

Quasiperiodic to bursting oscillations transition in the model of the Belousov–Zhabotinsky reaction

Oleg V. Noskov,^a Aleksandr D. Karavaev,^{*a} Valery P. Kazakov^a and Semen I. Spivak^b

^aInstitute of Organic Chemistry, Ufa Scientific Centre of the Russian Academy of Sciences, 450054 Ufa, Russian Federation.
Fax: +7 347 235 6066; e-mail: lang@chemorg.bashkiria.su

^bDepartment of Mathematics, Bashkirian State University, 450074 Ufa, Russian Federation.

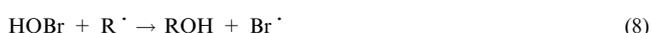
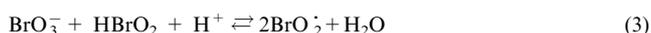
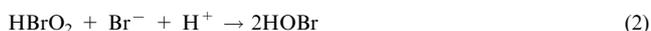
Complex dynamics of the BZ-oscillator: the transition from quasiperiodic to bursting regimes was simulated based on an eleven-stage reaction scheme; the transition was realized in two forms: (i) smooth deformation of the T²-tore due to its interaction with an unstable stationary point; (ii) a sharp change over between the regimes.

Experimental studies of the auto-oscillation Belousov–Zhabotinsky (BZ) reaction have revealed a variety of dynamics: from simple quasiperiodic oscillations to quasiperiodic, bursting, complex periodic and various chaotic ones.^{1–4}

Some types of these regimes were reproduced with more or less success in various models^{1,5,6} based on oversimplified schemes. However, studies of the finer effects (in particular transitions between regimes) were faced with some difficulties. Thus⁶ quasiperiodic and mixed-mode oscillations were simulated and studied in detail, but the continuous transition between them was not seen. Recently^{7,8} we showed that, based on the detailed eleven-stage scheme of Ruoff and Noyes, not only can all known types of BZ oscillations be modelled but the experimentally observable alternation order could also be reproduced:

SS – QS – QP – B – CP – B – QP – QS – SS,
where SS = stationary state; QS = quasiperiodic oscillations of low amplitudes; QP = quasiperiodic regimes; B = bursting oscillations; CP = a complex succession of chaotic and periodic regimes. This allowed us to study the mechanisms of the BZ dynamic transformations in more detail. This paper concerns the transition from quasiperiodicity to bursting.

The reaction scheme under consideration⁹ is as follows:



where RH stands for malonic acid, and Mⁿ⁺ for metal ions.

As in refs. 7–9, modification of the closed system with permanent the initial concentrations of BrO₃⁻ and RH were calculated which allow one to obtain stationary regimes. Calculations were carried out by the m,k-method¹⁰ with a relative precision of ε_r = 10⁻⁵–10⁻⁶ and an integration step of 0.1–1 s. The constant k₈ was used as a variable parameter which determines the rate of Br⁻ formation. The other parameters used are as follows:⁸

$$k_1 = 2.1 \text{ dm}^9 \text{ mol}^{-3} \text{ s}^{-1}; k_{-1} = 1 \times 10^4 \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1};$$

$$k_2 = 3 \times 10^6 \text{ dm}^6 \text{ mol}^{-2} \text{ s}^{-1};$$

$$k_3 = 42 \text{ dm}^6 \text{ mol}^{-2} \text{ s}^{-1}; k_{-3} = 4.2 \times 10^7 \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1};$$

$$k_4 = 8 \times 10^4 \text{ dm}^6 \text{ mol}^{-2} \text{ s}^{-1}; k_{-4} = 8.9 \times 10^3 \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1};$$

$$k_5 = 3 \times 10^3 \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1};$$

$$k_6 = 8 \times 10^9 \text{ dm}^6 \text{ mol}^{-2} \text{ s}^{-1}; k_{-6} = 110 \text{ s}^{-1};$$

