

## A Novel Approach to Carbacycline

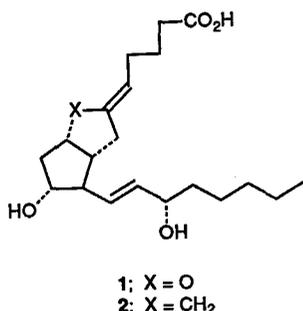
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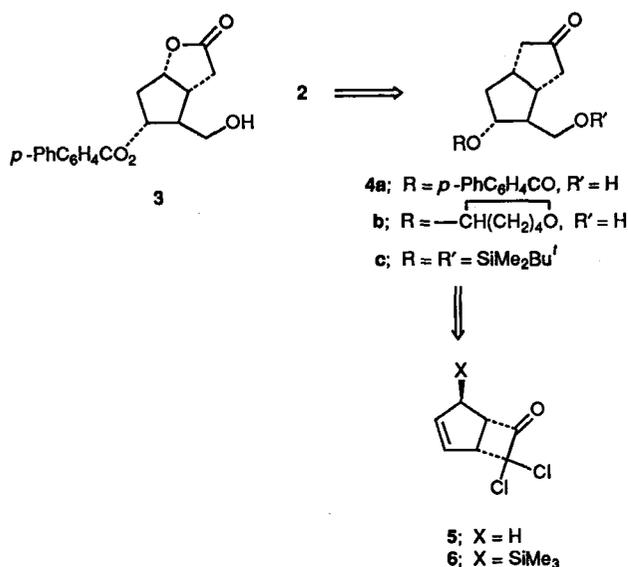
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A scheme for ( $\pm$ )-carbacycline synthesis has been developed, based on 7,7-dichlorobicyclo[3.2.0]hept-3-en-7-one.

The 6,9- $\alpha$ -carbanalogues of the highly labile prostacycline **1**, usually called carbacyclines (*e.g.* **2**), are chemically more stable than **1** and are similar to prostacycline in the nature of their bioactivity.<sup>1</sup> After purposeful modification of the side chains and ring framework of the carbacycline, analogues with similar and even greater activities than that of prostacycline have been found in antiaggregation dough. A series of intensive investigations in that field was undertaken to develop carbacycline cardiovascular and antithrombosis preparations.<sup>2–5</sup>



The chemical synthesis of the carbacyclines has been published elsewhere.<sup>1</sup> Most of the ways proposed for the preparation of carbacyclines may be considered as adaptations of a versatile prostaglandin scheme devised by Corey<sup>6</sup> using the key lactone **3**, and they were carried out with perhydropentalene equivalents of **3**: the functionalized bicyclo[3.3.0]octanes **4**. However, the known methods for the synthesis of carbacycline synthons **4a, b** are multistage and rather laborious.<sup>7–9</sup> A



Scheme 1

novel solution of this problem, based on the available bicycloheptenones **5** and **6**, is described in this communication. As shown in the retrosynthetic scheme (Scheme 1), we planned to transform **5** and **6** into the desired bicyclo[3.3.0]octane intermediates **4** in two stages: (i) regio- and stereo-controlled 'vicinal oxy-oxymethylation' of the double bonds using the Prins reaction, and (ii) insertion of methylene into the cyclobutane rings.

Initially, according to the established strategy, the possibility of functionalization of alkenes **5** and **6** by the Prins reaction was studied. Boiling a solution of alkene **5** and 15–20 equiv. CH<sub>2</sub>O in AcOH and concentrated H<sub>2</sub>SO<sub>4</sub> (50:1 by volume) for 70 h produced an oily mixture of compounds **7**, **8**, **9** and **10** in a 20:10:1:1 ratio, respectively, which were isolated separately by crystallization and column chromatography on SiO<sub>2</sub>, in a total yield of 90% (Scheme 2).<sup>††</sup>

The Prins reaction of allylsilane **6** proceeded in a different way. Under similar conditions as for **5** (10 h boiling), the main reaction products were monoacetates **11** and **12** in a 4:1 ratio (40%), alkenes **5** (20%) and polycyclic compounds **10** (15%); the minor diacetates **14** and **15** were formed in an 8:2 ratio in insignificant amounts (5%). It should be noted that in an alternative approach, electrophilic oxymethylation of **6** gave only the tricyclic adduct **16** in 50% yield on treatment with 1,3,5-trioxane–TiCl<sub>4</sub> (CH<sub>2</sub>Cl<sub>2</sub>, –78 °C→20 °C, 1 h) (Scheme 3).

