

**Magnetically recyclable Cu-BTC@Fe₃O₄-catalyzed synthesis of aryl
 α -chlorobenzyl ketones**

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General Experimental

Unless otherwise noted, all chemicals were purchased from commercial suppliers (Aladdin) and used without further purification. ¹H NMR and ¹³C NMR spectra were recorded at ambient temperature on a Bruker AVANCE III 300 spectrometer. Chemical shifts are reported in δ units, parts per million (ppm), and were referenced to CDCl₃ (7.26 or 77.0 ppm) as the internal standard. The coupling constants *J* are given in Hz. Column chromatography was performed using EM Silica gel 60 (300-400 mesh). Chromato-mass spectrometry (LCMS) was performed on a Shimadzu LCMS-2020 instrument with an ESI source. Fourier transform (FT) Infrared data were recorded on an AVATAR-370 (Nicolet) spectrometer by transmission through the sample deposited on a KBr pellet. Powder X-ray diffraction (PXRD) patterns of the as-synthesized and recycled samples were obtained using a Rigaku D/max 2500 PC X-ray diffractometer with Cu K α (1.5406 Å) radiation at 10 min⁻¹. Thermogravimetric analysis (TGA) experiments were carried out on a TG/DTA 6300 thermoanalyzer from room temperature to 800 °C under nitrogen atmosphere at a heating rate of 10 °C/min. Field-emission Scanning Electron Microscopy (SEM) images of the samples were taken at 30 kV with a ZEISS Supra 55 microscope. The compositions of samples were analyzed by ICP (inductively coupled plasma) analysis (Varian Vista-AX). N₂ adsorption isotherms were obtained using a Micromeritics ASAP 2460 instrument.

Preparation of Cu-BTC@Fe₃O₄ composite

Cu-BTC@Fe₃O₄ was prepared through a secondary growth strategy according to the literature.^{S1} PVP (0.2 g, 0.005 mmol) and Cu(OAc)₂·H₂O (0.1 g, 0.5 mmol) were uniformly dissolved in a mixed solution of EtOH/H₂O (2:1, 90 mL) in a 150 ml round-bottom flask under mechanical stirring (600 rpm). Then 0.2 g carboxyl functionalized Fe₃O₄ was gradually added to the mixed solution with high-speed mechanical stirring of 900 rpm and kept for 10 min. The reaction system was stirred for another 12 h upon the addition of trimesic acid (0.3 g, 1.428 mmol) and Cu(OAc)₂·H₂O (0.1 g, 0.5 mmol). The product was obtained by centrifugation and then washed with EtOH/H₂O (2:1, 3 × 20 ml) and EtOH (3 × 20 ml). Finally, the obtained product was dried at 60 °C under vacuum for 10 h and stored in a desiccator.

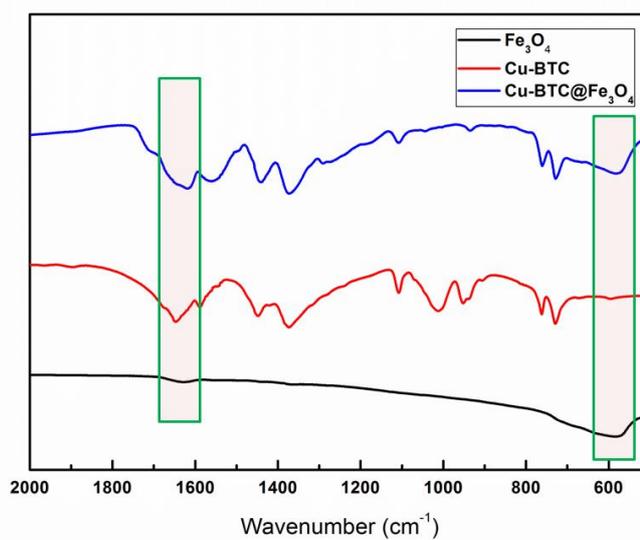
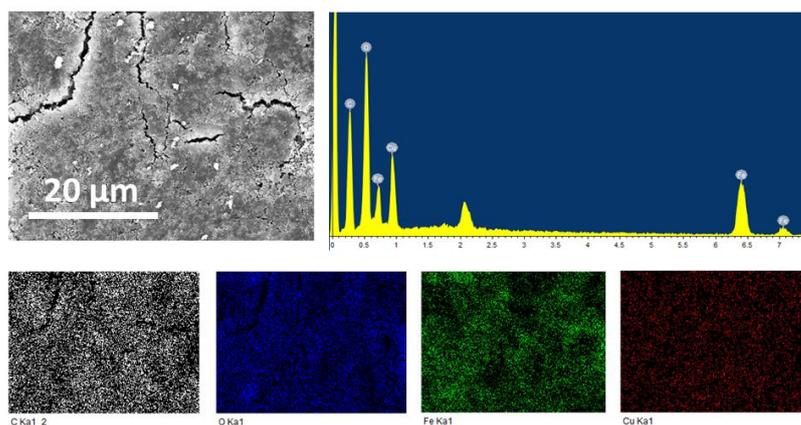


Figure S1. IR spectra of carboxyl functionalized Fe_3O_4 (black), Cu-BTC (red) and Cu-BTC@ Fe_3O_4 (blue).



Element	C	O	Fe	Cu
Weight Percentage (%)	39.51	35.42	12.89	12.18
Atomic Percentage (%)	55.51	37.36	3.90	3.24

Figure S2. The EDS element analysis of Cu-BTC@ Fe_3O_4 , the scale bar is 20 μm .

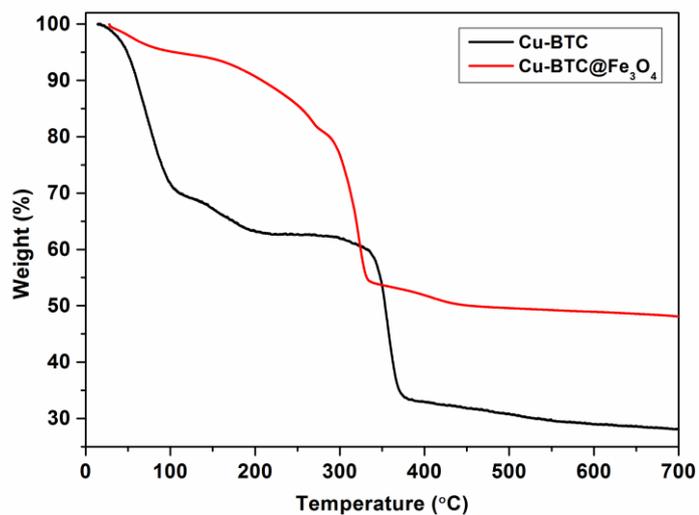


Figure S3. The TG curves of Cu-BTC and Cu-BTC@Fe₃O₄.

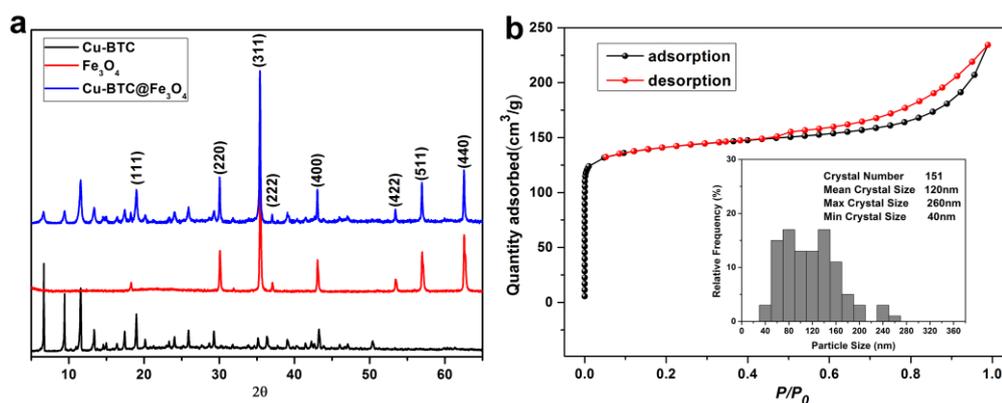


Figure S4. The PXRD patterns of Cu-BTC, Fe₃O₄ and Cu-BTC@Fe₃O₄ samples. (b) N₂ adsorption/desorption isotherms and particle size distribution of Cu-BTC@Fe₃O₄

