

Co-crystallization of isomers of iron(II) complex of a tridentate mixed amine/imine ligand in one unit cell

Carl B. Hollandsworth, Blayne M. Griffin, John Raymon Pruden,
Nikolay Gerasimchuk and Matthew Zeller

S1

Experimental Details

Synthesis of $[\text{Fe}(\text{trans-L})_2](\text{PF}_6)_2$ 1, mixture of stereoisomers: In a typical experiment, ammonium iron(II) sulfate (392 mg, 1.00 mmol) was dissolved in 5.0 mL of de-ionized water in a 50 mL round bottom flask, after which 15.0 mL of 95% EtOH was added. Pyridine-2-carboxaldehyde (214 mg, 2 mmol) was added in one portion while stirring. Subsequently, racemic *trans*-1,2-diaminocyclohexane (228 mg, 2 mmol) was added dropwise with stirring, immediately giving an indigo solution. The solution was allowed to stir at room temperature for 2 h, after which most of the ethanol and water were removed under reduced pressure. Further purification was achieved by extraction of the hexafluorophosphate salt into CH_2Cl_2 followed by removal of the solvent at ambient temperature and under reduced pressure. Yield after drying (39%). Elemental microanalysis was performed by Micro-Analysis, Inc. laboratory in Wilmington, DE, USA. *It should be noted that upon growing single crystals we observed incorporation of solvents of the process (propionitrile – from synthesis, and diethyl ether – from vapor diffusion method) into the asymmetric unit (ASU) content. This topic analyzed further in S10, S11.*

Elemental analysis for bulk powder: calculated for $\text{C}_{24}\text{H}_{34}\text{F}_{12}\text{FeN}_6\text{P}_2$: C - 38.31%; H - 4.55%; N - 11.17%; found: C - 38.84%; H - 4.68%; N - 10.91%. Complex decomposes at $\sim 190^\circ\text{C}$ with elimination of PF_5 .

Target dry complex represents dichroic solid with crystals being blue-black on reflected light and dark-red on transient light. The complex is soluble in most organic solvents and water, but not in hydrocarbons.

The IR, ^1H NMR, $^{13}\text{C}\{^1\text{H}\}$ NMR spectra, and electronic spectral data for the target complex $[\text{Fe}(\text{trans-}(1R,2R)\text{-L})_2](\text{PF}_6)_2$ were recorded and presented below.

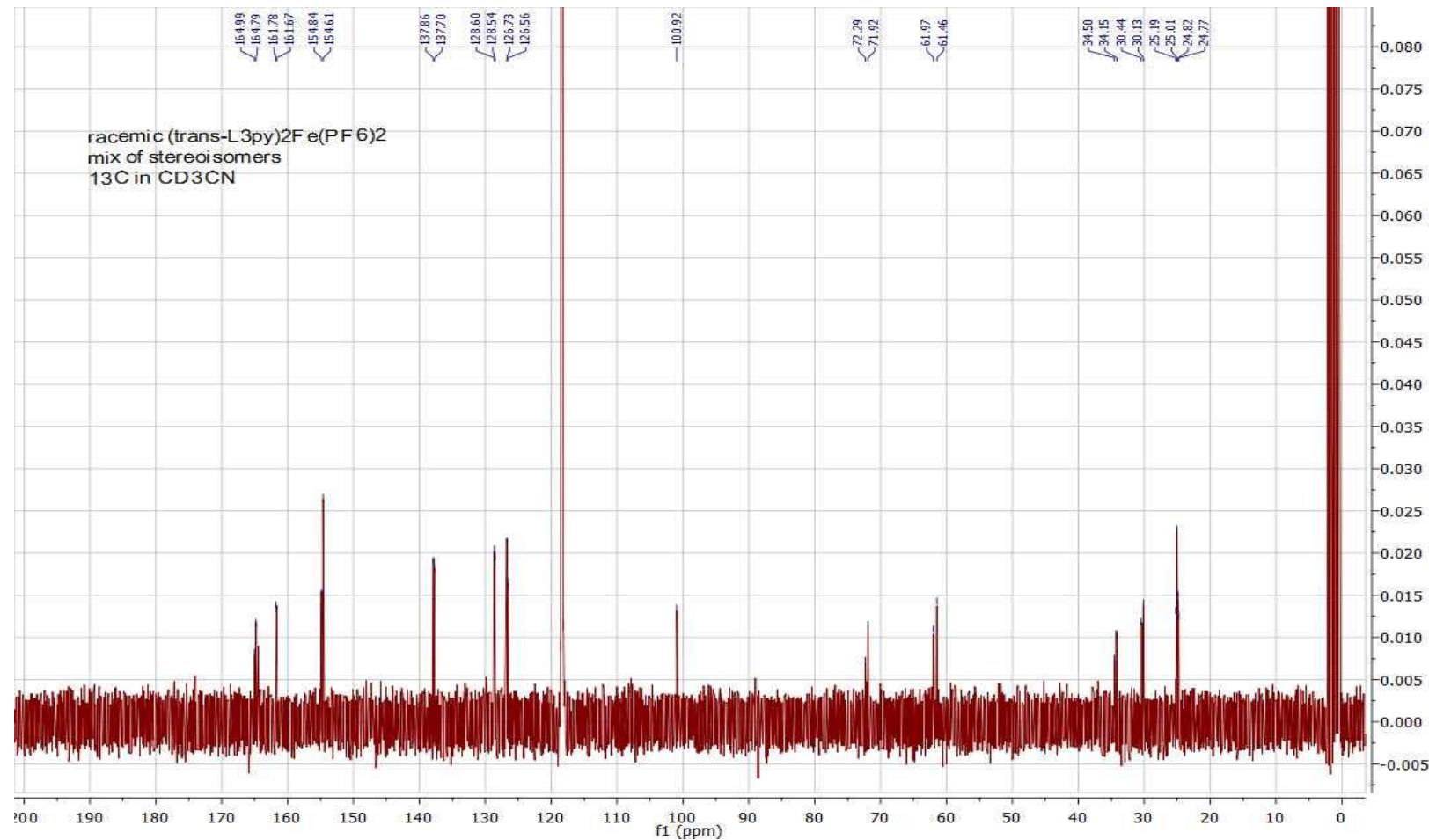
Physical methods and instrumentation used.

