

# Ring switching tricomponent synthesis of pyrano[2,3-*b*]pyridine multifunctional derivatives

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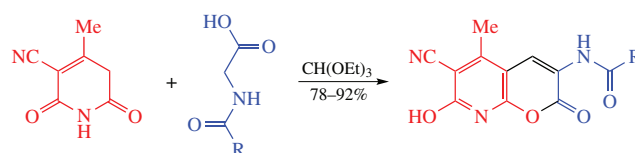
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**New 3-acylamino-7-hydroxy-5-methyl-2-oxo-2*H*-pyrano[2,3-*b*]pyridine-6-carbonitriles were synthesized via a tricomponent condensation of 3-cyano-4-methyl-1,2,5,6-tetrahydropyridine-2,6-dione, triethyl orthoformate and *N*-acylglycines under the Erlenmeyer–Plöchl reaction conditions. According to X-ray data, in the solid phase they exist as hydroxypyridine tautomer form.**



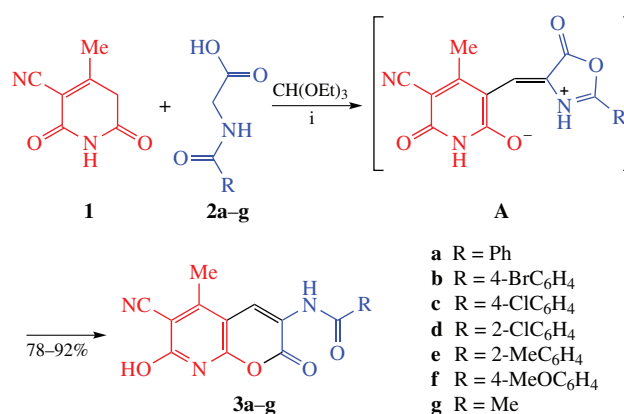
**Keywords:** 2*H*-pyrano[2,3-*b*]pyridinones, 1,2,5,6-tetrahydropyridine-2,6-dione, hippuric acid, triethyl orthoformate, tricomponent condensation, ring switching, Erlenmeyer–Plöchl reaction.

The 8-aza- and 7-hydroxycoumarin (umbelliferone) derivatives are regarded effective ‘blue fluorescence’ agents. Even more interesting and promising are coumarin derivatives simultaneously containing both 8-aza- and 7-hydroxy-units in one molecule.<sup>1–3</sup> Previously, the classical syntheses of *N*-aroyl-3-amino-2*H*-chromen-2-ones and their pyrano-fused hetarene analogues involving sequential condensations of phenols, naphthols or hydroxy-substituted heterocycles with dimethylformamide dimethyl acetal and hippuric acid esters were performed;<sup>4–7</sup> the final products did not depend on the sequence of condensations between the starting reactants. The synthesis of such heterocyclic systems in three-component condensation reactions has been described for phenols, naphthols, or hydroxy-substituted heterocycles but not for azaheterocycles containing amido- or imido-moieties.<sup>8–11</sup>

Herein, we have proposed a convenient synthesis of substituted 8-aza-7-hydroxycoumarins based on the abovementioned general synthetic approach. Continuing the study of the Erlenmeyer–Pöchl reaction<sup>12</sup> and multicomponent condensations with 3-cyano-4-methylpyrimidine-2,6-dione and orthoformates,<sup>13</sup> we have performed the three-component assembling of 3-cyano-4-methylpyridine-2,6-dione **1** with triethyl orthoformate and a number of *N*-acylglycines **2a–g** under the Erlenmeyer–Plöchl conditions (Scheme 1). When the starting reagents were heated in acetic anhydride, brightly colored, red-orange betain-like intermediates **A** precipitated. Upon further heating, the colour of the precipitate quickly disappeared, and 3-acylamino-7-hydroxy-5-methyl-2-oxo-2*H*-pyrano[2,3-*b*]pyridine-6-carbonitriles **3a–g** were formed.<sup>†</sup>

Unfortunately, confirmation of the intermediate structure **A** using NMR spectroscopy turned out unsuccessful. The red intermediate underwent rapid recyclization into white product **3** when dissolved in DMSO, DMF or pyridine, even on cooling. Testing other solvents looked not valuable for NMR studies.

Attempts to carry out the reaction in acetic acid or in DMF did not bring about any products, and only the starting reactants were recovered. An attempt to introduce *N*-phthaloylglycine instead of hippuric acid was also unsuccessful. This fact may be an indirect evidence in favor of the intermediate structure **A** formed at the first stage of the Erlenmeyer–Pöchl process.<sup>14</sup>



**Scheme 1** Reagents and conditions: i, Ac<sub>2</sub>O, 120 °C, 1 h.

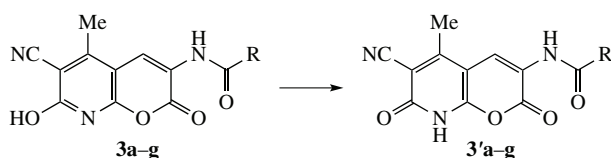
<sup>†</sup> **General procedure.** A mixture of 3-cyano-4-methyl-1,2,5,6-tetrahydropyridine-2,6-dione **1** (300 mg, 2 mmol), triethyl orthoformate (385 mg, 2.6 mmol) and hippuric acid **2a** (360 mg, 2 mmol) was heated in Ac<sub>2</sub>O (1.5 ml) for 1 h. The red-coloured intermediate appears and quickly turns into a white precipitate. The final product was filtered off and washed

with ethyl alcohol to give 3-benzoylamino-7-hydroxy-5-methyl-2-oxo-2*H*-pyrano[2,3-*b*]pyridine-6-carbonitrile **3a**. Yield 86% (550 mg). White solid, mp > 300 °C.

