

Photoluminescence and thermoluminescence studies of Dy³⁺-activated LaCePO₄ phosphor

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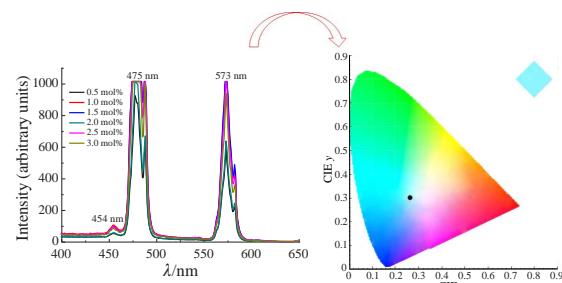
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The LaCePO₄:Dy³⁺ luminescence experiments revealed the excitation peak at 275 nm which gives the emission spectra in visible region with intensive peaks centered at 475 and 573 nm. The emission spectra showed the transition from ⁴F_{9/2} to the ground energy state. The prepared phosphor may be useful for display applications in white light emission and thermoluminescence dosimeters.



Keywords: LaCePO₄:Dy³⁺ phosphors, solid state lighting, XRD, SEM, CIE, white light emission.

White light emitting diodes (WLEDs) have a lot of advantages over the existing incandescent and halogen lamps in terms of power, efficiency, reliability, and long lifetime.^{1,2} The remarkable progress has been made in the development of WLEDs using InGaN chips whose emission bands shift to the near-ultraviolet (near-UV) range around 400 nm.^{3,4} Since UV-LEDs can offer highly efficient solid state light, more attention has been paid to the development of new phosphors that can be excited in the range of near-UV (350–420 nm) due to the necessity to increase the efficiency of white light emitting solid-state devices.⁵ Alkaline earth silicates R₂MgSi₂O₇ (R = Ca, Sr, and Ba) co-doped with Eu²⁺ and Dy³⁺ having akermanite-based structures attract great attention in the development of persistent lighting.⁶ The CaMgSi₂O₆, and CaMgSi₂O₇ phosphors activated with Eu²⁺, Dy³⁺ and Nd³⁺ with afterglow characteristics were prepared by Jiang *et al.*⁷ through solid-state reaction in a reducing atmosphere. Long-persistence phosphorescence in Ce³⁺-doped Ca₂Al₂SiO₇ with a melilite structure has been reported by Kodama *et al.*^{8,9} Phosphors like BaMg₂Si₂O₇ doped with Eu²⁺- and Mn²⁺, CdSiO₃:Mn²⁺:RE (RE = Y, La, Gd, and Lu), Eu and Dy-doped R₃MgSi₂O₈ (R = Ca, Sr, and Ba), blue-emitting Eu²⁺- and Dy³⁺-doped Sr₂ZnSi₂O₇, and CaAl₂Si₂O₈ doped with Eu²⁺ and Dy³⁺ are reported as efficient long-lasting phosphors.^{10–14}

In this work the spectroscopic parameters of Dy³⁺-activated LaCePO₄ phosphor and the formation of nanoparticles of the synthesized phosphor are reported for the first time. The phosphor was synthesized by the modified solid-state reaction method and phase composition and morphology of the samples were determined by X-ray diffraction (XRD) and scanning electron microscopy (SEM). Photoluminescence (PL) analysis of the prepared phosphors was performed at various doping ion concentrations.

The LaCePO₄:Dy³⁺ phosphors were prepared by solid-state reaction technique with varying concentration of Dy³⁺ ions

(0.5 to 3.0 mol%). La₂O₃, CeO₂, NH₄H₂PO₄, and Dy₂O₃ were used as starting materials at a proper stoichiometric ratio; H₃BO₃ (1.6 mol%) was used as a flux. All materials were thoroughly ground for ~ 2 h (using acetone for liquid grinding) to obtain a homogeneous powder using mortar and pestle. The ground mixture was transferred to the alumina crucible and heated at 800 °C for 2 h in a muffle furnace. The calcined sample was ground again and heated at 1150 °C for 4 h. The final calcined samples were ground to form fine powder and used for further characterization.

XRD diffractograms were recorded using a Bruker AXS D8 Advance X-ray Powder Diffractometer with Cu-K α radiation (1.5406 Å) in 2 θ range of 4–80°. SEM images were recorded using a JEOL JSM-6390LV scanning electron microscope, and PL excitation and emission spectra were monitored using a Shimadzu spectrophotofluorimeter. The Commission internationale de l'éclairage (CIE) chromaticity coordinates were calculated from the emission spectra of the phosphor samples using LUMPAC software. Thermoluminescence (TL) glow curves were recorded using a TLD I1009 reader supplied by Nucleonix Systems Pvt. Ltd.