The FTIR spectra were obtained on a Thermo Nicolet iS5 spectrophotometer and analyzed with OMNIC software (thermofisher.com). IR spectra were obtained on solid samples *via* ATR using a diamond transmission window. The UV-Visible spectra were obtained using HP 8453 spectrophotometer in dried organic solvents received from commercial vendors and used without further purification.

The NMR spectra were recorded in deuterated MeCN on a 300 MHz JEOL ECX-300 FT at 298 K with chemical shifts referenced to signals from the deuterated solvent. Magnetic susceptibility measurements were performed on anhydrous solids of $[\text{Fe}(\text{trans-}(1R,2R)\text{-L})_2](\text{PF}_6)_2$ using a Johnson-Matthey MSB Mk1 magnetic susceptibility balance.

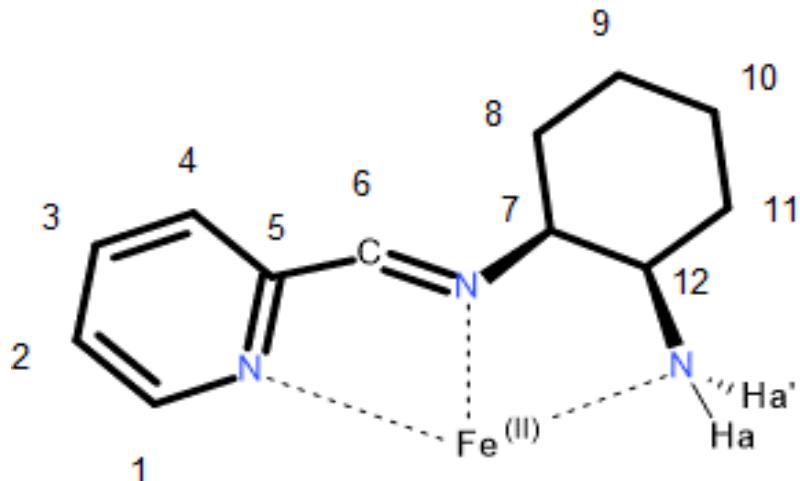
S2

The $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of stereoisomers of $[\text{Fe}(\text{trans-}\text{L})_2](\text{PF}_6)_2$ **1** in CD_3CN at 290 K showing clearly seen characteristic “doubling” of each resonance due to presence of the mixture.



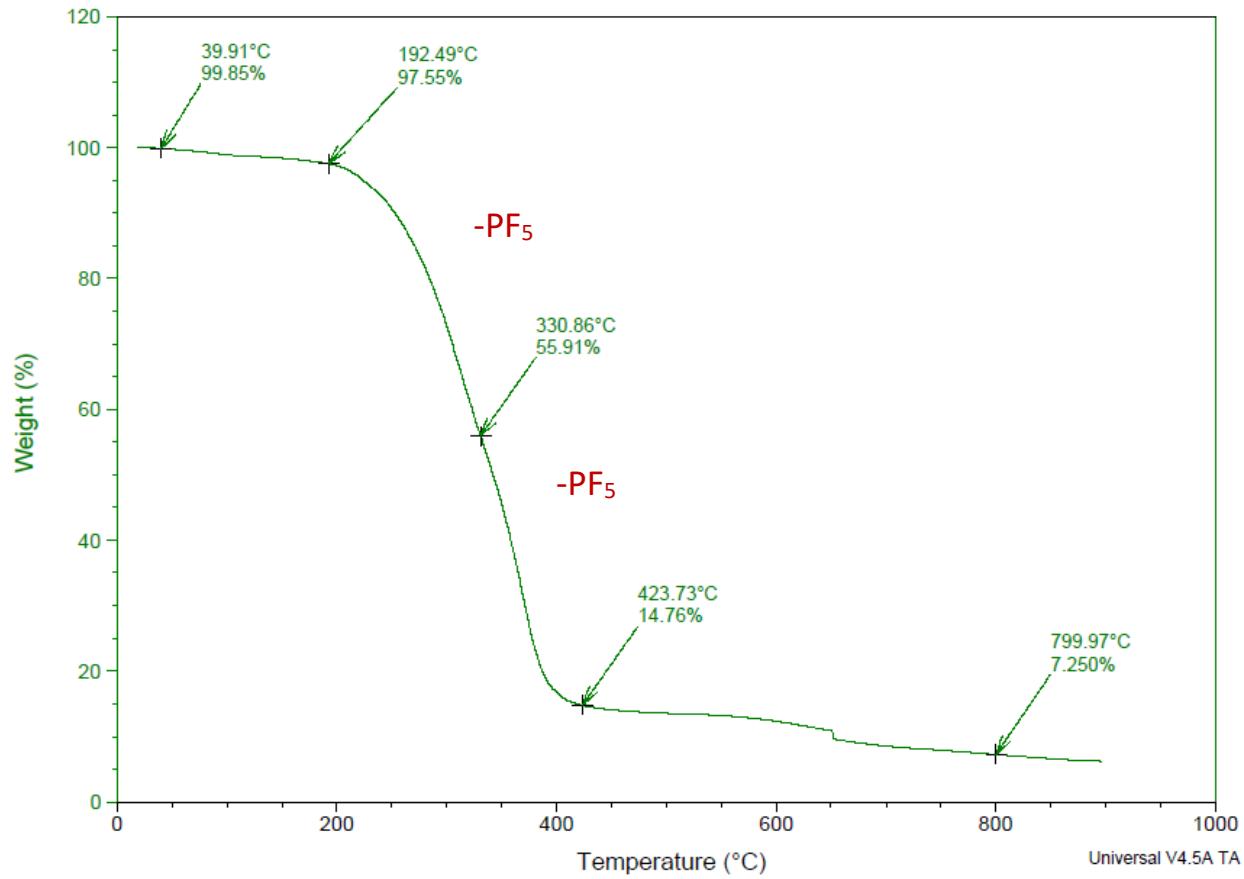
The NMR Characterization of $[\text{Fe}(\text{trans-L})_2](\text{PF}_6)_2$, mixture of stereoisomers in detail.

