

Ammonium iodide-mediated electrosynthesis of unsymmetrical thiosulfonates from arenesulfonohydrazides and thiols

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General Information

NMR spectra were registered on Bruker Avance II 300 MHz instrument. Chemical shifts were measured relative to solvent peaks as an internal standard set to δ 7.25 and δ 77.0 (CDCl₃), δ 2.50 and δ 39.51 (DMSO-d₆). The TLC analyses were carried out on standard silica-gel chromatography plates. The melting points were determined on a Kofler hot-stage apparatus. Chromatography was performed on silica gel (60-200 mesh).

p-Toluenesulfonohydrazide (**1a**), thiophenol (**2a**), 4-methylbenzenethiol (**2b**), 4-chlorobenzenethiol (**2c**), 2-methylpropane-1-thiol (**2d**), butane-1-thiol (**2e**), 4-methoxybenzenesulfonyl chloride, 4-fluorobenzenesulfonyl chloride, 4-chlorobenzenesulfonyl chloride, 4-bromobenzenesulfonyl chloride, 4-iodobenzenesulfonyl chloride, 4-nitrobenzenesulfonyl chloride, naphthalene-2-sulfonyl chloride, 2,4,6-trimethylbenzenesulfonyl chloride, tetrabutylammonium perchlorate, KI, NH₄I, TBAI, NH₄Br, NH₄Cl, Na₂SO₄, THF, dioxane, EtOH, DMSO, petroleum ether (PE, 40/70), ethyl acetate (EA) were used as purchased. 4-Methoxybenzenesulfonohydrazide (**1b**), 4-fluorobenzenesulfonohydrazide (**1c**), 4-chlorobenzenesulfonohydrazide (**1d**), 4-bromobenzenesulfonohydrazide (**1e**), 4-iodobenzenesulfonohydrazide (**1f**), 4-nitrobenzenesulfonohydrazide (**1g**), naphthalene-2-sulfonohydrazide (**1h**), 2,4,6-trimethylbenzenesulfonohydrazide (**1i**) were synthesized according to the literature.^[S1]

Cyclic voltammetry (CV) was implemented on an IPC-Pro computer-assisted potentiostat manufactured by Econix (scan rate error 1.0%, potential setting 0.25 mV). The experiments were performed in a 10ml five-necked glass conic electrochemical cell with a water jacket for thermostating. CV curves were recorded using a three-electrode scheme. The working electrode was a disc glassy-carbon electrode (d = 1.7 mm). A platinum wire served as an auxiliary electrode. A saturated calomel electrode was used as the reference electrode and was linked to the solution by a bridge with a porous ceramic diaphragm filled with background electrolyte. The tested solutions were thermostated at 25±0.5 °C. In a typical case, 5 ml solution was used and the depolarizer concentration was 3 mmol·dm⁻³. The working electrode was polished before recording each CV curve.

General procedure 1.

Optimization of the reaction conditions for synthesis of 3a from sulfonohydrazide 1a and thiophenol 2a (see Table 1). An undivided cell was equipped with a graphite plate anode (5 cm²) and a stainless steel plate cathode (5 cm²) and connected to a DC regulated power supply. The solution of *p*-toluenesulfonohydrazide **1a** (1-3 mmol, 186-558 mg), thiophenol **2a** (1-3 mmol, 110-330 mg) and supporting electrolyte KI, NH₄I, TBAI, NH₄Br, NH₄Cl (0.5-3 mmol) in H₂O-THF (1:1), H₂O-dioxane (1:1), EtOH-THF (1:1), EtOH or DMSO (30 ml each) was electrolyzed at constant current (50 mA·cm⁻²) at 30 °C under magnetic stirring. During the electrosynthesis the reaction mixture is heated, that is why external cooling is needed. Then the electrodes were washed with EtOH (2×30 ml). The solvent from combined organic phases was removed under reduced pressure. The residue was diluted with EtOAc (50 ml) and washed with saturated aqueous solution of Na₂S₂O₃ (8 ml), brine (2×8 ml) and water (2×8 ml), dried over Na₂SO₄, concentrated under reduced pressure and dried at 10-20 Torr. The desired product **3a** was isolated by chromatography on SiO₂ with elution using PE-EA in a linear gradient of the latter from 5 to 20 vol %.

General procedure 2.

Synthesis of thiosulfonates 3a-m (see Scheme 1). An undivided cell was equipped with a graphite plate anode (5 cm²) and a stainless steel plate cathode (5 cm²) and connected to a DC regulated power supply. The solution of sulfonyl hydrazide **1a-i** (1 mmol), thiol **2a-e** (2

mmol) and supporting electrolyte NH₄I (1 mmol) in THF-H₂O mixture (1:1, 30 ml) was electrolyzed at constant current (50 mA·cm⁻²) at 30 °C under magnetic stirring. During the electrosynthesis the reaction mixture is heated, that is why external cooling is needed. Then the electrodes were washed with EtOH (2×30 ml). The solvent from combined organic phases was removed under reduced pressure. The residue was diluted with EtOAc (50 ml) and washed with saturated aqueous solution of Na₂S₂O₃ (8 ml), brine (2×8 ml) and water (2×8 ml), dried over Na₂SO₄, concentrated under reduced pressure and dried at 10-20 Torr. The products **3a-m** were isolated by chromatography on SiO₂ with elution using PE-EA in a linear gradient of the latter from 5 to 20 vol %.

Scale up experiment for electrosynthesis of 3a. An undivided cell was equipped with a graphite plate anode (10 cm²) and a stainless steel plate cathode (10 cm²) and connected to a DC regulated power supply. The solution of *p*-toluenesulfonylhydrazide **1a** (10 mmol, 1.86g), thiophenol **2a** (20 mmol, 2.2g) and supporting electrolyte NH₄I (10 mmol) in H₂O-THF mixture (1:1, 150 ml) was electrolyzed at constant current (50 mA·cm⁻²) at 30 °C under magnetic stirring. During the electrosynthesis the mixture is heated, and external cooling is needed. Then electrodes were washed with EtOH (2×50 ml). The solvent from combined organic phases was removed under reduced pressure. The residue was diluted with EtOAc (250 ml) and washed with saturated aqueous solution of Na₂S₂O₃ (30 ml), brine (2×30 ml) and water (2×30 ml), dried over Na₂SO₄, concentrated under reduced pressure and dried at 10-20 Torr. The desired product **3a** was isolated in 64% yield (6.4 mmol, 1962 mg) by chromatography on SiO₂ with elution using PE-EA in a linear gradient of the latter from 5 to 20 vol %.

Analytical data for thiosulfonates **3a-m**.

S-Phenyl 4-methylbenzenesulfonothioate (**3a**). ^[S2]White solid, m.p. = 70.5-72.5°C. Yield 70%. R_f = 0.24 (TLC, PE:EA, 10:1). ¹H NMR (CDCl₃), δ: 2.40 (s, 3H), 7.19 (d, J = 8.1 Hz, 2H), 7.30-7.37 (m, 4H), 7.42-7.48 (m, 3H). ¹³C NMR (CDCl₃), δ: 21.6, 127.5, 128.0, 128.8, 129.3, 131.2, 136.5, 140.3, 144.7.

