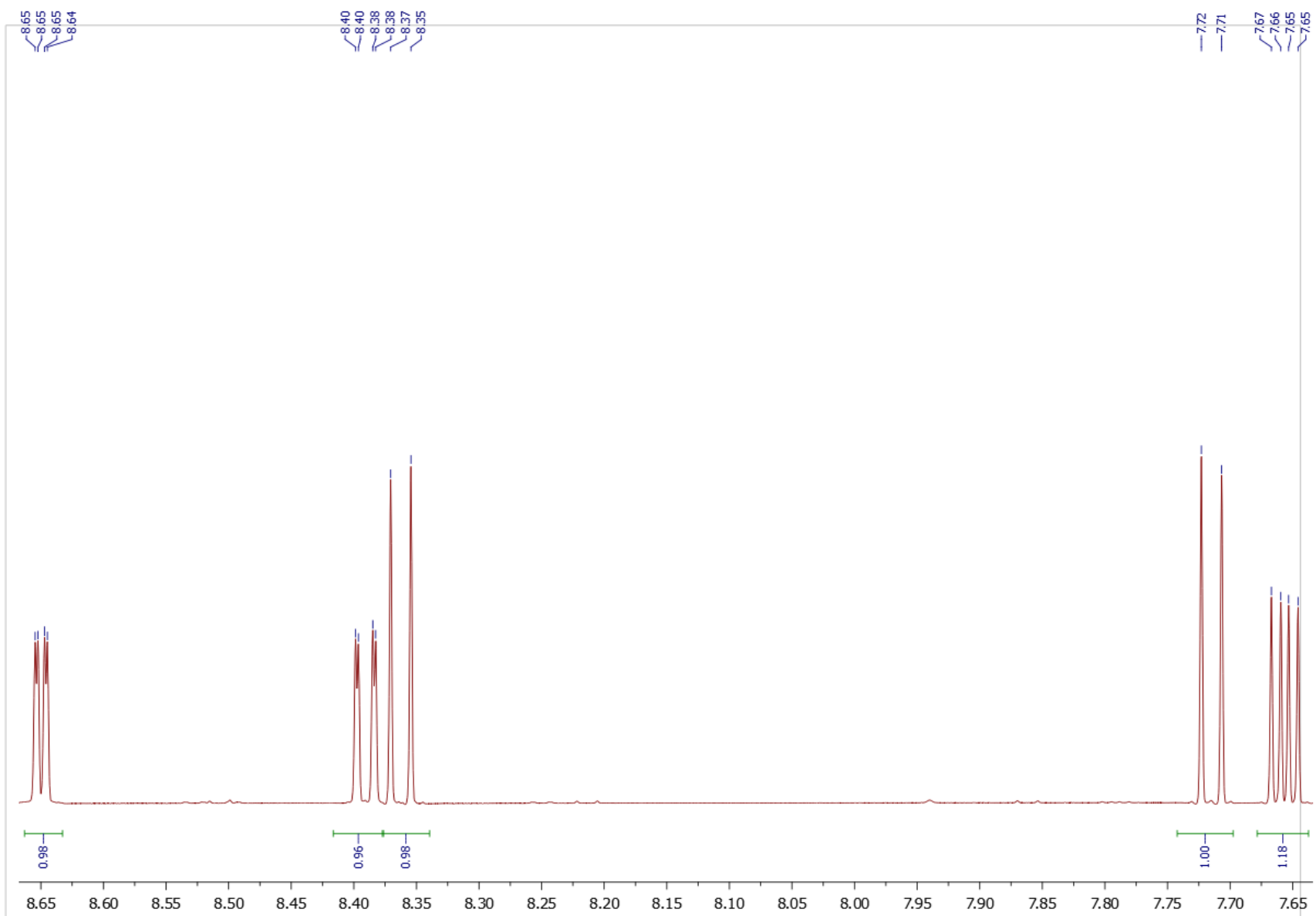


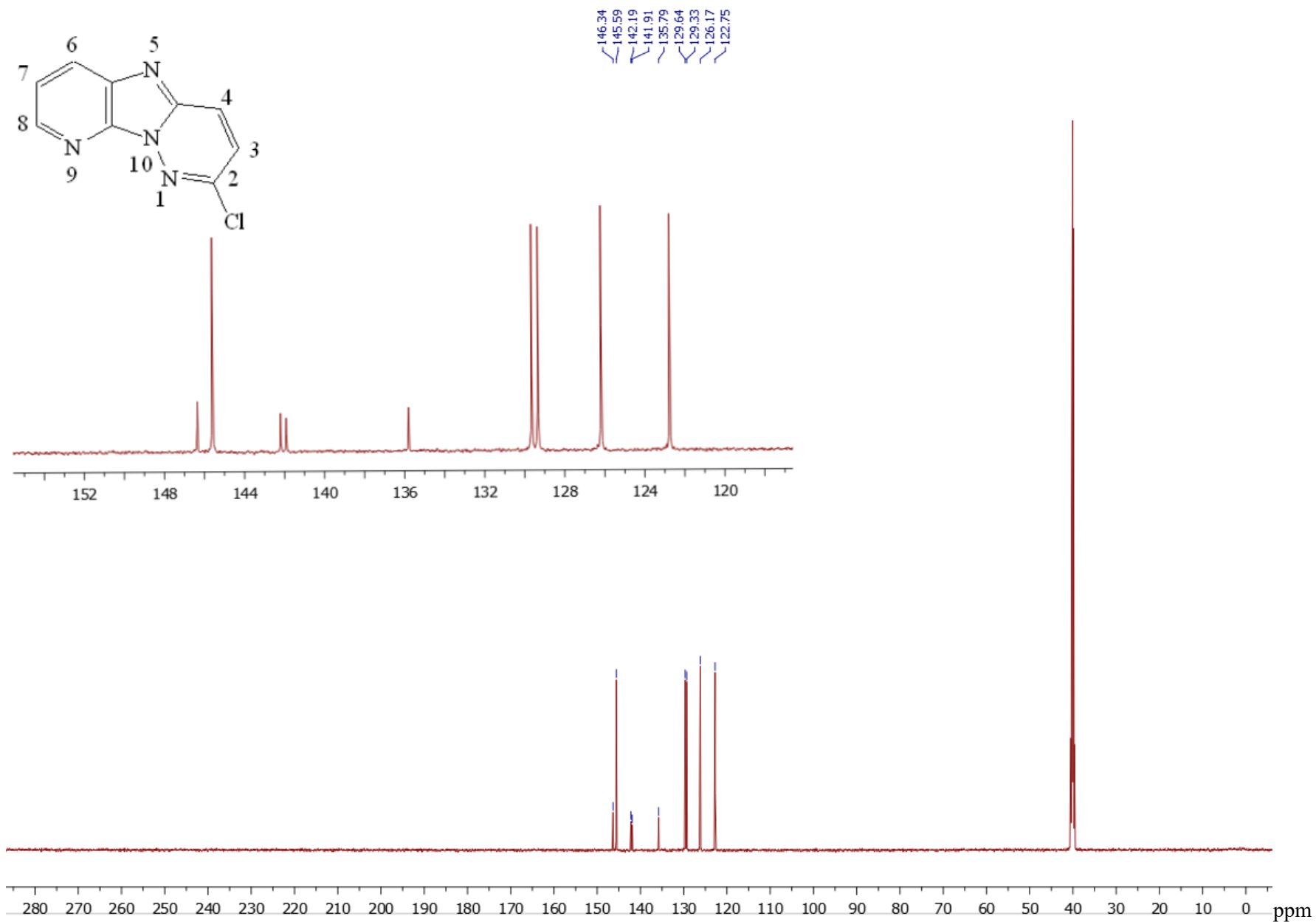
2-(3-Aminopyridin-2-yl)-6-chloropyridazin-3(2H)-one (5).