Used numbering scheme for ^{13}C -NMR signals assignments for $[\text{Fe}(\text{trans-L})_2](\text{PF}_6)_2$.



Resonances labeled for “Complex A” are assigned to the set of enantiomers $[\text{Fe}(\text{trans-}(1R,2R)-\text{L})_2](\text{PF}_6)_2$ and $[\text{Fe}(\text{trans-}(1S,2S)-\text{L})_2](\text{PF}_6)_2$ which are indistinguishable by NMR and have identical chemical shifts to the compound made from pure *trans*-(1*R*,2*R*)-diaminocyclohexane: $[\text{Fe}(\text{trans-}(1R,2R)-\text{L})_2](\text{PF}_6)_2$. Resonances labeled for “Complex B”, arise from the diastereomer assigned formula $[\text{Fe}(\text{trans-}(1R,2R)-\text{L})(\text{trans-}(1S,2S)-\text{L})](\text{PF}_6)_2$. The $^{13}\text{C}\{^1\text{H}\}$ -NMR (75.6 MHz, CD_3CN , ppm; carbon atom labels are as assigned in **Figure S2**) 165.0, 164.8 (C6A, C6B), 161.8, 161.7 (C5A, C5B), 154.8, 154.6 (C1A, C1B), 137.9, 137.7 (C4A, C4B), 128.6, 128.5 (C3B, C3A), 126.7, 126.6 (C2B, 2A), 72.3, 72.0 (C7B, C7A), 62.0, 61.5 (C12A, C12B), 34.5, 34.2 (C8B, C8A), 30.4, 30.1 (C11B, C11A), 25.2, 25.0 (C9 or C10A, C9 or C10B), 24.82, 24.77 (C9 or C10B and C9 or C10A).

Weight view of the TG/DSC experiment for the $[\text{Fe}(\text{trans-L})_2](\text{PF}_6)_2$ complex showing two consecutive PF_5 weight losses.

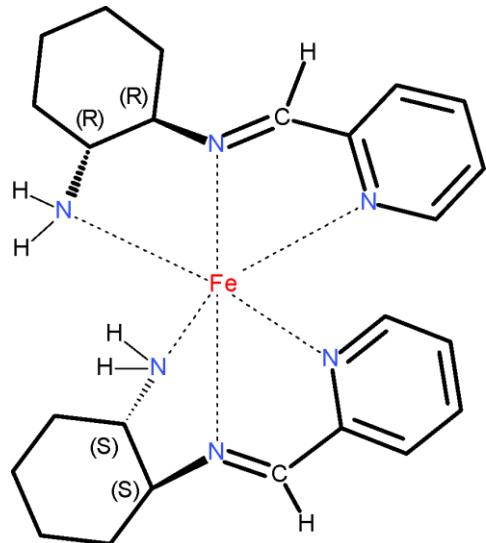


The TG analysis for $[\text{Fe}(\text{L})_2](\text{PF}_6)_2$ was performed on a TA Instrument Q-600 (Delaware, USA) in an alumina crucible in the range between 30 – 1000°C under nitrogen flow at 100 ± 1 mL/min and a heating rate of 10 °C/min. The crucible was calcined prior to each experiment using a propane torch. Data of the TG analyses were processed using the TA Universal Analysis software package.

Schematic drawing of stereoisomers of $[\text{Fe}(\text{trans-}\mathbf{L})_2]^{2+}$ and their manifestation in the NMR spectra.

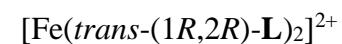
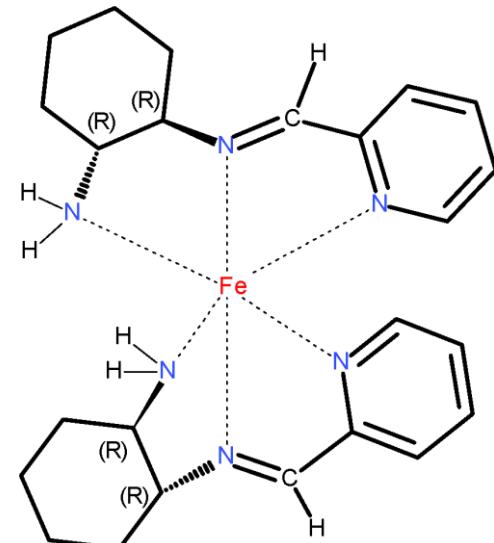
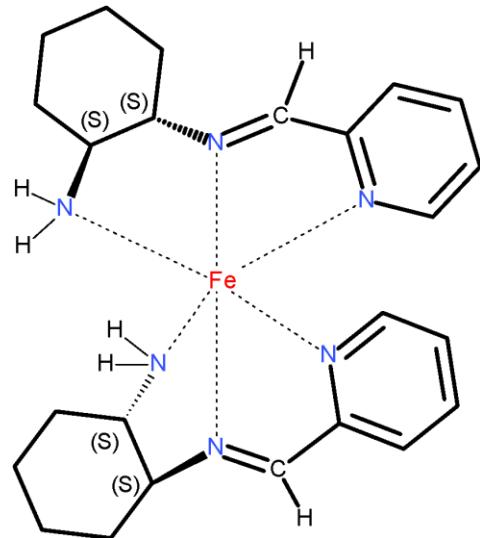
Diastereomer

(one set of ^{13}C NMR signals: well observable)