S-Phenyl 4-methoxybenzenesulfonothioate (**3b**). ^[S2]White solid, m.p. = 48.8-50.8°C. Yield 61%. R_f = 0.38 (TLC, PE:EA, 5:1). ¹H NMR (CDCl₃), δ: 3.84 (s, 3H), 6.83 (d, J = 8.9 Hz, 2H), 7.29-7.36 (m, 4H), 7.42-7.48 (m, 3H). ¹³C NMR (CDCl₃), δ: 55.7, 113.8, 128.1, 129.3, 129.8, 131.2, 134.8, 136.5, 163.6.

S-Phenyl 4-fluorobenzenesulfonothioate (**3c**). ^[S2]White solid, m.p. = 75.0-77.0°C. Yield 55%. R_f = 0.33 (TLC, PE:EA, 10:1). ¹H NMR (CDCl₃), δ: 7.07 (t, 2H, J = 8.4 Hz), 7.31-7.36 (m, 4H), 7.45-7.50 (m, 1H), 7.53-7.58 (m, 2H). ¹³C NMR (CDCl₃), δ: 116.0 (d, J = 22.6 Hz), 127.7, 129.5, 130.4 (d, J = 10.0 Hz), 131.5, 136.5, 139.0 (d, J = 3.3 Hz), 165.5 (d, J = 256.6 Hz).

S-Phenyl 4-chlorobenzenesulfonothioate (**3d**). ^[S2]White solid, m.p. = 77.5-79.5°C. Yield 46%. R_f = 0.38 (TLC, PE:EA, 10:1). ¹H NMR (CDCl₃), δ: 7.32-7.38 (m, 6H), 7.46-7.51 (m, 3H). ¹³C NMR (CDCl₃), δ: 127.5, 128.9, 129.0, 129.6, 131.6, 136.5, 140.2, 141.4.

S-Phenyl 4-bromobenzenesulfonothioate (**3e**). ^[S2]White solid, m.p. = 67.0-68.0°C. Yield 66%. R_f = 0.25 (TLC, PE:EA, 10:1). ¹H NMR (CDCl₃), δ: 7.32-7.40 (m, 6H), 7.46-7.50 (m, 1H), 7.54 (d, J = 8.8 Hz, 2H). ¹³C NMR (CDCl₃), δ: 127.5, 128.8, 128.9, 129.6, 131.6, 132.1, 136.5, 142.0.

S-Phenyl 4-iodobenzenesulfonothioate (**3f**). White solid, m.p. = 77.5-79.0°C. Yield 57%. R_f = 0.21 (TLC, PE:EA, 10:1). ¹H NMR (CDCl₃), δ: 7.26 (d, 2H, J = 8.5 Hz), 7.36-7.42 (m, 4H), 7.49-7.55 (m, 1H), 7.79 (d, 2H, J = 8.5 Hz). ¹³C NMR (CDCl₃), δ: 101.5, 127.5, 128.7, 129.6, 131.6, 136.5, 138.0, 142.7.

S-Phenyl 4-nitrobenzenesulfonothioate (**3g**). ^[S2]Yellow solid, m.p. = 137.0-139.0°C. Yield 25%. R_f = 0.27 (TLC, PE:EA, 10:1). ¹H NMR (CDCl₃), δ: 7.34-7.43 (m, 4H), 7.49-7.57 (m, 1H), 7.71 (d, J = 8.8 Hz, 2H), 8.25 (d, J = 8.8 Hz, 2H). ¹³C NMR (CDCl₃), δ: 124.1, 127.5, 128.7, 129.8, 132.0, 136.5, 147.9, 150.4.

S-Phenyl naphthalene-2-sulfonothioate (**3h**). ^[S3]White solid, m.p. = 61.0-62.5°C. Yield 43%. R_f = 0.21 (TLC, PE:EA, 10:1). ¹H NMR (CDCl₃), δ: 7.28-7.37 (m, 4H), 7.47 (t, J = 6.9 Hz, 1H), 7.61 (t, J = 7.3 Hz, 1H), 7.67-7.71 (m, 2H), 7.82 (d, J = 8.1 Hz, 1H), 7.93 (d, J = 8.8 Hz, 2H), 7.98 (s, 1H). ¹³C NMR (CDCl₃), δ: 122.3, 127.7, 127.9, 128.0, 129.2, 129.4, 129.4, 129.4, 131.4, 131.5, 135.1, 136.6, 139.6.

S-Phenyl 2,4,6-trimethylbenzenesulfonothioate (**3i**). White solid, m.p. = 72.5-74.5°C. Yield 30%. R_f = 0.78 (TLC, PE:EA, 5:1). ¹H NMR (CDCl₃), δ: 2.29 (s, 3H), 2.31 (s, 6H), 6.85 (s, 2H), 7.29-7.33 (m, 4H), 7.42-7.47 (m, 1H). ¹³C NMR (CDCl₃), δ: 21.0, 22.6, 127.9, 129.2, 131.2, 131.7, 137.0, 140.1, 143.6.

S-*p*-Tolyl 4-methylbenzenesulfonothioate (**3j**). ^[S2]White solid, m.p. = 70.5-72.5°C. Yield 68%. R_f = 0.62 (TLC, PE:EA, 5:1). ¹H NMR (CDCl₃), δ: 2.36 (s, 3H), 2.44 (s, 3H), 7.15 (d, J = 8.1 Hz, 2H), 7.24 (t, J = 8.1 Hz, 4H), 7.47 (d, J = 8.1 Hz, 2H). ¹³C NMR (CDCl₃), δ: 21.4, 21.6, 124.5, 127.5, 129.3, 130.2, 136.4, 140.4, 142.0, 144.6.

S-(4-Chlorophenyl) 4-methylbenzenesulfonothioate (**3k**). ^[S2]White solid, m.p. = 80.5-82.5°C. Yield 80%. R_f = 0.22 (TLC, PE:EA, 10:1). ¹H NMR (CDCl₃), δ: 2.45 (s, 3H), 7.24-7.32 (m, 6H), 7.48 (d, J = 8.1 Hz, 2H). ¹³C NMR (CDCl₃), δ: 21.6, 126.5, 127.6, 129.5, 129.6, 137.7, 138.1, 140.2, 145.0.

S-Isobutyl 4-methylbenzenesulfonothioate (**3l**). Colorless liquid. Yield 35%. R_f = 0.46 (TLC, PE:EA, 10:1). ¹H NMR (CDCl₃), δ: 0.89 (d, J = 6.6 Hz, 6H), 1.77-1.91 (m, 1H), 2.43 (s, 3H), 2.85 (d, J = 6.7 Hz, 2H), 7.31 (d, J = 8.1 Hz, 2H), 7.78 (d, J = 8.1 Hz, 2H). ¹³C NMR (CDCl₃), δ: 21.6, 21.6, 28.1, 44.3, 126.9, 129.7, 142.1, 144.5.

