

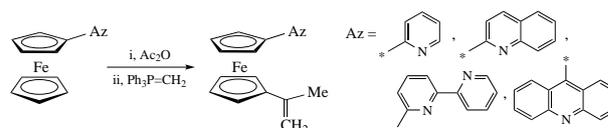
Regioselective synthesis of 1-azinyl-1'-isopropenylferrocenes

 Alexandra A. Musikhina,^{a,b} Irina A. Utepova,^{*a,b} Oleg N. Chupakhin,^{a,b}
Anna I. Suvorova^a and Elena Yu. Zyryanova^a
^a Institute of Chemical Engineering, Ural Federal University, 620002 Ekaterinburg, Russian Federation.
E-mail: i.a.utepova@urfu.ru

^b I. Ya. Postovsky Institute of Organic Synthesis, Ural Branch of the Russian Academy of Sciences,
620108 Ekaterinburg, Russian Federation

DOI: 10.1016/j.mencom.2020.03.026

The Friedel–Crafts acetylation (Ac₂O/AlCl₃) reaction of azinylferrocenes occurs regioselectively at position 1'. Acetylated derivatives upon reaction with Ph₃P=CH₂ give the corresponding 1-azinyl-1'-isopropenylferrocenes.



Keywords: ferrocenes, isopropenylferrocene, regioselective synthesis, Friedel–Crafts acylation, Wittig reaction, azines.

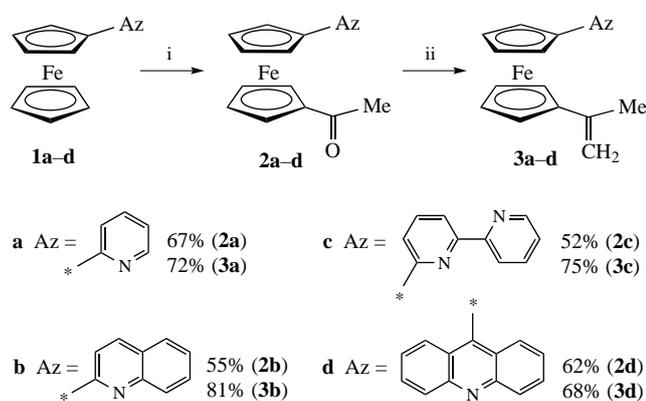
Owing to the exceptional electrochemical properties, substituted ferrocenes have been used in a wide range of material science applications.¹ Redox properties of the ferrocene moiety open up a new dimension for the construction of switchable motif for host–guest complexation,² nonlinear optical materials,³ sol–gel⁴ and micelle⁵ actuation, biological recognition,⁶ recyclable catalysts⁷ and switchable dyes.⁸ Olefins appended in ferrocene system allow one to create a highly conjugated two-dimensional p-electron system⁹ providing a wide scope of useful materials.¹⁰ Vinylferrocene, a monomer for metal-containing polymers,^{10,11} is used for production of self-healing^{4(a)} materials and composite colloidal crystal films.¹² Metal-containing polymers can participate in redox-responsive release from patchy nanocapsules,¹³ catalysis modulating activity¹⁴ and ion-selective membrane-gating through nanopores.¹⁵ The presence of lipophilic ferrocene moiety in the drug structure modulates their physiological activity.¹⁶

To date, the Wittig reaction of phosphoranes and aldehydes/ketones¹⁷ is most often employed for the synthesis of alkenylferrocenes. Alternatively, the McMurry coupling¹⁸ (TiCl₄/Zn) between the appropriate benzophenones and acetylferrocenes was used in the synthesis of anticancer drugs as well as for

reduction of aldehydes to obtain alcohol suitable for the following dehydration.¹⁹

The choice of azinylferrocenes is directly connected with the high efficiency of ferrocene derivatives in initiating mixtures for free-radical bulk polymerization of vinyl monomers.²⁰ Therefore, the synthesis of hetaryl-containing vinylferrocenes and study of their properties are of particular interest. To our knowledge, a protocol for the synthesis of non-symmetric azinyl-containing 1,1'-disubstituted vinylferrocenes has not been reported so far. In this work we have combined olefin and azine fragments on the ferrocene matrix by regioselective acylation followed by the Wittig reaction (Scheme 1).

