

## Synthesis of 2-Arylcarboxylic Acids and C-Norbenzomorphans via $\eta^6$ -Arenetricarbonylchromium Complexes

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A method for the regioselective carboxylation of ( $\eta^6$ -alkylarene)tricarbonylchromium complexes at the benzylic position has been developed via the reaction of the corresponding lithium derivatives with carbon dioxide in tetrahydrofuran; a method for the preparation of 5-acetoxy-3-benzyl-1,4-methano-2,3-4,5-tetrahydro-1H-3-benzazepine from 1-tetralol is described.

Complexation of  $\text{Cr}(\text{CO})_3$  with alkylarenes produces a large increase in the CH-acidity of the hydrogen atoms at the benzylic position of the aromatic ligands.<sup>1</sup> This allows selective activation of the benzylic CH-bonds via  $\alpha$ -metallation of ( $\eta^6$ -alkylarene)tricarbonylchromium complexes with lithium amides in tetrahydrofuran (THF)<sup>2</sup> or sodium amide in liquid ammonia.<sup>3</sup> These are useful methods for organic synthesis. We describe here the use of benzylic lithium derivatives of ( $\eta^6$ -alkylarene)tricarbonylchromium complexes for the preparation of 2-arylcarboxylic acids and C-norbenzomorphans.

We have found that benzylic lithium derivatives of ( $\eta^6$ -alkylarene)tricarbonylchromium complexes react easily with carbon dioxide giving, after acidification the corresponding tricarbonylchromium complexes of  $\alpha$ -arylcarboxylic acids in good yields (Scheme 1).

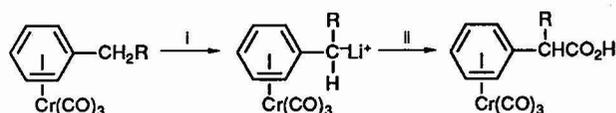
The carboxylation reaction is applicable to the preparation of various  $\alpha$ -arylcarboxylic acid complexes and their esters, e.g. 1–8.

The free acids can be produced from the complexes thus obtained by the usual methods.<sup>1†</sup>

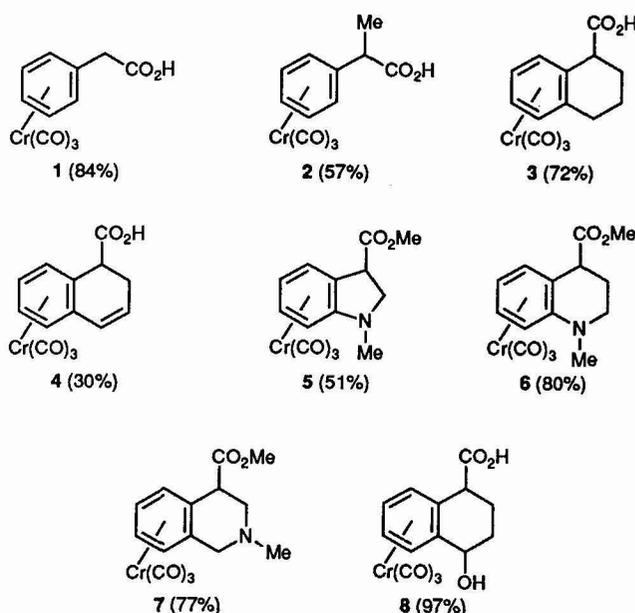
† Typical experimental procedure. An alkylarene tricarbonylchromium complex (1.0 mmol) was added to a solution of  $\text{Et}_2\text{NLi}$  (2.0 mmol) in 10 ml THF (prepared from 2.0 mmol of  $\text{Et}_2\text{NH}$  and 2.0 mmol of  $\text{Bu}^n\text{Li}$ ). The mixture was stirred at 20°C for 5 min. After cooling the solution down to –20°C,  $\text{Bu}^n\text{Li}$  (1.0 mmol) was added again. After 5 min stirring the mixture was poured onto an excess of solid  $\text{CO}_2$ . After the  $\text{CO}_2$  had escaped, the solvent was evaporated under reduced pressure and the residue dissolved in 5% aq. NaOH (30 ml). The solution was washed with diethyl ether (3 × 10 ml), acidified with 37% aq. HCl to pH 1.0 and extracted with diethyl ether (3 × 10 ml). The diethyl ether extracts were combined and dried over  $\text{Na}_2\text{SO}_4$ . Diethyl ether was evaporated and the product recrystallized from diethyl ether–hexane (2 : 1).