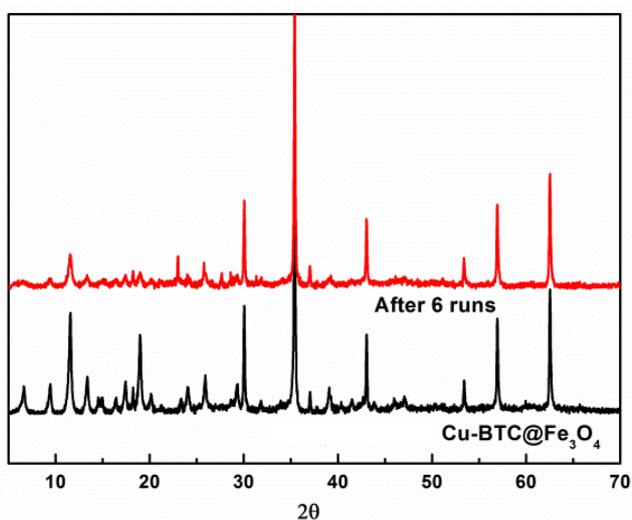
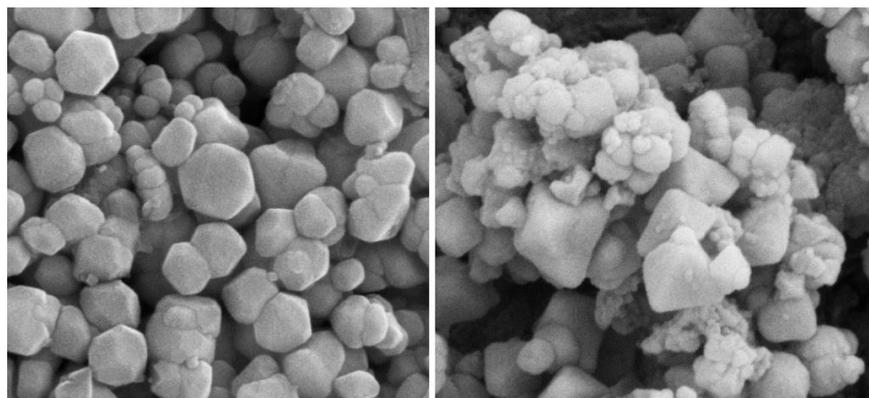


Figure S5. The PXRD patterns of Cu-BTC@Fe₃O₄ before and after catalysis reaction.



Fresh catalyst

Recovered catalyst after 6 runs

Figure S6. SEM of Cu-BTC@Fe₃O₄ before and after reaction

Experimental Procedures, Spectral and Analytical data

General procedure for the synthesis of 3a~l

A sealed tube equipped with a magnetic stirrer bar was charged with arenediazonium salt **1** (0.5 mmol), arylacetylene **2** (0.6 mmol), aqueous HCl (1 mmol, *conc.* 37%), Cu-BTC@Fe₃O₄ composite (15 mg) and acetonitrile (2 ml). The reaction mixture was stirred under oxygen atmosphere overnight. After reaction, the Cu-BTC@Fe₃O₄ was recovered with an external magnet, and washed with EtOAc and water. The reaction mixture was extracted with EtOAc (3 × 5 ml) and dried over Na₂SO₄. The organic phase was then concentrated under reduced pressure to give the crude product, which was further purified by column chromatography using petroleum ether/ethyl acetate (10:1, *v/v*) as eluent to afford product **3a~l**.

Characterization data of product

2-Chloro-2-(4-nitrophenyl)-1-phenylethan-1-one (3a).^{S2} Yellow solid (111 mg, 81%). ¹H NMR (300 MHz, CDCl₃) δ 8.21 (d, J = 8.8 Hz, 2H), 7.98 (d, J = 7.2 Hz, 2H), 7.67 (d, J = 8.7 Hz, 2H), 7.60 (t, J = 7.4 Hz, 1H), 7.47 (t, J = 7.6 Hz, 2H), 6.37 (s, 1H). ¹³C NMR (75 MHz, CDCl₃) δ 190.5, 148.0, 142.4, 134.3, 133.6, 129.6, 129.1, 128.9, 123.9, 59.2.

1-(4-Chlorophenyl)-2-chloro-2-(4-nitrophenyl)ethan-1-one (3b).^{S2} Yellow solid (113 mg, 73%). ¹H NMR (300 MHz, CDCl₃) δ 8.27 – 8.22 (m, 2H), 7.95 – 7.90 (m, 2H), 7.68 – 7.64 (m, 2H), 7.48 – 7.43 (m, 2H), 6.27 (s, 1H). ¹³C NMR (75 MHz, CDCl₃) δ 189.5, 148.1, 142.0, 141.0, 131.9, 130.5, 129.6, 129.4, 124.1, 59.1.

1-(4-Bromophenyl)-2-chloro-2-(4-nitrophenyl)ethan-1-one (3c).^{S2} Yellow solid (118 mg, 67%). ¹H NMR (300 MHz, CDCl₃) δ 8.23 (d, J = 8.8 Hz, 2H), 7.84 (d, J = 8.7 Hz, 2H), 7.67 – 7.60 (m, 4H), 6.27 (s, 1H). ¹³C NMR (75 MHz, CDCl₃) δ 189.7, 148.1, 142.0, 132.6, 132.3, 130.56, 129.7, 129.6, 124.0, 59.1.

4-(2-Chloro-2-(4-nitrophenyl)acetyl)benzonitrile (3d).^{S2} Yellow solid (112 mg, 75%). ¹H NMR (300 MHz, CDCl₃) δ 8.28 (d, J = 6.6 Hz, 2H), 8.08 (d, J = 6.4 Hz, 2H), 7.80 (d, J = 6.4 Hz, 2H), 7.67 (d, J = 6.5 Hz, 2H), 6.25 (s, 1H). ¹³C NMR (75 MHz, CDCl₃) δ 189.4, 148.4, 141.3, 136.7, 132.7, 129.6 (doublet), 124.2, 117.4 (doublet), 59.1.

Methyl 4-(2-chloro-2-(4-nitrophenyl)acetyl)benzoate (3e).^{S2} Yellow solid (113 mg, 68%). ¹H NMR (300 MHz, CDCl₃) δ 8.24 (d, J = 8.7 Hz, 2H), 8.13 (d, J = 8.4 Hz, 2H), 8.03 (d, J = 8.4 Hz, 2H), 7.67 (d, J = 8.7 Hz, 2H), 6.32 (s, 1H), 3.94 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) δ 190.2, 165.7, 148.2, 141.9, 137.0, 134.9, 130.1, 129.6, 129.1, 124.1, 59.3, 52.6.

2-Chloro-2-(4-nitrophenyl)-1-(m-tolyl)ethan-1-one (3f).^{S2} Yellow solid (94 mg, 65%). ¹H NMR (300 MHz, CDCl₃) δ 8.23 – 8.19 (m, 2H), 7.79 – 7.75 (m, 2H), 7.69 – 7.65 (m, 2H), 7.38 (dt, J = 15.1, 7.5 Hz, 2H), 6.36 (s, 1H), 2.40 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) δ 190.7, 148.0, 142.5, 139.0, 135.1, 133.7, 129.6, 129.5, 128.8, 126.2, 123.9, 59.2, 21.3.

2-Chloro-1-(2-chlorophenyl)-2-(4-nitrophenyl)ethan-1-one (3g).^{S2} Yellow solid (94 mg, 61%). ¹H NMR (300 MHz, CDCl₃) δ 8.20 (d, J = 8.8 Hz, 2H), 7.62 (d, J = 8.8 Hz, 2H), 7.44 – 7.39 (m, 2H), 7.38 – 7.27 (m, 2H), 6.28 (s, 1H). ¹³C NMR (75 MHz, CDCl₃) δ 193.8, 148.1, 141.4, 136.3, 132.7, 130.7, 130.5, 129.7, 129.6, 127.1, 123.9, 62.5.

4-(1-Chloro-2-oxo-2-phenylethyl)benzonitrile (3h).^{S2} Yellow solid (102 mg, 80%). ¹H NMR (300 MHz, CDCl₃) δ 8.00 – 7.95 (m, 2H), 7.70 – 7.65 (m, 2H), 7.62 – 7.57 (m, 3H), 7.50 – 7.45 (m, 2H), 6.31 (s, 1H). ¹³C NMR (75 MHz, CDCl₃) δ 190.6, 140.6, 134.2, 133.7, 132.6, 129.3, 129.1, 128.9, 118.1, 112.9, 59.7.

Ethyl 4-(1-chloro-2-oxo-2-phenylethyl)benzoate (3i).^{S2} White solid (99 mg, 66%). ¹H NMR (300 MHz, CDCl₃) δ 8.06 – 8.01 (m, 2H), 7.97 – 7.93 (m, 2H), 7.58 – 7.52 (m, 3H), 7.45 – 7.39 (m, 2H), 6.34 (s, 1H), 4.35 (q, J = 7.1 Hz, 2H), 1.36 (t, J = 7.1 Hz, 3H). ¹³C NMR (75 MHz, CDCl₃) δ 191.0, 165.8, 140.3, 134.2, 133.6, 131.0, 130.4, 129.9, 129.3, 128.8, 61.1, 60.9, 14.2.

2-Chloro-2-(4-chlorophenyl)-1-phenylethan-1-one (3j).^{S2} White solid (67 mg, 53%). ¹H NMR (300 MHz, CDCl₃) δ 7.98 – 7.92 (m, 2H), 7.61 – 7.54 (m, 1H), 7.48 – 7.40 (m, 4H), 7.38 – 7.32 (m, 2H), 6.28 (s, 1H). ¹³C NMR (75 MHz, CDCl₃) δ 191.1, 135.2, 134.3, 134.0, 133.9, 129.8, 129.3, 129.1, 128.8, 60.8.

