

Synthesis of σ -Heterocyclic Iron Complexes Based on [3+3] Cycloaddition to 1,3-Dipolar Reagents

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Treatment of iron complex ($\eta^5\text{-C}_5\text{H}_5$)Fe(CO)₂CH₂C≡CPh **1** with nitrogen-containing 1,3-dipolar reagents (nitrile oxides and azomethine ylide) yields six-membered σ -heterocyclic iron complexes, due to cycloaddition and subsequent 1,2-migration of the ($\eta^5\text{-C}_5\text{H}_5$)₂Fe(CO) group.

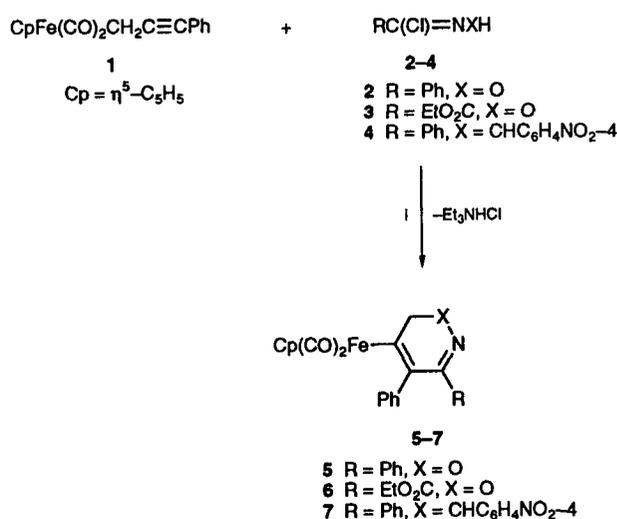
Cycloaddition reactions of transition metal σ -complexes with unsaturated ligands are known to provide useful methods for the synthesis of organometallic compounds with σ -carbo- and σ -heterocyclic ligands.^{1–5}

It has been found previously that σ -propargyl iron complex ($\eta^5\text{-C}_5\text{H}_5$)Fe(CO)₂CH₂C≡CPh **1** in reactions with 1,2-dipoles gives the products of [3 + 2] cycloaddition with subsequent 1,2-migration of the organometallic group.¹ The reactions of transition metal propargyl complexes with 1,3-dipoles were previously unknown.

We have carried out the reactions of **1** with nitrogen-containing 1,3-dipolar compounds generated *in situ*: benzonitrile oxide (from **2**), ethoxycarbonylnitrile oxide (from **3**) and benzonitrile(4-nitrobenzyl) ylide (from **4**). We have found that all these reactions proceed as [3 + 3] cycloadditions giving six-membered heterocycles, σ -bonded with the iron atom (Scheme 1).

One may suggest that migration of the CpFe(CO)₂ group proceeds on formation of the products and, apparently, π -allene intermediate **A** is formed and subsequently cyclizes into σ -oxazines **5** and **6** or σ -dihydropyridine **7** derivatives of iron (Scheme 2).

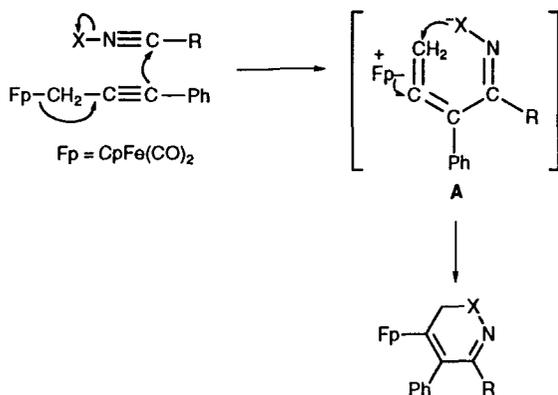
We failed to observe the formation of five-membered rings as possible isomeric products in these reactions; this would be characteristic of [3 + 2] interactions of 1,3-dipolar reagents with the triple bond in **1**, as is well-known for simple alkynes.⁶ It is known that complex CpFe(CO)₂C≡CPh, in which migration of the Cp(CO)₂Fe group is impossible, reacts with nitrile oxides to give isoxazole derivatives of iron.^{7,8}