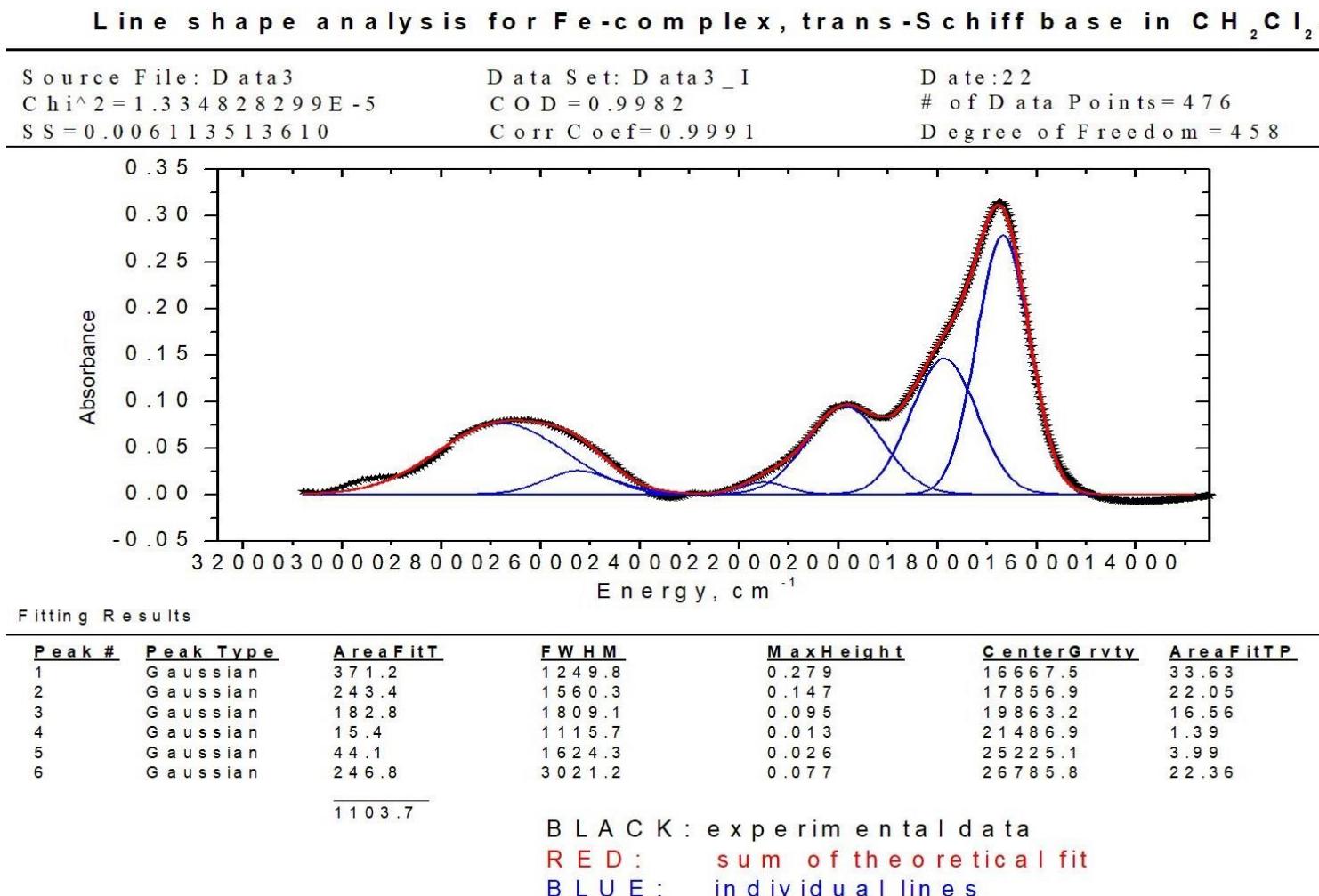
Set of *Enantiomers*

(second set of ^{13}C NMR signals, unseparable)



S6

Full line shape analysis of the UV-visible spectrum of $[\text{Fe}(\text{trans-}\mathbf{L})_2](\text{PF}_6)_2$. Only six Gaussian-type bands are sufficient for analysis.



X-ray Analysis.

A single crystal (dark red plate on transient light, but blue-black on reflected light) containing a mixture of stereoisomers of racemic $[\text{Fe}(\text{trans-}\mathbf{L})_2](\text{PF}_6)_2$ was grown using vapor diffusion of diethyl ether anti-solvent into a saturated solution of the complex in propionitrile. *Both those solvents were found to be present in the crystal lattice, and both were observed to be disordered. Thus, propionitrile $\text{C}_2\text{H}_5\text{CN}$ and $(\text{C}_2\text{H}_5)_2\text{O}$ were successfully modelled by two positions with respective site occupancy factors (SOFs) tied to one (see S10, S11 down below).*

Suitable single crystals were inspected, selected, and handled in NVH parathon oil, with subsequent mounting on a thin glass fiber, or placed into a CryoLoop and then attached to a copper pin positioned on the goniometer head of a diffractometer equipped with a CCD area detector. All data sets were measured at low temperatures as specified in the crystal structure and refinement data in the table below. Data were collected in ω scan mode using the Mo tube ($\text{K}\alpha$ radiation; $\lambda = 0.71073 \text{ \AA}$) with a highly oriented graphite monochromator. The intensities for the latter radiation were integrated from four series of 364 exposures each, covering 0.5° in ω , with the total data set being a sphere. The space group determination was done with the aid of XPREP software.¹ The absorption correction was performed using values from face-indexed crystals and numerical values obtained from the set of images recorded with a video-microscope, followed by the SADABS program that was included in the Bruker AXS software package.^{2,3}

All structures were solved by direct methods and refined by least squares on weighted F^2 values for all reflections using SHELXS-2013. The structures reported herein are well refined and without apparent errors. All hydrogen atoms in the structures were placed to their hosting atoms (C, N) and refined isotropically. The crystal structures and packing diagrams were drawn using the ORTEP 3v2^{4,5} and Mercury software packages.⁶ Thermal ellipsoids appearing in figures are drawn at their 50% probability level.

As we stated at the beginning of the Supplementary Materials section the bulk powder of the complex **1** did not contain solvents of inclusion. However, upon crystal growth we detected disordered molecules of two solvents used in the system: propionitrile and diethyl ether. Analytical data for the powdery sample were of satisfactory level, and no attempts to record powder diffraction pattern (PXRD) for bulk sample were made. Having fairly large solvent molecules occluded into the single crystal that had been analyzed using X-ray crystallography certainly changed the atomic content and undoubtedly unit cell dimensions. Therefore, no recording and comparison of PXRD patterns for the bulk powder and single crystal were done.

