

Rapid metal free construction of 3-positioned 2-pyridyl substituent in indoles

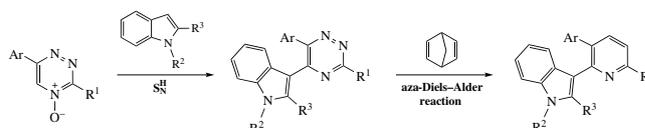
Maria I. Savchuk,^a Igor S. Kovalev,^a Vladimir L. Rusinov,^{a,b} Dmitry S. Kopchuk,^{*a,b} Alexey P. Krinochkin,^{a,b} Grigory V. Zyryanov,^{a,b} Oleg N. Chupakhin^{a,b} and Valery N. Charushin^{a,b}

^a Institute of Chemical Engineering, Ural Federal University, 620002 Ekaterinburg, Russian Federation

^b I. Ya. Postovsky Institute of Organic Synthesis, Ural Branch of the Russian Academy of Sciences, 620990 Ekaterinburg, Russian Federation. Fax: +7 343 374 1189; e-mail: dkopchuk@mail.ru

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New access to substituted 2-(indol-3-yl)pyridines involves two stage protocol, namely, a reaction of deoxygenative nucleophilic substitution of hydrogen in 1,2,4-triazine-4-oxides under the action of indoles followed by aza-Diels–Alder reaction of thus obtained 5-indolyl-1,2,4-triazines with 2,5-norbornadiene in a pressure vessel. The reactions sequence provides good yields and is suitable for wide scope of substituted 1,2,4-triazines.



Keywords: 2-(indol-3-yl)pyridines, nucleophilic substitution of hydrogen, aza-Diels–Alder reaction, 1,2,4-triazines, pyridines.

Indole fragment as a typical pharmacophore is met in a number of natural and synthetic compounds possessing cytotoxic,¹ antiviral, antimicrobial² and some other³ activities. On the other hand, many bioactive compounds are represented in the pyridine series.⁴ Compounds including fragments of both heterocycles, for example 2-(indol-3-yl)pyridines, are also bioactive,^{5–7} in particular, exhibit anti-inflammatory, anti-infection and anti-androgenic properties.

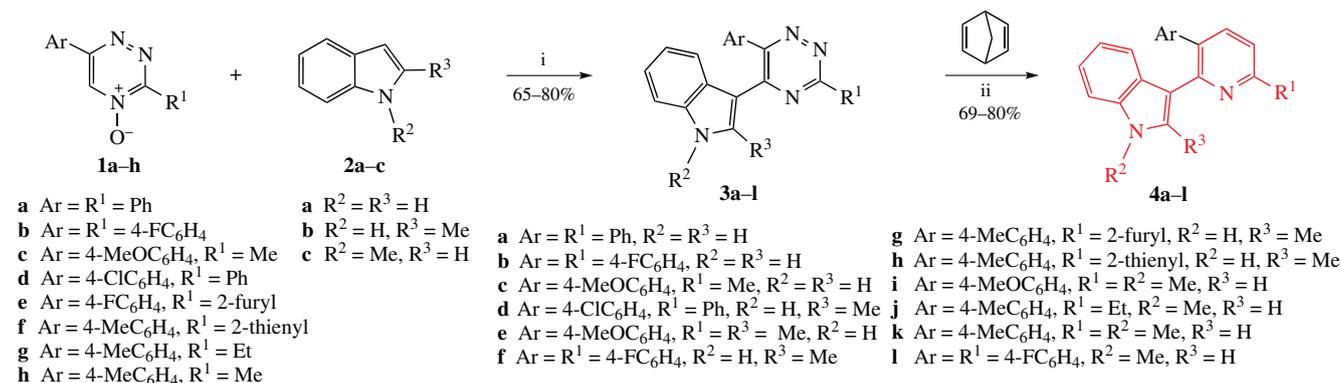
Such compounds are commonly prepared by cross-coupling of 3-indolyl and 2-pyridyl components.^{8–10} Also, a catalytic^{11–15} and non-catalytic^{16–18} construction of indole substituent at the position 2 of the pyridine core is documented. The corresponding pyridine *N*-oxides were used as the starting compounds when the final products were obtained *via* the deoxygenative nucleophilic substitution of hydrogen with the use of benzoyl chloride^{19,20} or acetic anhydride²¹ as *O*-acylating agents. In addition, the strategy of constructing a 2-pyridyl substituent at the position 3 of the indole core, including tandem reaction in unsubstituted indole²² leading to pyridine incrementing at the indole system²³ was reported (for graphical depiction of the previous syntheses, see Online Supplementary Materials, Scheme S1).

