

Does the ^{57}Fe Mössbauer isomer shift depend on the oxidation state of iron?

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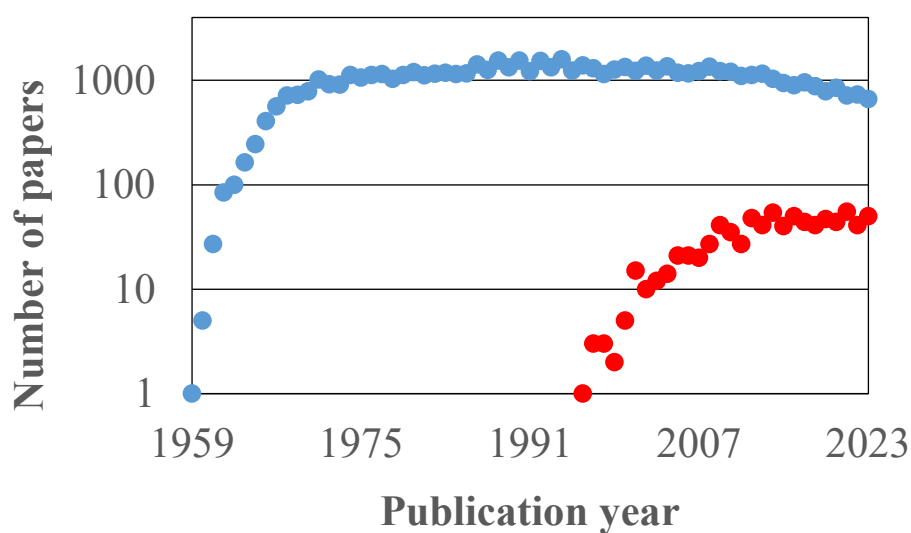


Figure S1 Publication dynamics of journal papers containing the words ‘Mössbauer’ (blue dots) or ‘Mössbauer + DFT’ (red dots) in their abstracts or keywords, based on data from CAS SciFinder (<https://scifinder-n.cas.org/>).

Table S1 Room-temperature isomer shifts for selected iron oxo compounds.^a

Formula	ICSD ^b deposition number	IS ^c /mm s ⁻¹
K ₃ Na(FeO ₄) ₂	159909	−0.89
K ₂ FeO ₄	32756	−0.90
Cs ₂ FeO ₄	33861	−0.87
K ₃ FeO ₄	65977	−0.56
Na ₄ FeO ₄	59585	−0.22
Na ₅ FeO ₄	2485	0.12
Cs ₉ FeO ₄	174308	0.10
LiLa ₂ FeO ₆	252554	−0.41
Sr ₂ FeO ₄	69849	−0.01
SrFeO ₃	91062	0.05
α-NaFeO ₂	420380	0.33
α-Fe ₂ O ₃ (hematite)	245851	0.37
Ca ₃ Fe ₂ (SiO ₄) ₃ (andradite)	203209	0.41
β-NaFeO ₂	14779	0.18
Sr ₂ FeO ₃	251020	0.42
Sr ₃ Fe ₂ O ₅	261512	0.46
SrFeO ₂	418603	0.50
BaFeSi ₄ O ₁₀ (gillespite)	31202	0.77
FeCr ₂ O ₄ (chromite)	250575	0.94
FeO (wustite)	82233	1.08
Fe ₃ Al ₂ (SiO ₄) ₃ (almandine)	80671	1.28

^a The table is adapted from reference S1. ^b ICSD is the Inorganic Crystal Structure Database.^{S2} ^c Isomer shifts are presented relative to α-iron, their uncertainties do not exceed 0.02 mm s⁻¹.

Table S2 Isomer shifts at liquid nitrogen temperature for selected organoiron compounds.^a

