

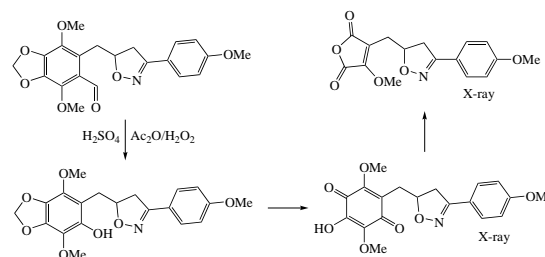
Baeyer–Villiger rearrangement of polyalkoxybenzaldehydes with benzene ring opening

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The Baeyer–Villiger oxidation of apiolaldehyde bearing *o*-(3-*p*-anisylisoxazolin-5-yl)methyl substituent proceeds first with the formation of the anticipated phenol. The subsequent oxidation of phenol with destruction of methylenedioxy ring leads to *p*-quinone derivative which would undergo opening of the benzene ring to finally produce maleic anhydride moiety. The structure of new compounds was proved by X-ray diffraction analysis.



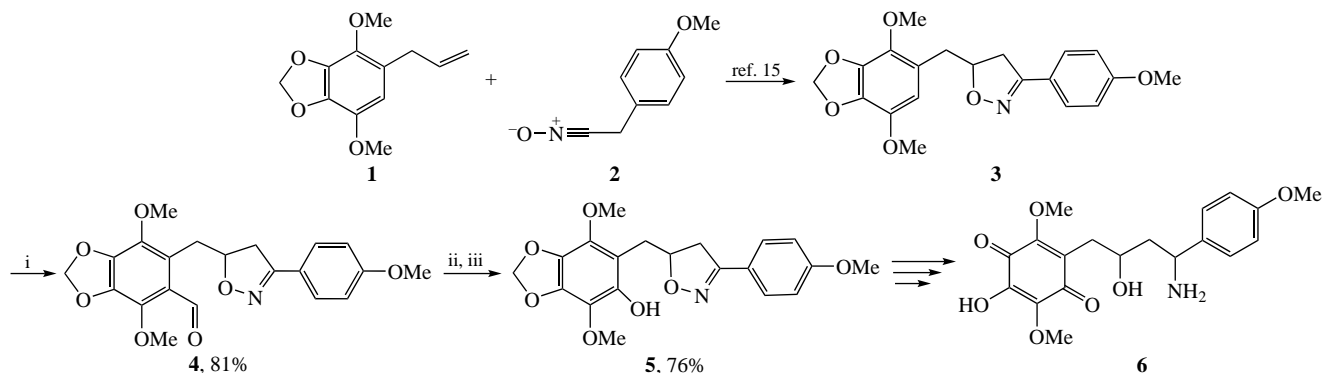
Keywords: Baeyer–Villiger rearrangement, apiol, quinone, isoxazole, maleic anhydride, antioxidant.

Polyhydroxy and polyalkoxy 1,4-naphthoquinones and hydroquinones are widespread in plants, animals and marine organisms.^{1–4} They are of interest to medicine due to their diverse biological activity^{1,5–8} such as antioxidant, antitumor or antibacterial. We have recently developed a convenient method for their synthesis using an oxidative Baeyer–Villiger rearrangements of polymethoxy(methylenedioxy)phenols^{9,10} which are readily available from parsley and dill seed metabolites.^{11,12} The transformation proceeded in two steps: first, the classical rearrangement of benzaldehydes into phenols occurred, which were further oxidized to quinones with the opening of the methylenedioxy moiety.¹⁰ However, the selectivity of the reactions depended on the nature of the alkyl substituent *ortho* to the aldehyde group.

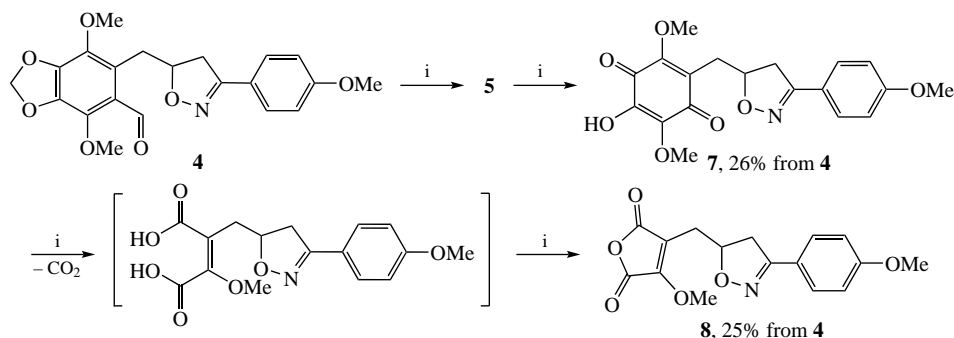
In the present work, such a rearrangement was studied using the example of apiolaldehyde bearing an *o*-(3-*p*-anisylisoxazolin-5-yl)methyl substituent (Scheme 1), which could be useful for the synthesis of quinones (like **6**) with biologically active α -amino- γ -hydroxyalkyl fragments. Quinones with such moieties as juglomycin C-amide isolated from the strain *S. coelicolor* A3(2) M145 belong to a family

of promising juglomycin-type antibiotics.^{13,14} Based on the reaction of dipolar cycloaddition of apiol **1** and nitrile oxide **2**, isoxazoline **3** was synthesized,¹⁵ which was further formylated into aldehyde **4** according to the reported procedure.¹⁰

