

Synthesis and X-ray structure of potent anticancer 4-(4-methoxyphenyl)-3-(3,4,5-trimethoxyphenyl)isoxazole

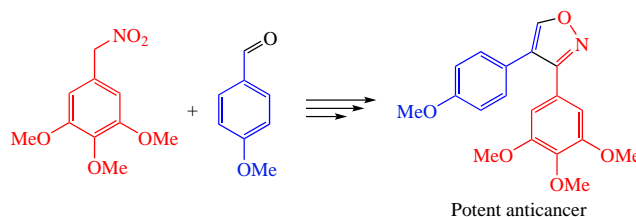
Victor P. Kislyi,^a Anna S. Maksimenko,^a Aida I. Samigullina,^a Marina N. Semenova^b and Victor V. Semenov^{*a}

^a N. D. Zelinsky Institute of Organic Chemistry, Russian Academy of Sciences, 119991 Moscow, Russian Federation. E-mail: vs@zelinsky.ru

^b N. K. Koltzov Institute of Developmental Biology, Russian Academy of Sciences, 119334 Moscow, Russian Federation

DOI: 10.1016/j.mencom.2024.06.013

The three-step synthesis of potent anticancer 4-(4-methoxyphenyl)-3-(3,4,5-trimethoxyphenyl)isoxazole starting from 1,2,3-trimethoxy-5-(nitromethyl)benzene and anisaldehyde was developed. In phenotypic sea urchin embryo assay, this compound exhibited antimitotic antitubulin activity comparable to that of the natural cytostatic combretastatin A4 (CA4). The title isoxazole inhibited *in vitro* growth of A549 human lung cancer cells and PC-3 human prostate cancer cells with IC₅₀ values of 8 and 6 nM, respectively, exceeding the effect of CA4.



Keywords: *o*-diarylisoxazoles, nitromethylarenes, sea urchin embryo, cancer cells, X-ray.

ortho-Diaryl substituted five-membered *N*-heterocycles, referred to as combretazoles, feature *cis*-biaryl topology of the A and B rings similar to that of the natural antimitotic combretastatin A4 (CA4), and exhibit strong anticancer activity both *in vitro* and *in vivo*.^{1–13} Within combretazoles, *o*-diarylisoxazoles have been identified as the most potent antimitotic compounds compared to their analogues of the pyrrole, pyrazole and 1,2,3-triazole series, both in sea urchin embryo tests and in NCI60 anticancer drug screen.^{4,14–19} Some of these compounds exhibited cytotoxicity against human cancer cell lines in low nanomolar concentration (GI₅₀ of 1–10 nM).

In the diarylisoxazole series, the most active antimitotic compound **1** (Figure 1) contained 3,4,5-trimethoxyphenyl pharmacophore as the ring A connected to position 5 of the isoxazole ring and 4-methoxyphenyl group as the ring B. It showed consistently high activity in NCI60 screen with

GI₅₀ < 10 nM.¹ Particularly, GI₅₀ values for HeLa human cervical carcinoma cells, HepG2 human liver carcinoma cells, and OVCAR-3 human ovarian carcinoma cells were 4.7, 4.4 and 10 nM, respectively.⁴ Compound **1** was included into the Tubulin list of standards at The NCI Development Therapeutics Program Database.²⁰ *In vitro* studies at the P. A. Herzen Moscow Oncology Research Institute demonstrated extremely high cytotoxicity of isoxazole **1** against PC-3 human prostate cancer cell line with IC₅₀ < 1 nM.²¹ This compound with improved formulation was studied *in vivo* on different human and mouse xenografts using the nude mouse model before publication of synthetic route.²² It was reported previously^{1,21} that in the series of 4,5- and 3,4-diarylisoxazole isomers cytotoxicity and antitubulin activity could differ by 1–2 orders. The position of the 3,4,5-trimethoxyphenyl substituent close to the heteroatom of isoxazole linker, as in **1** (see Figure 1), was identified as favourable for the effect.¹ Therefore, it was important to develop the synthetic procedure to obtain the corresponding isomer **2** that was previously inaccessible due to the lack of starting 1,2,3-trimethoxy-5-(nitromethyl)benzene **3** (Scheme 1).

