

Pyridinium bromide as a new mediator for electrochemical transformations involving CH-acids

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All melting points were measured with a Gallenkamp melting point apparatus and are uncorrected. ^1H and ^{13}C NMR were recorded with a Bruker AM300 at ambient temperature in $\text{DMSO-}d_6$ or CDCl_3 solutions with working frequency of 300.13 MHz. Chemical shifts values are given in δ scale relative to Me_4Si . IR spectra were recorded with a Bruker ALPHA-T FT-IR spectrometer in KBr pellets. Mass-spectra (EI = 70 eV) were recorded with a Finnigan MAT INCOS 50 spectrometer. High-resolution mass spectrometry (HRMS) (electrospray ionization, ESI) was measured on a Bruker micrOTOF II instrument; external or internal calibration was done with an Electrospray Calibrant Solution (Fluka). Alkylidenemalononitriles **1** were synthesized by the condensation of the corresponding carbonyls and malononitrile *via* the Knoevenagel condensation.^[S1,S2]

General procedure. An undivided cell was equipped with a graphite anode (area 5 cm^2) and Fe cathode (area 5 cm^2) and connected to a DC regulated power supply. The cell was charged with the corresponding alkylidenemalononitrile **1** (10 mmol), C-H acid **2** (10 mmol), PyHBr (7.5 mmol), and acetonitrile (20 ml). The mixture was electrolyzed (2.2 F mol^{-1}) using constant current conditions (50 mA cm^{-2}) at 20 $^\circ\text{C}$ under magnetic stirring (TLC control). When the reaction was complete, the solvent was removed under reduced pressure. The residue was treated with water, and the product was extracted with dichloromethane ($3 \times 20 \text{ ml}$), the extract was dried over Na_2SO_4 , and concentrated in vacuum. The residue was purified by crystallization from methanol to afford the desired cyclopropane **3**.

3-Phenyl-1,1,2,2-tetracyanocyclopropane (3a)

White solid. Yield 1.57 g (72%); mp 226-228 $^\circ\text{C}$ (lit.^[S3] mp 227-230 $^\circ\text{C}$). ^1H NMR ($\text{DMSO-}d_6$) δ : 5.10 (1H, s, CH), 7.48-7.80 (5H, m, Ar).

3-(4-Methylphenyl)-1,1,2,2-tetracyanocyclopropane (3b)

White solid. Yield 1.53 g (66%); mp 226-227 $^\circ\text{C}$ (lit.^[S3] mp 227-230 $^\circ\text{C}$). ^1H NMR ($\text{DMSO-}d_6$) δ : 2.31 (3H, s, CH_3), 5.20 (1H, s, CH), 7.28 (2H, d, $J = 8.0 \text{ Hz}$, Ar), 7.65 (2H, d, $J = 8.0 \text{ Hz}$, Ar).

3-(4-Chlorophenyl)-1,1,2,2-tetracyanocyclopropane (3c)

White solid. Yield 1.61 g (64%); mp 248-251 $^\circ\text{C}$ (lit.^[S3] mp 250-251 $^\circ\text{C}$). ^1H NMR ($\text{DMSO-}d_6$) δ : 5.28 (1H, s, CH), 7.59 (2H, d, $J = 8.5 \text{ Hz}$, Ar), 7.88 (2H, d, $J = 8.5 \text{ Hz}$, Ar).

3-(3-Bromophenyl)-1,1,2,2-tetracyanocyclopropane (3d)

White solid. Yield 1.72 g (58%); mp 187-189 $^\circ\text{C}$ (lit.^[S4] mp 186-187 $^\circ\text{C}$). ^1H NMR ($\text{DMSO-}d_6$) δ : 5.31 (1H, s, CH), 7.44 (1H, t, $J = 8.5 \text{ Hz}$, Ar), 7.68 (1H, d, $J = 8.5 \text{ Hz}$, Ar), 7.87 (1H, d, $J = 8.5 \text{ Hz}$, Ar), 8.22 (1H, s, Ar).

3-(4-Nitrophenyl)-1,1,2,2-tetracyanocyclopropane (3e)

White solid. Yield 1.26 g (48%); mp 232-234 °C (lit. ^[S3] mp 232-235 °C). ¹H NMR (DMSO-*d*₆) δ: 5.52 (1H, s, CH), 8.20 (2H, d, *J* = 8.8 Hz, Ar), 8.35 (2H, d, *J* = 8.8 Hz, Ar).

1,1,2,2-Tetracyanospiro[2.5]octane (3f)

White solid. Yield 1.26 g (60%); mp 176-178 °C (lit. ^[S5] mp 177-179 °C). ¹H NMR (DMSO-*d*₆) δ: 1.45-1.55 (2H, m, CH₂), 1.60-1.75 (4H, m, 2CH₂), 1.80-1.95 (4H, m, 2CH₂).

Dimethyl 2,2-dicyano-3-phenylcyclopropane-1,1-dicarboxylate (3g)

White solid. Yield 1.90 g (67%); mp 126-128 °C. (lit. ^[S6] mp 126-128 °C). ¹H NMR (DMSO-*d*₆) δ: 3.75 (s, 3H, OMe), 3.92 (s, 3H, OMe), 4.41 (s, 1H, CH), 7.35-7.50 (m, 5H, Ph).