$$k_7 = 4.6 \times 10^{-3} \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1};$$

$$k_8 = 10^6\text{--}10^7 \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1};$$

$$k_9 = 10^6 \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1};$$

$$k_{10} = 0.2 \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1};$$

$$k_{11} = 3.2 \times 10^9 \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1};$$

$$[\text{Br}^-]_0 = 10^{-5} \text{ mol dm}^{-3}; [\text{M}^{n+}]_0 = 5 \times 10^{-4} \text{ mol dm}^{-3};$$

$$\text{constants: } [\text{BrO}_3^-] = 0.08 \text{ mol dm}^{-3}; [\text{RH}] = 0.2 \text{ mol dm}^{-3};$$

$$[\text{H}^+] = 1 \text{ mol dm}^{-3}.$$

There are two areas of each kind of oscillation's existence in the model, corresponding to the oxidized (with low k₈ values) and reduced (higher k₈ values) form of the catalyst. Accordingly, two areas of quasiperiodic oscillations (with toroidal attractor — T²-tore¹¹ — in a phase space) and two forms of transition to bursting oscillations — 'mild' and 'rigid' are observed.

Figures 1 and 2 describe the first case. Quasiperiodic regimes, arising in the model as a result of secondary Hopf bifurcation, are considered to be two-frequency and look like high frequency oscillations modulated by a low frequency component on the kinetic curves. Moving from the bifurcation point (when k₈ increases) the amplitude of a second frequency grows which leads to an increase in the modulation depth (Figure 1). Thus, phase portraits demonstrate a decrease of the T²-tore internal diameter. Simultaneously, the tore increases and approaches an unstable saddle focus stationary point (Figure 2). This evolution is caused by their interaction ('attraction' of phase trajectories by a stable manifold of the saddle focus and 'rejection' by a two-dimensional unstable one). The approach of the phase point to the stationary point is accompanied by a slowing down of its movement, and it begins to spend more and more time in the vicinity of the latter. Subsequently, sloping areas are observed on kinetic curves whose length rises as k₈ increases up to the appearance of pronounced bursting oscillations (Figure 3). This process coincides with the formation of an homoclinic structure¹² in the system, causing complex organization of phase space and chaotic character of bursting attractors (Figure 3c). The continuous character of the quasiperiodic regimes evolution described above is in good agreement with experimental observations.⁴

Thus, the left hand side of the diagram shows a smooth transition without abrupt changes of the attractor's structure. Similar T²-tore evolution was observed previously in simpler model.⁶ However, as the authors of ref. 6 noted, they could not monitor the transition to bursting oscillations due to calculation difficulties.