The direct C–H functionalization of heteroarenes with ferrocenyllithium as a nucleophile (S_N^H reaction) generated azinylferrocenes **1**.^{15(a)} The Friedel–Crafts acetylation (Ac₂O/AlCl₃) of compounds **1a–d** occurs exclusively at the 'free' Cp fragment to afford 1,1'-disubstituted products **2a–d** in good yields (see Scheme 1).[†] Apparently, an electron-withdrawing azine substituent would somewhat reduce the nucleophilicity of cyclopentadienyl ring, which is responsible for the regioselectivity of the process.²¹ The steric bulkiness of these azine substituents



Scheme 1 Reagents and conditions: i, Ac₂O, AlCl₃, CH₂Cl₂, 0 °C; ii, Ph₃P=CH₂ (from Ph₃PMe⁺F and BuLi), Et₂O, room temperature.

[†] 1-Acetyl-1'-(pyridin-2-yl)ferrocene **2a**. Acetic anhydride (0.283 ml, 6 mmol) was added to a solution of azinylferrocene **1a** (0.263 g, 1 mmol) in anhydrous CH₂Cl₂ (20 ml) at 0 °C followed by portionwise addition of AlCl₃ (0.399 g, 1.1 mmol) within 2 h, while the colour changed from red-orange to deep purple. After stirring at room temperature for 24 h, the mixture was poured into cold water and extracted with CH₂Cl₂. The extract was dried over sodium sulfate and concentrated under reduced pressure. The residue was purified by column chromatography (Al₂O₃, hexane/EtOAc, 9:1). The appropriate fractions were concentrated *in vacuo*. Yield of **2a**, 0.204 g (67%). Mp, 78 °C. R_f = 0.15. ¹H NMR (400 MHz, CDCl₃) δ: 2.08 (s, 3 H, Me), 4.40 (m, 2H, C₅H₄), 4.47 (m, 2H, C₅H₄), 4.59 (m, 2H, C₅H₄), 5.04 (m, 2H, C₅H₄), 7.19–7.22 (m, 1H, 5'-H), 7.54–7.56 (d, 1H, 3'-H, ³J 7.9 Hz), 7.70–7.74 (m, 1H, 4'-H), 8.47–8.48 (d, 1H, 6'-H, ³J 4.1 Hz). ¹³C NMR (100 MHz, CDCl₃) δ: 27.13 (Me), 68.35, 70.51, 71.12, 73.21, 80.28, 85.45 (Fc), 120.43, 121.28, 136.27, 149.11, 156.56 (Py), 200.44 (CO). IR (ν/cm⁻¹): 785.52, 1151.81, 1585.44, 1650.51 (CO). HRMS, *m/z* (%): 312.0662 [M+Li]⁺ (100). Found (%): C, 66.98; H, 5.18; N, 4.60. Calc. for C₁₇H₁₅FeNO (%): C, 66.89; H, 4.92; N, 4.59.

should also be considered. In previous reports, acylation of methyl- or bromoferrocenes with smaller substituents afforded mixtures of 1,2-, 1,3- and 1,1'-disubstituted derivatives.^{1–3} In our case of azinylferrocenes, the regioselectivity is also attributed to the nature of the acetylating agent. For instance, on application of acetyl chloride or ethyl chloroformate, descent in the regioselectivity was observed. No products were formed when (CF₃CO)₂O/AlCl₃ and AcOH/Al₂O₃ systems were used. It should be noted that treatment of (quinoxalin-2-yl)ferrocene with Ac₂O/AlCl₃ system did not give the target 1'-acetyl derivative but occurred sluggishly at annealed benzene ring of the quinoxaline moiety. As for (phenanthrolin-2-yl)ferrocene, the target 1'-acetyl derivative was unstable towards column chromatography.