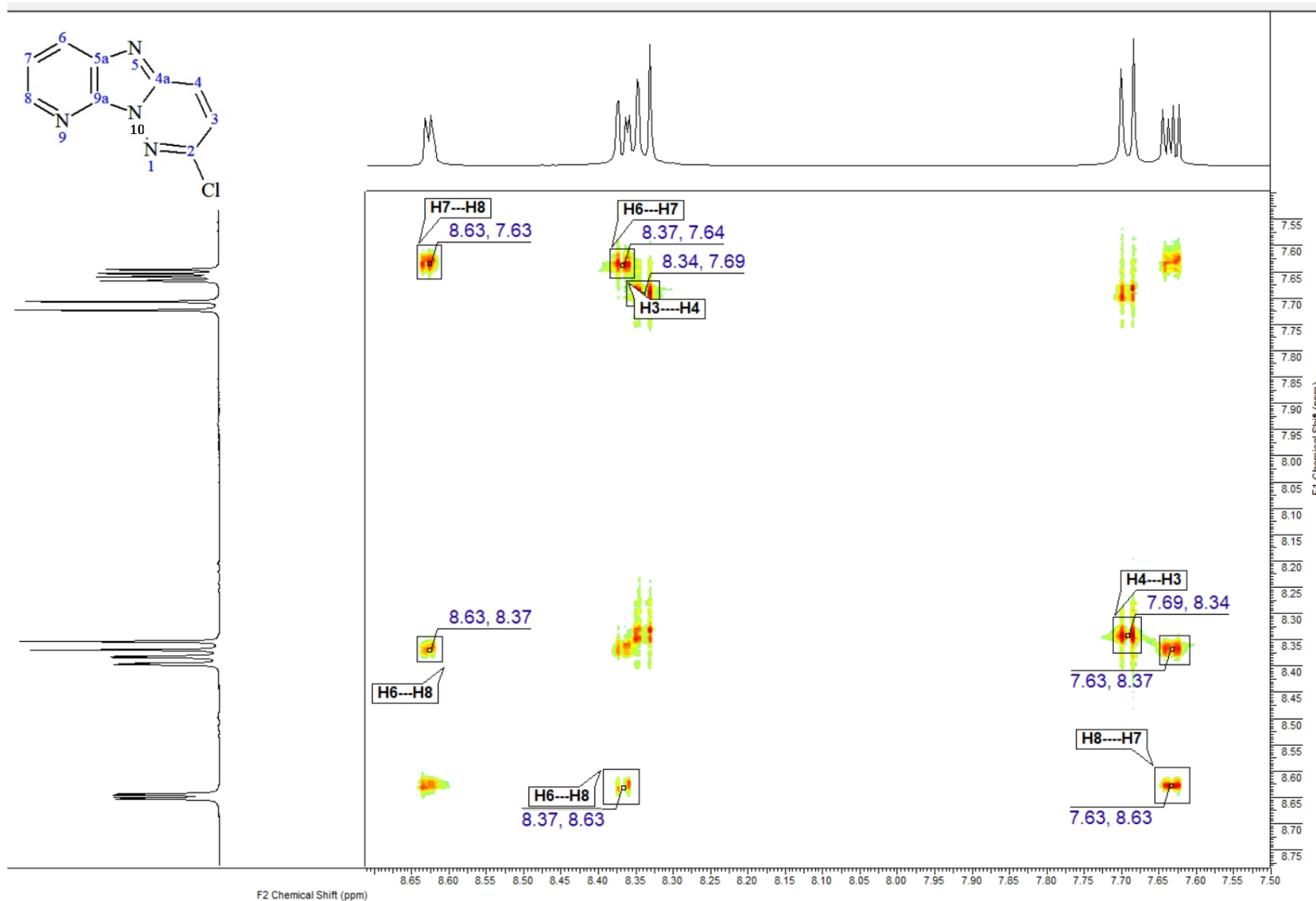
2-Chloropyrido[3',2':4,5]imidazo[1,2-*b*]pyridazine (6).