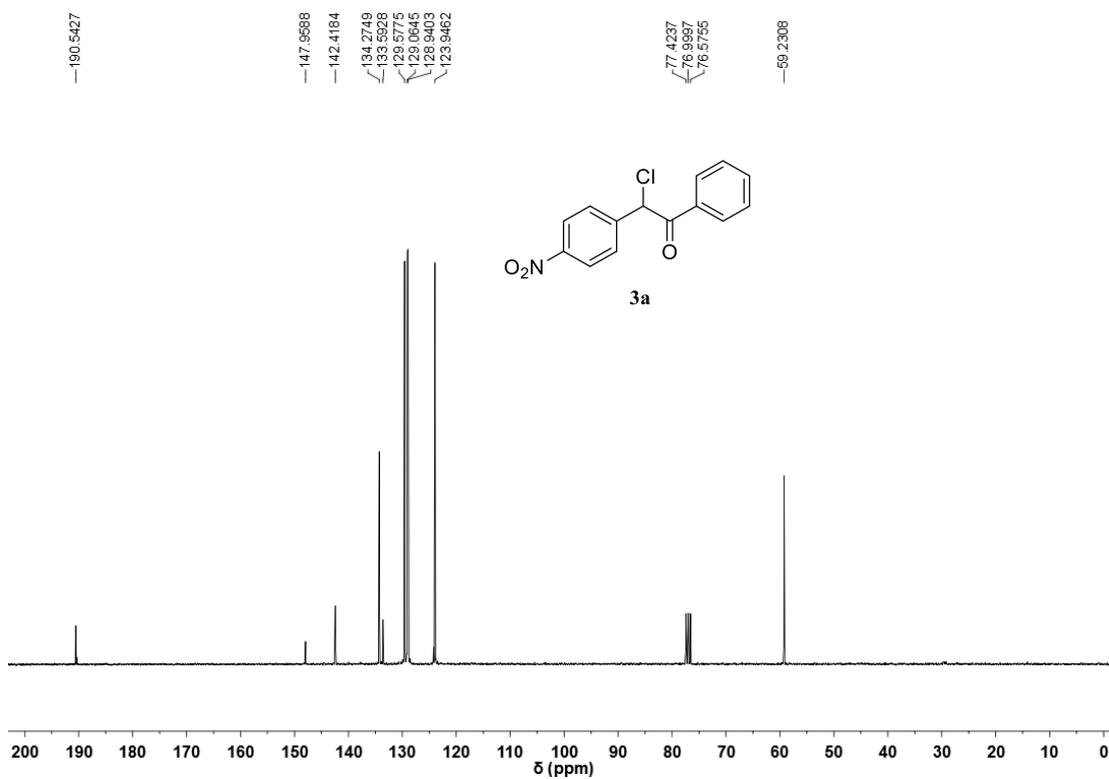
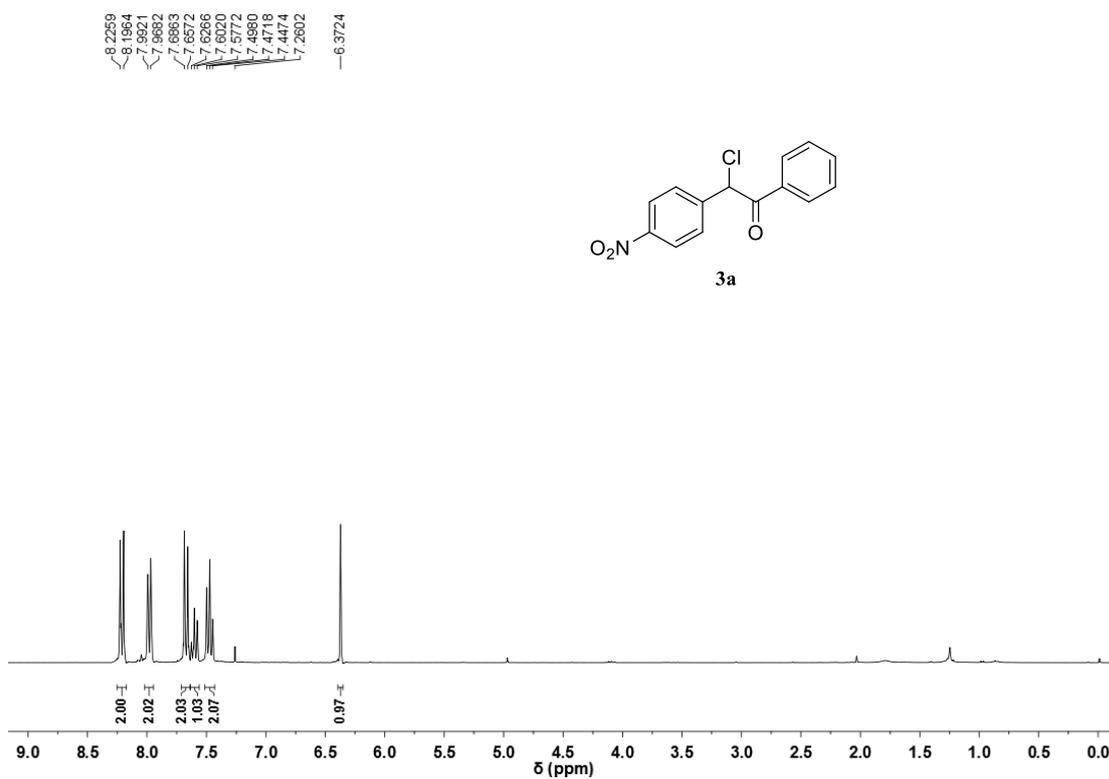
2-(4-Acetylphenyl)-2-chloro-1-phenylethan-1-one (**3k**).^{S2} Yellow solid (82 mg, 60%). ¹H NMR (300 MHz, CDCl₃) δ 7.97 – 7.95 (m, 2H), 7.94 – 7.91 (m, 2H), 7.61 – 7.52 (m, 3H), 7.48 – 7.38 (m, 2H), 6.35 (s, 1H), 2.57 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) δ 197.3, 190.9, 140.5, 137.3, 134.0, 133.9, 129.0, 128.9, 128.8, 128.7, 60.8, 26.6.

2-Chloro-2-(3-nitrophenyl)-1-phenylethan-1-one (**3l**).^{S2} Yellow solid (99 mg, 72%). ¹H NMR (300 MHz, CDCl₃) δ 8.37 (t, J = 2.0 Hz, 1H), 8.23 – 8.19 (m, 1H), 8.02 – 7.98 (m, 2H), 7.87 – 7.81 (m, 1H), 7.65 – 7.56 (m, 2H), 7.52 – 7.46 (m, 2H), 6.36 (s, 1H). ¹³C NMR (75 MHz, CDCl₃) δ 190.6, 148.3, 137.6, 134.7, 134.3, 133.6, 129.9, 129.1, 129.0, 124.0, 123.7, 58.8.

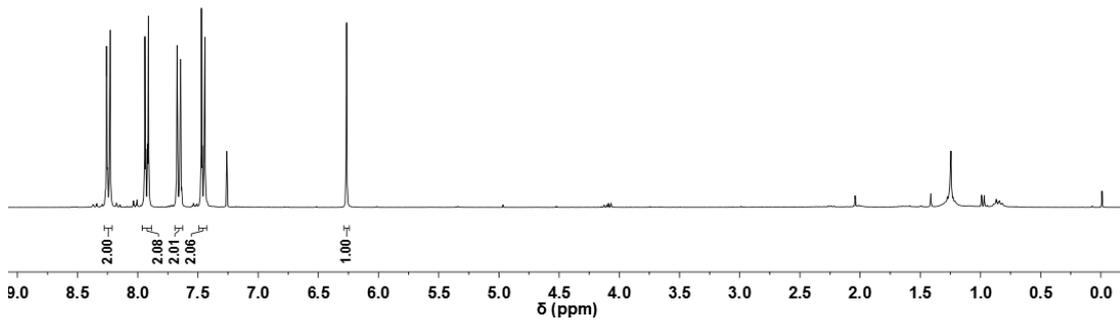
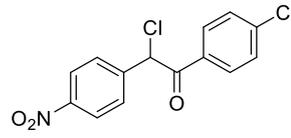
References

- S1. Y. F. Chen, X. Q. Huang, S. H. Zhang, S. Q. Li, S. J. Cao, X. K. Pei, J. W. Zhou, X. Feng, B. Wang, *J. Am. Chem. Soc.*, 2016, **138**, 10810.
- S2. T. Niu, D. Jiang, S. Li, B. Ni, L. Wang, *Chem. Commun.*, 2016, **52**, 13105.

