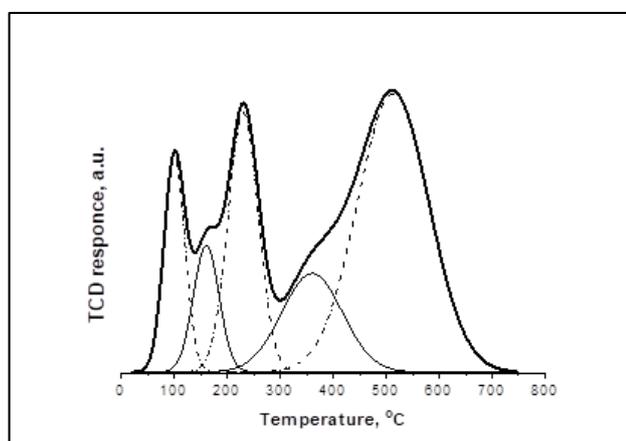
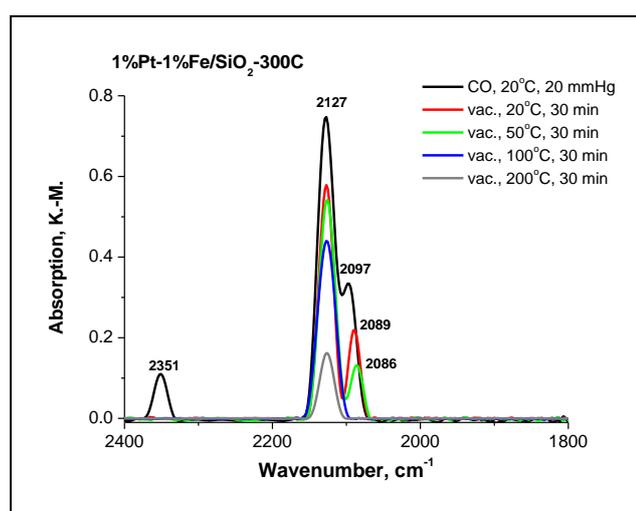


**Influence of the electronic state of the metals in Fe-Pt/SiO<sub>2</sub> catalysts on the performance of hydrogenation of phenylacetylene**

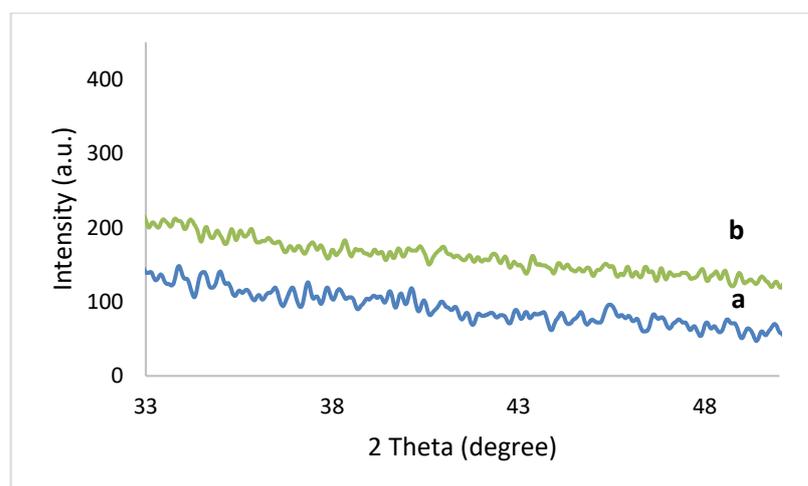
Anastasiya A. Shesterkina, Olga P. Tkachenko, Elena V. Shuvalova,  
Gennady I. Kapustin, Vladimir B. Kazansky and Leonid M. Kustov



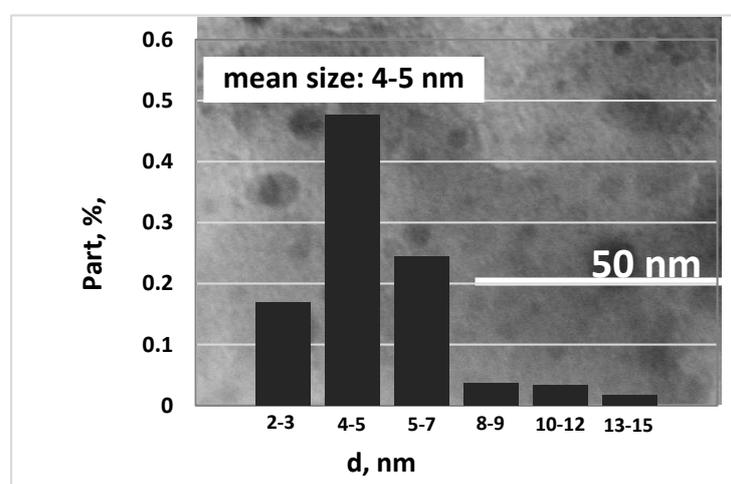
**Figure S1** Deconvolution of the TPR curve for the calcined 1%Pt-1%Fe/SiO<sub>2</sub>-300C catalyst.



**Figure S2** DRIFT-CO spectra of the calcined 1%Pt-1%Fe/SiO<sub>2</sub>-C catalyst.



**Figure S3** XRD patterns of 1%Pt/SiO<sub>2</sub> (a), 1%Pt-1%Fe/SiO<sub>2</sub>-C-H (b).



**Figure S4** TEM images of the 1%Pt-1%Fe/SiO<sub>2</sub>-C-H catalyst.

**Table S1** XPS results for monometallic Pt and bimetallic Pt-Fe catalysts.

Sample	Binding energy, eV		Surface atomic ratio			Volume atomic ratio		
	Fe 2p <sub>3/2</sub>	Pt 4f <sub>7/2</sub>	Fe/Si	Pt/Si	Pt/Fe	Fe/Si	Pt/Si	Pt/Fe
1%Pt/SiO <sub>2</sub> -H	-	71.2	-	0.0022	-	-	0.0036	-
1%Pt-1%Fe/SiO <sub>2</sub> -C-H	710.9	70.8	0.0058	0.0017	0.29	0.014	0.0036	0.26