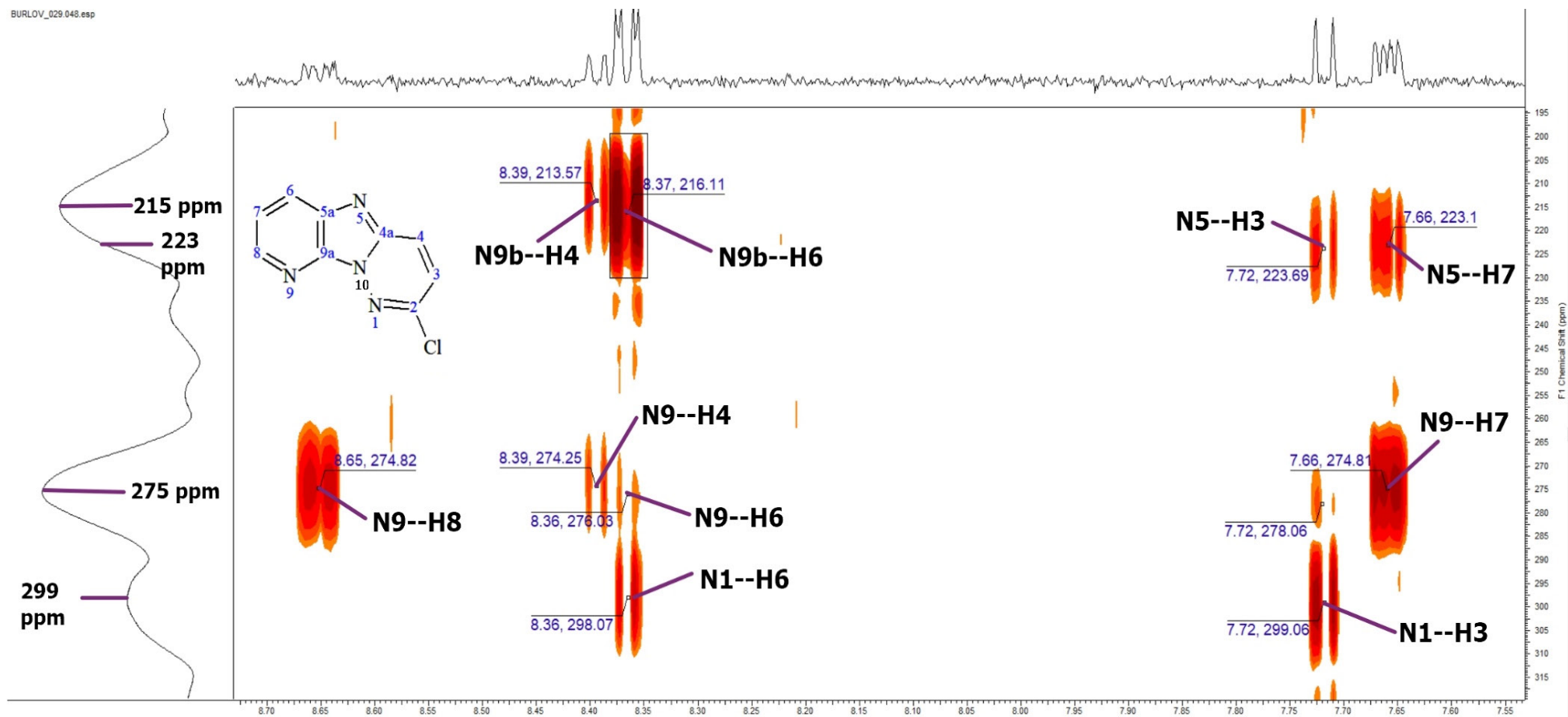


2D NMR Cosy ^1H — ^1H

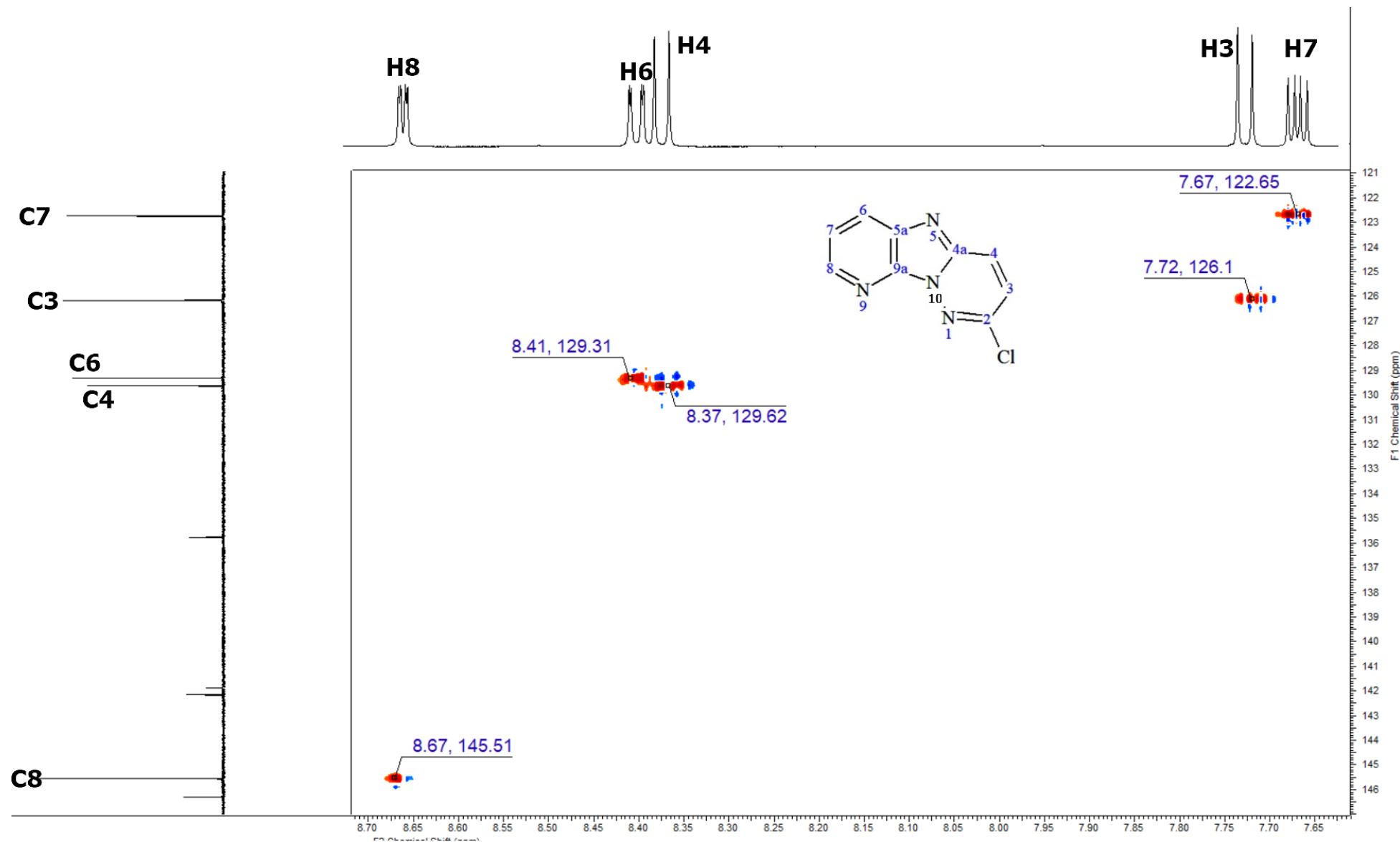


HMBC ^{15}N – ^1H