Aldehyde **4** normally undergoes the Baeyer–Villiger rearrangement in the presence of SeO₂ in H₂O₂/Bu^tOH system to form phenol **5** (see Scheme 1). However, phenol **5** being isolated could not be further oxidized to quinone **7** (Scheme 2) under standard conditions (H₂O₂–Ac₂O), similarly to most phenols with an apiol fragment.^{9,10} Under these conditions, according to HRMS data, more than five products are formed, among which only traces of maleic anhydride derivative **8** (a decomposition product of quinone **7**) were detected. Fortunately, the *in situ* Baeyer–Villiger oxidation of compound **4** in the same medium (H₂O₂–Ac₂O) could proceed further with the formation of quinone **7** and maleic anhydride **8** derivatives in acceptable yields (totally 51%). Apparently, in the first case, at a high initial concentration of polymethoxyphenol **5**, a side interaction of the resulting highly reactive quinone **8** with the initial **5** occurs.



Scheme 1 Reagents and conditions: i, HCO₂Et, PCl₅, SnCl₄, CH₂Cl₂, room temperature, 5 h (ref. 10); ii, H₂O₂, SeO₂, Bu^tOH, 50–60 °C, 3 h; iii, MeOH, Et₃N, 20 °C, 1 h, then HCl/CH₂Cl₂, 20 °C.



Scheme 2 Reagents and conditions: i, H_2O_2 , H_2SO_4 , Ac_2O , room temperature, 6 h.

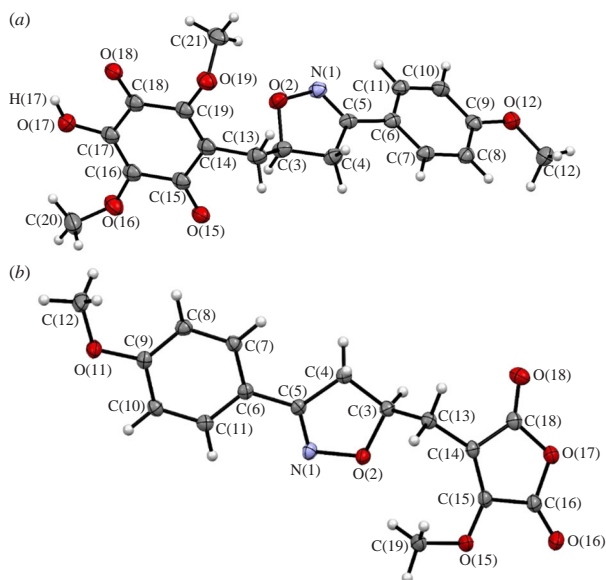


Figure 1 Structures of (a) compound **7** and (b) compound **8** in crystals. The thermal vibrations of atoms are represented by ellipsoids in anisotropic approximation ($p = 50\%$).

Apparently, phenol **5** upon formation is further oxidized with methylenedioxy ring destruction to produce quinone **7**. Further oxidative degradation of the benzene ring and decarboxylation occur to form a maleic acid derivative, which would cyclize in an acid medium to maleic anhydride **8** (see Scheme 2). The structure of new products **7** and **8** was proved by X-ray diffraction analysis (Figure 1).[†]

[†] Crystal data for **7**. $\text{C}_{19}\text{H}_{19}\text{NO}_7$ ($M = 373.35$), orthorhombic, space group $P2_12_12_1$, at $T = 100(1)$ K, $a = 4.79523(7)$, $b = 8.20587(11)$ and $c = 43.0852(6)$ Å, $\beta = 90^\circ$, $V = 1695.36(4)$ Å³, $Z = 4$, $d_{\text{calc}} = 1.463$ g cm⁻³, $F(000) = 784$, $\mu = 0.949$ mm⁻¹. 13607 reflections (3493 independent reflections, $R_{\text{int}} = 0.0435$) were measured and used in the refinement. The refinement converged to $R_1 = 0.0435$ for 3277 observed reflections with $I > 2\sigma(I)$ and $wR_2 = 0.1137$ for all independent reflections, GOF = 1.075.

Crystal data for **8**. $\text{C}_{16}\text{H}_{15}\text{NO}_6$ ($M = 317.29$), monoclinic, space group $P2_1/n$, at $T = 100(1)$ K, $a = 8.29006(9)$, $b = 14.54498(18)$ and $c = 11.91274(13)$ Å, $\beta = 94.3701(10)^\circ$, $V = 1432.25(3)$ Å³, $Z = 4$, $d_{\text{calc}} = 1.471$ g cm⁻³, $F(000) = 664$, $\mu = 0.962$ mm⁻¹. 16647 reflections (3102 independent reflections, $R_{\text{int}} = 0.0311$) were measured and used in the refinement. The refinement converged to $R_1 = 0.0345$ for 2886 observed reflections with $I > 2\sigma(I)$ and $wR_2 = 0.0935$ for all independent reflections, GOF = 1.046.

CCDC 2291864 (**7**) and 2291865 (**8**) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via <http://www.ccdc.cam.ac.uk>.

In summary, polyalkoxybenzaldehydes in the Baeyer–Villiger rearrangement can undergo further oxidation with the decomposition of the benzene ring.

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Online Supplementary Materials

Supplementary data associated with this article can be found in the online version at doi: 10.1016/j.mencom.2024.04.026.

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