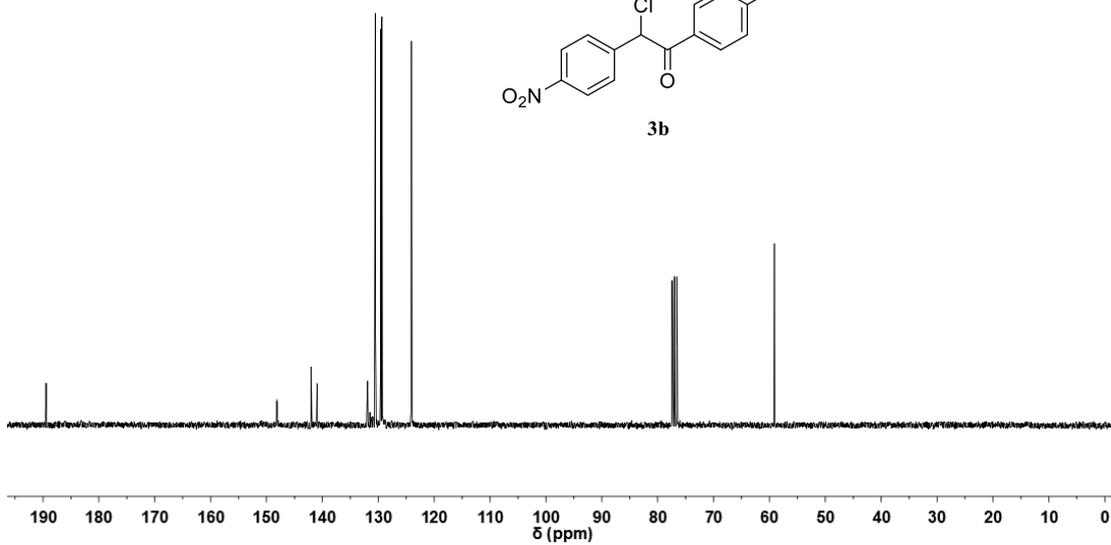
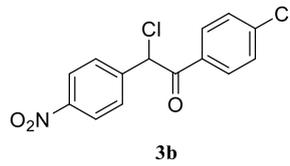
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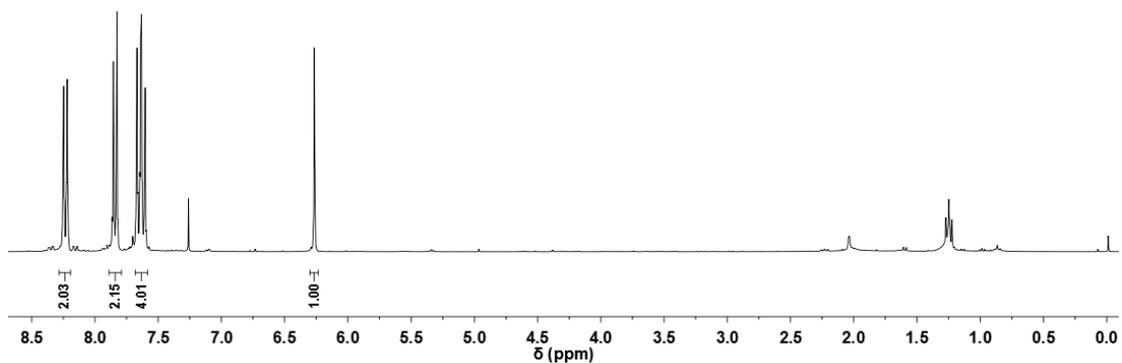
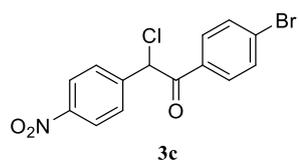