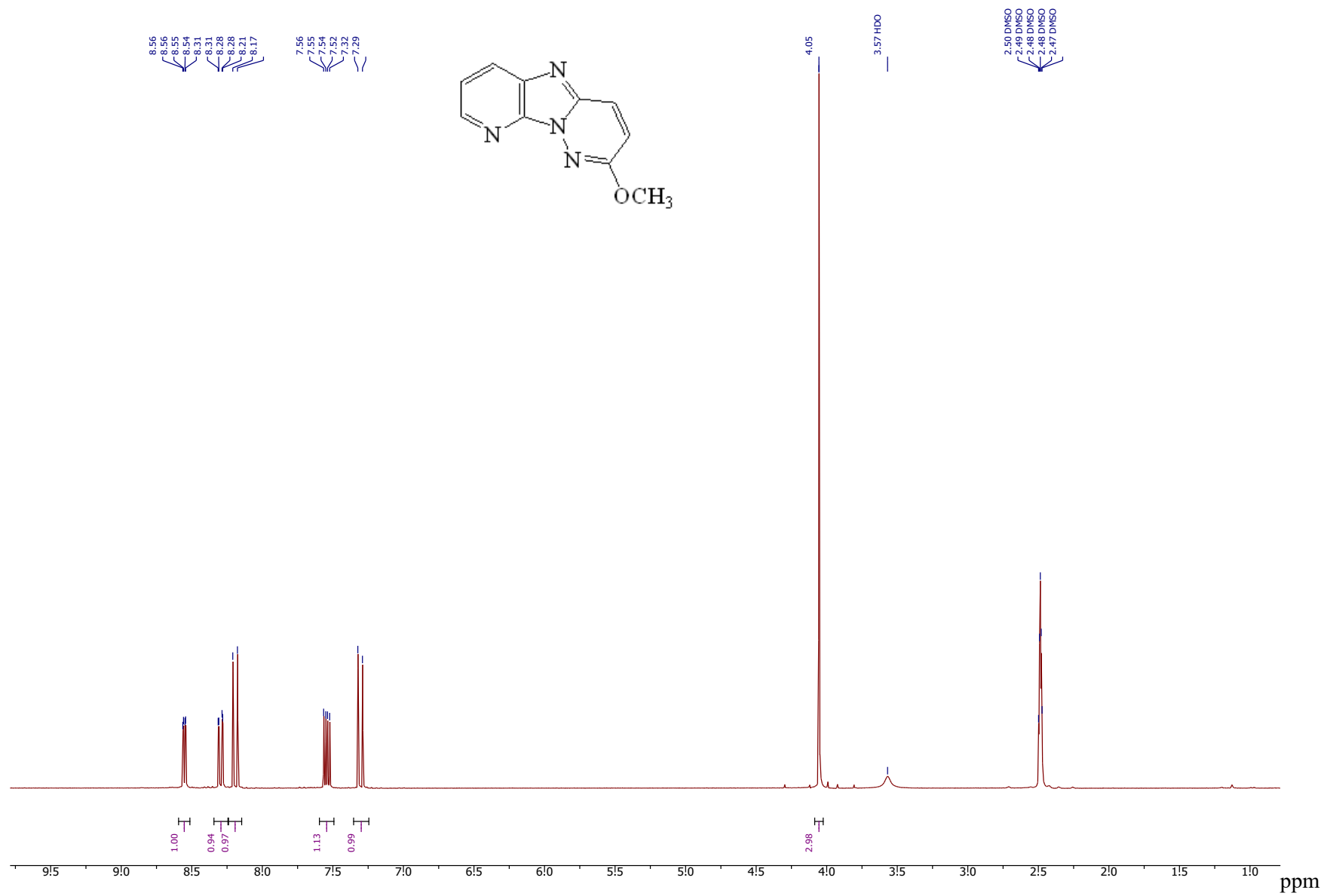
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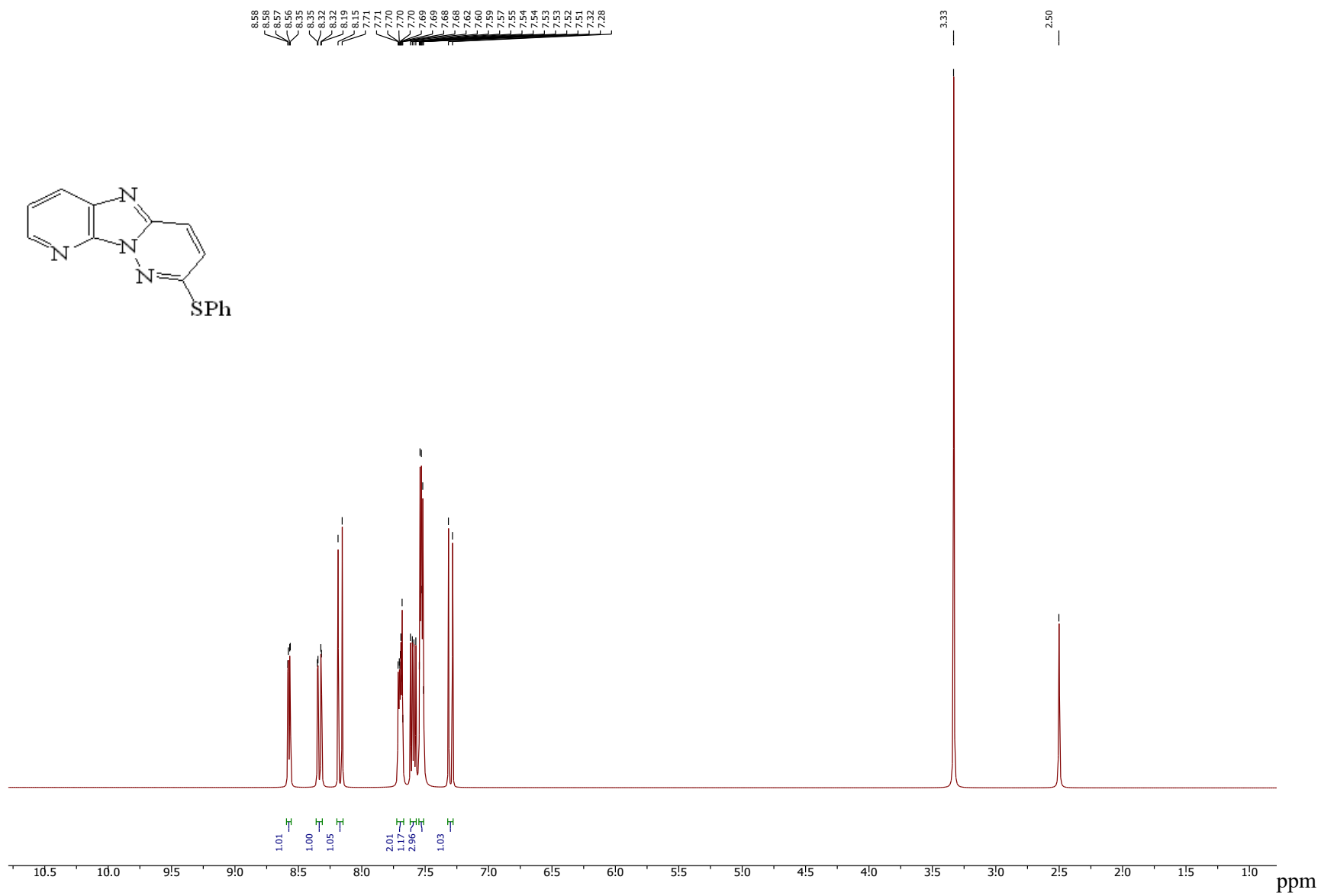
HSQC