Dimethyl 2,2-dicyano-3-(4-methylphenyl)cyclopropane-1,1-dicarboxylate (3h)

White solid. Yield 1.87 g (63%); mp 137-139 °C. (lit. ^[S6] mp 137-139 °C). ¹H NMR (CDCl₃) δ: 2.39 (s, 3H, CH₃), 3.79 (s, 3H, OMe), 3.96 (s, 1H, CH), 3.98 (s, 3H, OMe), 7.18-7.30 (m, 4H, Ar).

Dimethyl 2,2-dicyano-3-(3-methoxyphenyl)cyclopropane-1,1-dicarboxylate (3i)

White solid. Yield 2.10 g (67%); mp 92-94 °C (methanol). ¹H NMR (CDCl₃) δ: 3.78 (s, 3H, OMe), 3.80 (s, 3H, OMe), 3.92 (s, 1H, CH), 3.95 (s, 3H, OMe), 6.92-7.02 (m, 2H, Ar), 7.24-7.40 (m, 2H, Ar). ¹³C NMR (75.47 MHz, CDCl₃) δ: 16.6, 40.6, 46.6, 53.9, 54.8, 56.0, 110.1, 111.8, 112.4, 113.9, 119.1, 124.6, 146.2, 151.1, 161.9, 163.4. IR, $\nu_{\max}/\text{cm}^{-1}$: 2252, 1748, 1732, 1438, 1272. MS (EI), *m/z* (%): 344 [M]⁺ (5), 313 (17), 285 (27), 225 (16), 200 (13), 181 (15), 156 (11), 140 (13), 114 (10), 59 (100). HRMS *m/z* (ESI) calcd for C₁₆H₁₄N₂O₅ (M+H)⁺ 315.0980, found 315.0976.

Dimethyl 2,2-dicyano-3-(3-bromophenyl)cyclopropane-1,1-dicarboxylate (3j)

White solid. Yield 2.32 g (64%); mp 110-111 °C (lit. ^[S2] mp 110-111 °C). ¹H NMR (CDCl₃) δ: 3.80 (s, 3H, CH₃O), 3.95 (s, 1H, CH), 3.97 (s, 3H, CH₃O), 7.26-7.35 (m, 2H, Ar), 7.50-7.55 (m, 2H, Ar).

Dimethyl 2,2-dicyano-3-(3-nitrophenyl)cyclopropane-1,1-dicarboxylate (3k)

White solid. Yield 1.44 g (44%); mp 113-115 °C (methanol). ¹H NMR (CDCl₃) δ: 3.85 (s, 3H, OMe), 4.03 (s, 3H, OMe), 4.07 (s, 1H, CH), 7.55-7.65 (m, 1H, Ar), 7.70-7.84 (m, 1H, Ar), 8.18-8.32 (m, 2H, Ar). ¹³C NMR (75.47 MHz, CDCl₃) δ: 16.5, 40.0, 46.2, 54.5, 55.2, 109.8, 111.4, 123.6, 123.9, 130.3, 131.6, 134.8, 149.3, 162.3, 163.8. IR, $\nu_{\max}/\text{cm}^{-1}$: 2256, 1748, 1524, 1438, 1266. MS (EI), *m/z* (%): 329 [M]⁺ (4), 298 (11), 270 (57), 239 (16), 200 (11), 173 (15), 153 (11), 140 (13), 114 (12), 59 (100). HRMS *m/z* (ESI) calcd for C₁₅H₁₁N₃O₆ (M+H)⁺ 330.0726, found 330.0732.

Dimethyl 2,2-dicyano-3-(4-nitrophenyl)cyclopropane-1,1-dicarboxylate (3l)

White solid. Yield 1.25 g (38%); mp 140-142 °C. (lit. ^[S6] mp 140-142 °C). ¹H NMR (CDCl₃) δ: 3.85 (s, 3H, OMe), 4.02 (s, 3H, OMe), 4.05 (s, 1H, CH), 7.62 (2H, d, *J* = 8.0 Hz, Ar), 8.30 (2H, d, *J* = 8.0 Hz, Ar).

References

- [S1] J. Mirek, M. Adamczyk and M. Mokrosz, *Synthesis*, 1980, 296.
- [S2] A. N. Vereshchagin, M. N. Elinson, A. S. Dorofeev and G. I. Nikishin, *Russ. Chem. Bull., Int. Ed.*, 2009, **58**, 902 (*Izv. Akad. Nauk, Ser. Khim.*, 2009, 883).
- [S3] Y. C. Kim and H. Hart, *Tetrahedron*, 1969, **25**, 3869.
- [S4] M. N. Elinson, S. K. Feducovich, N. O. Stepanov, A. N. Vereshchagin and G. I. Nikishin, *Tetrahedron*, 2008, **64**, 708.
- [S5] H. Hart and Y. C. Kim, *J. Org. Chem.*, 1966, **31**, 2784.
- [S6] M. N. Elinson, S. K. Feducovich, T. A. Zaimovskaya, A. N. Vereshchagin, S. V. Gorbunov and G. I. Nikishin, *Russ. Chem. Bull., Int. Ed.*, 2005, **54**, 1593 1593 (*Izv. Akad. Nauk, Ser. Khim.*, 2005, 1547).