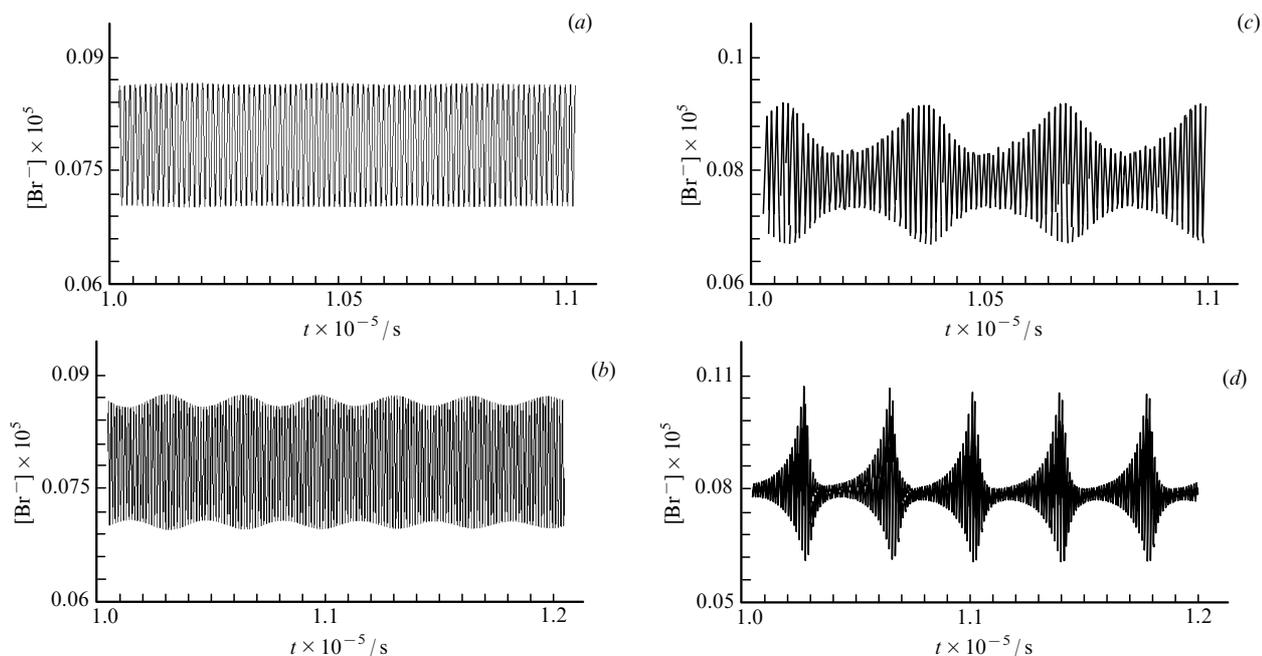


Figure 1 Kinetic curves of quasiperiodic oscillations. $k_8 \times 10^{-6} = 2.117$ (a); 2.118 (b); 2.119 (c); 2.120 (d) $\text{dm}^3 \text{mol}^{-1} \text{s}^{-1}$.

