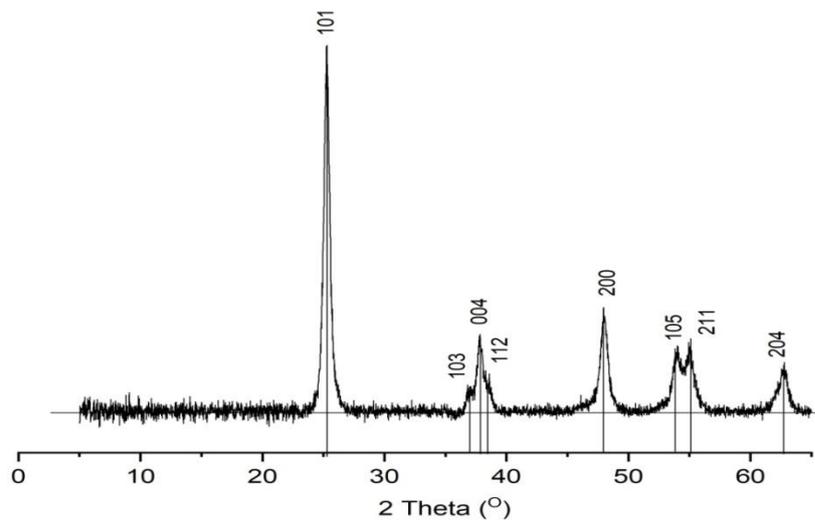


**Effects of Sn<sup>4+</sup> dopant ions located either in the bulk or at crystallite surfaces on the ultraviolet photocatalytic activity of anatase**

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Routine X-ray diffraction (XRD) measurements were performed on a powder sample ARL X'TRA Thermo Scientific diffractometer using Cu  $K_{\alpha}$  radiation (wavelength  $\lambda = 1.5418 \text{ \AA}$ ). All studied catalysts were found to be anatase type (space group  $I4_1/amd$ ) single-phase polycrystalline materials. An XRD pattern of a hydrogen-annealed sample containing 0.6 at% Sn<sup>4+</sup> on surface sites is shown in Figure S1. Its specific surface area  $S_{\text{BET}}$  determined by the BET method was found to be nearly  $56 \text{ m}^2\text{g}^{-1}$  and virtually not affected by the presence of the dopant used. Both <sup>119</sup>Sn and <sup>121</sup>Sb Mössbauer spectra were recorded on a MS-1104 spectrometer and analyzed by a least-square fitting program. To perform <sup>119</sup>Sn Mössbauer measurements a Ca<sup>119m</sup>SnO<sub>3</sub> ( $E_{\gamma} = 23.88 \text{ keV}$ ) source along with a thin NaI(Tl) scintillator was used. To perform <sup>121</sup>Sb Mössbauer measurements a Ca<sup>121\*</sup>SnO<sub>3</sub> source ( $E_{\gamma} = 37.15 \text{ keV}$ ), producing in the same NaI(Tl) scintillator the 8.5 keV escape peak was used. During the measurements both Ca<sup>121\*</sup>SnO<sub>3</sub> source and studied powder sample (absorber) were introduced into the hole of a copper bar immersed in a Dewar flask filled with liquid nitrogen. Under these conditions, the temperature of both the source and absorber was close to 100 K. To perform *in situ* <sup>119</sup>Sn Mössbauer measurements with hydrogen-annealed powders a cylindrical quartz reactor equipped with a lateral thin-window sample cell was used. During the measurements both the sample cell filled with studied powder and the source were again introduced in the copper bar immersed in liquid nitrogen.



**Figure S1** XRD pattern of hydrogen-annealed 0.6 Sn at% anatase TiO<sub>2</sub>.