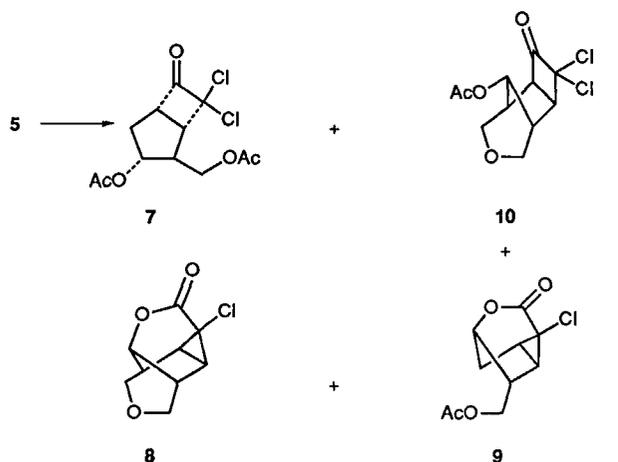
<sup>†</sup> Characterization data for 3 $\alpha$ -acetoxy-2 $\beta$ -acetoxymethylene-7,7-dichlorobicyclo[3.2.0]heptan-6-one **7**: IR  $\nu_{\max}/\text{cm}^{-1}$  1240, 1720–1750, 1805; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.95 (s, 3H, OAc), 2.95 (t, 1H, *J* 6.35 Hz, C<sup>5</sup>H), 3.25 (dd, 1H, *J* 6.28, 2 Hz, C<sup>1</sup>H), 4.10–4.40 (m, 2H, CH<sub>2</sub>O), 5.20 (m, 1H, C<sup>3</sup>H); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  20.77 (Me), 20.90 (Me), 34.93 (C<sup>4</sup>), 48.15 (C<sup>5</sup>), 55.20 (C<sup>2</sup>), 60.35 (C<sup>1</sup>), 63.89 (CH<sub>2</sub>O), 78.73 (C<sup>3</sup>), 86.92 (C<sup>7</sup>), 169.85 (OAc), 170.57 (OAc), 195.52 (C=O).

7,9-Carbolactone-3-oxa-7-chlorotricyclo[3.3.1.0<sup>6,8</sup>]nonane **8**: m.p. 122–123 °C; IR  $\nu_{\max}/\text{cm}^{-1}$  1750, 3010, 3070; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  2.40–2.43 (m, 4H, C<sup>1</sup>H, C<sup>5</sup>H, C<sup>6</sup>H, C<sup>8</sup>H), 3.58 (d, 2H, C<sup>2</sup>H, C<sup>4</sup>H, <sup>3</sup>J<sub>AB</sub> 12 Hz), 3.98 (dd, 2H, C<sup>2</sup>H, C<sup>4</sup>H, <sup>2</sup>J<sub>BA</sub> 12, <sup>3</sup>J<sub>RX</sub> 2.5 Hz), 4.34–4.37 (m, 1H, C<sup>9</sup>H); <sup>13</sup>C NMR (CDCl<sub>3</sub>):  $\delta$  29.48 (C<sup>1</sup>, C<sup>5</sup>), 39.22 (C<sup>6</sup>, C<sup>8</sup>), 40.46 (C<sup>7</sup>), 67.74 (C<sup>2</sup>, C<sup>4</sup>), 73.37 (C<sup>9</sup>), 167.65 (CO<sub>2</sub>); MS *m/z* 200 (M<sup>+</sup>), 172 (M–CO), 170 (M–CH<sub>2</sub>O), 165 (M–Cl), 142 (M–CH<sub>2</sub>CO<sub>2</sub>), 126, 107, 100, 81, 79, 77, 55, 44 (CO<sub>2</sub>).