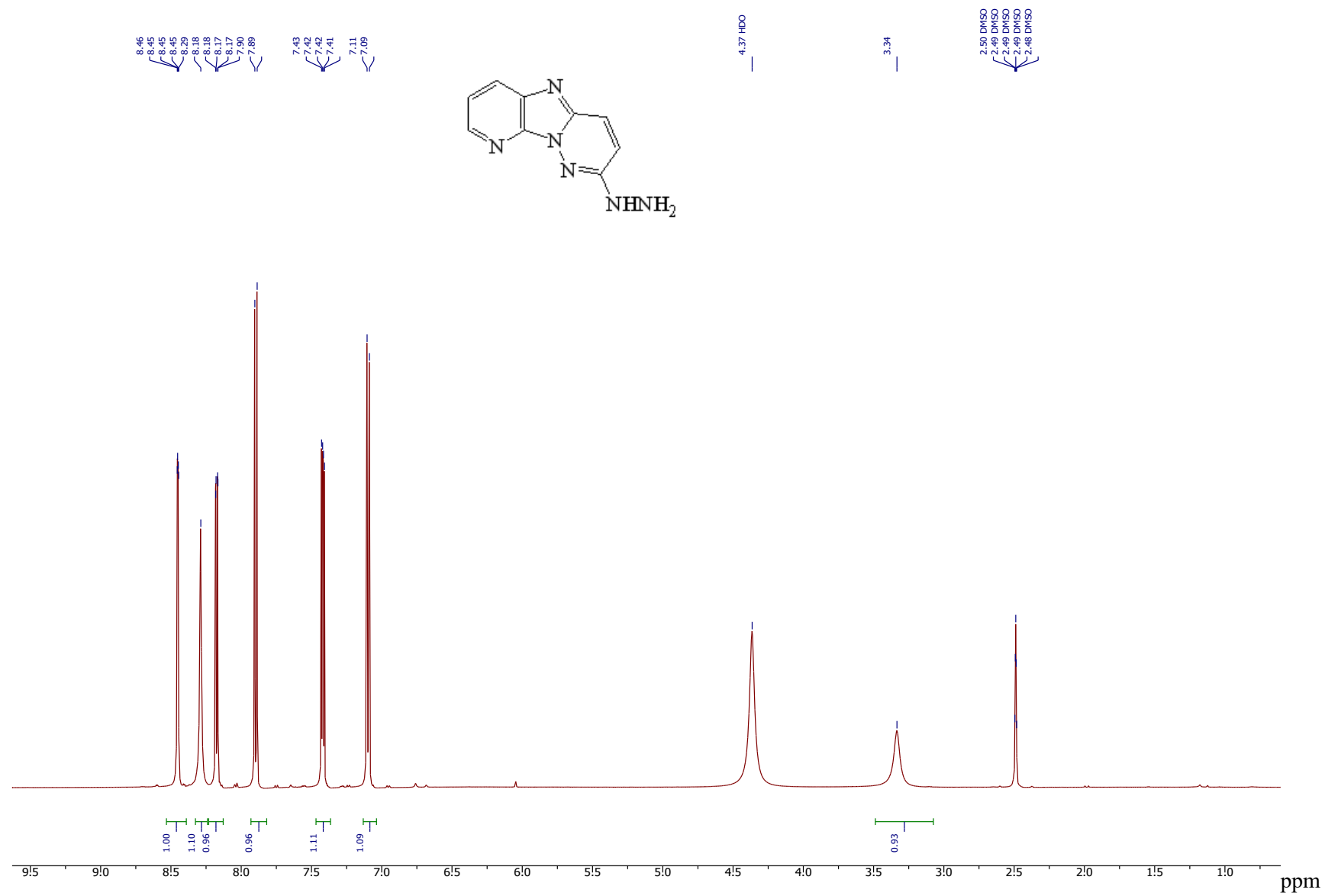
2-Methoxypyrido[3',2':4,5]imidazo[1,2-*b*]pyridazine (7a).

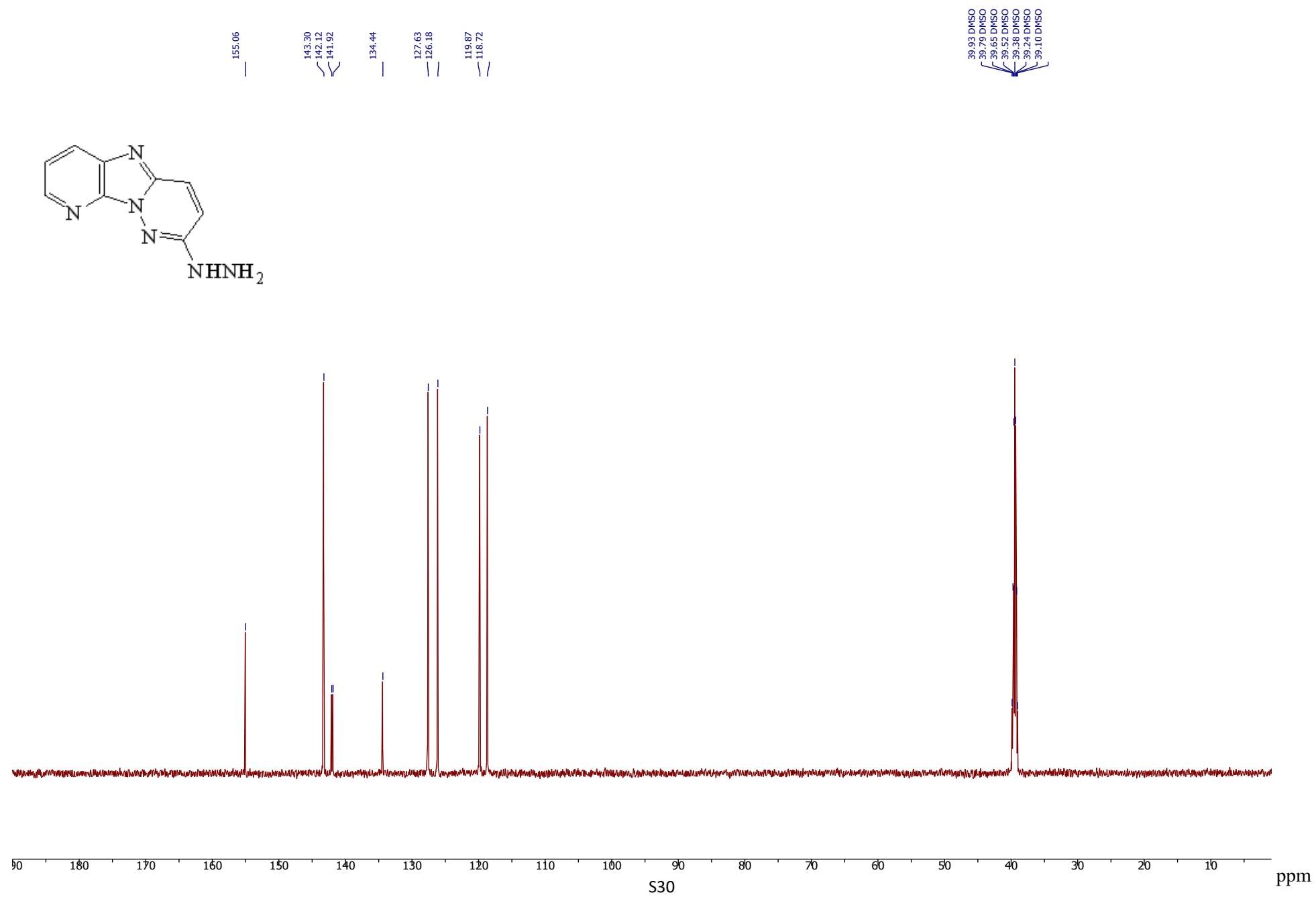
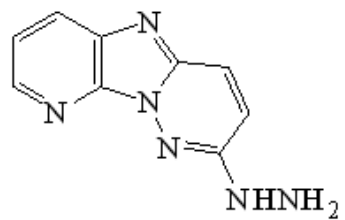


2-(Phenylthio)pyrido[3',2':4,5]imidazo[1,2-*b*]pyridazine (7b).