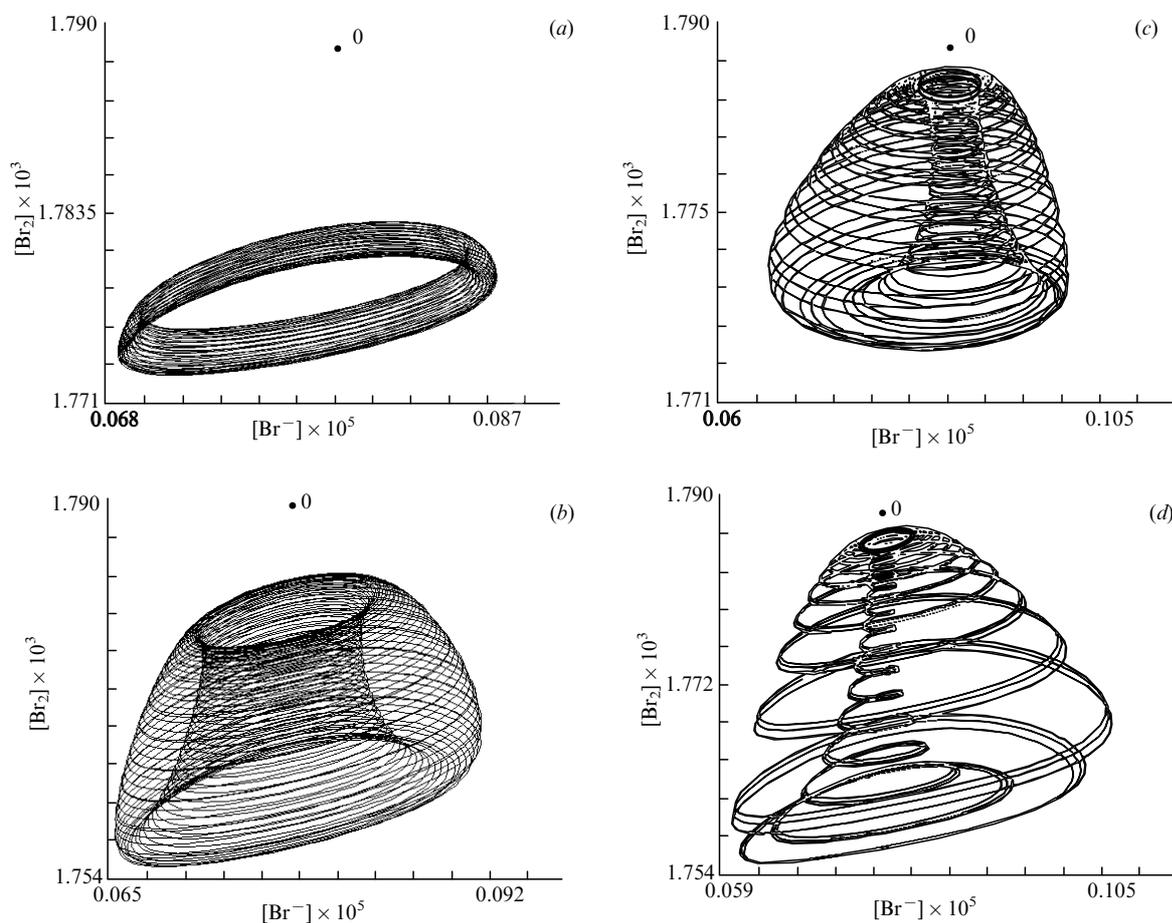


Figure 2 Tore transformation during 'mild' transition to bursting oscillations. $k_8 \times 10^{-6} = 2.117$ (a); 2.118 (b); 2.119 (c); 2.120 (d) $\text{dm}^3 \text{mol}^{-1} \text{s}^{-1}$. 0 is an unstable stationary point (saddle focus).

In the right hand side of the diagram this transition is very sharp. The quasiperiodic and bursting regimes in Figure 4 were obtained with k_8 values differing by less than $10^{-3}\%$. Probably, these oscillating states coexist in a range of k_8 values, and when the parameter changes the sharp transition from one regime to another occurs. The existence area for

quasiperiodic regimes was calculated to be much less (by 1–2 orders of magnitude) than that for bursting oscillations. Apparently, this is why the direct transition from bursting to stationary state, by passing quasiperiodicity, is often observed in experiments.^{3,4}