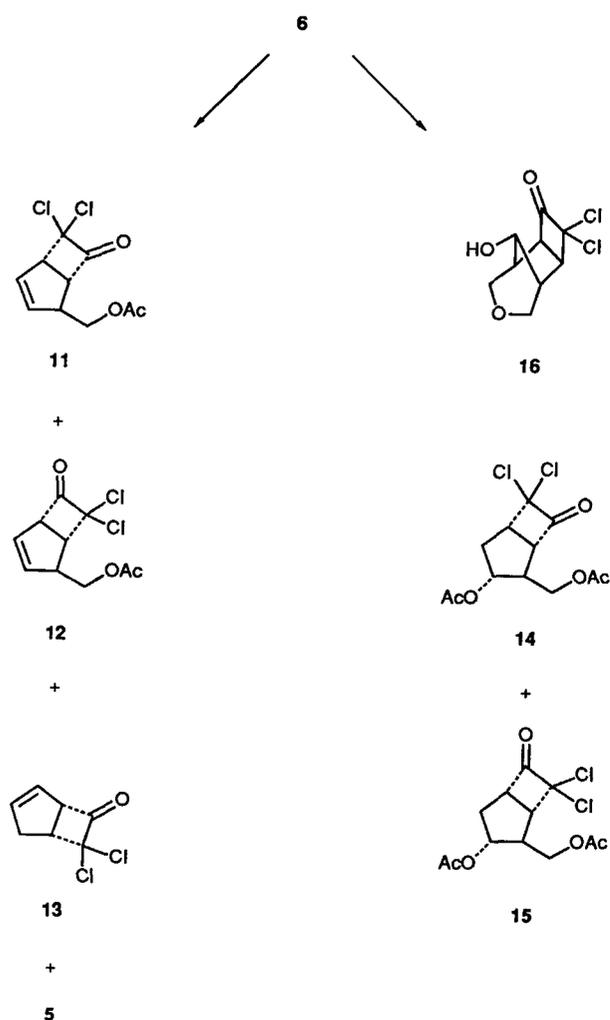
6,3-Carbolactone-2 $\beta$ -acetoxymethylene-6-chlorobicyclo[3.1.0]hexane **9**: IR  $\nu_{\max}/\text{cm}^{-1}$  1240, 1730, 3020, 3080; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  2.05 (s, 3H, OAc), 4.55 (m, 1H, OCHO); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  20.69 (Me), 29.57 (C<sup>1</sup>), 29.57 (C<sup>4</sup>), 31.46 (C<sup>5</sup>), 42.95 (C<sup>6</sup>), 43.41 (C<sup>2</sup>), 67.14 (CH<sub>2</sub>O), 76.63 (C<sup>3</sup>), 167.11 (CO), 170.56 (OAc).

2 $\beta$ -Acetoxymethylene-6,6-dichlorobicyclo[3.2.0]hept-3-en-7-one **11**: IR  $\nu_{\max}/\text{cm}^{-1}$  1270, 1750, 1810, 3080; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.96 (s, 3H, OAc), 3.26 (m, 1H, C<sup>2</sup>H), 3.86 (dd, 1H, *J* 11.1, 6.6 Hz, C<sup>8</sup>H), 3.93 (dd, 1H, *J* 11.1, 5.9 Hz, C<sup>9</sup>H), 4.02 (m, 1H, C<sup>5</sup>H), 4.09 (d, 1H, *J* 7.06 Hz, C<sup>1</sup>H), 5.82 (m), 5.90 (m, 2H, CH=CH); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  20.53 (OAc), 47.95 (C<sup>2</sup>), 58.77 (C<sup>5</sup>), 61.67 (C<sup>1</sup>), 88.44 (C<sup>6</sup>), 64.95 (CH<sub>2</sub>O), 130.65 (C<sup>4</sup>), 136.75 (C<sup>3</sup>), 170.34 (OAc), 195.72 (C=O). MS *m/z*: 248 (M<sup>+</sup>), 206 (M–COCH<sub>2</sub>), 188 (M–AcOH).

