

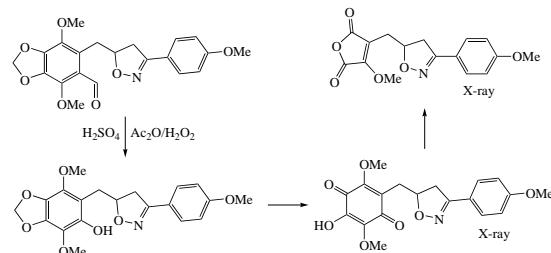
## Baeyer–Villiger rearrangement of polyalkoxybenzaldehydes with benzene ring opening

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**The Baeyer–Villiger oxidation of apiolaldehyde bearing *o*-(3-*p*-anisylisoxazolin-5-yl)methyl substituent proceeds first with the formation of the anticipated phenol. The subsequent oxidation of phenol with destruction of methylenedioxy ring leads to *p*-quinone derivative which would undergo opening of the benzene ring to finally produce maleic anhydride moiety. The structure of new compounds was proved by X-ray diffraction analysis.**



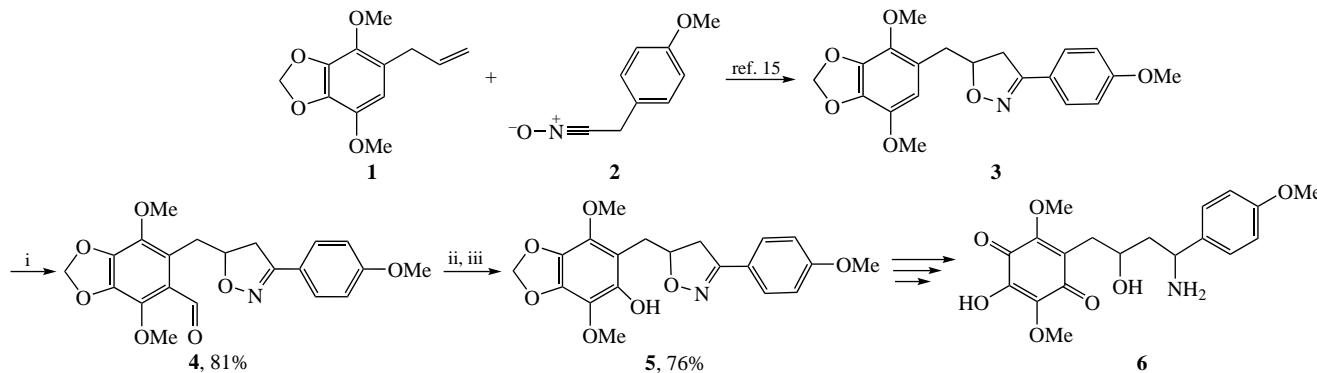
**Keywords:** Baeyer–Villiger rearrangement, apiol, quinone, isoxazole, maleic anhydride, antioxidant.

Polyhydroxy and polyalkoxy 1,4-naphthoquinones and hydroquinines are widespread in plants, animals and marine organisms.<sup>1–4</sup> They are of interest to medicine due to their diverse biological activity<sup>1,5–8</sup> such as antioxidant, antitumor or antibacterial. We have recently developed a convenient method for their synthesis using an oxidative Baeyer–Villiger rearrangements of polymethoxy(methylenedioxy)phenols<sup>9,10</sup> which are readily available from parsley and dill seed metabolites.<sup>11,12</sup> The transformation proceeded in two steps: first, the classical rearrangement of benzaldehydes into phenols occurred, which were further oxidized to quinones with the opening of the methylenedioxy moiety.<sup>10</sup> However, the selectivity of the reactions depended on the nature of the alkyl substituent *ortho* to the aldehyde group.