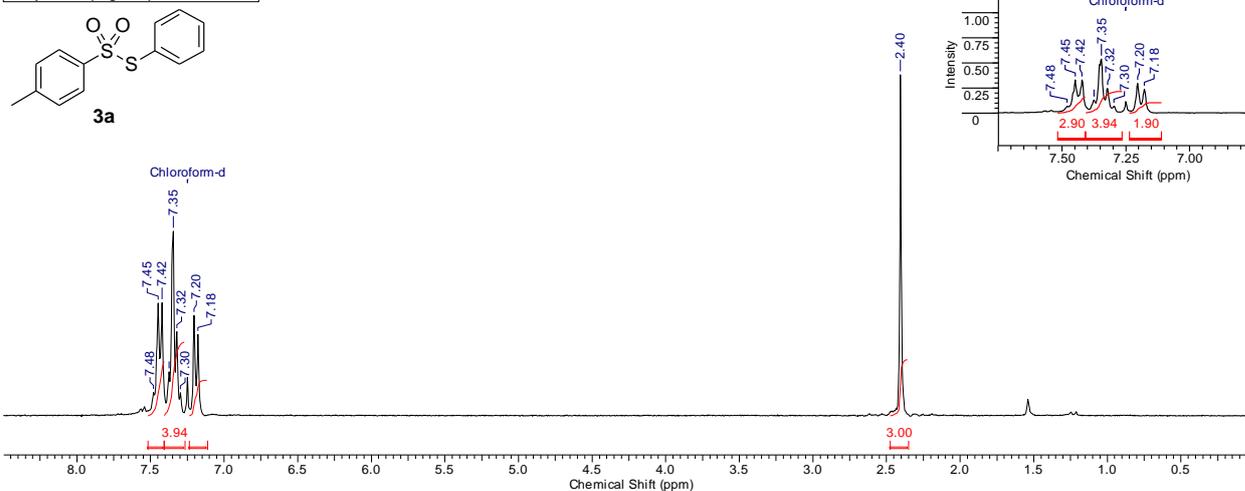
S-Butyl 4-methylbenzenesulfonothioate (**3m**). ^[S4]Colorless liquid. Yield 55%. R_f = 0.46 (TLC, PE:EA, 10:1). ¹H NMR (CDCl₃), δ: 0.83 (t, J = 7.3 Hz, 3H), 1.25-1.37 (m, 2H), 1.51-1.61 (m, 2H), 2.43 (s, 3H), 2.96 (t, J = 7.3 Hz, 2H), 7.32 (d, J = 8.2 Hz, 2H), 7.79 (d, J = 8.2 Hz, 2H). ¹³C NMR (CDCl₃), δ: 13.3, 21.6, 30.6, 35.6, 127.0, 129.8, 142.0, 144.6.

References

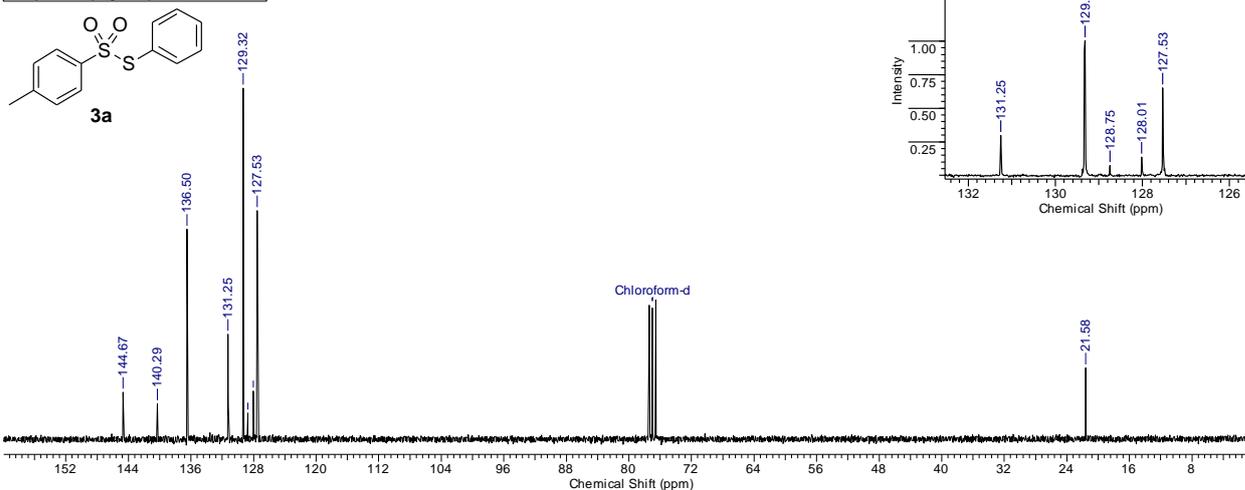
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- [S3] Z. Peng, X. Zheng, Y. Zhang, D. An and W. Dong, *Green Chem.*, 2018, **20**, 1760.
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NMR Spectra of Synthesized Compounds