A general synthetic route to vinylferrocenes is based on the Wittig reaction of ferrocene carboxaldehyde with ylides.^{17(c)} In this work, acetylferrocenes **2a–d** reacted at room temperature under an argon atmosphere with ylide obtained *in situ* from excess triphenylmethylphosphonium iodide and BuLi (see Scheme 1).[‡] When reactivities of 1-acetyl-1'-azinylferrocenes **2a–d** and parent acetylferrocene in the Wittig olefination were compared, we indicated that isopropenyl derivatives **3a–d** were formed at room temperature, while the similar transformation of acetylferrocene required heating to 45 °C. Apparently, electron-withdrawing effect of hetaryl substituents in compounds **2** could enhance the electrophilicity of their carbonyl group. All obtained azinyl-substituted ferrocenes **2a–d** and **3a–d** are stable solid compounds coloured from orange to dark red. Their structures were confirmed by ¹H, ¹³C NMR and IR spectroscopy, high resolution mass spectrometry (HRMS), and elemental analyses.

In conclusion, our study provides a method for the regioselective 1'-acetylation of mono-substituted azinylferrocenes. The obtained acylated derivatives **2a–d** react with phosphorus ylides under mild conditions with formation of olefins **3a–d** enabling the design of redox and vinyl polymers.

This study was supported by the Russian Science Foundation (grant no. 19-73-00243, synthesis of derivatives **2**, **3**), the Ministry of Education and Science of the Russian Federation (no. 0836-2020-0058), and the Government of the Russian Federation (Foundation Act 211, contract no. 02.A03.21.0006).

Online Supplementary Materials

Supplementary data associated with this article can be found in the online version at doi: 10.1016/j.mencom.2020.03.026.