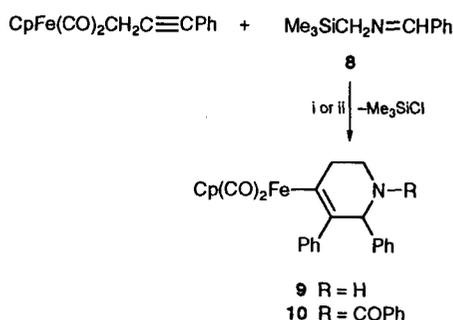
Scheme 1 Reagents and conditions: i, Et₃N, C₆H₆, 40–45 °C, 1 h.

We have also found that complex **1** reacts with 1,3-dipolar ylide reagents (generated *in situ* from *N*-[(trimethylsilyl)methyl]imine **8** under the action of HCl or PhCOCl) to give σ -tetrahydropyridine complexes of iron (Scheme 3).

Recently it has been shown⁹ that this ylide reacts with simple alkynes to form five-membered *N*-substituted heterocycles.



Scheme 2



Scheme 3 Reagents and conditions: i, HCl/Et₂O, THF, 20 °C, 0.5 h; ii, PhCOCl, THF, 20 °C, 0.5 h.

The ability of the σ -propargylic ligand to play the role of a three-carbon component in reactions with 1,3-dipolar compounds depends on the high tendency of the organometallic group to migration and π -complexation. The reactions investigated are the first examples of [3+3] cycloaddition for transition-metal propargyl complexes.

All the new complexes gave satisfactory analytical and spectroscopic data.[†]

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[†] **5**: yield 15%; m.p. 114–115 °C; MS: m/z 306 [M – CO, Ph]⁺; IR ν/cm^{-1} : 2050, 2005 (C≡O); ¹H NMR (200 MHz) (C₆D₆): δ 2.15 (s, 2 H, CH₂), 4.04 (s, 5 H, C₅H₅), 7.2–7.6 (m, 10 H, 2 Ph).

6: yield 21%; m.p. 110–111 °C; MS: m/z 306 [M – CO, CO₂Et]⁺; IR ν/cm^{-1} : 2053, 2005 (C≡O), 1740 (C=O); ¹H NMR (200 MHz) (C₆D₆): δ 0.85 (t, 3 H, Me, *J* 7.3 Hz), 3.95 (q, 2 H, CH₂, *J* 7.3 Hz), 2.90 (s, 2 H, CH₂), 4.04 (s, 5 H, C₅H₅), 7.10–7.40 (m, 5 H, Ph).

7: yield 27%; m.p. 117–118 °C; MS: m/z 410 [M – C₆H₄NO₂]⁺; IR ν/cm^{-1} : 2050, 2010 (C≡O), ¹H NMR (200 MHz) (C₆D₆): δ 3.99 (m, 1 H, CH), 4.13–4.21 (m, 2 H, CH₂), 4.04 (s, 5 H, C₅H₅), 6.8–7.9 (m, 14 H, 3Ph).

9: yield 45%; m.p. 102–103 °C; MS: m/z 410 [M – H]⁺; IR ν/cm^{-1} : 2050, 2005 (C≡O), ¹H NMR (200 MHz) (C₆D₆): δ 1.44 (m, 2 H, CH₂C=C), 2.84 (s, 1 H, CH), 3.08 (m, 2 H, CH₂N), 4.02 (s, 5 H, C₅H₅), 5.65 (s, sh, 1 H, NH), 7.19–7.50 (m, 10–2 Ph).

10: yield 50%; m.p. 122–123 °C; MS: m/z 410 [M – C(O)Ph]⁺; IR ν/cm^{-1} : 2060, 2010 (C≡O), 1730 (C=O); ¹H NMR (200 MHz) (C₆D₆): δ 1.87 (m, 2 H, CH₂C=C), 2.69 (s, 1 H, CH), 2.89 (m, 2 H, CH₂N), 4.04 (s, 5 H, C₅H₅), 7.2–7.8 (m, 15 H, 3 Ph).