Recently,²³ the nitromethylarene **3** was synthesized by the Victor Meyer reaction with 35% yield. In the present work, the corresponding nitrostilbene **4**, and isoxazoline oxide **5** were obtained using modified methods¹⁸ in good yields (see Scheme 1).[†] The intermediate isoxazoline *N*-oxide **5** was

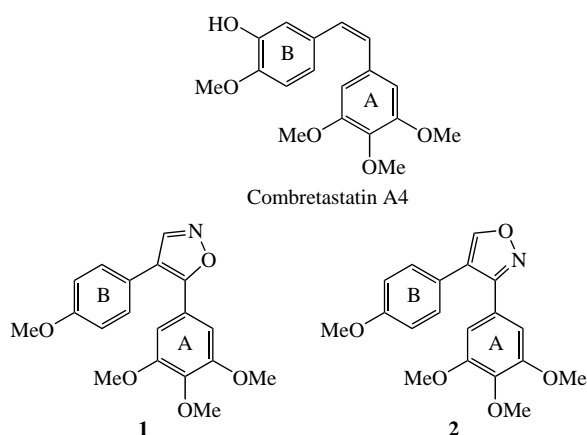
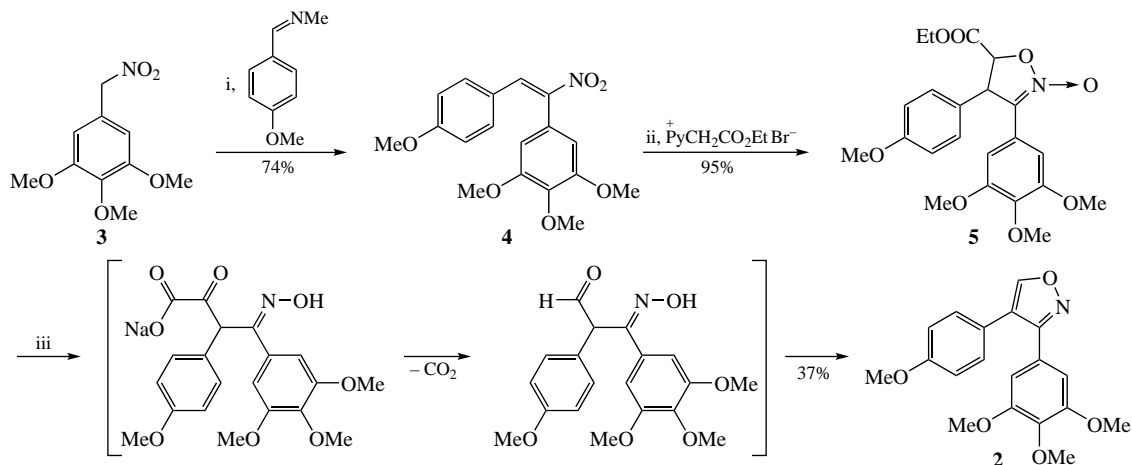


Figure 1 Structures of combretastatin A4 (CA4) and its diarylisoxazole analogues.

[†] 5-Ethoxycarbonyl-4-(4-methoxyphenyl)-3-(3,4,5-trimethoxyphenyl)-4,5-dihydroisoxazole *N*-oxide **5**. Pyridinium salt Py⁺CH₂CO₂Et Br[−] (0.738 g, 3 mmol) was added to a solution of nitrostilbene **4** (0.518 g, 1.5 mmol) in MeOH (10 ml) at room temperature and stirring, then Et₃N (1.075 g, 4.35 mmol) was added, and the mixture was stirred for 1 h (TLC control). The solvent was evaporated *in vacuo*, the residue was washed with water and extracted with CH₂Cl₂. The extract was dried over MgSO₄ and evaporated *in vacuo* to give 4,5-dihydroisoxazole *N*-oxide **5**. Yellow oil (95%); ¹H NMR (500 MHz, CDCl₃) δ: 7.28 (d, *J* = 8.2 Hz, 2 H,



Scheme 1 Reagents and conditions: i, AcOH–MeCN, 20 °C, overnight; ii, Et₃N–MeCN, 40 °C, 5 h; iii, 2% NaOH, EtOH–H₂O, 60 °C, 6 h.

isomerized with simultaneous hydrolysis and decarboxylation to afford the target diarylisoxazole **2**.