Compounds **3b–g** were prepared similarly. For their characteristics, see Online Supplementary Materials.

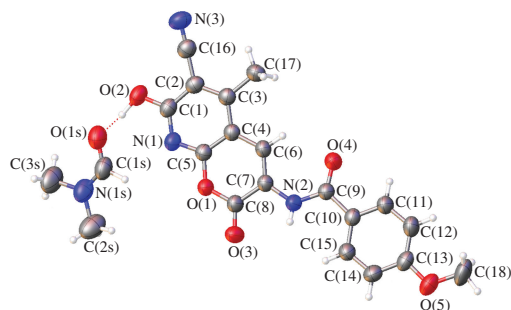
Subsequently, intermediate **A** underwent a ring switching transformation<sup>15</sup> similar to the Zimmer's recycling<sup>16,17</sup> to afford final products **3a–g**.

The <sup>1</sup>H NMR spectra of compounds **3** showed a singlet for the protons of the methyl group of the pyridinedione ring, signals for the aromatic protons (compounds **3a–f**) or the acetyl protons (compound **3g**). Two singlets and one broad singlet with an integral intensity of one proton each also appear for compounds **3a–g** in the low field area. Although the <sup>1</sup>H NMR spectra (DMSO-*d*<sub>6</sub>) contained only one set of signals, it was difficult to make a right choice between 2-hydroxypyridine **3a–g** and pyridin-2-ones **3'a–g** tautomeric forms.



The product structure was ultimately elucidated from X-ray diffraction study<sup>‡</sup> for compound **3f**. Single crystals suitable for such purpose were isolated with DMF as a lattice solvent in a ratio of 1 : 1 (Figure 1), assisted by a rather strong hydrogen bond O–H...O [O...O 2.528(2) Å, OHO 167.42(12)°] with the hydroxy group of the product **3f** being a 7-hydroxy tautomer. The molecule is nearly planar with an angle between mean-square planes of the pyranopyridine and phenyl moieties joined by an amide group equal to 7.19(6)°.

As a result, two types of intermolecular stacking interactions are produced, one between two symmetry-equivalent pyridine rings and one between the pyridine ring and the phenyl group; the corresponding centroid–centroid distances, angles between the planes and their shifts are 3.7685(16) and 3.7440(13) Å,



**Figure 1** General view of compound **3f** in representation of non-hydrogen atoms *via* thermal ellipsoids with a probability level of 50%.

<sup>‡</sup> Crystal data for **3f**. C<sub>21</sub>H<sub>20</sub>N<sub>4</sub>O<sub>6</sub> (*M* = 424.41), triclinic, space group *P*1̄, at 296 K: *a* = 8.5205(4), *b* = 10.9920(5) and *c* = 11.7307(5) Å, *α* = 104.150(2), *β* = 98.285(3) and *γ* = 99.863(3)°, *V* = 1029.51(8) Å<sup>3</sup>, *Z* = 2, *d*<sub>calc</sub> = 1.369 g cm<sup>−3</sup>, *μ*(MoK $\alpha$ ) = 1.02 cm<sup>−1</sup>, *F*(000) = 444. Intensities of 17013 reflections were measured with a Bruker D8 Quest CCD diffractometer [*λ*(MoK $\alpha$ ) = 0.71073 Å,  $\omega$ -scans, 2 $\theta$  < 52°], and 4059 independent reflections (*R*<sub>int</sub> = 0.0572) were used in the further refinement. Using Olex2,<sup>18</sup> the structure was solved with the ShelXT<sup>19</sup> structure solution program using Intrinsic Phasing and refined with the XL<sup>20</sup> refinement package using Least Squares minimization. Hydrogen atoms of NH and OH groups were located from difference Fourier synthesis, positions of others were calculated, and they all were refined in the isotropic approximation using the riding model. The refinement converged to *wR*<sub>2</sub> = 0.1553 and GOF = 1.009 for all the independent reflections [*R*<sub>1</sub> = 0.0501 was calculated against *F* for 2582 observed reflections with *I* > 2 $\sigma$ (*I*)].

CCDC 2054855 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre *via* <http://www.ccdc.cam.ac.uk>.

0.0(2) and 7.19(6)°, 1.660(2) and 1.407(2) Å, respectively. The amino group is involved in a rather weak N–H...O bond [N...O 3.312(2) Å, NHO 134.58(11)°] with the oxo group at the pyranopyridine moiety, thus producing a centrosymmetric dimer from the neighbouring molecules of **3f**. Other weak interactions, such as C–H...O, C–H...N and C–H... $\pi$ , complete the formation of the 3D framework.

In summary, the one-pot three-component procedure for the synthesis of new 3-acylamino-7-hydroxy-5-methyl-2-oxo-2H-pyran[2,3-*b*]pyridine-6-carbonitriles as new 8-aza-7-hydroxy-coumanine derivatives was accomplished with high yields under the Erlenmeyer–Plöchl reaction conditions. The further study of this procedure for similar condensation with other ortho esters, methylene active compounds, and hippuric acids is in progress.

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#### Online Supplementary Materials

Supplementary data associated with this article can be found in the online version at doi: 10.1016/j.mencom.2021.05.041.

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