The XRD patterns of the prepared pure and 2.5 mol% Dy³⁺-doped LaCePO₄ phosphors and the phase composition are shown in Figure 1. The crystallite size of the sample was calculated from the full width at half maximum (FWHM) of all peaks in the XRD pattern using the Scherrer formula.^{15–24} The crystallite size for the intense peak in the XRD pattern of the prepared LaCePO₄:Dy³⁺ (2.5 mol%) phosphor was found to be 41.81 nm.

Figures 2(a)–(d) show the SEM images of the Dy³⁺ (2.5 mol%)-activated LaCePO₄ phosphor at different magnifications. The prepared phosphor has a cube-like structure clearly visible in the 1 μ m to 100 nm size range. This nanocube type of the

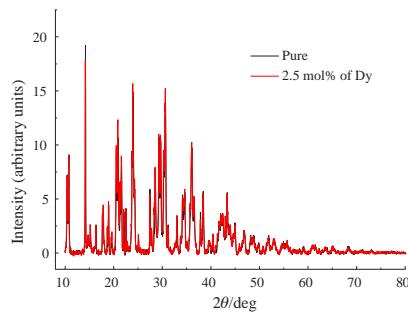


Figure 1 XRD patterns of pure and Dy^{3+} (2.5 mol%)-activated LaCePO_4 phosphors.

structure minimizes light scattering on the crystallite surface and enhances luminescence efficiency of the prepared phosphor. It is confirmed that the particles are uniformly distributed and should be useful for different practical applications in optoelectronics.

The PL excitation spectrum is recorded for the LaCePO_4 phosphor doped with 2.5 mol% of Dy^{3+} [Figure 3 (a)]. The broad emission peak centered at 275 nm is due to the charge transfer phenomenon of the O^{2-} – Dy^{3+} ion. The emission spectra were monitored at 275 nm excitation and several distinct peaks in the visible region centered at 475 nm (blue emission) and 573 nm (yellow one) having excellent PL emission intensity were revealed [Figure 3(b)]. The emission spectra show interesting features of the Dy^{3+} ion in the host lattice displaying blue and yellow emission simultaneously. The emission originates from the transition from the $4\text{F}_{9/2}$ level to the ground state and other excited energy levels of the Dy^{3+} ion. Overall composed peaks resulted in white light emission confirmed by CIE coordinates.

The CIE chromaticity coordinates of the $\text{LaCePO}_4:\text{Dy}^{3+}$ (2.5 mol%) phosphor shown in Figure 3(c) are calculated from the corresponding emission spectrum. The CIE coordinates of the phosphor are expressed in (x, y) and the values are found in the deep bluish-white region (touching the white light boundary). This clearly shows that the Dy^{3+} -doped LaCePO_4 sample can be used for white light emitting applications such as solid-state lighting and display device applications and its chromaticity coordinates are $x = 0.26$ and $y = 0.30$.²⁵

Figure 4(a) shows the TL glow curves of the LaCePO_4 phosphor doped with Dy^{3+} (2.5 mol%) after UV exposure for various time. The TL glow curve intensity increases with increasing UV exposure time and the linear response to dose is found (see Online Supplementary Materials). Herein, the broad peak centered at 178°C shows the composite nature, its deconvolution shows five distinct peaks, and the corresponding trap parameters are calculated using computerized glow curve deconvolution (CGCD) technique.^{26–28} The trap parameters such as trap depth (activation energy), order of kinetics using shape

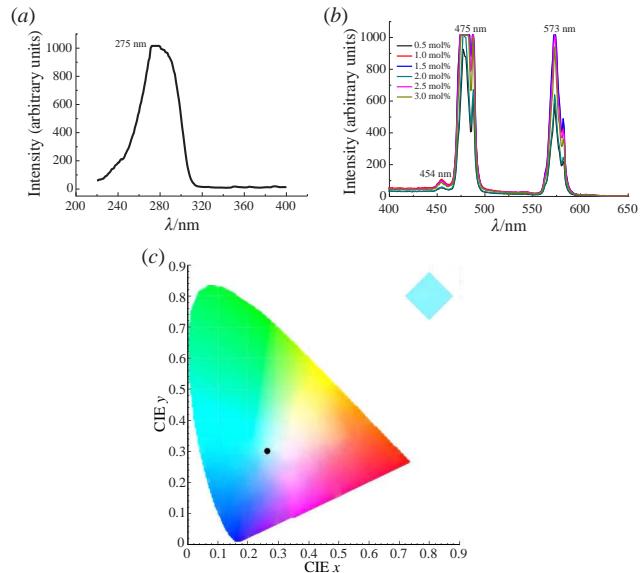


Figure 3 (a) PL excitation spectrum of $\text{LaCePO}_4:\text{Dy}^{3+}$ (2.5 mol%) phosphors, (b) PL emission spectra of $\text{LaCePO}_4:\text{Dy}^{3+}$ (0.5–3.0 mol%) phosphors, and (c) CIE 1931 coordinates of $\text{LaCePO}_4:\text{Dy}^{3+}$ (2.5 mol%) phosphors ($x = 0.26$ and $y = 0.30$).