The structure of new diarylisoxazole **2** was proved by X-ray diffraction analysis (Figure 2).[‡] The compound crystallizes in the *P* $\bar{1}$ space group (no. 2) with one molecule in the asymmetric part of unit cell without solvent in the crystal structure. Conformation of the molecule is characterized by dihedral angles showing the turning of aromatic fragments relative to the central isoxazole ring. The values of these dihedral angles are 45.83 and 27.51° for methoxyphenyl and trimethoxyphenyl fragments, respectively.

The phenotypic sea urchin embryo assay¹ was used to confirm antimitotic microtubule destabilizing mechanism of action of the target isoxazole **2**. Compound CA4 and previously synthesized isomer **1** were selected as positive controls. This *in vivo* assay allows for rapid identification of antimitotic activity by cleavage alteration/arrest after the application of a test molecule to

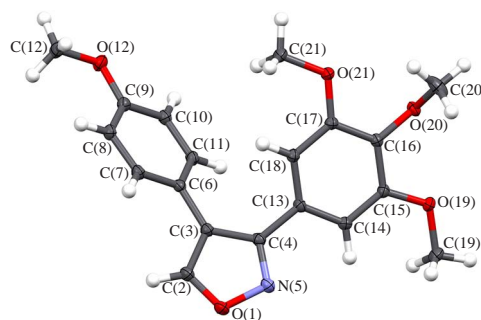


Figure 2 Structure of 4-(4-methoxyphenyl)-3-(3,4,5-trimethoxyphenyl)isoxazole **2** in a crystal. The thermal vibrations of atoms are represented by ellipsoids in anisotropic approximation (*p* = 50%).

fertilized eggs. A specific modification of swimming pattern observed after hatching, *i.e.* spinning at the bottom of the culture vessel instead of forward swimming near the surface of the seawater, is an evidence of the microtubule destabilizing mode of action (Table 1). Compound **2** was more potent antimitotic agent than CA4, and its effect was related to microtubule destabilization as followed from its ability to induce embryo spinning. Both compounds **1** and **2** strongly inhibited growth of A549 and PC-3 cancer cells with the IC₅₀ values from 0.08 to 10 nmol dm^{−3} depending on the cell line, which significantly exceeded the antiproliferative effect of CA4.²¹

In conclusion, the synthesis of potent anticancer 4-(4-methoxyphenyl)-3-(3,4,5-trimethoxyphenyl)isoxazole **2** was accomplished. Its molecular structure was confirmed by X-ray crystallography. The study using *in vivo* sea urchin embryo assay and *in vitro* cytotoxicity test demonstrated that isoxazole **2** is strong tubulin-targeting antimitotic agent even more potent than CA4.

Table 1 Antimitotic microtubule destabilizing activity and cytotoxicity of compounds **1**, **2** and CA4.

Compound	Sea urchin embryo effects, ^a EC/nM			IC ₅₀ ^b /nM	
	Cleavage alteration	Cleavage arrest	Embryo spinning	A549 ^c	PC-3 ^d
CA4	2	10	50	86 ± 2 ²¹	40 ± 8 ²¹
1 ^e	2	10	100	7 ± 2 ²¹	0.08 ± 0.01 ²¹
2	1	5	50	8 ± 1 ²¹	6 ± 1 ²¹

^aThe sea urchin embryo assay was conducted as described previously.¹ Fertilized eggs and hatched blastulae were exposed to 2-fold decreasing concentrations of compounds. Duplicate measurements showed no differences in effective threshold concentration (EC) values. ^bIC₅₀: concentration required for 50% cell growth inhibition (data from ref. 21). ^cA549: human lung cancer cell line. ^dPC-3: human prostate cancer cell line. ^eEC values from ref. 1.

H-2'6'), 7.14 (s, 2H, H-2,6), 6.91 (d, *J* = 8.7 Hz, 2H, H-3',5'), 4.99 (d, *J* = 2.8 Hz, 1H, H-4 iso), 4.81 (d, *J* = 2.6 Hz, 1H, H-5 iso), 4.38–4.28 (m, 2H, OCH₂), 3.82 (s, 3H, OMe), 3.79 (s, 3H, OMe), 3.76 (s, 6H, 2OMe), 1.35 (t, *J* = 7.2 Hz, 3H, Me); ¹³C NMR (126 MHz, CDCl₃) δ: 169.1, 159.8, 153.1 (2C), 139.2, 130.4, 128.3 (2C), 128.2, 120.7, 115.0 (2C), 104.4 (2C), 78.7, 62.5, 60.8, 56.1, 55.3 (2C), 54.2, 14.1.