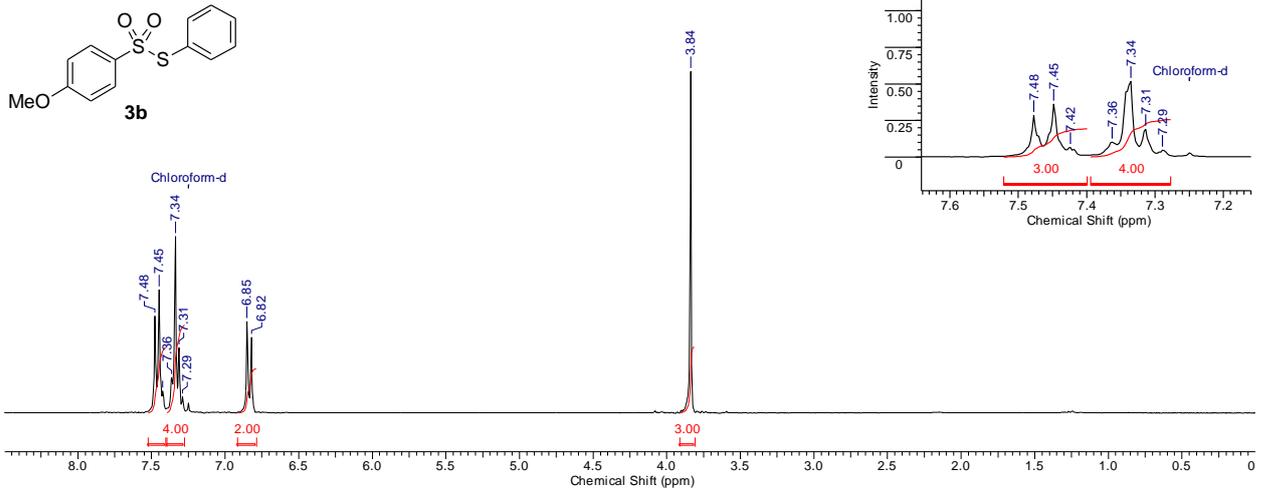
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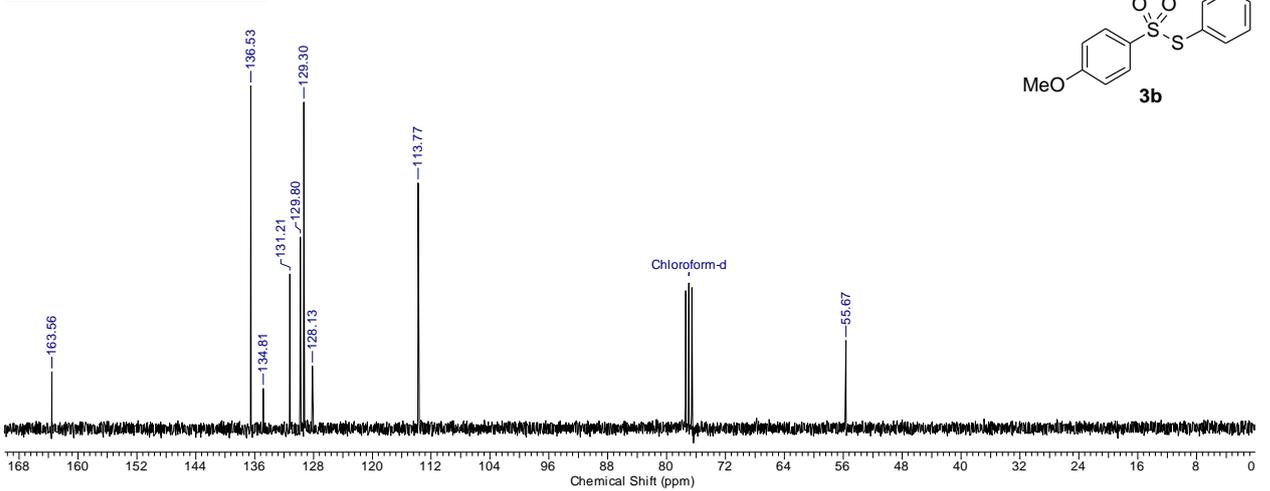
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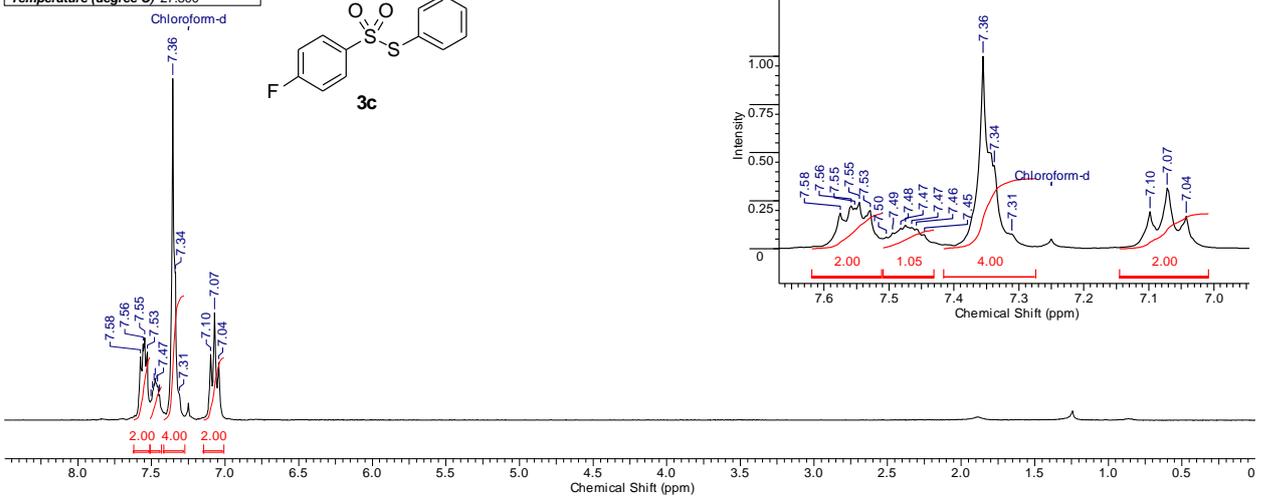
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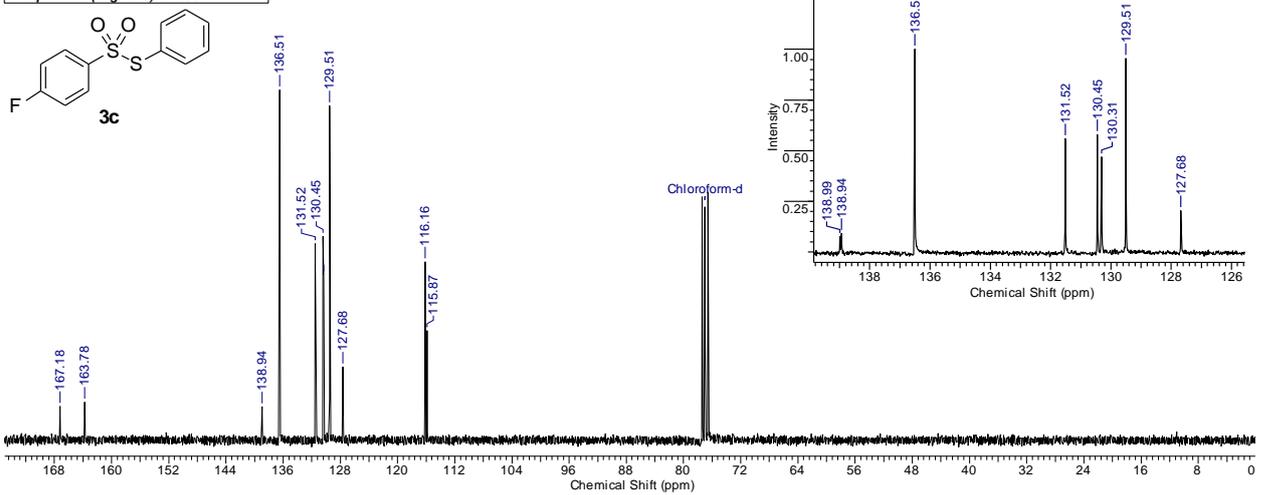
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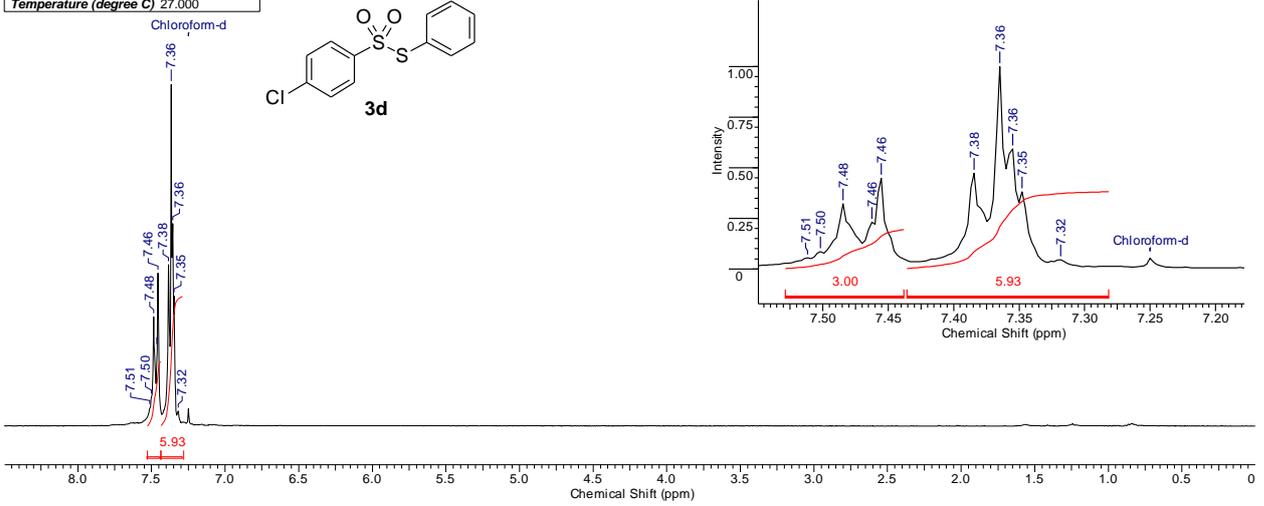
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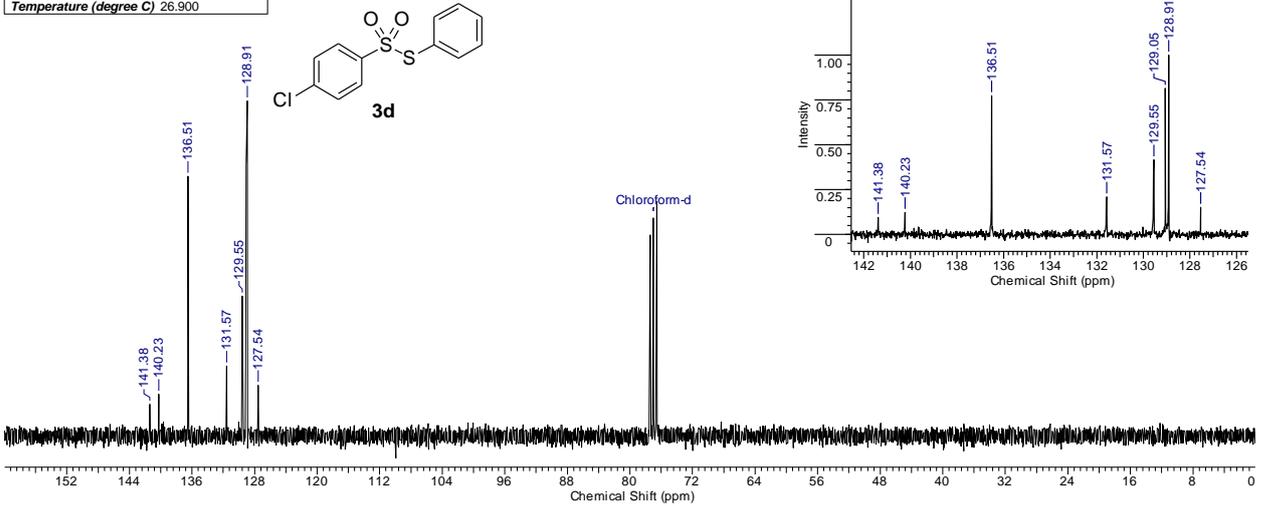