Complex	Formula	CCDC ^b deposition number	CCDC identifier	IS ^c /mm s ⁻¹
5	$[(\text{TIMMN}^{\text{Mes}*})\text{Fe}(=\text{N}^*)(\text{NCMe})](\text{PF}_6)_2(\text{MF}_6)$, M=Mo, Re	2258914	XOJPAP	-0.16
1	$[(\text{TIMMN}^{\text{Mes}})\text{Fe}(\text{N})](\text{PF}_6)$	2020254	AKEJEI	-0.35
2	$[(\text{TIMMN}^{\text{Mes}})\text{Fe}(\text{N})](\text{PF}_6)_2$	2020255	AKEJIM	-0.47
6	$[(\text{TIMMN}^{\text{Mes}*})\text{Fe}(\text{NH}^*)(\text{F})](\text{PF}_6)_3$	2297600	XOJPEV	<i>no data</i>
3	$[(\text{TIMMN}^{\text{Mes}})\text{Fe}(\text{N})(\text{F})](\text{PF}_6)_2$	2258913	XOJNUJ	-0.60
4	$[(\text{TIMMN}^{\text{Mes}})\text{Fe}(\text{N})(\text{F})](\text{PF}_6)_2(\text{PF}_6/\text{MF}_6)$, M=Mo, Re	<i>no structural data</i>		-0.72
	$[(\text{ImMes-})_3\text{-BPh}]\text{Fe}(\text{N-adamantyl})$	616728	MOBNEX	-0.11
	$\{[(\text{ImMes-})_3\text{-BPh}]\text{Fe}(\text{N-adamantyl})\}[\text{BPh}_4]$	631683	MOBNUN	-0.17
	$[(\text{Im}^t\text{Bu-})_3\text{-BPh}]\text{Fe}(\text{N})$	1549741	HAQLAP	-0.32
	$[(\text{ImMes-})_3\text{-BPh}]\text{Fe}(\text{N})$	1549742	HAQLET	-0.35
	$[\{[(\text{ImMes}(\text{C}_2\text{H}_4\text{-}))_3\text{-N}]\text{Fe}(\text{N})\}][\text{BPh}_4]$	676550	TIYJOB	-0.27
	$\{[(\text{Im}^t\text{Bu-})_3\text{-BPh}]\text{Fe}(\text{N})\}[\text{BArF}_{24}]$	786059	APUQIM	-0.45
	$[(\text{ImMes-})_2\text{-BH}_2]\text{Fe}(\text{N-Mes})_2$	1990357	ZACWUZ	-0.25
	$\{[(\text{ImMes-})_2\text{-BH}_2]\text{Fe}(\text{N-Mes})_2\}[\text{BPh}_4]$	1990356	ZACXAG	-0.48
	$\{[(\text{ImMe-})_3\text{-BPh}]_2\text{Fe}\}[\text{PF}_6]_6$	1842079	BIRWAD	-0.05^d
	$\{[(\text{ImMe-})_3\text{-BPh}]_2\text{Fe}\}[\text{PF}_6]_2$	1961137	ERUXIB	-0.23

^a The table is adapted from references S3 and S4. ^b CCDC is the database of the Cambridge Crystallographic Data Centre.^{S5} ^c The isomer shifts are presented relative to α -iron at room temperature, their uncertainties do not exceed 0.02 mm s⁻¹. ^d The isomer shift of BIRWAD was calculated by adding 0.10 mm s⁻¹ to the room-temperature value.

Table S3 Structural information for the previously known compounds.^a

Complex	Formula	Interatomic distances ^b /pm							
		aID _{Fe-X}	Fe-C	Fe-C	Fe-C	Fe-N	Fe-N ^c	Fe-NC	Fe-F
5	[(TIMMN ^{Mes*})Fe(=N*)(NCMe)](PF ₆) ₂ (MF ₆), M=Mo, Re	190	186	198	198	168	<i>217</i>	203	
1	[(TIMMN ^{Mes})Fe(N)](PF ₆)	185	194	196	198	153	<i>233</i>		
2	[(TIMMN ^{Mes})Fe(N)](PF ₆) ₂	182	192	193	193	150	<i>243</i>		
6	[(TIMMN ^{Mes*})Fe(NH*)(F)](PF ₆) ₃	192	179	197	197	175	<i>210</i>		206
3	[(TIMMN ^{Mes})Fe(N)(F)](PF ₆) ₂	186	197	198	199	152	<i>235</i>		187
4	[(TIMMN ^{Mes})Fe(N)(F)](PF ₆) ₂ (PF ₆ /MF ₆), M=Mo, Re	<i>no data</i>							

^a The table is adapted from reference S3. ^b The interatomic distances are rounded to picometers. ^c The Fe-N distances marked in *italics* were not taken for the calculations of the aID_{Fe-X}.

Table S4 Mössbauer isomer shifts and structural information for the hemoglobin and related compounds.^a

Complex	RCSB PDB ^b entry ID or CCDC identifier	IS _{α-Fe} ^d / mm s ⁻¹	X	Interatomic distances ^c /pm						
				aID _{Fe-X}	Fe-X	Non-hem Fe-N	Fe-N in hem			
DeoxyHbA-α	2DN2	0.89		210		221	202	208	210	210
			213		220	208	211	212	212	
DeoxyHbA-b			208		216	204	206	207	208	
			212		219	208	209	210	212	
OxyHbA-α L	2DN1	0.23	O ₂	199	182	207	200	201	202	202
OxyHbA-b L			O ₂	199	178	206	198	199	200	211
CO-HbA-α L	2DN3	0.23	CO	198	171	212	196	203	203	203
CO-HbA-b L			CO	200	174	211	202	203	203	209
porphyrinato-iron(II)	DEDWUE	0.62		200			199	199	201	201
imidazol-porphyrinato-iron(II)	WENPAJ	0.89		210		216	207	208	209	209
pyrazol-porphyrinato-iron(II)	AMAJII	0.89		209		215	206	207	207	209
bis-imidazol-porphyrinato-iron(II)	EWURUL	0.38	N	199	199	199	198	198	199	200
bis-imidazol-porphyrinato-iron(III)	LEVHOK	0.35	N	198	200	201	197	197	197	198
imidazol-dioxygen-porphyrinato-iron	DOGWED	0.28	O	198	186	206	199	199	199	199

^a References S6–S14. ^b RCSB PDB is RCSB Protein Data Bank,^{S6} <https://www.rcsb.org/>. ^c The interatomic distances are rounded to picometers.

^d The isomer shifts correspond to temperatures close to the temperature of liquid nitrogen, with the exception of DEDWUE value (bold italic), which refers to the liquid-helium temperature. The uncertainties do not exceed 0.03 mm s⁻¹.

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