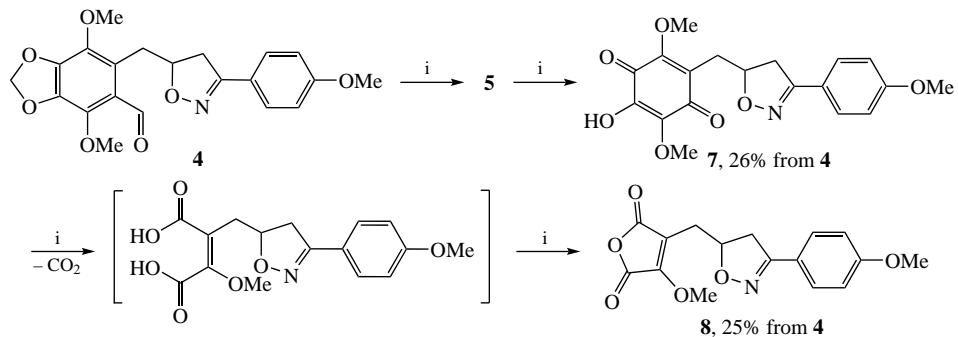
In the present work, such a rearrangement was studied using the example of apiolaldehyde bearing an *o*-(3-*p*-anisylisoxazolin-5-yl)methyl substituent (Scheme 1), which could be useful for the synthesis of quinones (like **6**) with biologically active  $\alpha$ -amino- $\gamma$ -hydroxyalkyl fragments. Quinones with such moieties as juglomycin C-amide isolated from the strain *S. coelicolor* A3(2) M145 belong to a family

of promising juglomycin-type antibiotics.<sup>13,14</sup> Based on the reaction of dipolar cycloaddition of apiol **1** and nitrile oxide **2**, isoxazoline **3** was synthesized,<sup>15</sup> which was further formylated into aldehyde **4** according to the reported procedure.<sup>10</sup>

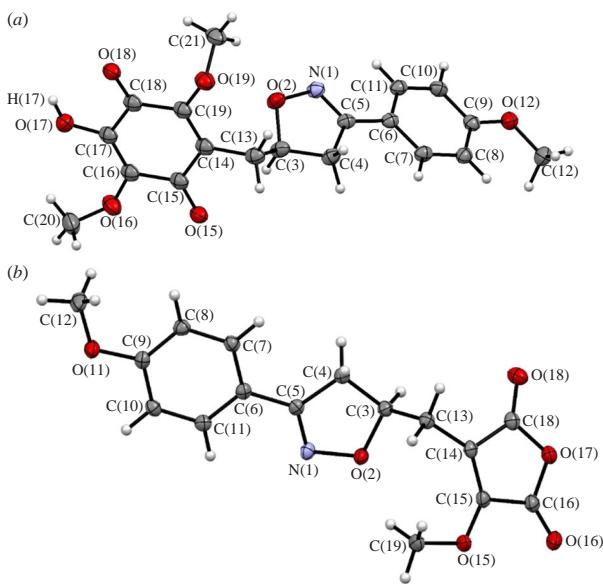
Aldehyde **4** normally undergoes the Baeyer–Villiger rearrangement in the presence of  $\text{SeO}_2$  in  $\text{H}_2\text{O}_2/\text{Bu}^t\text{OH}$  system to form phenol **5** (see Scheme 1). However, phenol **5** being isolated could not be further oxidized to quinone **7** (Scheme 2) under standard conditions ( $\text{H}_2\text{O}_2\text{–Ac}_2\text{O}$ ), similarly to most phenols with an apiol fragment.<sup>9,10</sup> Under these conditions, according to HRMS data, more than five products are formed, among which only traces of maleic anhydride derivative **8** (a decomposition product of quinone **7**) were detected. Fortunately, the *in situ* Baeyer–Villiger oxidation of compound **4** in the same medium ( $\text{H}_2\text{O}_2\text{–Ac}_2\text{O}$ ) could proceed further with the formation of quinone **7** and maleic anhydride **8** derivatives in acceptable yields (totally 51%). Apparently, in the first case, at a high initial concentration of polymethoxyphenol **5**, a side interaction of the resulting highly reactive quinone **8** with the initial **5** occurs.