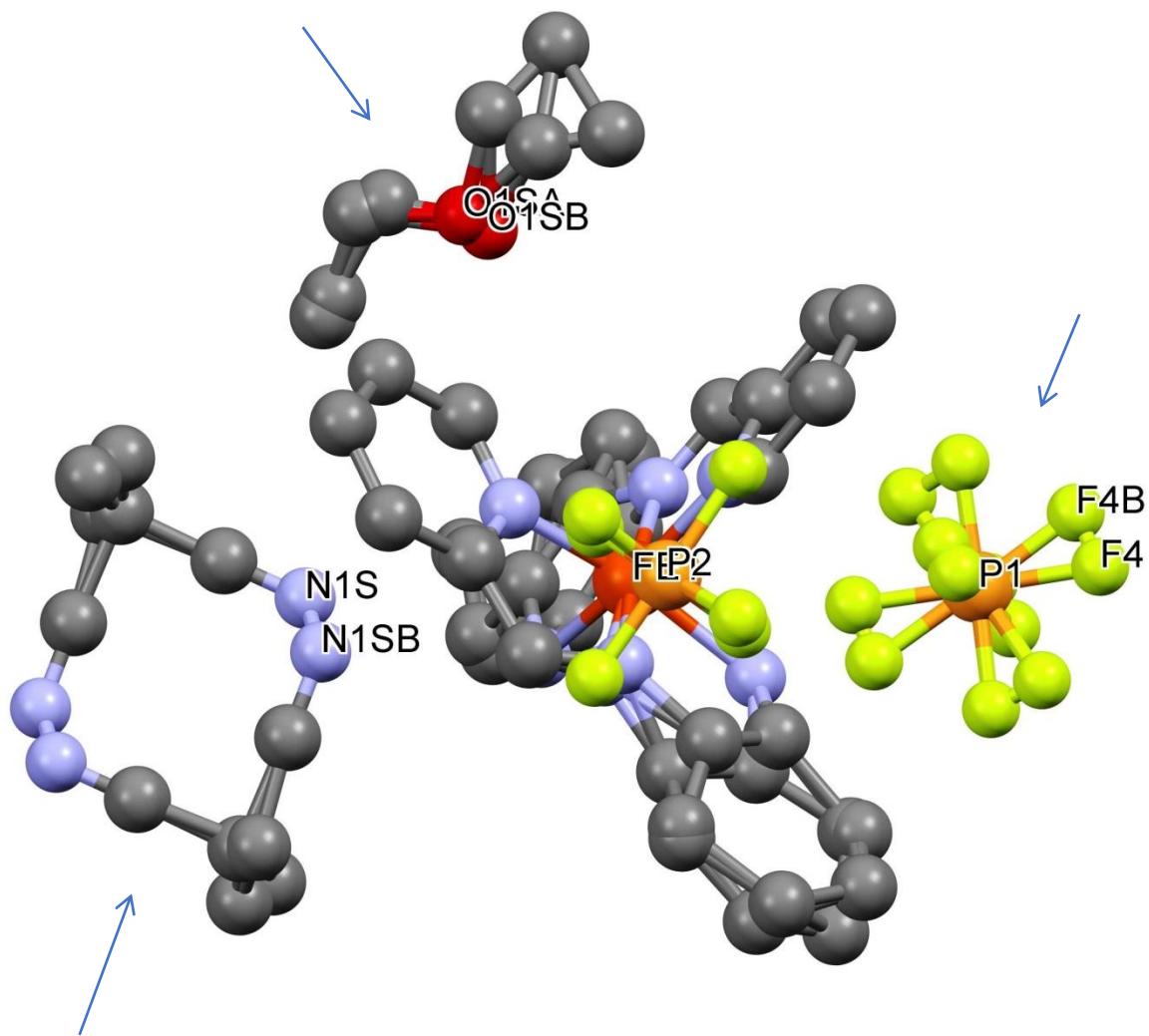
S8

Table 1. Crystal and refinement data for $[\text{Fe}(\text{trans-}\mathbf{L})_2](\text{PF}_6)_2$ **1** (which was a mixture of $[\text{Fe}(\text{trans-}(1R,2R)\text{-}\mathbf{L})_2](\text{PF}_6)_2$, $[\text{Fe}(\text{trans-}(1S,2S)\text{-}\mathbf{L})_2](\text{PF}_6)_2$, and $[\text{Fe}(\text{trans-}(1R,2R)\text{-}\mathbf{L})(\text{trans-}(1S,2S)\text{-}\mathbf{L})](\text{PF}_6)_2$ as ether/propionitrile solvate.

Parameter	Data for complex
Empirical formula	$\text{C}_{31}\text{H}_{49}\text{F}_{12}\text{FeN}_7\text{OP}_2$
Formula weight, g/M	881.09
Temperature, K	120(2)
Crystal system	monoclinic
Color /habitus	dark-red plate
Crystal size (mm)	$0.047 \times 0.226 \times 0.311$
Space group	$\text{P}2_1/\text{c}$, #14
Cell constants, $\text{\AA}/^\circ$	
a	9.3699(7)
b	20.0609(15)
c	21.0430(16)
α	90
β	100.187(1)
γ	90
Volume (\AA^3)	1102.5(6)
Z	4
ρ_{calc} (g/cm^3)	1.503
μ (mm^{-1})	0.564
F(000)	1822
2θ range for data ($^\circ$)	3.92 to 50.484
Index ranges:	
Reflections collected	-13 < h < 13
Independent	-28 < k < 28
Data/Restraints/Parameters	-29 < l < 29
Goodness-of-fit on F^2	61453
Final R indices:	4553 [$R_{\text{int}}=0.0288$; $R_{\sigma}=0.0226$]
All data:	11407 / 948 / 753
Peak/hole difference, e(\AA^{-3})	1.034
Final R indices:	9383 [$I>2\sigma(I)$]; $R1 = 0.0398$, $wR2 = 0.0526$
All data:	$R1 = 0.0917$; $wR2 = 0.0975$
Volume taken, \AA^3 (%)	+0.845 / -0.642
Volume taken, \AA^3 (%)	2740.1 (70.4)

The asymmetric unit (ASU) in the crystal structure of $[\text{Fe}(\text{trans-}\mathbf{L})_2](\text{PF}_6)_2$ **1** co-crystallized isomers.