References

- (a) G. Ricci, G. Leone, F. Masi and A. Sommacchi, in *Ferrocenes: Compounds, Properties and Applications*, ed. E. S. Phillips, Nova Science Publishers, 2011, ch. 7, pp. 273–313; (b) *Ferrocenes: Ligands, Materials and Biomolecules*, ed. P. Štěpnička, Wiley, Chichester, 2008; (c) M. Herberhold, in *Ferrocenes: Homogeneous Catalysis, Materials Science*, eds. A. Togni and T. Hayashi, Wiley-VCH, Weinheim, 1995, pp. 219–278.
- (a) D. Sobransingh and A. E. Kaifer, *Org. Lett.*, 2006, **8**, 3247; (b) L. Yang, A. Gomez-Casado, J. F. Young, H. D. Nguyen, J. Cabanas-Danés, J. Huskens, L. Brunsveld and P. Jonkhøj, *J. Am. Chem. Soc.*, 2012, **134**, 19199; (c) J. Guo, C. Yuan, M. Guo, L. Wang and F. Yan, *Chem. Sci.*, 2014, **5**, 3261; (d) Y. Ahn, Y. Jang, N. Selvapalam, G. Yun and K. Kim, *Angew. Chem., Int. Ed.*, 2013, **52**, 3140; (e) Q. Yan, A. Feng, H. Zhang, Y. Yin and J. Yuan, *Polym. Chem.*, 2013, **4**, 1216.
- A. Togni and G. Rihns, *Organometallics*, 1993, **12**, 3368.
- (a) M. A. Hempenius, C. Cirmi, F. L. Savio, J. Song and G. J. Vancso, *Macromol. Rapid Commun.*, 2010, **31**, 772; (b) M. Nakahata, Y. Takashima, H. Yamaguchi and A. Harada, *Nat. Commun.*, 2011, **2**, 511; (c) C. Yuan, J. Guo, M. Tan, M. Guo, L. Qiu and F. Yan, *ACS Macro Lett.*, 2014, **3**, 271.
- T. Saji, K. Hoshino and S. Aoyagui, *J. Am. Chem. Soc.*, 1985, **107**, 6865.
- (a) B. S. Aytar, J. P. E. Muller, S. Golan, Y. Kondo, Y. Talmon, N. L. Abbott and D. M. Lynn, *J. Colloid Interface Sci.*, 2012, **387**, 56; (b) B. S. Aytar, J. P. E. Muller, Y. Kondo, Y. Talmon, N. L. Abbott and D. M. Lynn, *J. Am. Chem. Soc.*, 2013, **135**, 9111.
- (a) C. K. A. Gregson, V. C. Gibson, N. J. Long, E. L. Marshall, P. J. Oxford and A. J. P. White, *J. Am. Chem. Soc.*, 2006, **128**, 7410; (b) V. C. Gibson, N. J. Long, P. J. Oxford, A. J. P. White and D. J. Williams, *Organometallics*, 2006, **25**, 1932; (c) C. A. Fleckenstein and H. Plenio, *Adv. Synth. Catal.*, 2006, **348**, 1058; (d) X. Wang, A. Thevenon, J. L. Brosmer, I. Yu, S. I. Khan, P. Mehrkhodavandi and P. L. Diaconescu, *J. Am. Chem. Soc.*, 2014, **136**, 11264; (e) P. Neumann, H. Dib, A. M. Caminade and E. Hey-Hawkins, *Angew. Chem., Int. Ed.*, 2015, **54**, 311; (f) Q. Zhang, X. Cui, L. Zhang, S. Luo, H. Wang and Y. Wu, *Angew. Chem., Int. Ed.*, 2015, **54**, 5210.
- (a) S. Nagashima, M. Murata and H. Nishihara, *Angew. Chem., Int. Ed.*, 2006, **45**, 4298; (b) W.-Y. Wang, N.-N. Ma, S.-L. Sun and Y.-Q. Qiu, *Organometallics*, 2014, **33**, 3341; (c) P. Aloukos, K. Iliopoulos, S. Couris, D. M. Guldi, C. Soobar and A. Mateo-Alonso, *J. Mater. Chem.*, 2011, **21**, 2524; (d) Z. Yu, L. Yang, H. Min, P. Zhang, C. Zhou and R. Riedel, *J. Mater. Chem. C*, 2014, **2**, 1057.
- L. Chen and R. G. Compton, *ACS Sens.*, 2019, **4**, 1716.
- (a) F. Li, V. M. Basile and M. J. Rose, *Langmuir*, 2015, **31**, 7712; (b) *Synthetic Metal-Containing Polymers*, ed. I. Manners, Wiley-VCH, Weinheim, 2004; (c) N. J. Long and K. Kowalski, in *Ferrocenes: Ligands, Materials and Biomolecules*, ed. P. Štěpnička, Wiley, Chichester, 2008, pp. 393–496.