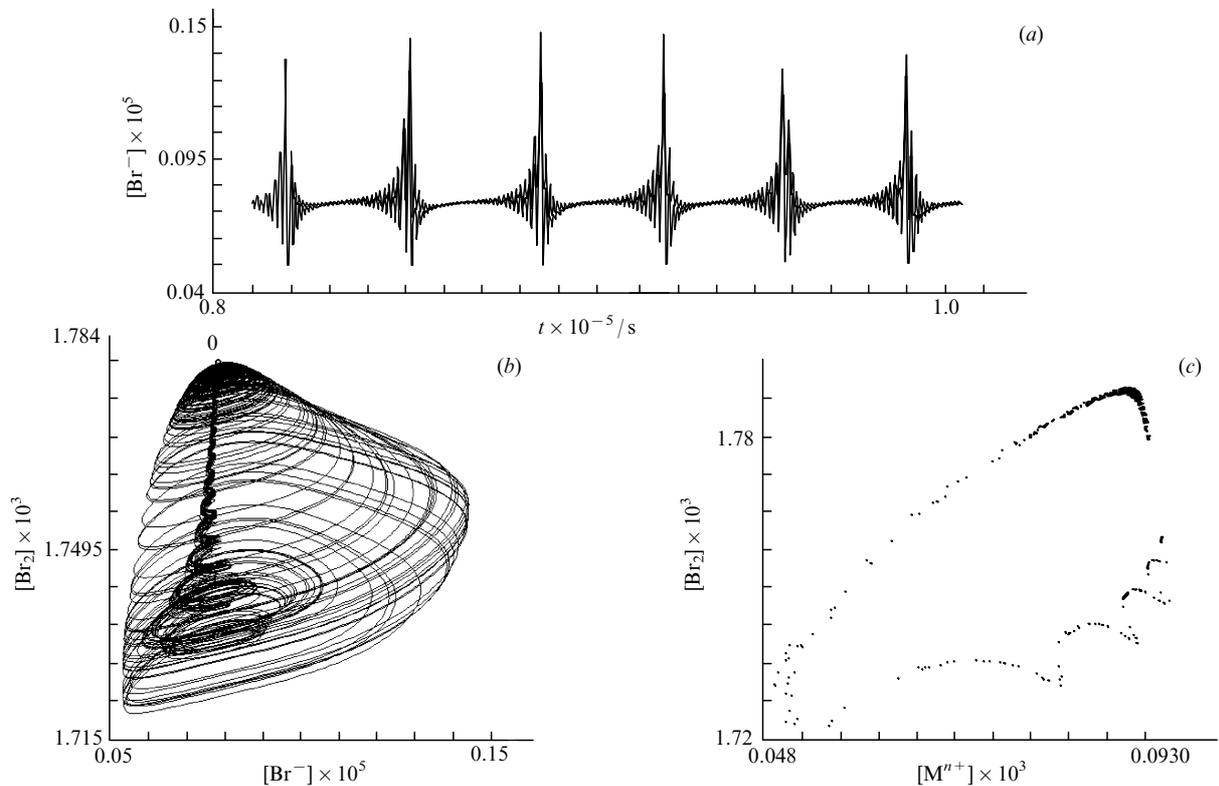


Figure 3 Kinetics (a), phase portrait (b), and Poincaré section (c) of bursting oscillations. ($k_8 = 2.14 \times 10^6 \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1}$, 0 is unstable stationary point).