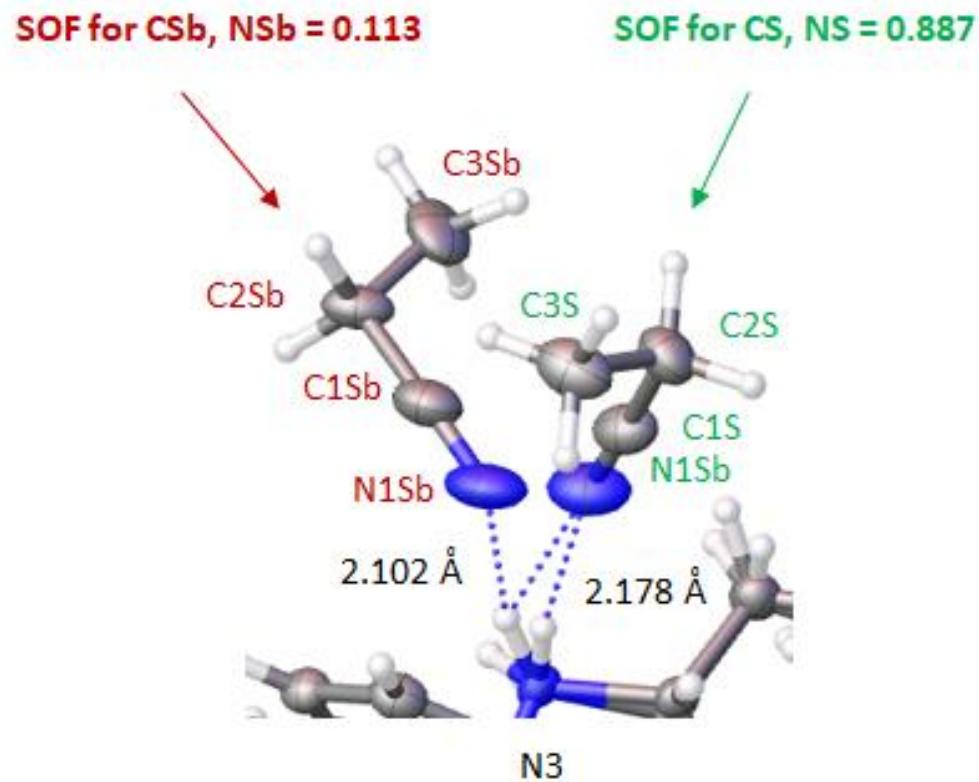
Disordered PF_6^- anion (at P1) and two stoichiometric molecules of solvents of crystallization $\text{C}_2\text{H}_5\text{CN}$ and diethyl ether are shown with arrows. Hydrogen-atoms are omitted for clarity.



Details of the disorder:

Explanation and presentation of disordered fragments in the structure of $[\text{Fe}(\text{trans-L})_2](\text{PF}_6)_2$: disordered by two positions molecules of the solvent of reaction - propionitrile. The geometry of long electrostatic C-H---N contacts is shown.

SOF = site occupancy factor with the sum being equal to 1 for modeled two-positional disorder.

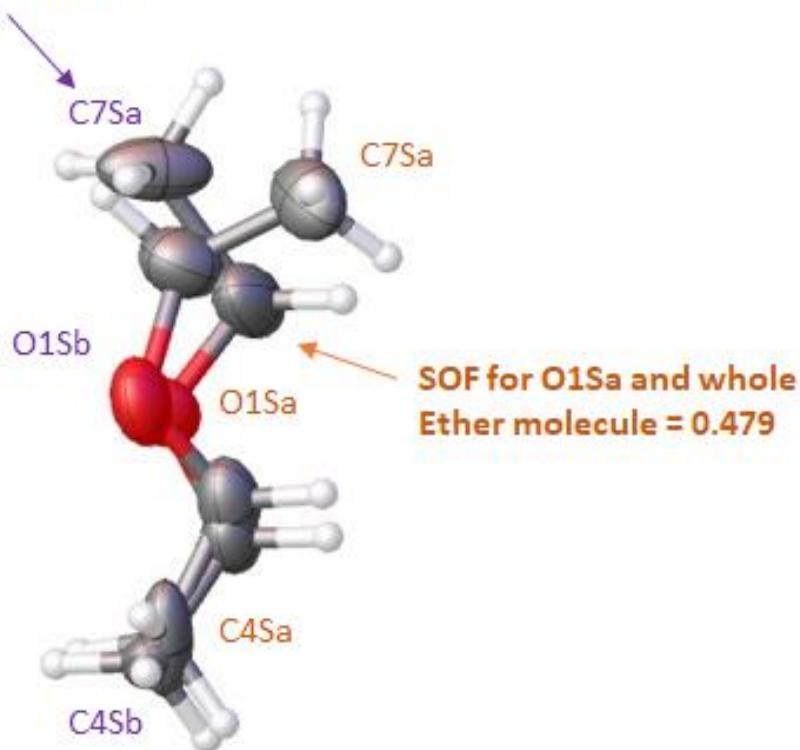


Details of the disorder:

Explanation and presentation of disordered fragments in the structure of $[\text{Fe}(\text{trans-L})_2](\text{PF}_6)_2$: disordered by two positions molecules of the “pushing” solvent of crystallization – diethyl ether.

SOF = site occupancy factor with the sum being equal to 1 for modeled two-positional disorder.