<sup>††</sup> Satisfactory elemental analyses were obtained for compounds **8** and **16**.



Scheme 2



Scheme 3

Thus, as a result of our investigation of the Prins reaction of alkenes **5** and **6**, a series of adducts was prepared, from which diacetate **8** and the isomeric monoacetates **11** and **12** were easily isolated from the reaction mixtures by single column chromatography on  $\text{SiO}_2$  in 40–50% yields.† These three compounds were of interest for further study.

The next stage of our investigation included a study of the homologation of the cyclobutane fragment of these com-

pounds. Because of the anomalously high reactivity of diacetate **7** towards  $\text{CH}_2\text{N}_2$  (the reaction is difficult to control), the direct transformation **7**→**17** was accomplished in only a moderate yield (40%), hence, at the ring-expansion stage the less reactive monochloro derivative **18** was used. After complete dechlorination of the intermediate **19**, the yield of the final diacetate **17** was more than 60% with respect to **7**. In contrast to diacetate **7**, the isomeric bicycloheptenones **11** and **12** reacted readily with  $\text{CH}_2\text{N}_2$  to form the corresponding homologues **16** and **20** which, after treatment with Zn in AcOH, led to the same acetate **22** in a total yield of 80%. Stereo- and regio-selective addition of an OH-group at position 3 of alkene **22** was carried out by an initial transformation of the latter to the bromhydrin **23** and then by radical dehalogenation with  $\text{Bu}_3\text{SnH}$  to monoacetate **24**, and further to diacetate **17** (Scheme 4).§

Of the two approaches developed, the preparation of the key diacetate **17** from **7** was shown to be more efficient. The diacetate **17** is considered to be a novel equivalent of retrones **4**, and its preparation facilitated the final stages of the synthesis by providing a solution to the problem of forming side chains. Further, we have described a short and chemorational way to transform diacetate **17** into carbacycline **2**. For this purpose, using versatile procedures, diacetate **17** was transformed into bis(triethylsilyl) ether **25** and an acid **27** (identified as the methyl ester **28**) was prepared by alkenation with phosphorane **26** (Scheme 5).<sup>10</sup> The application of our recently developed method for direct oxidation of silyl ethers by the  $(\text{COCl})_2$ -dimethylsulfoxide (DMSO) system<sup>11</sup> generated siloxyaldehyde **29**, which was then condensed with dimethyl-2-oxoheptylphosphonate<sup>12</sup> according to Emmons–Horner to form a 5(*Z,E*)-isomeric (2:1) mixture of enones **30** in a total yield of 20% based on ether **25**. After the removal of the silane protecting group, the isomeric oxenones **31** were separated by HPLC. Reduction of an individual enone **30a** by the Yamamoto reagent,<sup>13</sup> followed by saponification, gave carbacycline **2** and its 15 $\beta$ -isomer in a ratio of 7:3, which were separated by column chromatography on  $\text{SiO}_2$ .§