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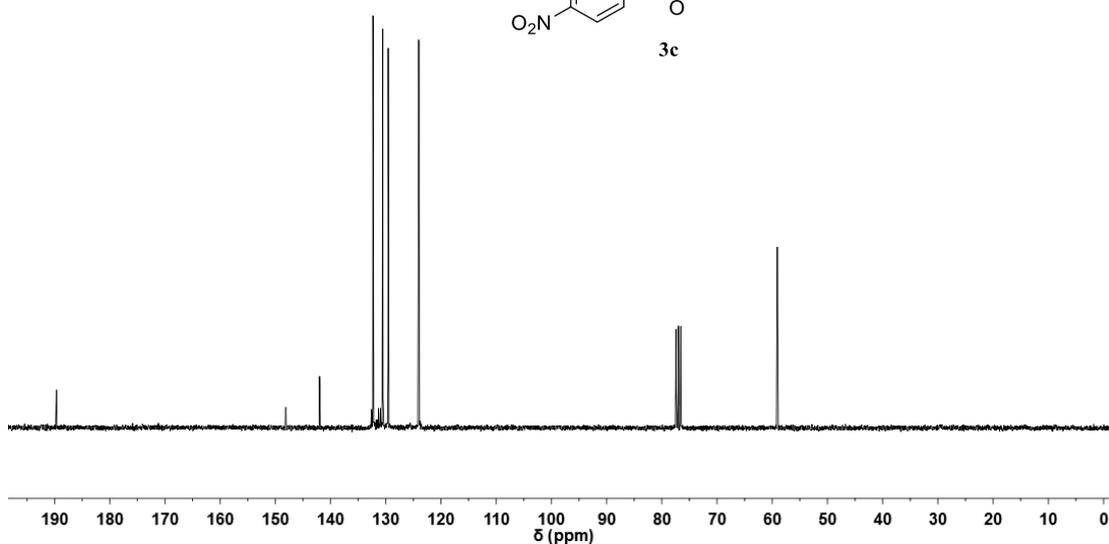
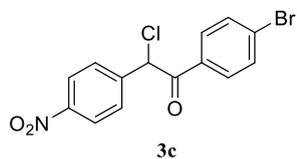
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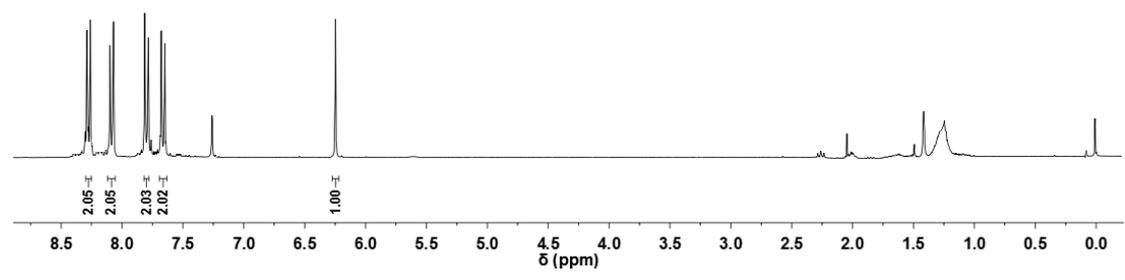
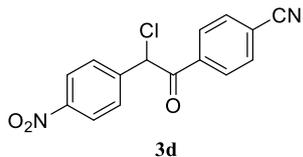
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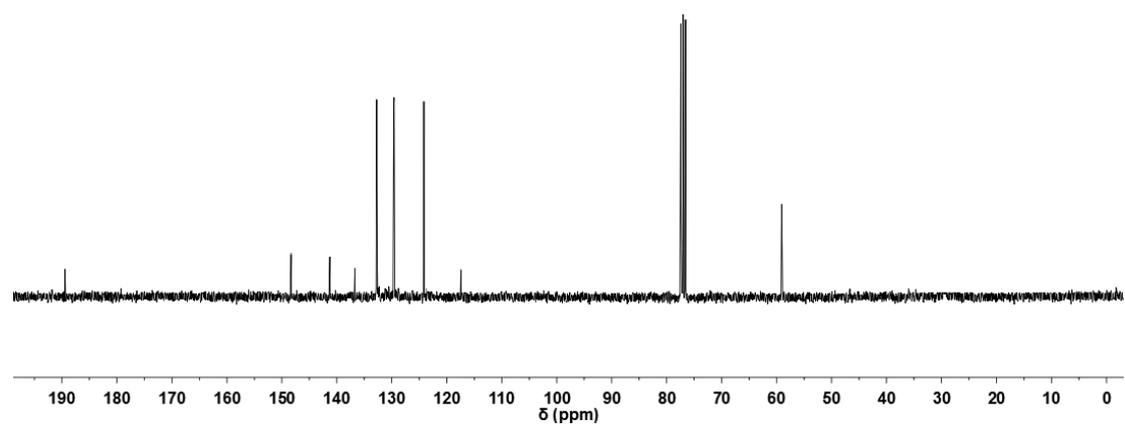
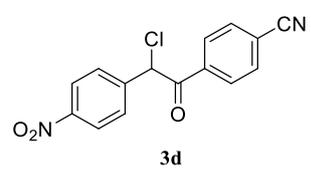
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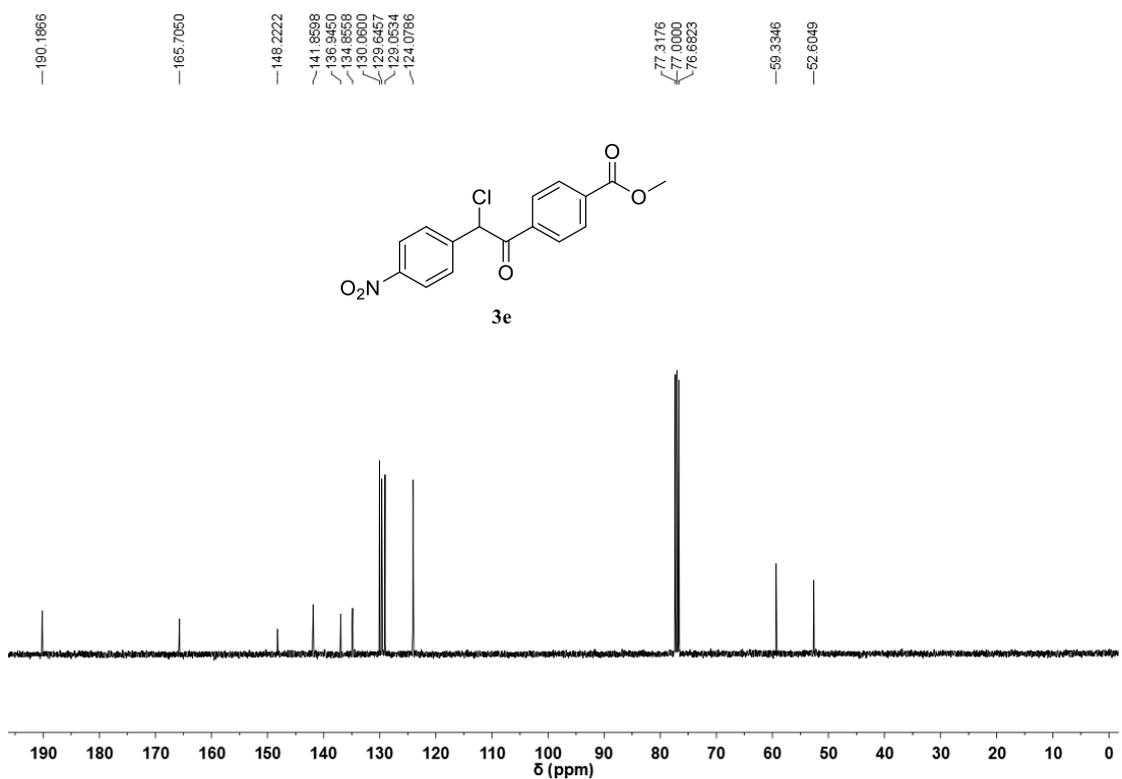
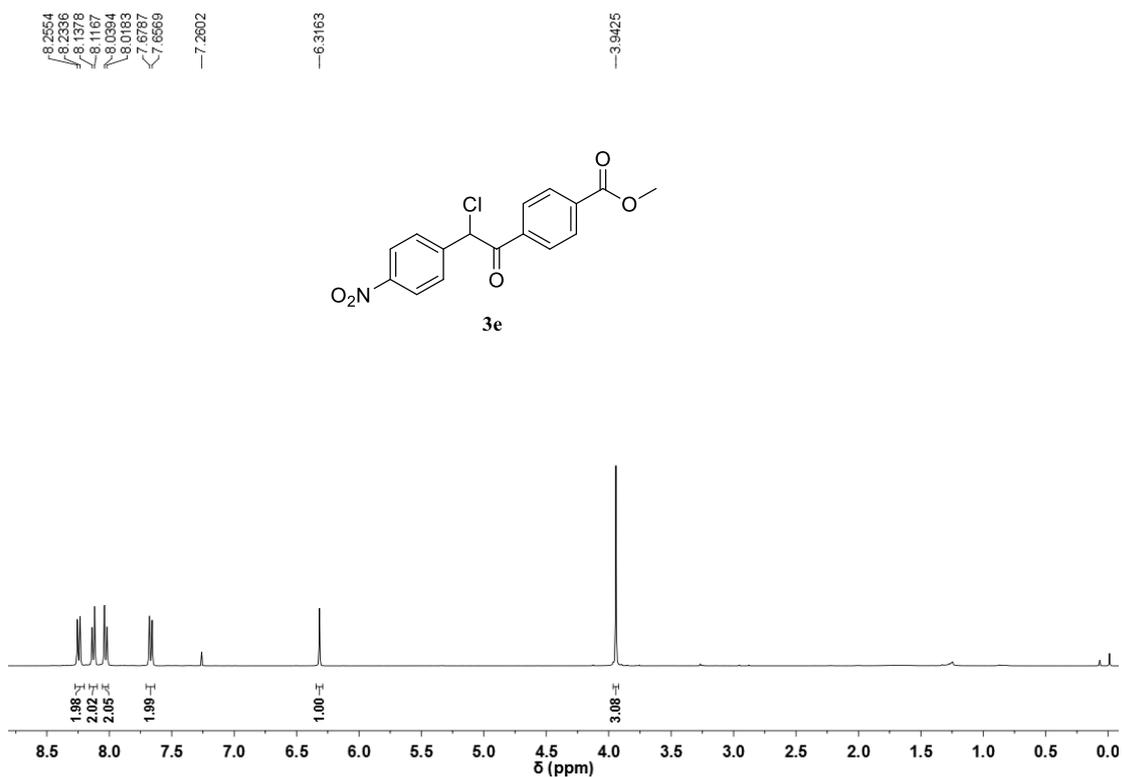
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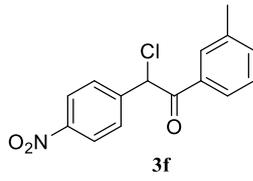
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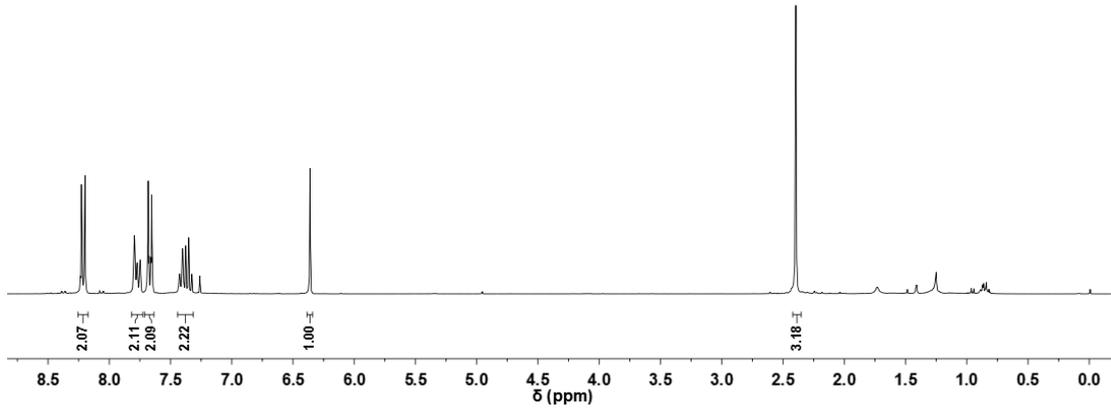