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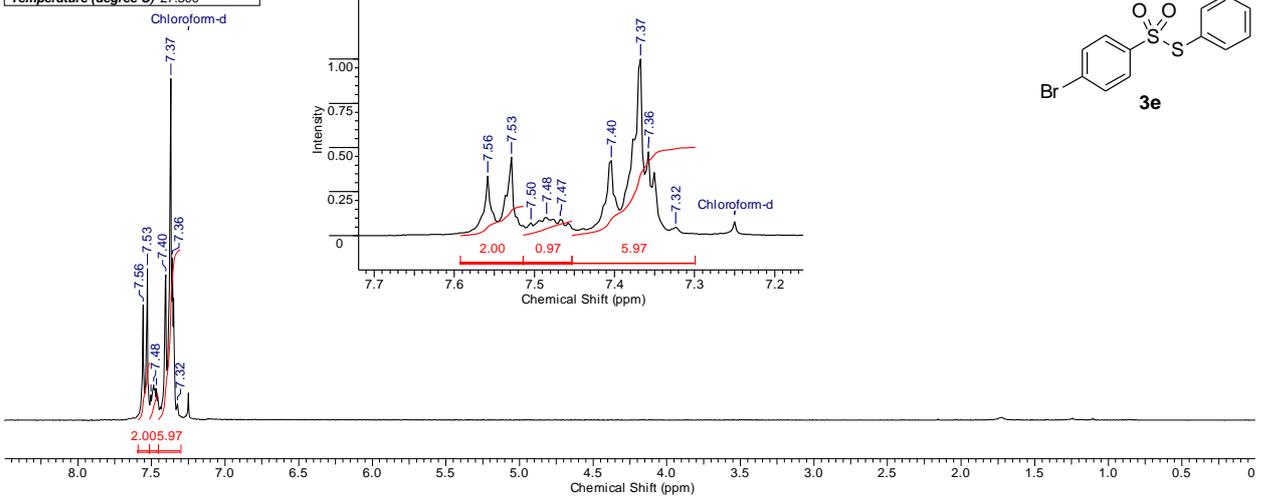
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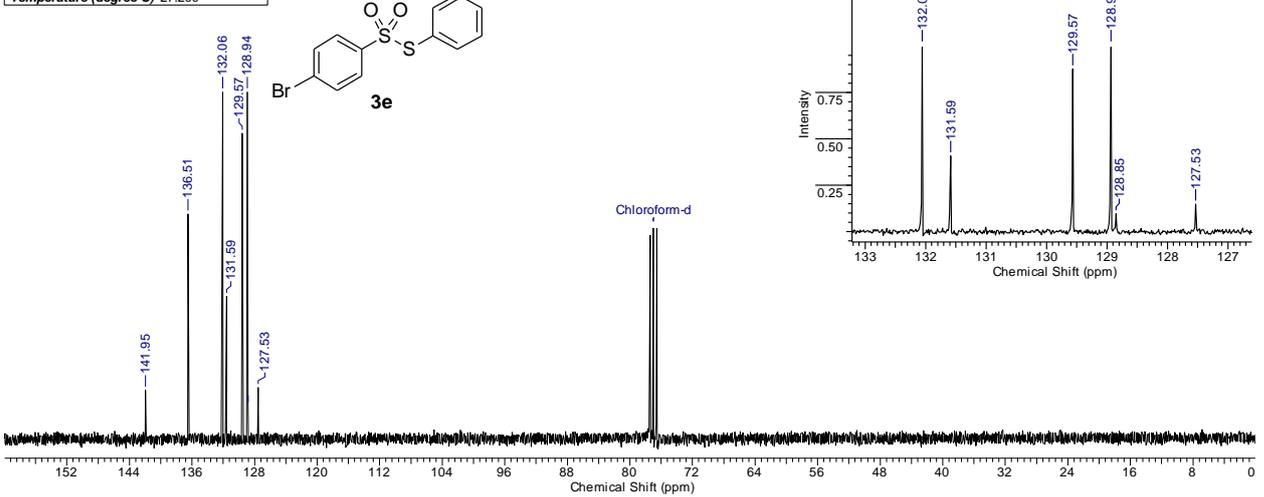
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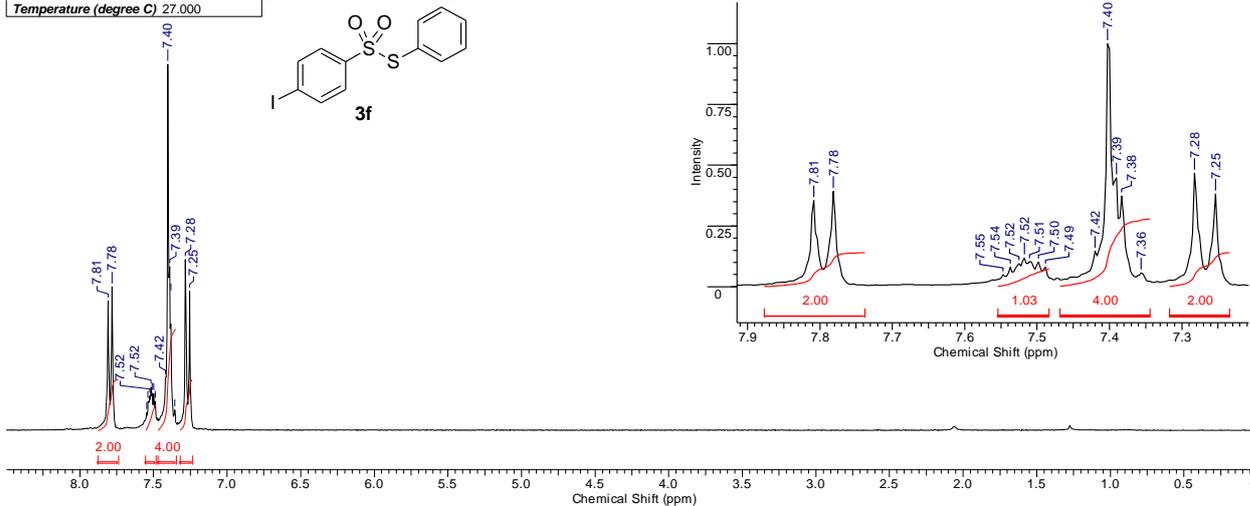
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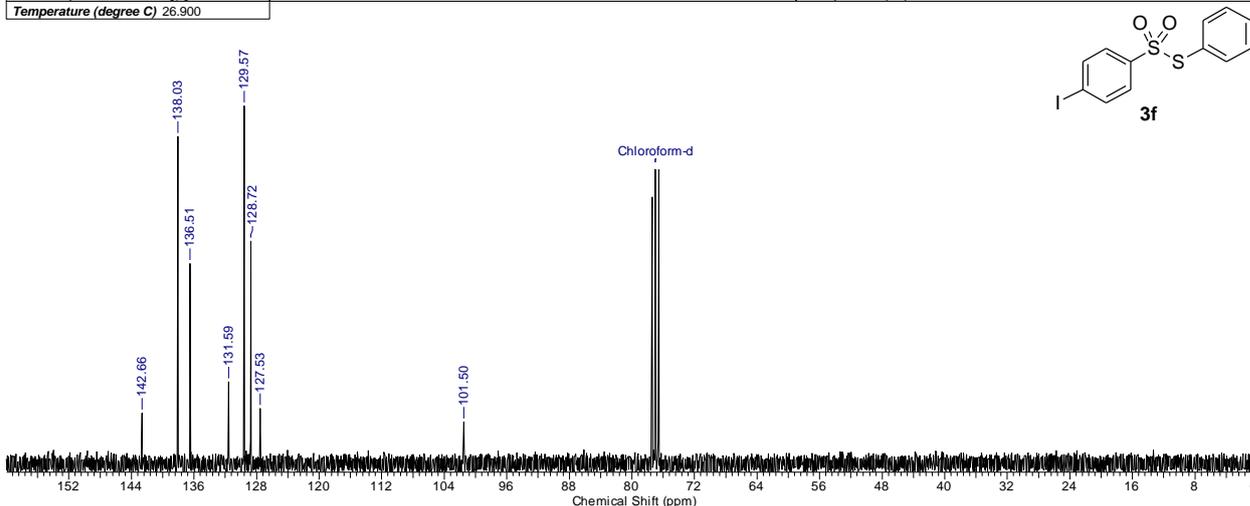
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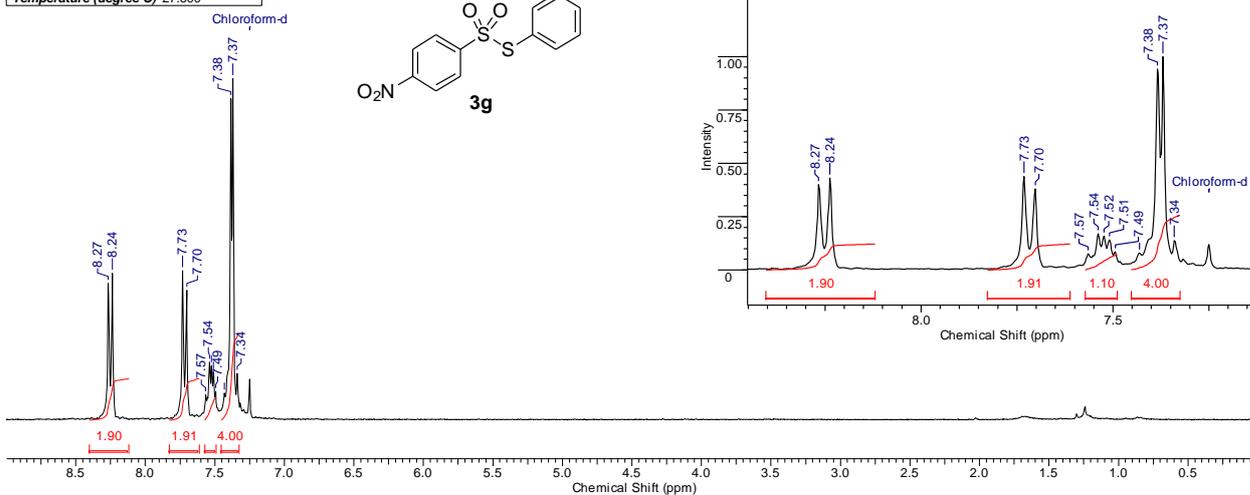
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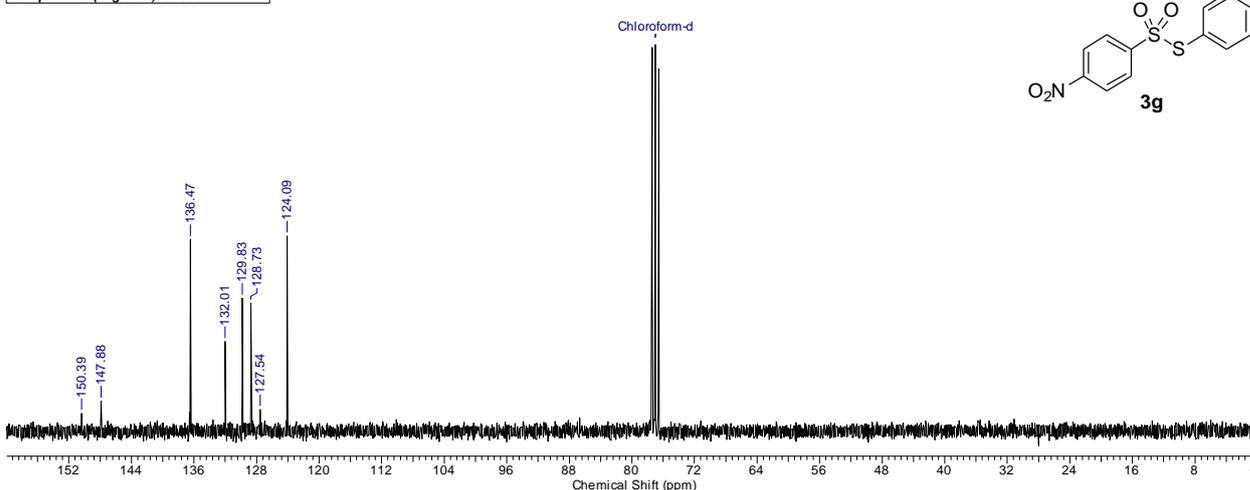