The possibility of synthesizing pyridine derivatives *via* their 1,2,4-triazine precursors should also be mentioned²⁴ in the this

aspect. Due to the higher electrophilicity of 1,2,4-triazines compared to pyridines, the possibilities of pre-functionalization of 1,2,4-triazines *via* the nucleophilic substitution of hydrogen seem topical.²⁵ Examples of using sequential S_N^H reactions, sometimes in combination with S_N^{PSO} reactions, and aza-Diels–Alder reactions to obtain a variety of multisubstituted (bi-)pyridines have been documented.²⁶ Here, we report a convenient synthetic approach to various 2-(indol-3-yl)pyridines using the convenient ‘1,2,4-triazine’ method which does not require the use of any catalysts and expensive reagents.

Direct metal-free introduction of indole residues at the position 5 of 1,2,4-triazines²⁷ or their 4-oxides²⁸ has been described previously. This approach is quite interesting due to wide possibilities of varying substituents at the positions 3 and 6 of 1,2,4-triazines or their oxides. This method was herein used to obtain 1,2,4-triazin-6-yl-substituted indole precursors (Scheme 1).

Starting 1,2,4-triazine 4-oxides **1a–h** were obtained as previously described.^{29,30} As a result of deoxygenative nucleophilic hydrogen substitution,²⁸ 5-(3-indolyl)-1,2,4-triazines **3a–l** were prepared in up to 80% yields. To further access the target 2-(indol-3-yl)pyridines **4**, we intended to employ the aza-



Scheme 1 Reagents and conditions: i, PhC(O)Cl, CH₂Cl₂, 20 °C, 2 h; ii, 1,2-dichlorobenzene, 215 °C, 20 h.

Diels–Alder reaction of triazines **3** with 2,5-norbornadiene as a dienophile.³¹ However, the application of typical conditions to these reactants, namely, boiling in *o*-xylene or 1,2-dichlorobenzene, provided low (5–10%) conversions even under the prolonged (72 h) processing. Luckily, carrying the reaction in a pressure vessel (which was previously implemented for 1,2,4-triazines with electron-donating substituents³²) at 215 °C for 20 h followed by flash chromatography afforded the desired pyridines **4a–l** in yields up to 79%. In this case, no influence of the nature of the reactant substituents on the reaction outcome was observed.

The structure of the obtained products **4** was confirmed based on the data of ¹H and ¹³C NMR spectroscopy, mass spectrometry, and elemental analysis. Thus, two characteristic doublets for protons of newly formed pyridine ring are observed in the ¹H NMR spectra. Also, the signals for protons of the substituents at the positions 3 and 6 of the former 1,2,4-triazine ring, as well as signals for protons of the indole fragments are presented. In most cases, a noticeable upfield shift for the signals of all the protons is observed upon the conversion of 1,2,4-triazines **3** into pyridines **4**. In particular, for the signal of proton of the indole NH group, a shift from 11.41–12.44 to 10.8–11.05 ppm can be noted.

In summary, we have reported herein a convenient metal-free synthesis of 2-(indol-3-yl)pyridines, which are of interest as promising biologically active candidates. Reaction sequence proceeds *via* the transformation of 1,2,4-triazine precursors, in the absence of expensive catalysts or reagents. The approach allows one to vary in a wide range a nature of substituents in the heterocyclic core of the target products (such as at the positions 3 and 6 of pyridine, as well as in the indolyl moiety) by using previously described convenient methods on the stage of the synthesis of starting 1,2,4-triazine-4-oxides and their derivatives.

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Online Supplementary Materials

Supplementary data associated with this article can be found in the online version at doi: 10.1016/j.mencom.2020.11.007.

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