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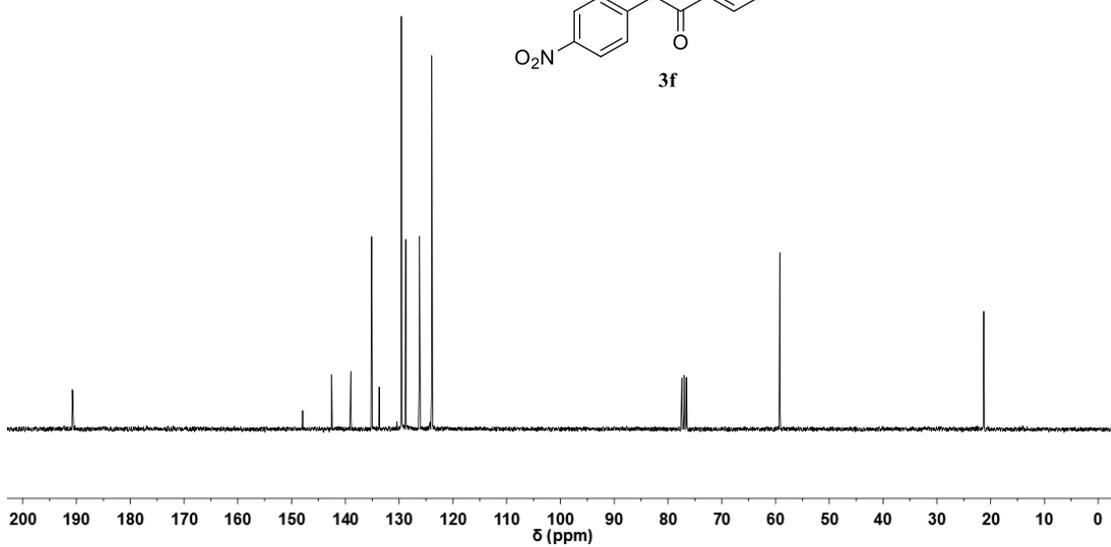
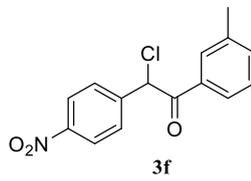
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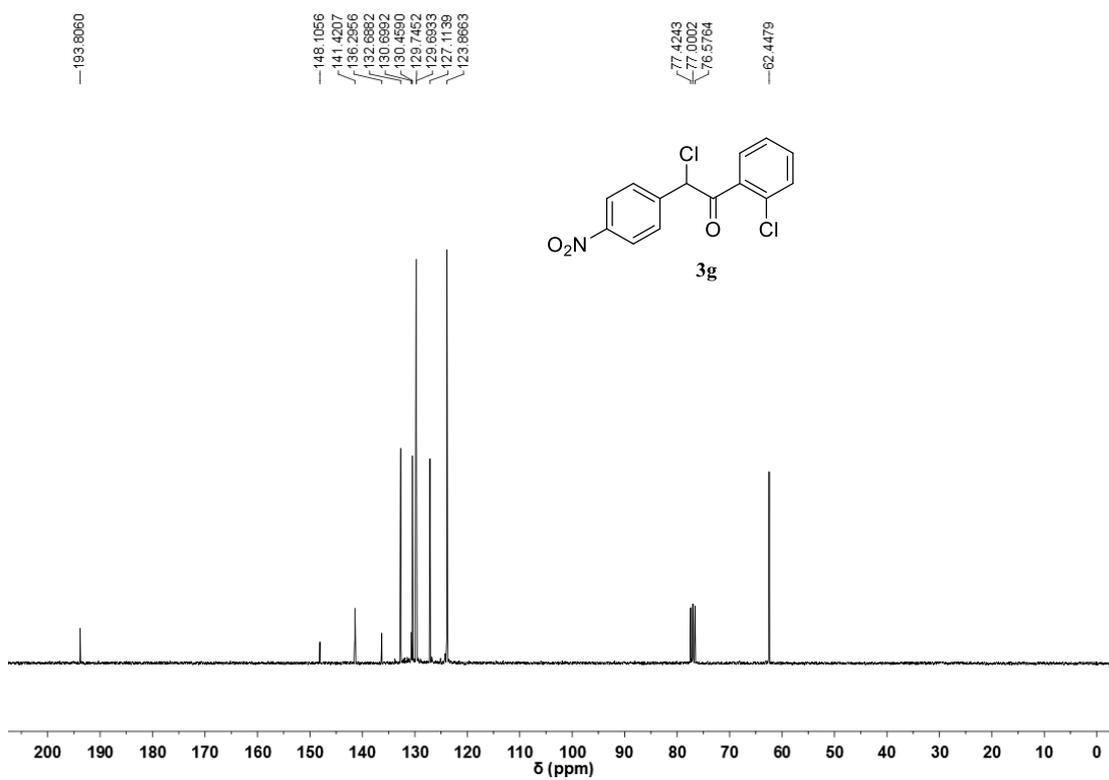
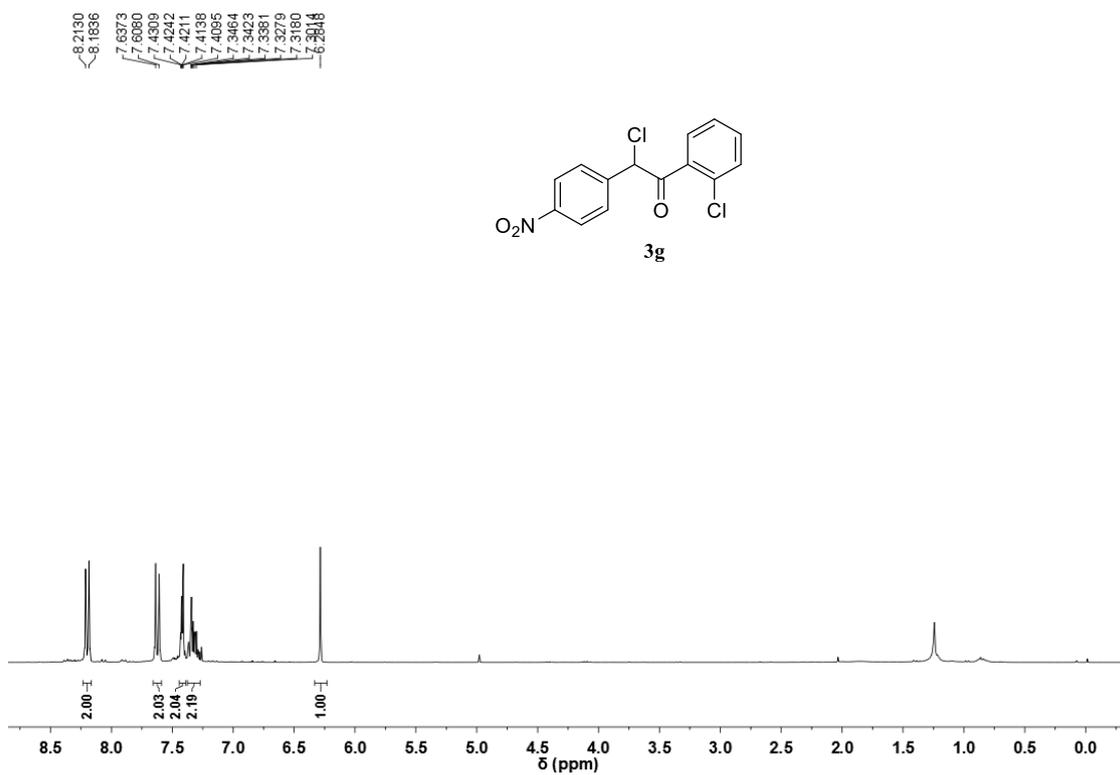
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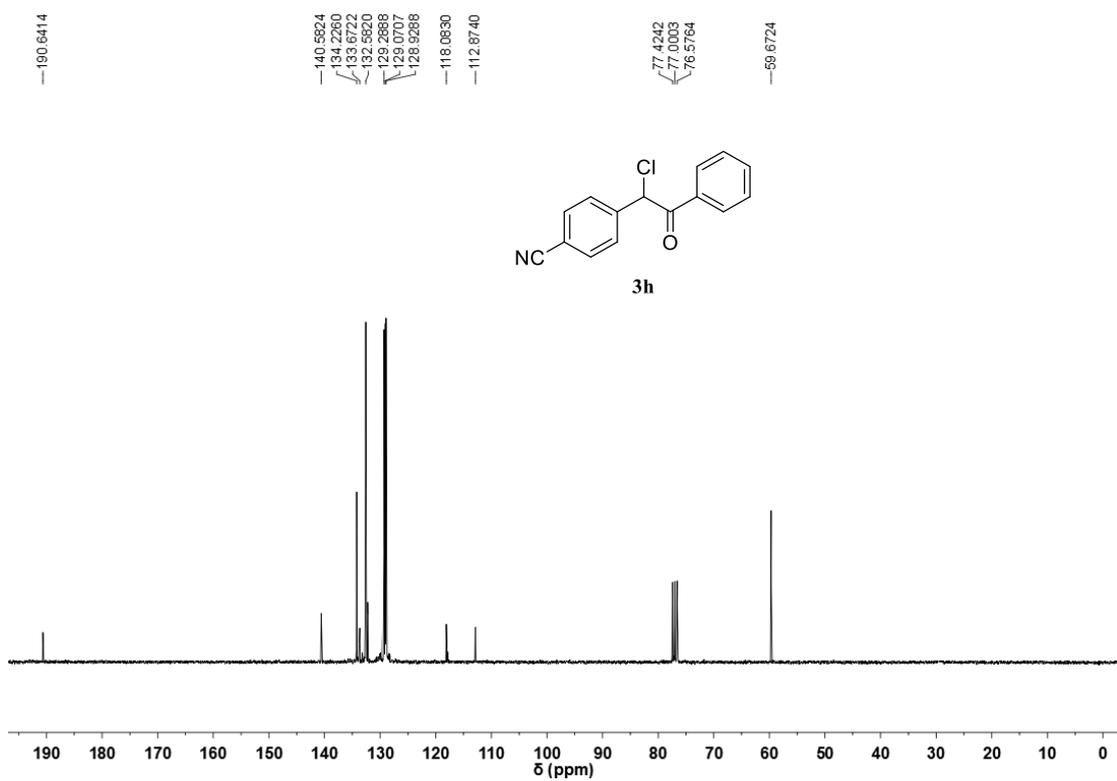
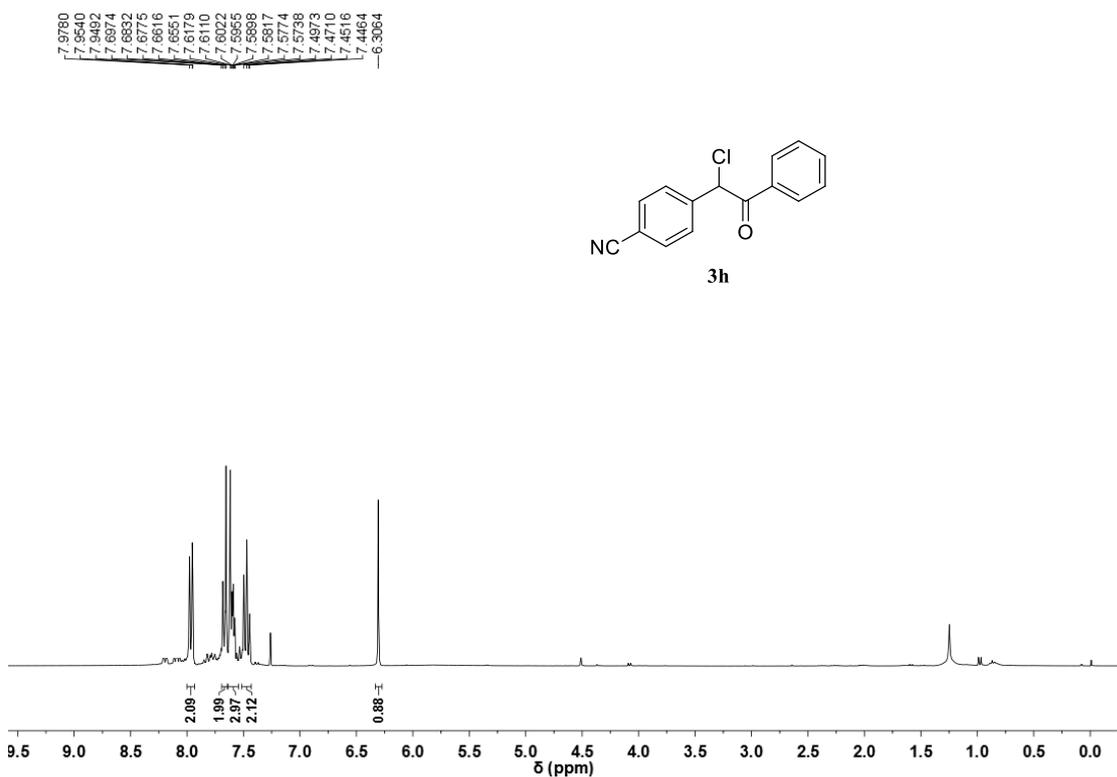
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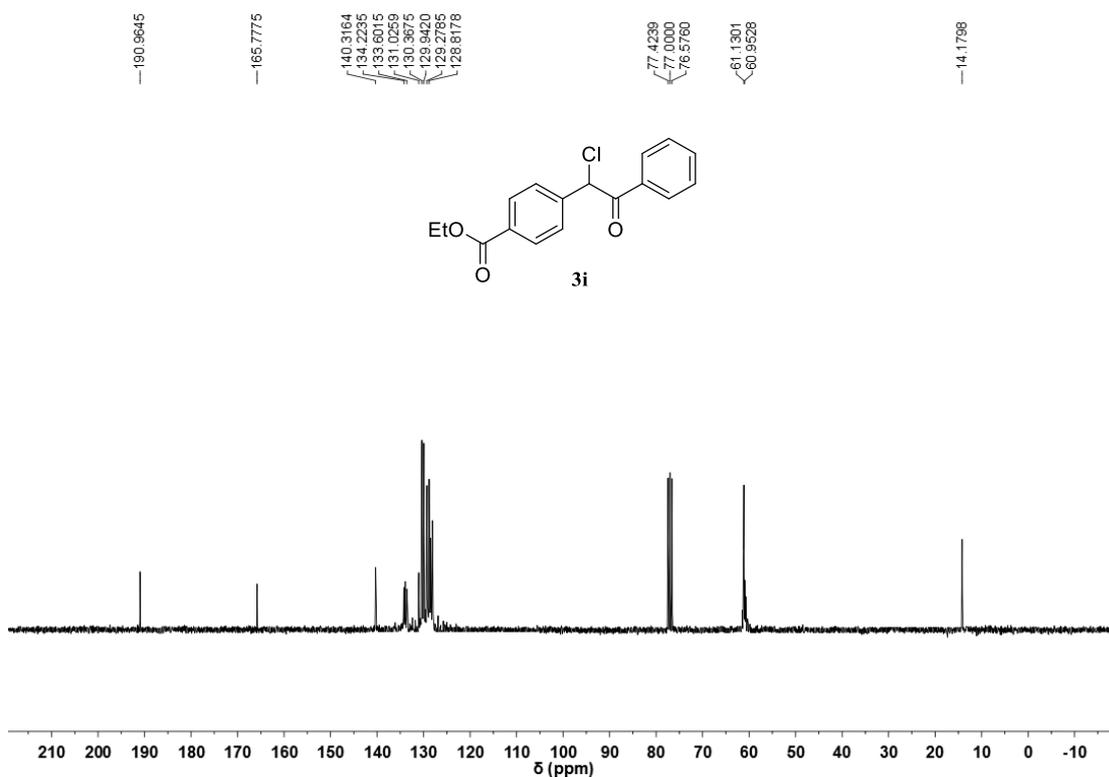
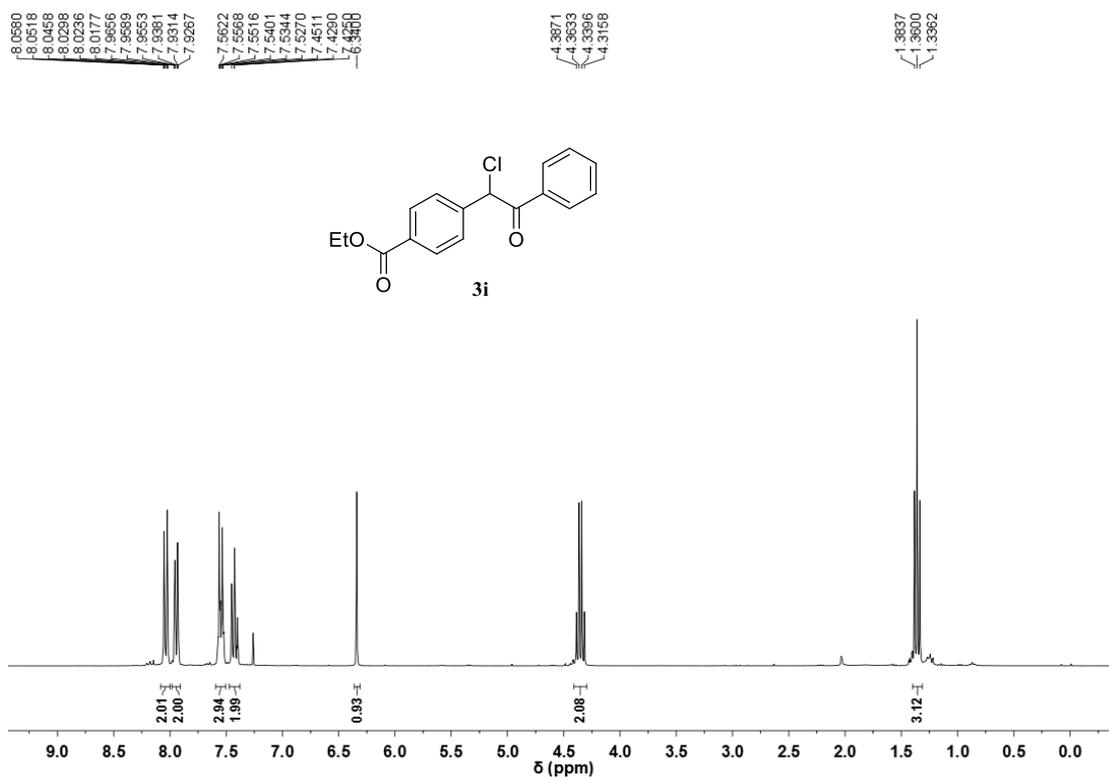
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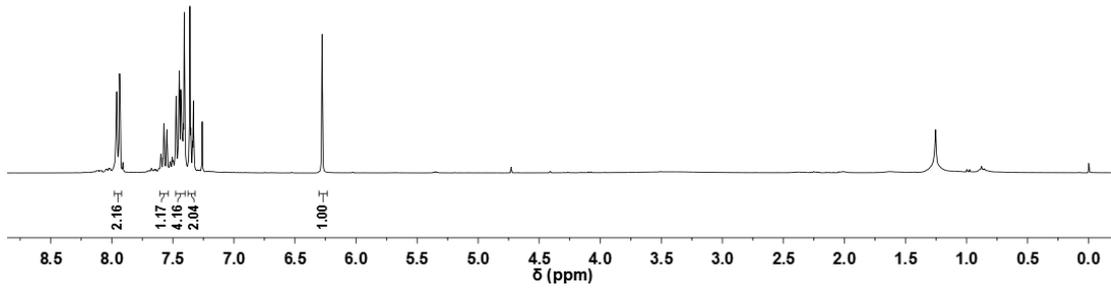
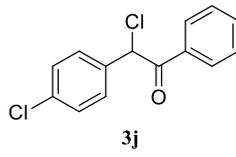