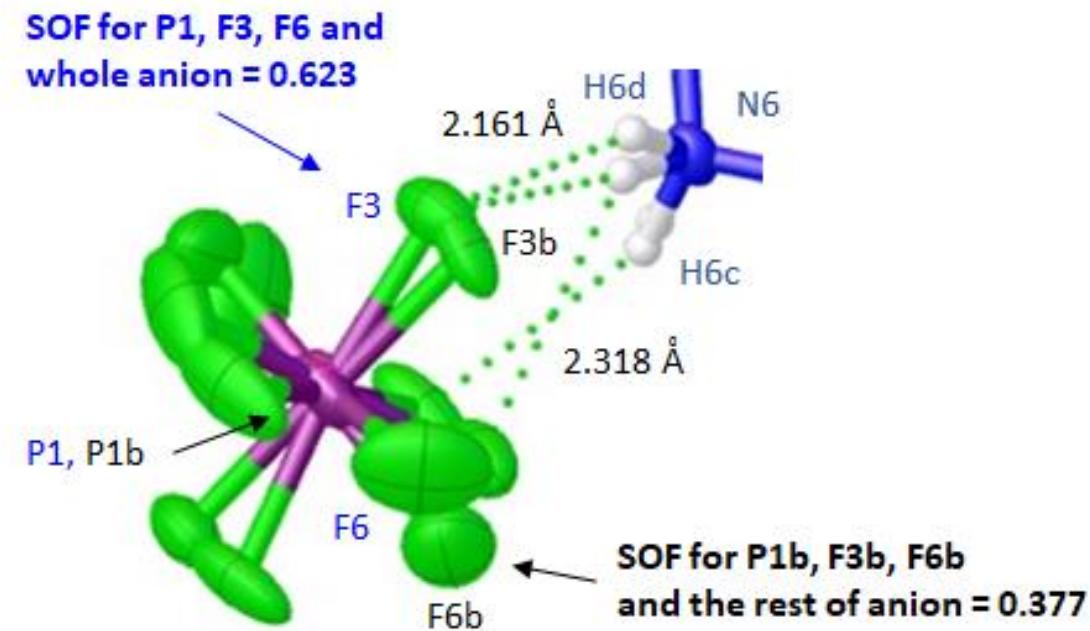
SOF for O1Sb and whole ether molecule = 0.521



Details of the disorder:

Explanation and presentation of disordered fragments in the structure of $[\text{Fe}(\text{trans-L})_2](\text{PF}_6)_2$: disordered by two positions one of the counter ions – PF_6^- . The geometry of long electrostatic N-H---F contacts is also shown.

SOF = site occupancy factor with the sum being equal to 1 for modeled two-positional disorder.



checkCIF/PLATON (basic structural check)

Structure factors have been supplied for datablock(s) steroisomers_mixture-Fe-2

No syntax errors found. [CIF dictionary](#)

Please wait while processing [Interpreting this report](#)

[Structure factor report](#)

Datablock: steroisomers_mixture-Fe-2

Bond precision:	C-C = 0.0024 Å	Wavelength=0.71073
Cell:	a=9.3699(7) b=20.0609(15) c=21.0430(16)	
	alpha=90 beta=100.187(1) gamma=90	
Temperature:	120 K	
	Calculated	Reported
Volume	3893.1(5)	3893.1(5)
Space group	P 21/n	P 21/n
Hall group	-P 2yn	-P 2yn
Moiety formula	C24 H34 Fe N6, 2(F6 P), C4 H10 O, C3 H5 N	C24 H34 Fe N6, 2(F6 P), C4 H10 O, C3 H5 N
Sum formula	C31 H49 F12 Fe N7 O P2	C31 H49 F12 Fe N7 O P2
Mr	881.56	881.09
Dx, g cm ⁻³	1.504	1.503
Z	4	4
μ (mm ⁻¹)	0.564	0.564
F000	1824.0	1822.0
F000'	1827.31	
h, k, lmax	13, 28, 29	13, 28, 29
Nref	11448	11407
Tmin, Tmax	0.857, 0.974	0.669, 0.745
Tmin'	0.839	
Correction method=	# Reported T Limits: Tmin=0.669	
Tmax=0.745	AbsCorr = MULTI-SCAN	
Data completeness=	0.996	Theta(max) = 30.083
R(reflections)=	0.0398 (9383)	wR2(reflections)= 0.0975 (11407)
S =	1.034	Npar= 753

The following ALERTS were generated. Each ALERT has the format

test-name_ALERT_alert-type_alert-level.

Click on the hyperlinks for more details of the test.

● Alert level C

PLAT068 ALERT 1 C	Reported F000 Differs from Calcd (or Missing)...	Please Check
PLAT250 ALERT 2 C	Large U3/U1 Ratio for Average U(i,j) Tensor	2.9 Note
PLAT767 ALERT 4 C	INS Embedded LIST 6 Instruction Should be LIST 4	Please Check
PLAT911 ALERT 3 C	Missing FCF Refl Between Thmin & STh/L= 0.600	2 Report

● Alert level G

PLAT002 ALERT 2 G	Number of Distance or Angle Restraints on AtSite	59 Note
PLAT003 ALERT 2 G	Number of Uiso or Uij Restrained non-H Atoms ...	58 Report
PLAT007 ALERT 5 G	Number of Unrefined Donor-H Atoms	8 Report
PLAT171 ALERT 4 G	The CIF-Embedded .res File Contains EADP Records	1 Report
PLAT175 ALERT 4 G	The CIF-Embedded .res File Contains SAME Records	6 Report
PLAT176 ALERT 4 G	The CIF-Embedded .res File Contains SADI Records	3 Report
PLAT178 ALERT 4 G	The CIF-Embedded .res File Contains SIMU Records	5 Report
PLAT188 ALERT 3 G	A Non-default SIMU Restraint Value has been used	0.0100 Report

And 4 other PLAT188 Alerts

More ...

PLAT231 ALERT 4 G	Hirshfeld Test (Solvent) P1 --F1 . 6.0 s.u.
PLAT231 ALERT 4 G	Hirshfeld Test (Solvent) P1 --F6 . 8.5 s.u.
PLAT244 ALERT 4 G	Low 'Solvent' Ueq as Compared to Neighbors of P2 Check
PLAT301 ALERT 3 G	Main Residue Disorder(Resd 1) 42% Note
PLAT302 ALERT 4 G	Anion/Solvent/Minor-Residue Disorder (Resd 2) 100% Note

And 5 other PLAT302 Alerts

More ...

PLAT304 ALERT 4 G	Non-Integer Number of Atoms in (Resd 2) 4.36 Check
-----------------------------------	---

And 5 other PLAT304 Alerts

More ...