For the heterocyclic derivatives 5–7 the reaction mixture was acidified after expulsion of  $\text{CO}_2$  with 37% aq. HCl (1.0 ml) and evaporated under reduced pressure. The dry residue was worked up with an excess of  $\text{CH}_2\text{N}_2$  in diethyl ether solution for 60 min and the diethyl ether evaporated. Chromatography of the residue on  $\text{Al}_2\text{O}_3$  [hexane–diethyl ether (3 : 1)] followed by recrystallization from hexane–diethyl ether (3 : 1) afforded a pure product as yellow crystals.

All the tricarbonylchromium complexes of carboxylic acids or their methyl esters gave satisfactory analytical and  $^1\text{H}$  NMR data.

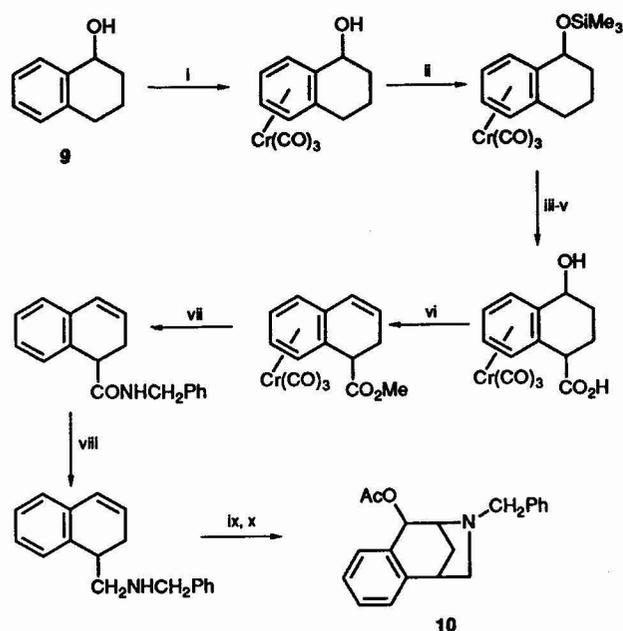


Scheme 1 Reagents and conditions: i,  $\text{LiNEt}_2$ , THF; ii,  $\text{CO}_2$  then  $\text{H}^+$



2-Arylcarboxylic acids are of interest as antiinflammatory agents and analgesics,<sup>4,5</sup> and much attention is paid to methods for their preparation.<sup>4,6</sup> The method of carboxylation described above permits easy introduction of a carboxyl group directly into the saturated hydrocarbon chain of acyclic, carbocyclic and heterocyclic fragments.

In addition to their pharmacological activity, 2-arylcarboxylic acids are also interesting as synthons in fine organic



**Scheme 2** Reagents and conditions: i,  $\text{Cr(CO)}_6$ ,  $\text{Bu}^n_2\text{O}$ , heat, 30 h (90%); ii,  $\text{Me}_3\text{SiCl}$ , py, THF,  $20^\circ\text{C}$ , 5 h (91%); iii,  $\text{LiNEt}_2$ , THF,  $20^\circ\text{C}$ , 10 min; iv,  $\text{CO}_2$ ; v,  $\text{H}^+$  (96%); vi,  $\text{MeOH}$ ,  $\text{H}_2\text{SO}_4$ , heat, 2 h (60%); vii,  $\text{PhCH}_2\text{NH}_2$ ,  $180^\circ\text{C}$ , 9 h (51%); viii,  $\text{LiAlH}_4$ ,  $\text{AlCl}_3$ , THF,  $\text{Et}_2\text{O}$ ,  $0^\circ\text{C}$ , 5 h (69%); ix,  $\text{Hg(OAc)}_2$ , THF,  $\text{H}_2\text{O}$ ,  $0^\circ\text{C}$ , 3 h; x,  $\text{NaBH}_4$ ,  $\text{KOH}$ ,  $0^\circ\text{C}$ , 1 h (35%); py = pyridine

synthesis. We have used this approach for the preparation of 5-acetoxy-3-benzyl-1,4-methano-2,3,4,5-tetrahydro-1H-3-benzazepine **10** (Scheme 2) from easily available 1-tetralol **9**. The compound is a representative of the major class of physiologically active compounds known as C-norbenzomorphans.<sup>7</sup>

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