- M. Nakahata, Y. Takashima, H. Yamaguchi and A. Harada, *Nat. Commun.*, 2011, **2**, 511.
- R. H. Staff, M. Gallei, M. Mazurowski, M. Rehahn, R. Berger, K. Landfester and D. Crespy, *ACS Nano*, 2012, **6**, 9042.
- J. Elbert, J. Mersini, N. Vilbrandt, C. Lederle, M. Kraska, M. Gallei, B. Stühn, H. Plenio and M. Rehahn, *Macromolecules*, 2013, **46**, 4255.
- J. Elbert, F. Krohm, C. Rüttiger, S. Kienle, H. Didzoleit, B. N. Balzer, T. Hugel, B. Stühn, M. Gallei and A. Brunsen, *Adv. Funct. Mater.*, 2014, **24**, 1591.
- (a) O. N. Chupakhin, I. A. Utepova, I. S. Kovalev, V. L. Rusinov and Z. A. Starikova, *Eur. J. Org. Chem.*, 2007, **5**, 857; (b) I. A. Utepova, A. E. Lakhina, M. V. Varaksin, I. S. Kovalev, V. L. Rusinov, P. A. Slepukhin, M. I. Kodess and O. N. Chupakhin, *Russ. Chem. Bull., Int. Ed.*, 2008, **57**, 2156 (*Izv. Akad. Nauk, Ser. Khim.*, 2008, 2116).
- (a) M. Dervisevic, E. Çevik and M. Şenel, *Enzyme Microbiol. Technol.*, 2015, **68**, 69; (b) J. Morsbach, J. Elbert, C. Rüttiger, S. Winzen, H. Frey and M. Gallei, *Macromolecules*, 2016, **49**, 3406; (c) M. Ge, H. Huang, X. Gou, C. Hua, B. Chen and J. Zhao, *Chem. Heterocycl. Compd.*, 2018, **54**, 951.
- (a) J. Jia, Y. Cui, Y. Li, W. Sheng, L. Han and J. Gao, *Dyes Pigm.*, 2013, **98**, 273; (b) A. Hildebrandt, K. Al Khalyfeh, D. Schaarschmidt and M. Korb, *J. Organomet. Chem.*, 2016, **804**, 87; (c) J. M. Osgerby and P. L. Pauson, *J. Chem. Soc.*, 1961, 4604.
- (a) Y. Wang, M. A. Richard, S. Top, P. M. Dansette, P. Pigeon, A. Vessières, D. Mansuy and G. Jaouen, *Angew. Chem., Int. Ed.*, 2016, **55**, 10431; (b) M. Görmén, P. Pigeon, S. Top, E. A. Hillard, M. Huché, C. G. Hartinger, F. de Montigny, M.-A. Plamont, A. Vessières and G. Jaouen, *ChemMedChem*, 2010, **5**, 2039.
- (a) R. Shi, H. Wang, P. Tang and Y. Bin, *Front. Chem. Sci. Eng.*, 2014, **8**, 171; (b) K. Plevová, B. Mudráková and R. Šebesta, *Synthesis*, 2018, **50**, 760.
- (a) R. M. Islamova, O. I. Golovochesova, Yu. B. Monakov, I. A. Utepova, A. A. Musikhina and O. N. Chupakhin, *Polym. Sci., Ser. B*, 2010, **52**, 637 (*Vysokomol. Soedin.*, 2010, **52**, 2184); (b) Yu. B. Monakov, R. M. Islamova, O. I. Golovochesova, I. A. Utepova, A. A. Musikhina and O. N. Chupakhin, *Dokl. Chem.*, 2010, **432**, 140 (*Dokl. Akad. Nauk*, 2010, **432**, 195).
- T.-Y. Dong, C.-H. Huang, C.-K. Chang, H.-C. Hsieh, S.-M. Peng and G.-H. Lee, *Organometallics*, 1995, **14**, 1776.

Received: 4th October 2019; Com. 19/6034

[‡] 1-Isopropenyl-1'-(pyridin-2-yl)ferrocene **3a**. A 1.6 M hexane solution of BuLi (0.71 ml, 1.14 mmol) was added to a suspension of methyl(triphenyl)phosphonium iodide (1.25 mmol) in anhydrous diethyl ether (5 ml) under argon at room temperature. After stirring for 15 min, a solution of acetylferrocene **2a** (1.0 mmol) in anhydrous diethyl ether (20 ml) was added. After stirring the reaction mixture at room temperature for 4 h, it was purified on Al₂O₃ using hexane–ethyl acetate mixture (9:1) as eluent. The eluate was concentrated to dryness *in vacuo*. Yield of **3a**, 0.218 g (72%). Mp, 84 °C. R_f = 0.5. For characteristics, see Online Supplementary Materials.