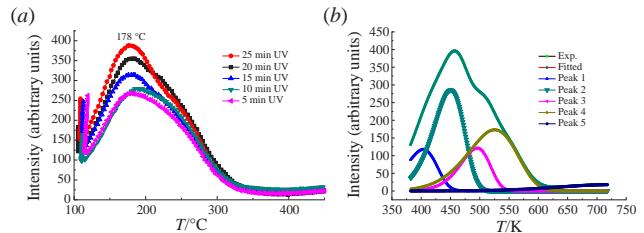


Figure 4 (a) TL glow curves of $\text{LaCePO}_4:\text{Dy}^{3+}$ (2.5 mol%) phosphor at various UV exposure times; (b) CGCD pattern of TL glow curve of $\text{LaCePO}_4:\text{Dy}^{3+}$ (2.5 mol%) phosphor after 25 min UV exposure.

parameters and frequency factors are given in Table 1. Here, most of the peaks show the first and general order of kinetics, and the values of shape factor μ from 0.37 to 0.46 indicate the position of trapped electrons in the luminescence center. The one electron trapped in the trap center for each peak for the UV irradiated phosphor shows the shallower trapping phenomenon. The CGCD pattern^{27,28} of the prepared phosphor for 25 min UV exposure is shown in Figure 4(b).

In summary, dysprosium-doped lanthanum cerium phosphate phosphor showed good PL spectra in the blue and yellow regions which may be applicable for white light emission in various electronic devices. The crystal structure and morphology indicated the practical usability of the prepared phosphors. The PL emission spectra contain peaks at 475 and 573 nm, and the corresponding CIE coordinates $x = 0.26$ and $y = 0.30$ are very close to white light. The TL glow curve showed a broad intense

Table 1 Various trap parameters calculated using CGCD technique of the prepared $\text{LaCePO}_4:\text{Dy}^{3+}$ (2.5 mol%) phosphor.^a

Peak	T_1/K	T_m/K	T_2/K	τ/s	δ/K	ω/K	$\mu = \delta/\omega$	Activation energy E/eV	Frequency factor s^{-1}
1	371	403	431	32	28	60	0.46	1.84	7×10^{14}
2	414	449	471	35	22	57	0.38	1.9	2×10^{14}
3	457	497	521	40	24	64	0.37	1.88	2×10^{13}
4	472	525	567	53	42	95	0.44	1.53	4×10^{10}
5	587	669	720	82	51	133	0.38	1.35	2×10^{14}

^a T_1 and T_2 are the low and high temperatures at the half maximum of the peak; T_m is the temperature at the peak maximum; τ is the mean lifetime; $\delta = T_2 - T_m$; $\omega = T_2 - T_1$.

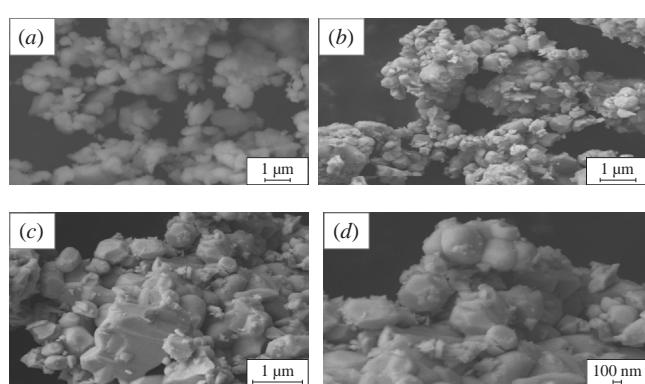


Figure 2 SEM images of Dy^{3+} (2.5 mol%)-activated LaCePO_4 phosphor at various magnifications.

peak centered at 178 °C, and the TL glow curve intensity increased with UV exposure time. The corresponding trap parameters were calculated using the CGCD technique. The trap depth varied from 1.35 to 1.9 eV, and the frequency factor was quite low varying from 4×10^{10} to 7×10^{14} s⁻¹. Therefore, the prepared phosphate-based phosphor may be useful for TL dosimetry applications.

Online Supplementary Materials

Supplementary data associated with this article can be found in the online version at doi: 10.1016/j.mencom.2024.06.041.

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