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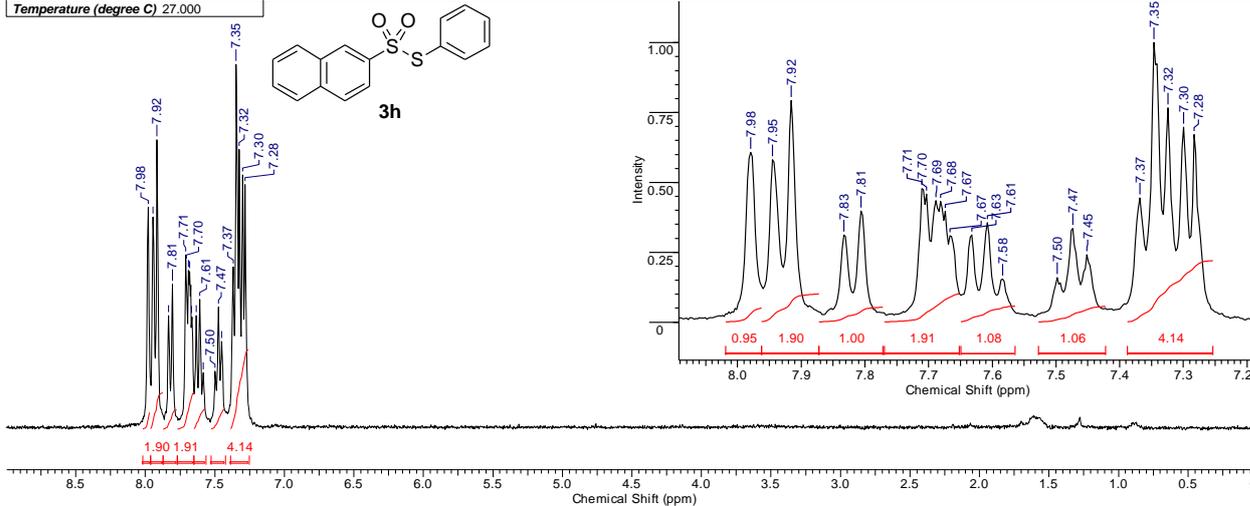
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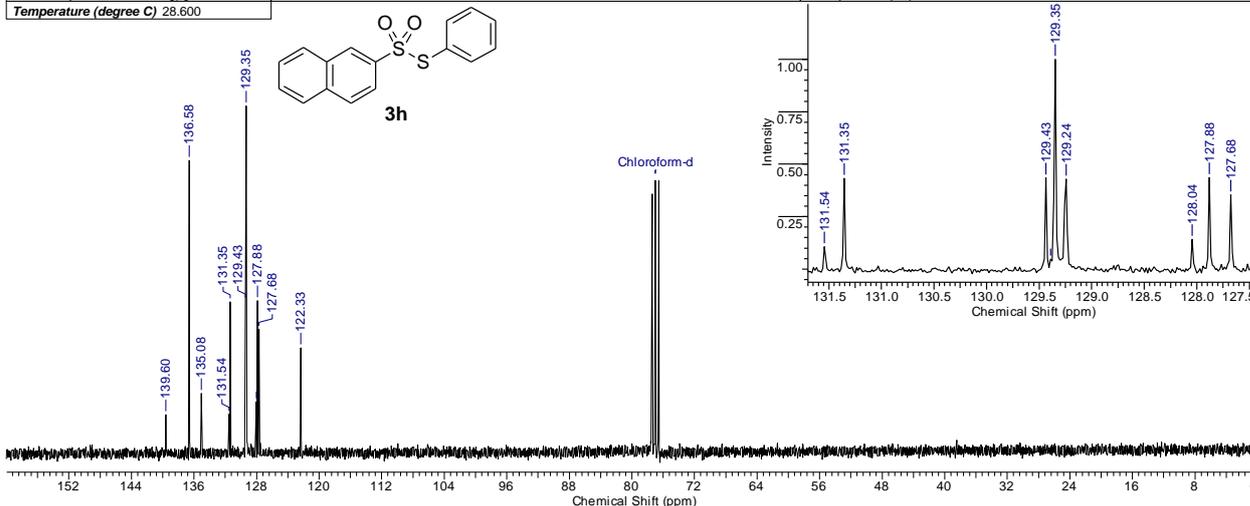
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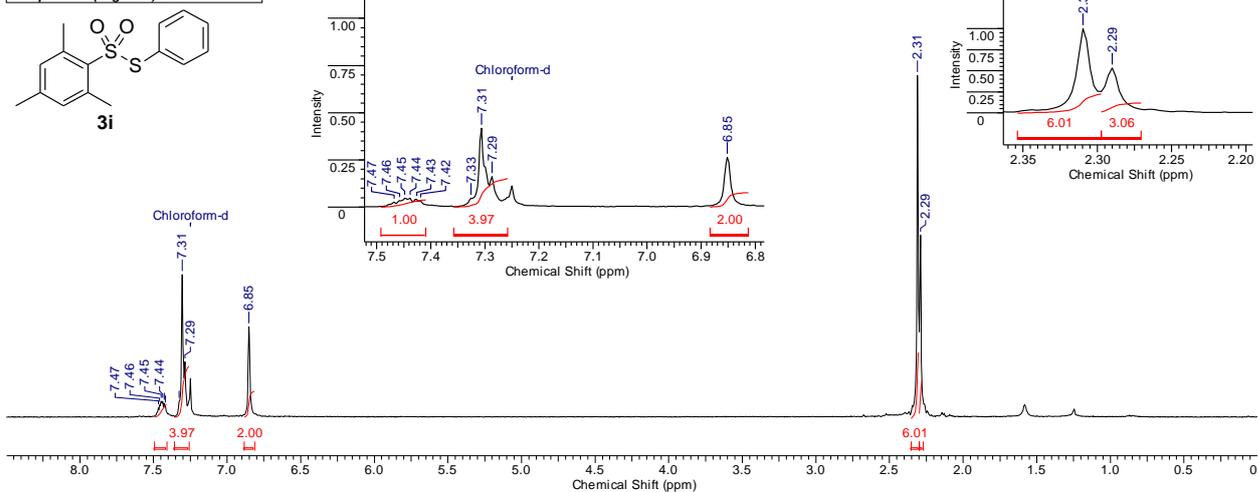
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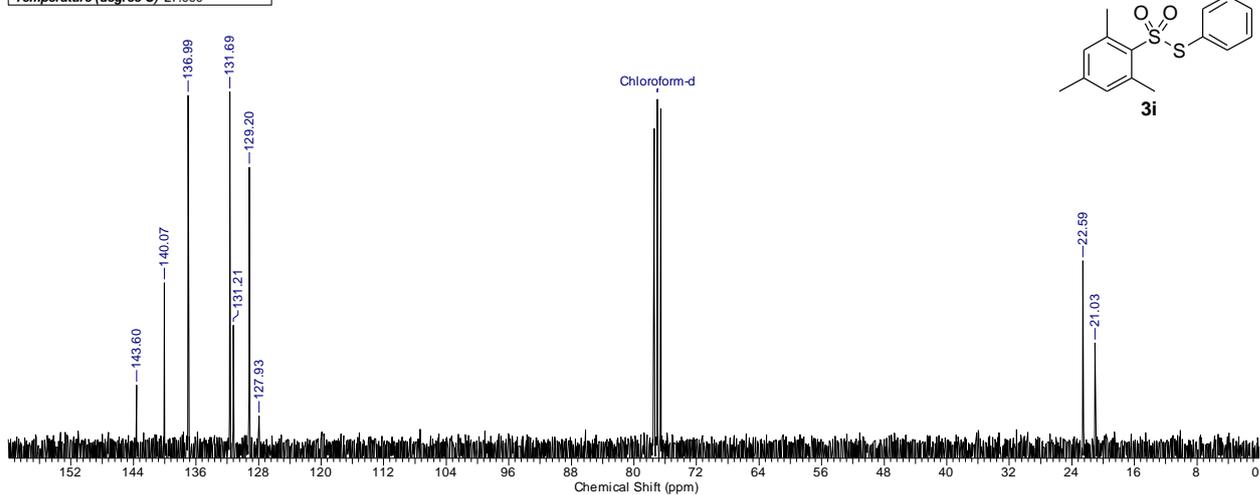
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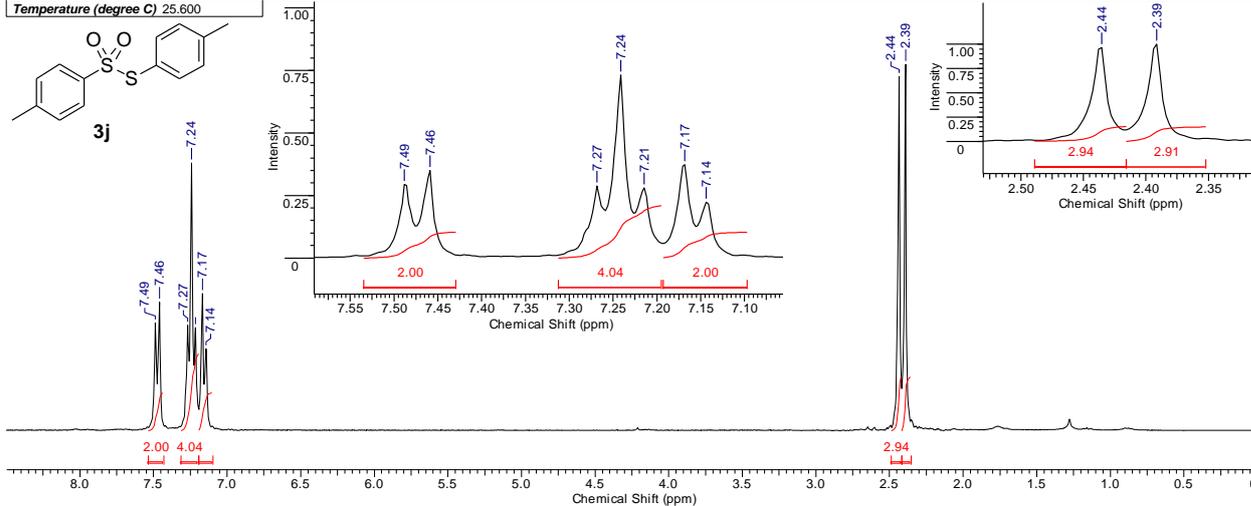
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Pulse Sequence	zg	Temperature (degree C)	28.300				