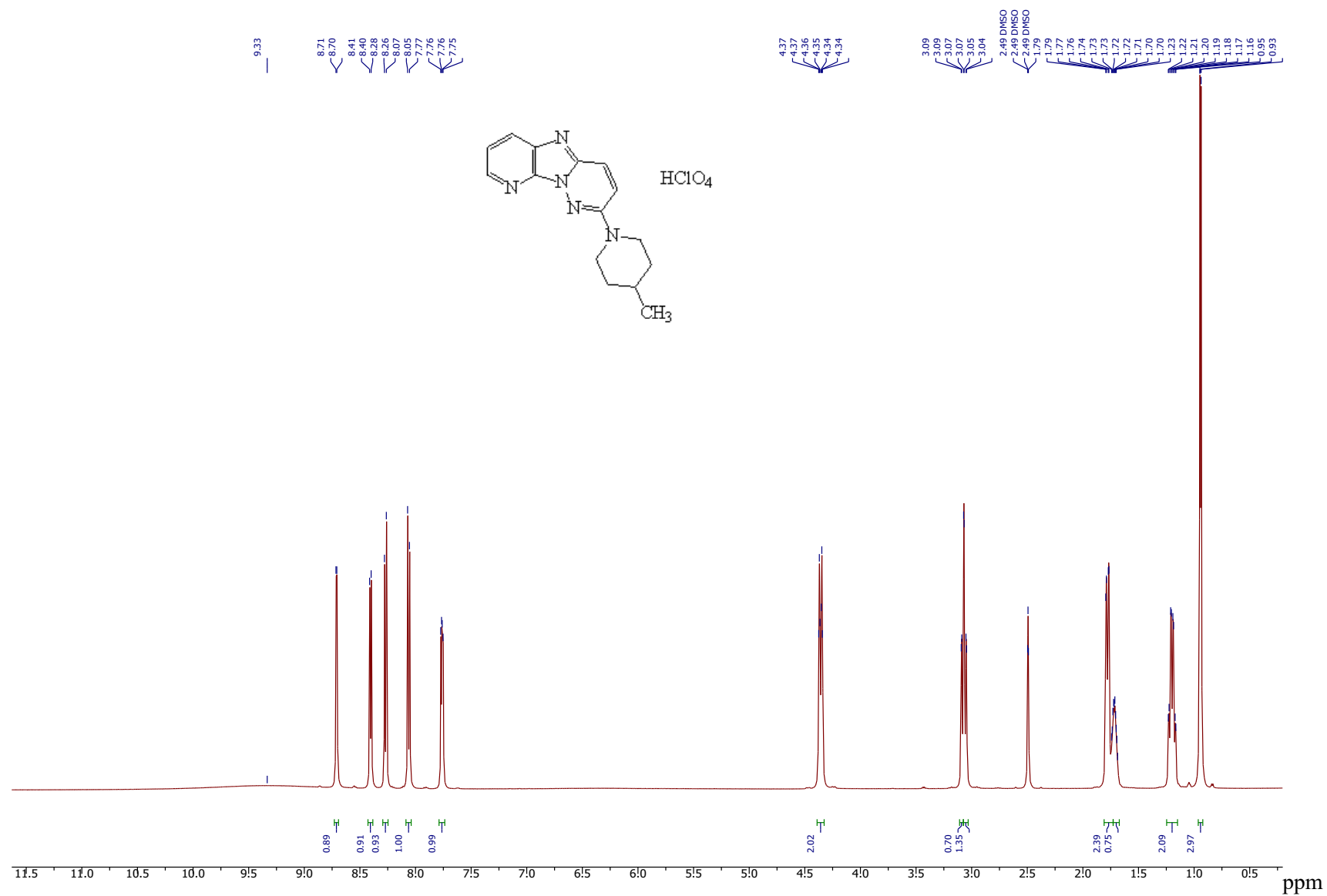


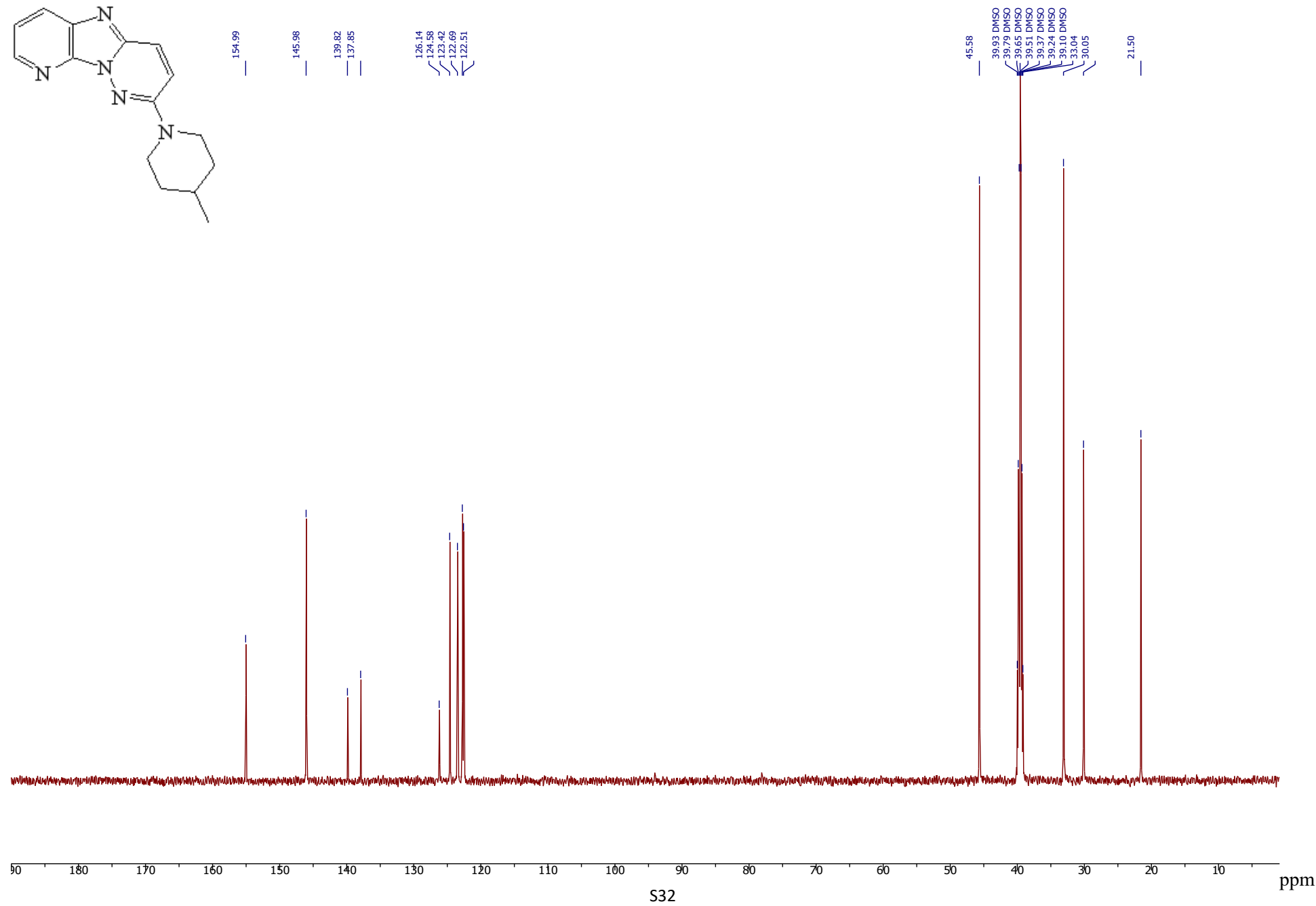
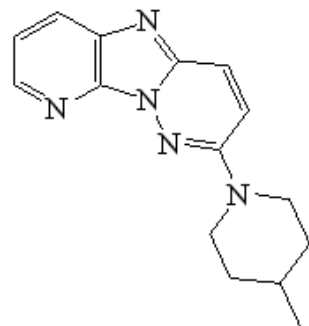
2-Hydrazinylpyrido[3',2':4,5]imidazo[1,2-*b*]pyridazine (7c).