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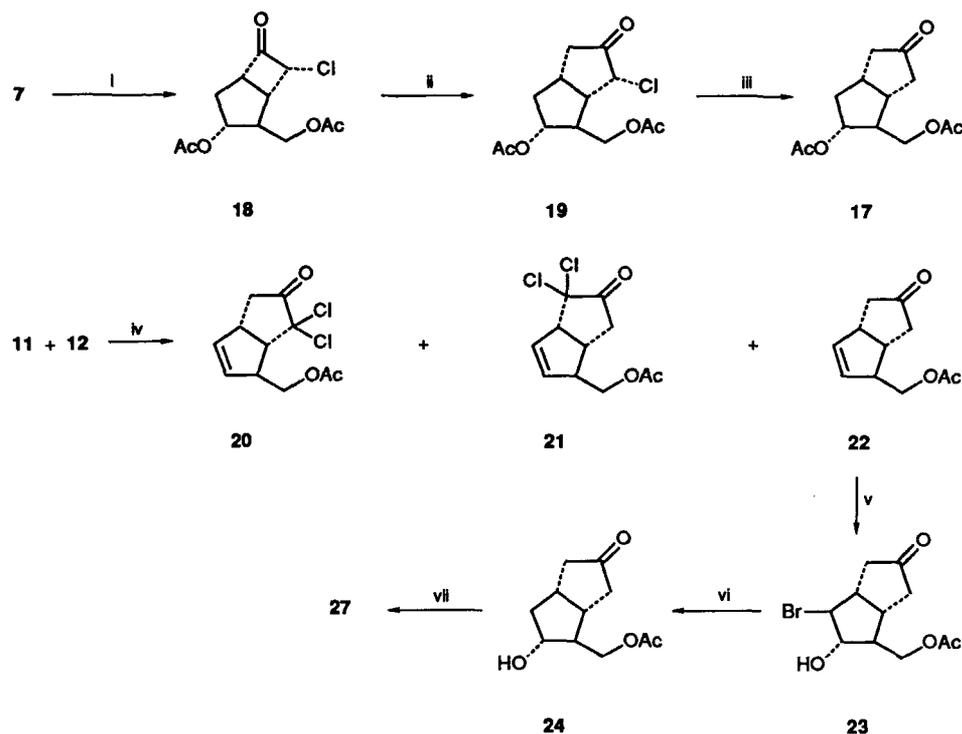
§ Characterization data for 3 $\alpha$ -acetoxy-2 $\beta$ -acetoxyethylene-7-*endo*-chlorobicyclo[3.2.0]heptan-6-one **18**: IR  $\nu_{\text{max}}/\text{cm}^{-1}$  1750, 1790; <sup>1</sup>H NMR ( $\text{CDCl}_3$ )  $\delta$  1.99 (s, 3H, OAc), 2.09 (3H, OAc), 4.10 (d, 2H, *J* 6.6 Hz,  $\text{CH}_2\text{O}$ ), 5.26 (m, 1H,  $\text{C}^3\text{H}$ ); <sup>13</sup>C NMR ( $\text{CDCl}_3$ )  $\delta$  20.30 (Me), 20.56 (Me), 34.40 ( $\text{C}^4$ ), 39.69 ( $\text{C}^5$ ), 44.72 ( $\text{C}^2$ ), 59.73 ( $\text{C}^1$ ), 63.78 ( $\text{CH}_2\text{O}$ ), 62.60 ( $\text{C}^7$ ), 78.64 ( $\text{C}^3$ ), 169.59 (OAc), 203.27 ( $\text{C}^6$ ).

(1*R*\*,5*S*\*,6*R*\*,9*S*\*,10*R*\*)-8,8-dichloro-10-hydroxy-3-oxatricyclo-[7.2.1<sup>1,5</sup>.0<sup>6,9</sup>]undecan-7-one **16**: m.p. 131–132 °C (hexane); IR  $\nu_{\text{max}}/\text{cm}^{-1}$  1790; <sup>1</sup>H NMR ( $\text{CDCl}_3$ )  $\delta$  2.97 (br s, 1H,  $\text{C}^9\text{H}$ ), 3.01 (br s, 1H,  $\text{C}^5\text{H}$ ), 3.58 (d, 1H, *J* 7.72 Hz,  $\text{C}^8\text{H}$ ), 3.93 (dd, 1H, *J* 11.1, 3.07 Hz,  $\text{C}^4\text{H}$ ), 3.78 (dd, 1H, *J* 11.1, 2.56 Hz,  $\text{C}^2\text{H}$ ), 3.62 (br s, 0.5H), 3.69 (br s, 1H), 3.79 (br s, 0.5H,  $\text{C}^9\text{H}_2$ ); <sup>13</sup>C NMR ( $\text{CDCl}_3$ )  $\delta$  48.95 ( $\text{C}^1$ ), 57.17 ( $\text{C}^6$ ), 55.51 ( $\text{C}^2$ ), 63.02 ( $\text{C}^3$ ), 65.06 ( $\text{C}^{10}$ ), 73.04 ( $\text{C}^7$ ), 74.74 ( $\text{C}^8$ ).