PLAT398 ALERT 2 G	Deviating C-O-C Angle From 120 for O1SA . 107.8 Degree
PLAT398 ALERT 2 G	Deviating C-O-C Angle From 120 for O1SB . 108.0 Degree
PLAT720 ALERT 4 G	Number of Unusual/Non-Standard Labels
PLAT773 ALERT 2 G	Check long C-C Bond in CIF: C4SB --C5SB 1.75 Ang.
PLAT811 ALERT 5 G	No ADDSYM Analysis: Too Many Excluded Atoms ! Info
PLAT860 ALERT 3 G	Number of Least-Squares Restraints 948 Note
PLAT883 ALERT 1 G	No Info/Value for _atom_sites_solution_primary . Please Do !
PLAT899 ALERT 4 G	SHELXL2018 is Deprecated and Succeeded by SHELXL 2019/3 Note
PLAT910 ALERT 3 G	Missing # of FCF Reflection(s) Below Theta(Min). 1 Note
PLAT912 ALERT 4 G	Missing # of FCF Reflections Above STh/L= 0.600 39 Note
PLAT933 ALERT 2 G	Number of HKL-OMIT Records in Embedded .res File 2 Note
PLAT978 ALERT 2 G	Number C-C Bonds with Positive Residual Density. 7 Info

0 **ALERT level A** = Most likely a serious problem - resolve or explain

0 **ALERT level B** = A potentially serious problem, consider carefully

4 **ALERT level C** = Check. Ensure it is not caused by an omission or oversight

40 **ALERT level G** = General information/check it is not something unexpected

2 ALERT type 1 CIF construction/syntax error, inconsistent or missing data

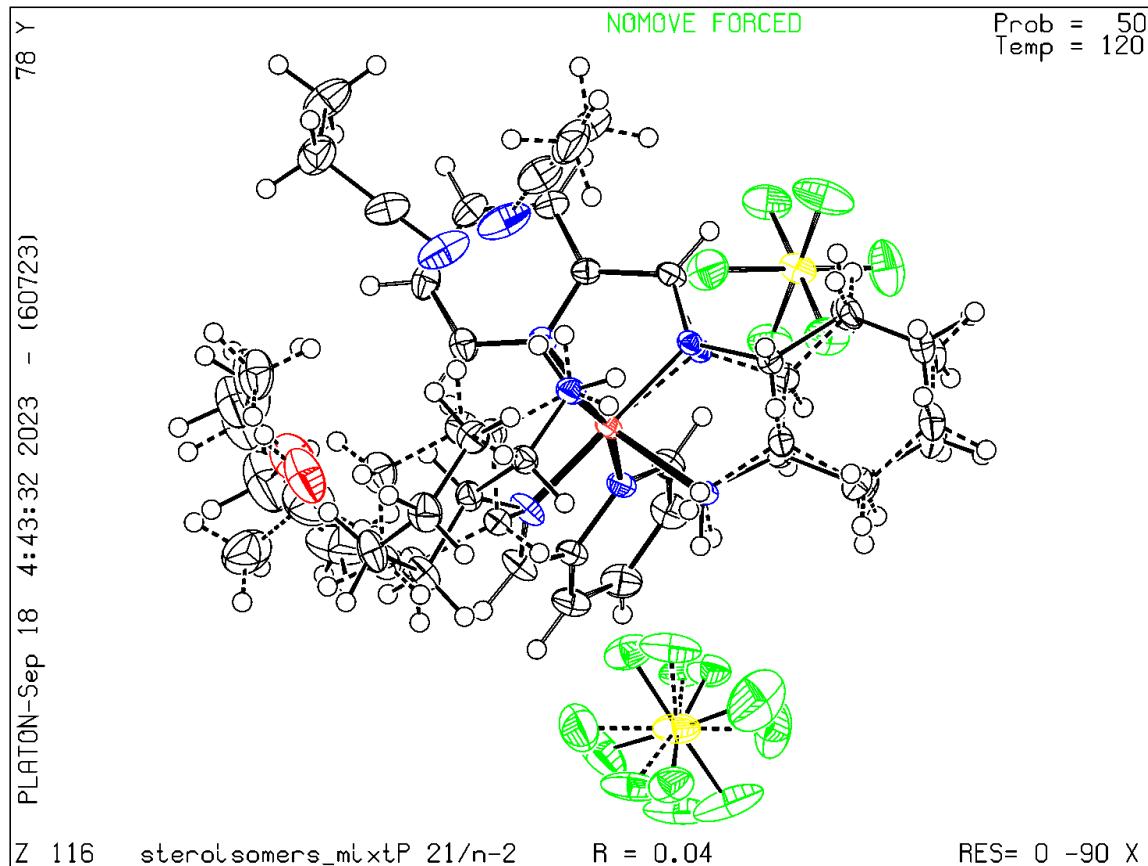
8 ALERT type 2 Indicator that the structure model may be wrong or deficient

9 ALERT type 3 Indicator that the structure quality may be low

23 ALERT type 4 Improvement, methodology, query or suggestion

2 ALERT type 5 Informative message, check

Datablock stereoisomers_mixture-Fe-2 - ellipsoid plot



Necessary references for Supplemental Materials section:

1. G. S. Girolami, A Guide to Using the SHELXTL Crystallographic Software Package. Department of Chemistry, University of Illinois at Urbana-Champaign; 2004.
2. R. H. Blessing, *Acta Cryst.*, 1995, **A51**, 33.
3. G. M. Sheldrick, Bruker APEX2 and APEX3 Software Suits; Bruker AXS, Madison, WI, USA.
4. L. J. Farrugia, *J. Appl. Cryst.*, 1997, **30**, 565.
5. M. N. Burnett, C. K. Johnson, ORTEP-III: Oak Ridge Thermal Ellipsoid Plot Program for Crystal Structure Illustrations., Oak Ridge National Laboratory, Oak Ridge, TN, USA, 1996.
6. Mercury Software, CCDC (Cambridge Crystal Data Centre), Cambridge, England, 4.2 edn., 2016.