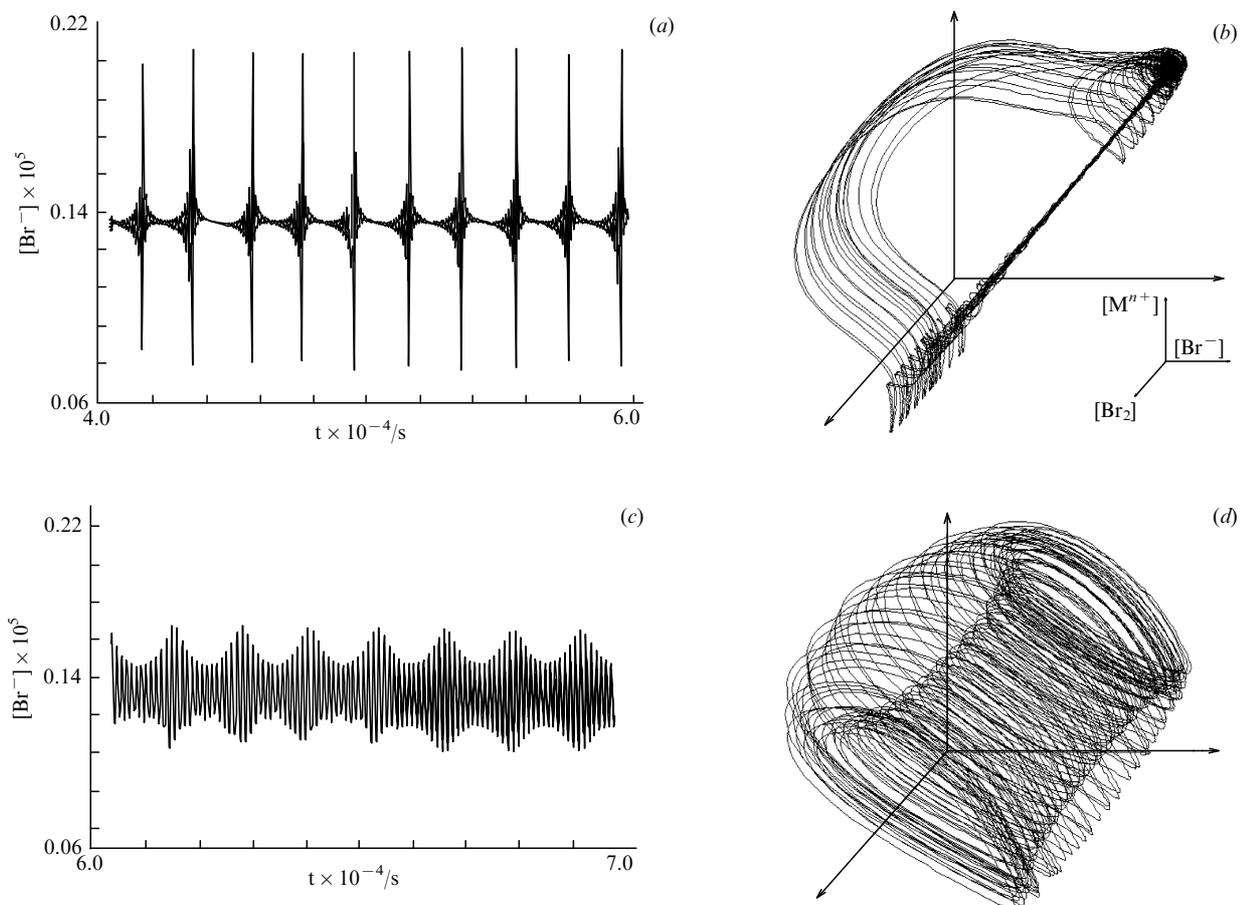


Figure 4 'Rigid' transition between bursting [(a), (b), $k_8 = 7.370306040 \times 10^6 \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1}$] and quasiperiodic [(c), (d), $k_8 = 7.370306041 \times 10^6 \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1}$] regimes. (a), (c) kinetic curves; (b), (d) phase portraits.

This research was supported by a grant from the Russian Foundation for Basic Research (no. 93-03-18165).

References

- 1 *Kolebaniya i begushchie volny v khimicheskikh sistemakh (Oscillations and traveling waves in chemical systems)*, eds. R. Field and M. Burger, Mir, Moscow, 1988 (in Russian).
- 2 F. Argoul, A. Arneodo and P. Richetti, *Acc. Chem. Res.*, 1987, **20**, 436.
- 3 F. W. Schneider and A. F. Munster, *J. Phys. Chem.*, 1991, **95**, 2130.
- 4 F. Argoul, A. Arneodo, P. Richetti and J. C. Roux, *J. Chem. Phys.*, 1987, **86**, 3325.
- 5 P. Richetti, J. C. Roux, F. Argoul and A. Arneodo, *J. Chem. Phys.*, 1987, **86**, 3339.
- 6 D. Barkley, J. Ringland and J. S. Turner, *J. Chem. Phys.*, 1987, **87**, 3812.
- 7 O. V. Noskov, A. D. Karavaev, S. I. Spivak and V. P. Kazakov, *Kinet. Katal.*, 1992, **33**, 704 [*Kinet. Catal. (Engl. Transl.)*, 1992, **33**, 567].
- 8 O. V. Noskov, A. D. Karavaev, V. P. Kazakov and S. I. Spivak, *Mendeleev Commun.*, 1994, 82.
- 9 P. Ruoff and R. M. Noyes, *J. Chem. Phys.*, 1986, **84**, 1413.
- 10 E. A. Novikov and Yu. A. Shitov, Reprint N 20, Computing Centre of the Siberian Branch of the Russian Academy of Sciences, Krasnoyarsk, 1988 (in Russian).
- 11 H. G. Schuster, *Deterministic chaos*, Physik-Verlag, Weinheim, 1984.
- 12 U. I. Neimark and L. P. Shil'nikov, *Dokl. Akad. Nauk SSSR*, 1965, **160**, 1261 (in Russian).

Received: Moscow, 10th January 1996
Cambridge, 22nd April 1996; Com. 6/00312E