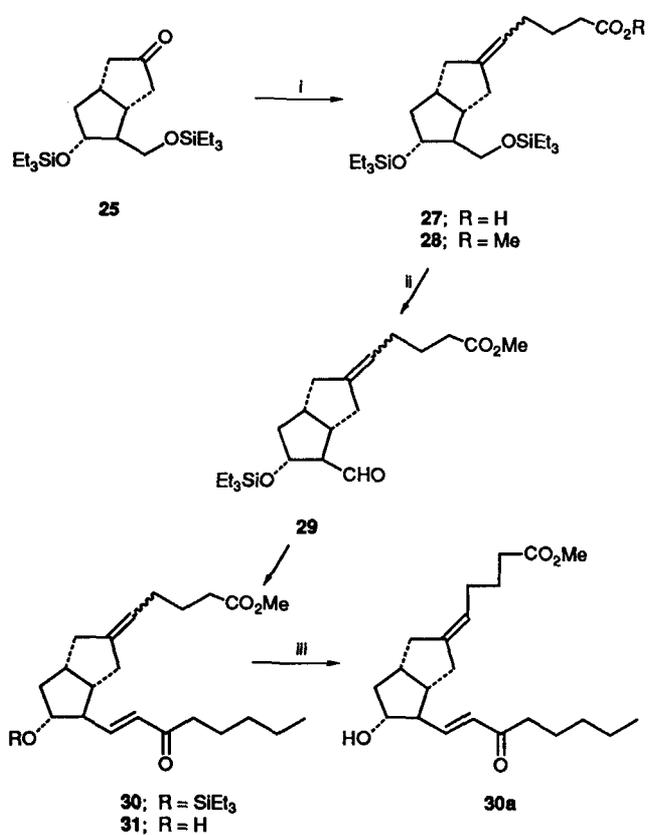
3 $\alpha$ -Acetoxy-2 $\beta$ -acetoxyethylenebicyclo[3.3.0]octan-7-one **17**: IR  $\nu_{\text{max}}/\text{cm}^{-1}$  1240, 1736; <sup>1</sup>H NMR ( $\text{CDCl}_3$ )  $\delta$  1.60 (dt, 1H,  $\text{C}^4\text{H}$ , <sup>2</sup>*J* 14, <sup>3</sup>*J* 6 Hz), 2.03 (s, 3H, OAc), 2.06 (s, 3H, OAc), 2.08–2.63 (m, 7H, 2 $\text{CH}_2$ ,  $\text{C}^4\text{H}$ ,  $\text{C}^1\text{H}$ ,  $\text{C}^3\text{H}$ ), 2.86 (m, 1H,  $\text{C}^2\text{H}$ ), 4.12 (d, 2H, *J* 6.2 Hz,  $\text{CH}_2\text{O}$ ), 5.38 (dt, 1H,  $\text{C}^3\text{H}$ , <sup>3</sup>*J* 6.1, <sup>3</sup>*J* 6.4 Hz); <sup>13</sup>C NMR ( $\text{CDCl}_3$ )  $\delta$  20.17 (Me), 20.43 (Me), 36.31 ( $\text{C}^5$ ), 37.85 ( $\text{C}^4$ ), 40.56 ( $\text{C}^1$ ), 43.25 ( $\text{C}^6$ ), 43.98 ( $\text{C}^8$ ), 51.47 ( $\text{C}^2$ ), 63.61 ( $\text{CH}_2\text{O}$ ), 77.01 ( $\text{C}^3$ ), 169.75 (OAc), 170.12 (OAc), 218.01 ( $\text{C}^7$ ). MS *m/z*  $\text{M}^+$  254.