**Scheme 1** Reagents and conditions: i,  $\text{HCO}_2\text{Et}$ ,  $\text{PCl}_5$ ,  $\text{SnCl}_4$ ,  $\text{CH}_2\text{Cl}_2$ , room temperature, 5 h (ref. 10); ii,  $\text{H}_2\text{O}_2$ ,  $\text{SeO}_2$ ,  $\text{Bu}^t\text{OH}$ , 50–60 °C, 3 h; iii,  $\text{MeOH}$ ,  $\text{Et}_3\text{N}$ , 20 °C, 1 h, then  $\text{HCl}/\text{CH}_2\text{Cl}_2$ , 20 °C.



**Scheme 2** Reagents and conditions: i,  $\text{H}_2\text{O}_2$ ,  $\text{H}_2\text{SO}_4$ ,  $\text{Ac}_2\text{O}$ , room temperature, 6 h.



**Figure 1** Structures of (a) compound 7 and (b) compound 8 in crystals. The thermal vibrations of atoms are represented by ellipsoids in anisotropic approximation ( $p = 50\%$ ).

Apparently, phenol 5 upon formation is further oxidized with methylenedioxy ring destruction to produce quinone 7. Further oxidative degradation of the benzene ring and decarboxylation occur to form a maleic acid derivative, which would cyclize in an acid medium to maleic anhydride 8 (see Scheme 2). The structure of new products 7 and 8 was proved by X-ray diffraction analysis (Figure 1).<sup>†</sup>

<sup>†</sup> Crystal data for 7.  $\text{C}_{19}\text{H}_{19}\text{NO}_7$  ( $M = 373.35$ ), orthorhombic, space group  $P2_12_12_1$ , at  $T = 100(1)$  K,  $a = 4.79523(7)$ ,  $b = 8.20587(11)$  and  $c = 43.0852(6)$  Å,  $\beta = 90^\circ$ ,  $V = 1695.36(4)$  Å $^3$ ,  $Z = 4$ ,  $d_{\text{calc}} = 1.463$  g cm $^{-3}$ ,  $F(000) = 784$ ,  $\mu = 0.949$  mm $^{-1}$ . 13607 reflections (3493 independent reflections,  $R_{\text{int}} = 0.0435$ ) were measured and used in the refinement. The refinement converged to  $R_1 = 0.0435$  for 3277 observed reflections with  $I > 2\sigma(I)$  and  $wR_2 = 0.1137$  for all independent reflections, GOF = 1.075.

Crystal data for 8.  $\text{C}_{16}\text{H}_{15}\text{NO}_6$  ( $M = 317.29$ ), monoclinic, space group  $P2_1/n$ , at  $T = 100(1)$  K,  $a = 8.29006(9)$ ,  $b = 14.54498(18)$  and  $c = 11.91274(13)$  Å,  $\beta = 94.3701(10)^\circ$ ,  $V = 1432.25(3)$  Å $^3$ ,  $Z = 4$ ,  $d_{\text{calc}} = 1.471$  g cm $^{-3}$ ,  $F(000) = 664$ ,  $\mu = 0.962$  mm $^{-1}$ . 16647 reflections (3102 independent reflections,  $R_{\text{int}} = 0.0311$ ) were measured and used in the refinement. The refinement converged to  $R_1 = 0.0345$  for 2886 observed reflections with  $I > 2\sigma(I)$  and  $wR_2 = 0.0935$  for all independent reflections, GOF = 1.046.

CCDC 2291864 (7) and 2291865 (8) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via <http://www.ccdc.cam.ac.uk>.

In summary, polyalkoxybenzaldehydes in the Baeyer–Villiger rearrangement can undergo further oxidation with the decomposition of the benzene ring.

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#### Online Supplementary Materials

Supplementary data associated with this article can be found in the online version at doi: 10.1016/j.mencom.2024.04.026.

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