

**Electrochemical synthesis and properties of polyporphyrin films  
based on 5,10,15,20-tetra(4-pyridyl)porphyrin**

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**Materials and methods**

**Starting 5,10,15,20-tetra(4-pyridyl)porphyrin 1** was prepared by the known procedure with modifications [S1,S2].

To a boiling mixture of acetic acid (500 ml), nitrobenzene (150 ml) and acetic anhydride (27 ml, 0.286 mol) was added gradually a mixture of pyrrole (10 ml, 0.144 mol) and pyridine-4-carboxaldehyde (13.7 ml, 0.144 mol). The mixture was boiled for 1.5 h and cooled down. After the acetic acid was removed by rotary distillation, the residue was diluted with water (150 ml), after which the nitrobenzene was steam-distilled off. The residue was filtered, washed with water and dried in air. The residue was then boiled with methanol (100 ml) under stirring; the solid was filtered, washed with methanol and dried. The material was extracted by chloroform in a Soxhlet apparatus until discoloration of the solvent. The extract was evaporated to 200 ml, and the resulting precipitate was filtered out. The solution was chromatographed on a column (3 × 50 cm) with aluminum oxide, Brockmann grade II. The eluate was evaporated up to the minimum volume and product **1** was precipitated with methanol (50 ml). The porphyrin was filtered, washed with methanol (50 ml) and dried at 70 °C until a constant weight. Yield: 6.3 g (25%).  $R_f$  0.80 (silufol,  $\text{CHCl}_3$  / MeOH 5:1).

UV-vis ( $\lambda_{\text{max}}$ , nm (lg $\epsilon$ ),  $\text{CHCl}_3$ ): 643 (3.43); 589 (3.82); 546 (3.79); 514 (4.30); 417 (5.62).

$^1\text{H}$  NMR ( $\text{CDCl}_3$ ): 8.87s (8H,  $\beta$ -H); 9.08d (8H, 2,6-H); 8.18d (8H, 3,5-H); -2.92s (2H, NH).

Mass spectrum,  $m/z$ : 619.27  $[\text{M}+\text{H}]^+$ . The calculated value was 618.70.

### **Electrochemical measurements**

The electrochemical measurements were made with an SP-150 potentiostat (Bio-Logic Science Instruments, France).

### **UV-Vis spectra.**

The electronic absorption spectra in the range of 350-800 nm of the porphyrin solution in dichloromethane and polyporphyrin film on an ITO electrode were registered with a Cary 50 spectrophotometer (Varian, USA).

### **IR-spectra.**

The IR-spectra of the studied compounds were obtained on a Bruker Vertex 80 spectrometer (Germany) in the region of 4500-350  $\text{cm}^{-1}$  with the resolution of 0.1  $\text{cm}^{-1}$ .

### **References**

- S1 A. D. Adler, F. R. Longo, J. D. Finarelli, J. Goldmacher, J. Assour and L. J. Korsakoff, *J. Org. Chem.*, 1967, **32**, 476.
- S2 Q. Lin, C. Mao, A. Kong, X. Bu, X. Zhao and P. Fen, *J. Mater. Chem. A*, 2017, **5**, 21189.