4-(4-Methoxyphenyl)-3-(3,4,5-trimethoxyphenyl)isoxazole **2**. 4,5-Dihydroisoxazole *N*-oxide **5** was added to a 2% solution of NaOH in EtOH (10 ml) and water (2 ml), and the mixture was stirred at 60 °C for 5–6 h. Ethanol was evaporated *in vacuo*, the residue was suspended in dichloromethane (60 ml) and washed with water. The organic layer was dried with MgSO₄ and evaporated. Isoxazole **2** was obtained as a yellowish oil (37%) after purification by column chromatography (benzene–benzene/ethyl acetate gradient, 1 → 10). Analytical sample was obtained by crystallization from MeOH. Yellow crystals; mp 93–94 °C (MeOH). ¹H NMR (500 MHz, CDCl₃) δ: 8.45 (s, 1H, H-5 iso), 7.23 (d, *J* = 7.2 Hz, 2H, H-2',6'), 6.91 (d, *J* = 6.9 Hz, 2H, H-3',5'), 6.76 (s, 2H, H-2,6), 3.87 (s, 3H, OMe), 3.82 (s, 3H, OMe), 3.70 (s, 6H, 2OMe); ¹³C NMR (126 MHz, CDCl₃) δ: 159.8, 159.6, 156.0 (2C), 153.2, 139.1, 130.5 (2C), 123.9, 121.2, 119.8, 114.2 (2C), 105.8 (2C), 60.9, 56.0 (2C), 55.4. HRMS (ESI/QTOF) *m/z*: [M+H]⁺ calc. for C₁₉H₂₀NO₅ 342.1338; found 342.1336.

[‡] Crystal data for **2**. C₁₉H₁₉NO₅, *M*_r = 341.35, triclinic, *P* $\bar{1}$ at 100.0(1) K, *a* = 7.60340(10), *b* = 10.6503(2) and *c* = 10.8878(2) Å, *α* = 74.5640(10), *β* = 89.5210(10) and *γ* = 75.7160(10)°, *V* = 822.05(2) Å³, *Z* = 2, *d*_{calc} = 1.379 g cm^{−3}, *μ*(MoK_α) = 0.100 mm^{−1}, *F*(000) = 360. Total of 40048 reflections were measured and 6112 independent reflections (*R*_{int} = 0.0485) were used. The refinement converged to *wR*₂ = 0.1089 and GOF = 1.014 for all independent reflections [*R*₁ = 0.0447 was calculated for 4657 observed reflections with *I* > 2σ(*I*)].

CCDC 2335715 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via <http://www.ccdc.cam.ac.uk>.

The biological experiments were conducted by M.N.S. under the IDB RAS Government Basic Research Program in 2024, no. 0088-2024-0015.

Online Supplementary Materials

Supplementary data associated with this article can be found in the online version at doi: 10.1016/j.mencom.2024.06.013.