3 $\alpha$ -Triethylsilyloxy-2 $\beta$ -triethylsilyloxymethylenebicyclo[3.3.0]octan-7-one **25**: IR  $\nu_{\text{max}}/\text{cm}^{-1}$  740, 800, 1000, 1240, 1410, 1460, 1740; <sup>1</sup>H NMR ( $\text{CDCl}_3$ )  $\delta$  4.05–4.15 (m, 1H,  $\text{C}^3\text{H}$ ), 3.59 (d, 2H, *J* 5 Hz,  $\text{CH}_2\text{O}$ ); <sup>13</sup>C NMR ( $\text{CDCl}_3$ )  $\delta$  4.50 (t,  $\text{SiCH}_2$ ), 4.83 (t,  $\text{SiCH}_2$ ), 14.16 (Me), 36.55 ( $\text{C}^4$ ), 40.14 ( $\text{C}^5$ ), 42.29 ( $\text{C}^1$ ), 44.51 ( $\text{C}^8$ ), 45.53 ( $\text{C}^6$ ), 57.70 ( $\text{C}^2$ ), 62.86 ( $\text{C}^9$ ), 75.26 ( $\text{C}^3$ ), 220.70 ( $\text{C}^7$ ).

(±)-6- $\alpha$ -Methanoprostaglandin **1**: <sup>1</sup>H NMR ( $\text{CDCl}_3$ )  $\delta$  0.87 (t, 3H, *J* 6.3 Hz, Me), 1.20–2.50 (m, 23H, 10 $\text{CH}_2$ , 3CH), 3.50–4.30 (m, 2H, 2CH), 5.16 (m, 1H,  $\text{C}^5\text{H}$ ), 5.40–5.60 (m, 2H,  $\text{CH}=\text{CH}$ ); <sup>13</sup>C NMR ( $\text{CDCl}_3$ )  $\delta$  177.01 ( $\text{C}^1$ ), 33.14 ( $\text{C}^2$ ), 24.68 ( $\text{C}^3$ ), 28.63 ( $\text{C}^4$ ), 121.01 ( $\text{C}^5$ ), 142.84 ( $\text{C}^6$ ), 36.03 ( $\text{C}^7$ ), 45.39 ( $\text{C}^8$ ), 37.63 ( $\text{C}^9$ ), 41.68 ( $\text{C}^{10}$ ), 77.30 ( $\text{C}^{11}$ ), 57.00 ( $\text{C}^{12}$ ), 133.07 ( $\text{C}^{13}$ ), 135.51 ( $\text{C}^{14}$ ), 73.49 ( $\text{C}^{15}$ ), 37.22 ( $\text{C}^{16}$ ), 25.29 ( $\text{C}^{17}$ ), 31.81 ( $\text{C}^{18}$ ), 22.69 ( $\text{C}^{19}$ ), 14.08 ( $\text{C}^{20}$ ), 38.26 ( $\text{C}^{21}$ ).



**Scheme 4** Reagents and conditions: i, Zn-AcOH, 20 °C; ii, 5 equiv.  $\text{CH}_2\text{N}_2$ ; iii, Zn-AcOH, 40 °C; iv, excess of  $\text{CH}_2\text{N}_2$ ; v, 1.2 equiv. *N*-bromosuccinimide (NBS), tetrahydrofuran (THF)- $\text{H}_2\text{O}$  (9:1); vi, 2 equiv.  $\text{Bu}_3\text{SnH}$ , azoisobutyronitrile,  $\text{C}_6\text{H}_6$ , 70%; vii,  $\text{Ac}_2\text{O}$ , py



**Scheme 5** Reagents and conditions: i, (1)  $\text{Ph}_3\text{P}=\text{CH}(\text{CH}_2)_3\text{CO}_2\text{Na}$  **26**, (2)  $\text{CH}_2\text{N}_2$ ; ii, see ref. 11; iii, HPLC

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