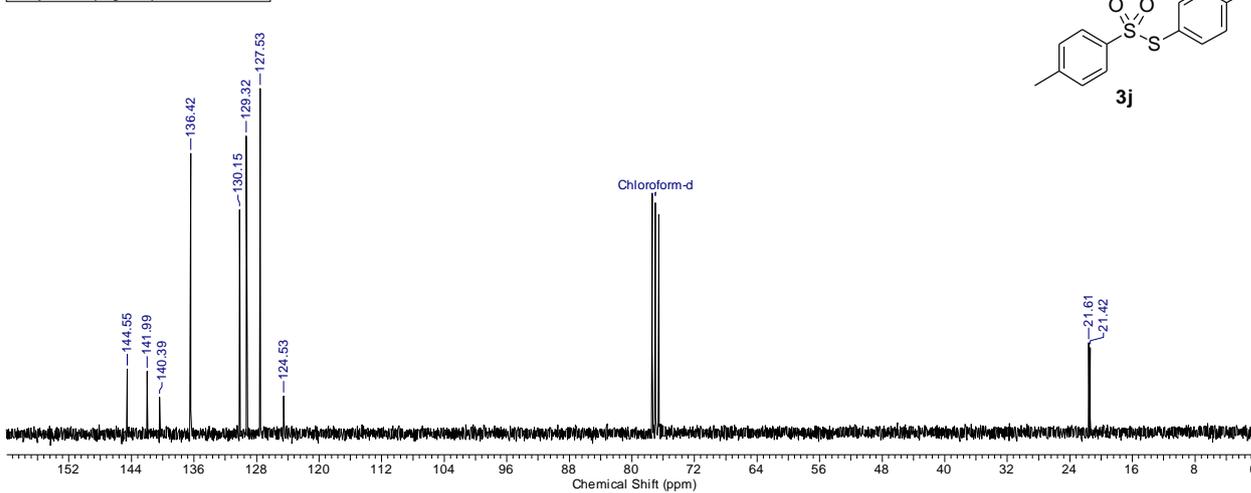
Acquisition Time (sec)	0.9006	Date	25 May 2018 09:14:40		Frequency (MHz)	75.48	
File Name		Number of Transients	122	Original Points Count	16316	Points Count	16384
Nucleus	¹³ C	Solvent	CHLOROFORM-D		Sweep Width (Hz)	18115.94	
Pulse Sequence	zgpg30	Temperature (degree C)	27.000				