References

- 1 M. N. Semenova, D. V. Demchuk, D. V. Tsyganov, N. B. Chernysheva, A. V. Samet, E. A. Silyanova, V. P. Kislyi, A. S. Maksimenko, A. E. Varakutin, L. D. Konyushkin, M. M. Raihstat, A. S. Kiselyov and V. V. Semenov, *ACS Comb. Sci.*, 2018, **20**, 700.
- 2 H. N. Pati, M. Wicks, H. L. Holt, Jr., R. LeBlanc, P. Weisbruch, L. Forrest and M. Lee, *Heterocycl. Commun.*, 2005, **11**, 117.
- 3 J. Kaffy, R. Pontikis, D. Carrez, A. Croisy, C. Monneret and J.-C. Florent, *Bioorg. Med. Chem.*, 2006, **14**, 4067.
- 4 C.-M. Sun, L.-G. Lin, H.-J. Yu, C.-Y. Cheng, Y.-C. Tsai, C.-W. Chu, Y.-H. Din, Y.-P. Chau and M.-J. Don, *Bioorg. Med. Chem. Lett.*, 2007, **17**, 1078.
- 5 Q. Zhang, Y. Peng, X. I. Wang, S. M. Keenan, S. Arora and W. J. Welsh, *J. Med. Chem.*, 2007, **50**, 749.
- 6 N. Xue, X. Yang, R. Wu, J. Chen, Q. He, B. Yang, X. Lu and Y. Hu, *Bioorg. Med. Chem.*, 2008, **16**, 2550.
- 7 B. Biersack, K. Effenberger, S. Knauer, M. Ocker and R. Schobert, *Eur. J. Med. Chem.*, 2010, **45**, 4890.
- 8 B. Burja, T. Čimbora-Zovko, S. Tomić, T. Jelušić, M. Kočevár, S. Polanc and M. Osmak, *Bioorg. Med. Chem.*, 2010, **18**, 2375.
- 9 R. Romagnoli, P. G. Baraldi, M. K. Salvador, D. Preti, M. Aghazadeh Tabrizi, A. Brancale, X.-H. Fu, J. Li, S.-Z. Zhang, E. Hamel, R. Bortolozzi, G. Basso and G. Viola, *J. Med. Chem.*, 2012, **55**, 475.
- 10 R. Zaninetti, S. V. Cortese, S. Aprile, A. Massarotti, P. L. Canonico, G. Sorba, G. Groza, A. A. Genazzani and T. Pirali, *ChemMedChem*, 2013, **8**, 633.
- 11 J.-X. Duan, X. Cai, F. Meng, L. Lan, C. Hart and M. Matteucci, *J. Med. Chem.*, 2007, **50**, 1001.
- 12 K. Jaroch, M. Karolak, P. Górski, A. Jaroch, A. Krajewski, A. Ilnicka, A. Sloderbach, T. Stefański and S. Sobiak, *Pharmacol. Rep.*, 2016, **68**, 1266.
- 13 H. Rajak, P. K. Dewangan, V. Patel, D. K. Jain, A. Singh, R. Veerasamy, P. C. Sharma and A. Dixit, *Curr. Pharm. Des.*, 2013, **19**, 1923.
- 14 K. D. Shin, Y. J. Yoon, Y.-R. Kang, K.-H. Son, H. M. Kim, B.-M. Kwon and D. C. Han, *Biochem. Pharmacol.*, 2008, **75**, 383.
- 15 C. B. L. Sun, C. Borella, H. Li, J. Jiang, S. Chen, K. Koya, T. Inoue, Z. Du, K. Foley, Y. Wu, M. Zhang and W. Ying, *Patent US 089177*, 2006.
- 16 E. A. Silyanova, V. I. Ushkarov, A. V. Samet, A. S. Maksimenko, I. A. Koblov, V. P. Kislyi, M. N. Semenova and V. V. Semenov, *Mendeleev Commun.*, 2022, **32**, 120.
- 17 D. V. Tsyganov, M. N. Semenova, L. D. Konyushkin, V. I. Ushkarov, M. M. Raihstat and V. V. Semenov, *Mendeleev Commun.*, 2019, **29**, 163.
- 18 V. S. Stroylov, I. V. Svitanko, A. S. Maksimenko, V. P. Kislyi, M. N. Semenova and V. V. Semenov, *Bioorg. Med. Chem. Lett.*, 2020, **30**, 127608.
- 19 G. L. Karetnikov, D. A. Skvortsov, A. A. Moseicheva, N. V. Zyk and O. B. Bondarenko, *Asian J. Org. Chem.*, 2023, **12**, e202300131.
- 20 National Cancer Institute, <https://dtp.cancer.gov/compsub/>.
- 21 A. D. Plyutinskaya, E. R. Nemtsova, A. A. Pankratov, P. V. Shegai, S. S. Krylov, V. N. Iskandarova, A. S. Maksimenko, D. V. Demchuk, T. S. Kuptsova, M. N. Semenova and V. V. Semenov, *Bull. Exp. Biol. Med.*, 2022, **174**, 221.
- 22 E. R. Nemtsova, N. B. Morozova, A. D. Plyutinskaya, A. N. Noev, A. A. Pankratov, P. V. Shegai, A. D. Kaprin, D. V. Demchuk, S. S. Krylov and V. V. Semenov, *Pharm. Chem. J.*, 2024, in press.
- 23 A. S. Maksimenko, P. A. Buikin, E. D. Daeva, V. P. Kislyi and V. V. Semenov, *Synthesis*, 2022, **54**, 2724.

Received: 26th February 2024; Com. 24/7403