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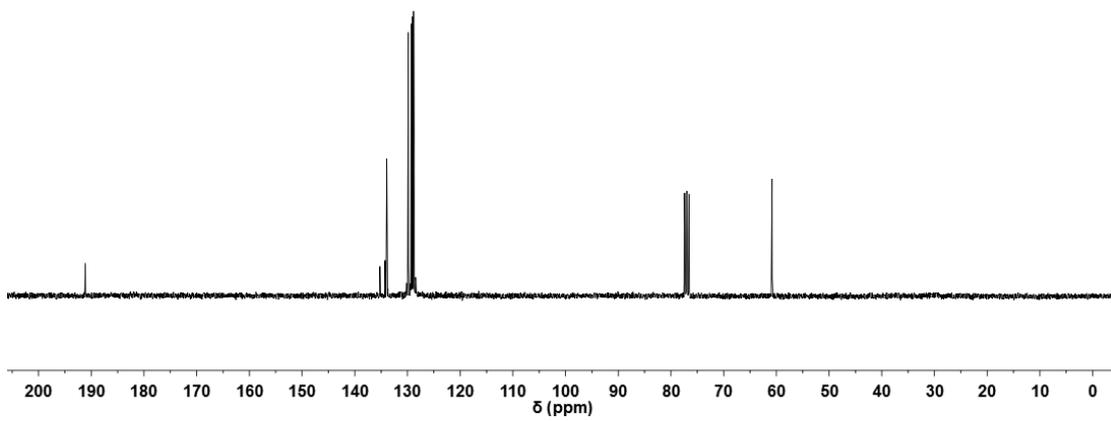
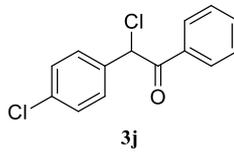