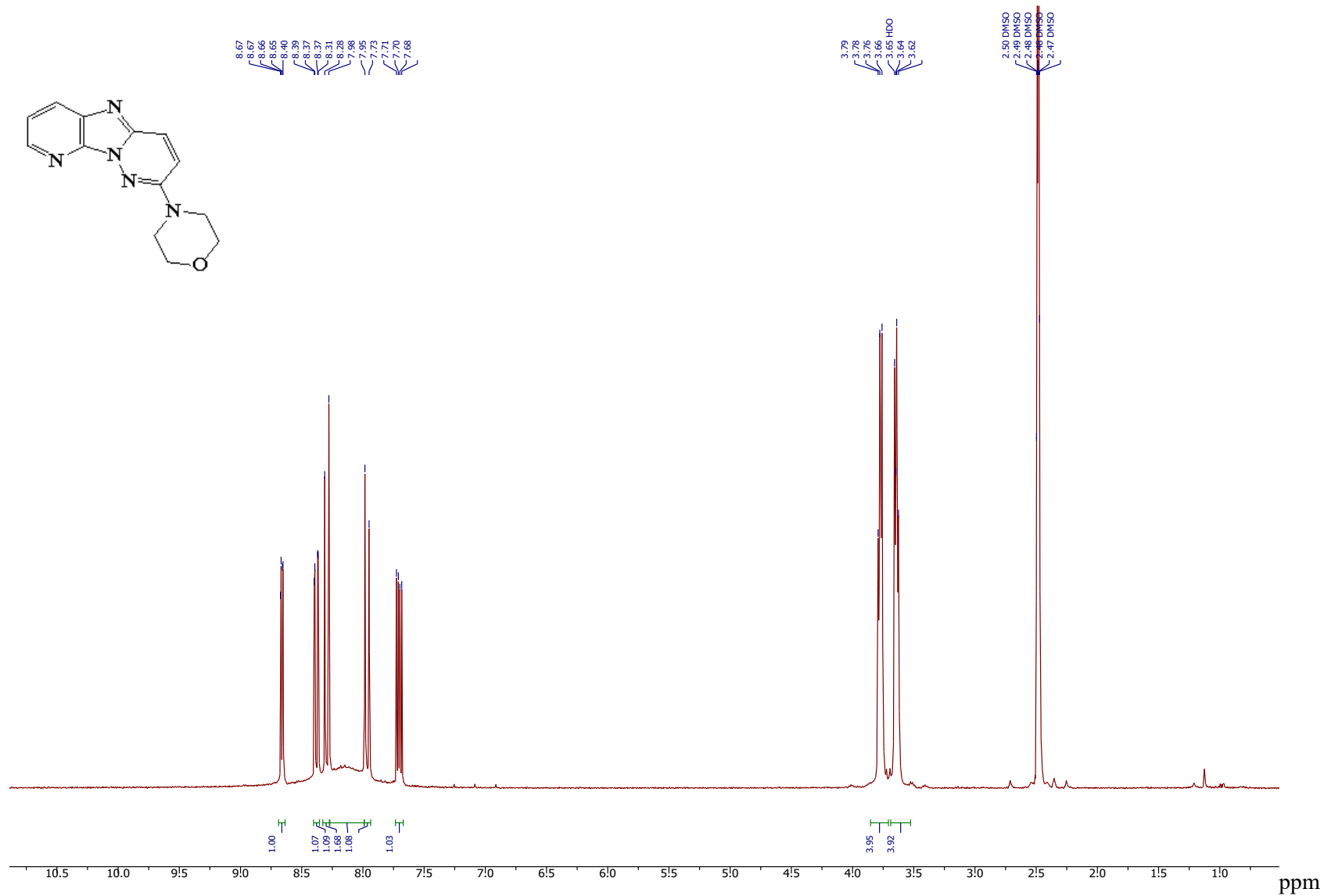


2-(4-Methylpiperidin-1-yl)pyrido[3',2':4,5]imidazo[1,2-*b*]pyridazine hydroperchlorate (7d·HClO₄).

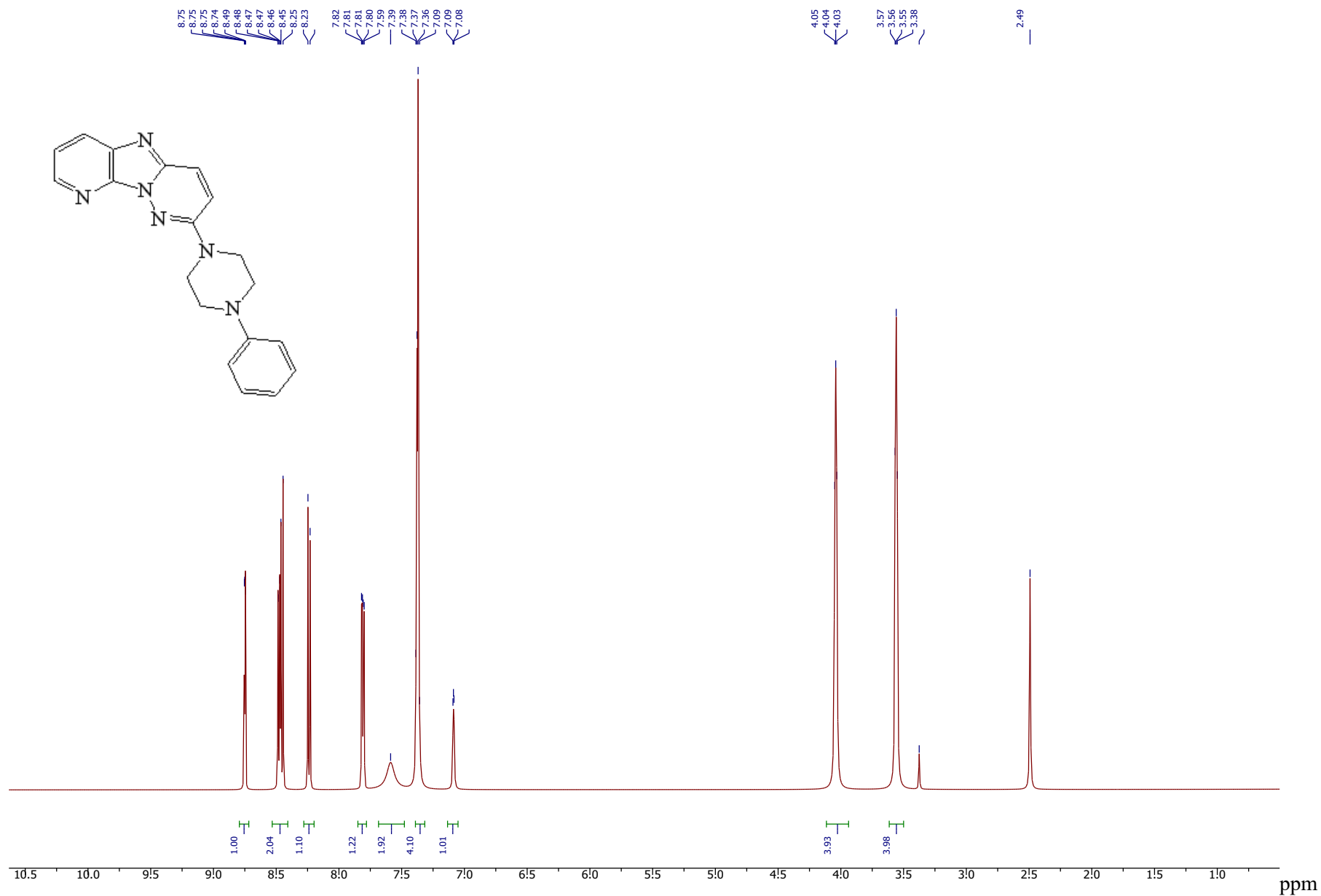


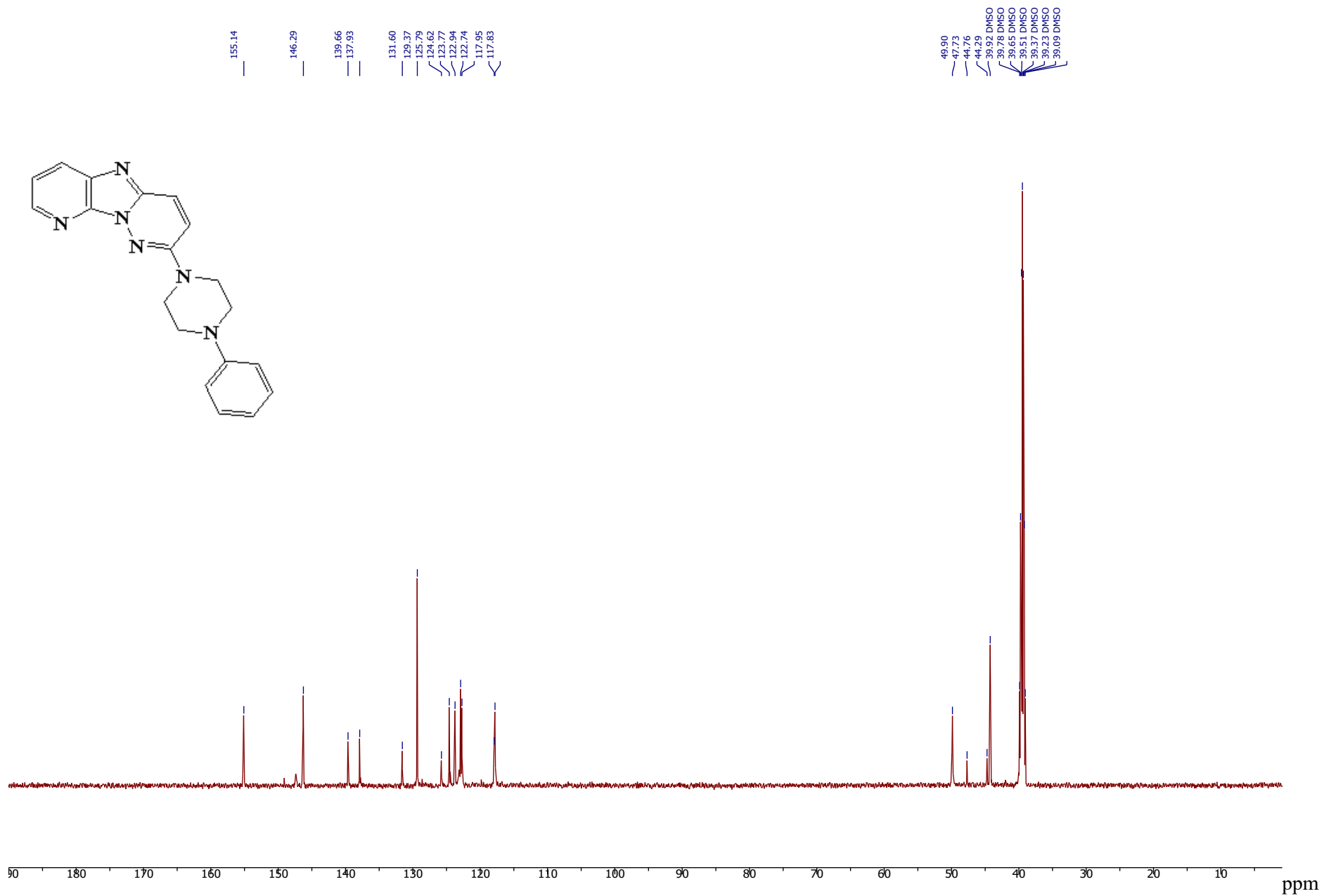


2-(Morpholin-4-yl)pyrido[3',2':4,5]imidazo[1,2-*b*]pyridazine dihydrosulfate (7e·H₂SO₄).



2-(4-Phenylpiperazin-1-yl)pyrido[3',2':4,5]imidazo[1,2-*b*]pyridazine dihydrobromide (7f·2HBr).





5. References

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- S4. M. W. Schmidt, K. K. Baldridge, J. A. Boatz, S. T. Elbert, M. S. Gordon, J. H. Jensen, S. Koseki, N. Matsunaga, K. A. Nguyen, S. Su, T. L. Windus, M. Dupuis and J. A. Montgomery, Jr., *J. Comput. Chem.*, 1993, **14**, 1347; <https://doi.org/10.1002/jcc.540141112>.