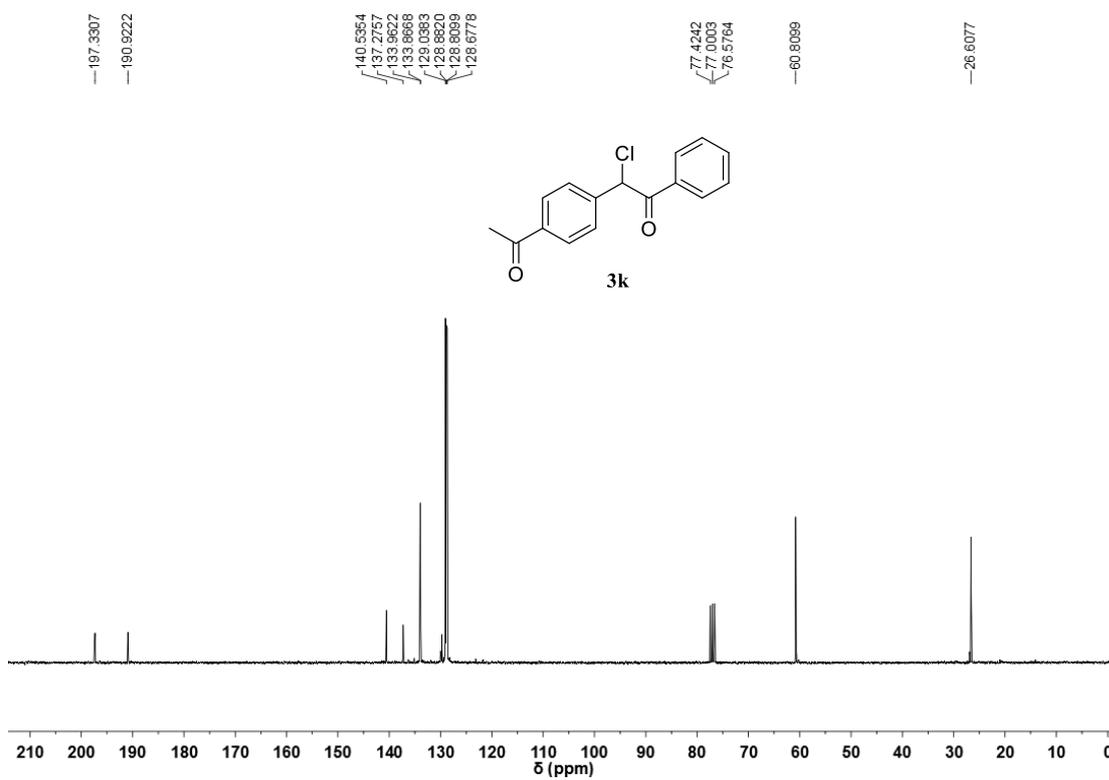
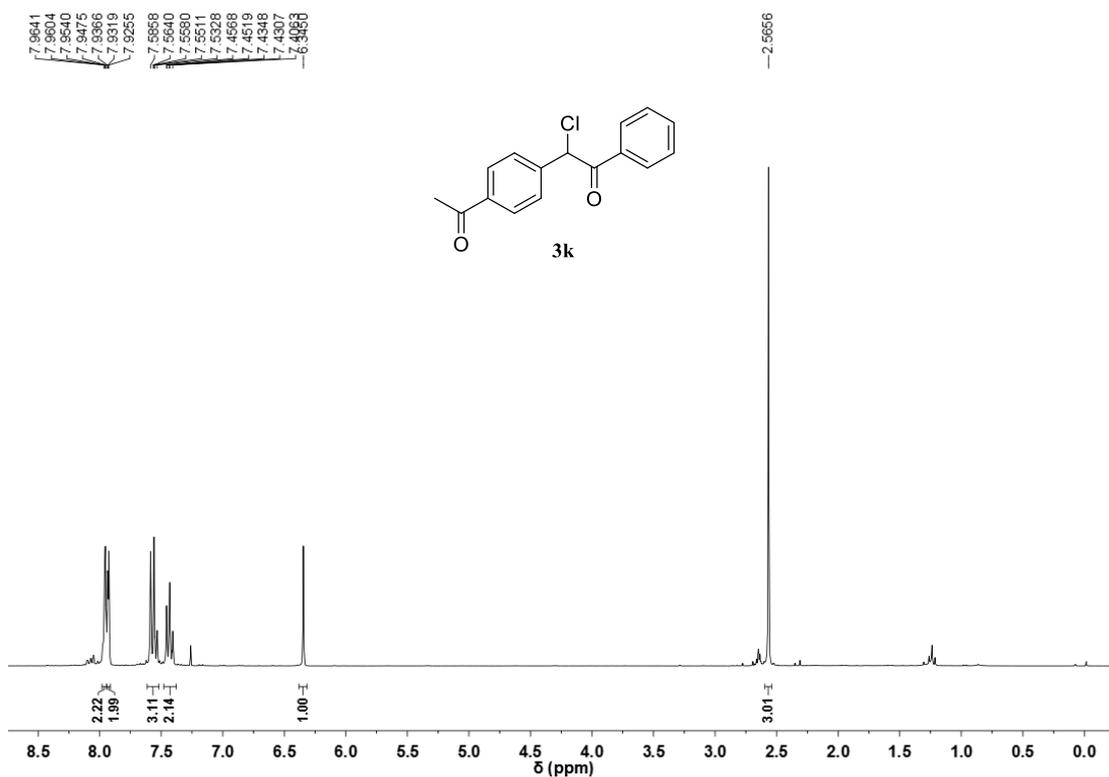
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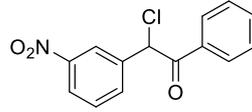
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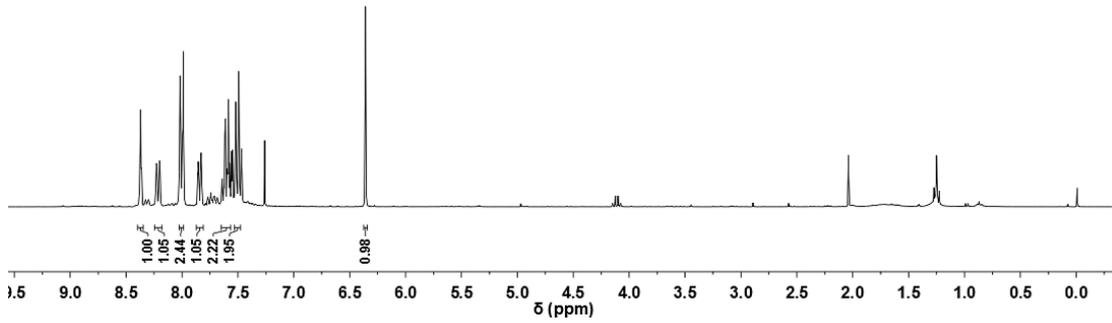




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31



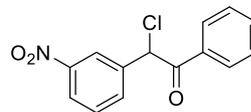
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31