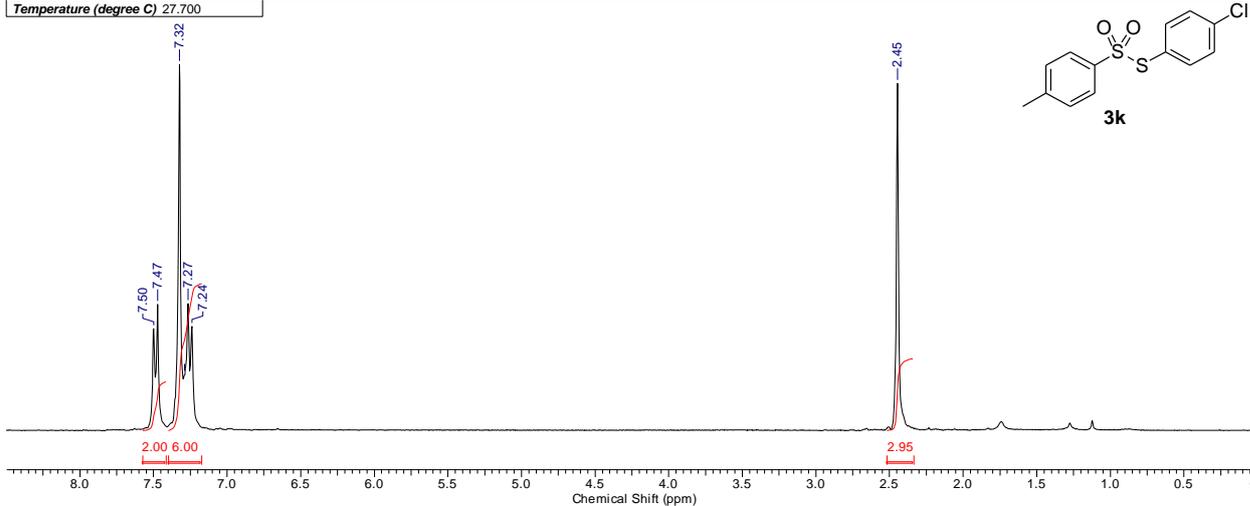
Acquisition Time (sec)	1.3518	Date	03 Jul 2018 12:54:24		Frequency (MHz)	300.13	
File Name		Number of Transients	1	Original Points Count	8124	Points Count	8192
Nucleus	¹ H	Solvent	CHLOROFORM-D		Sweep Width (Hz)	6009.62	
Pulse Sequence	zg	Temperature (degree C)	25.600				



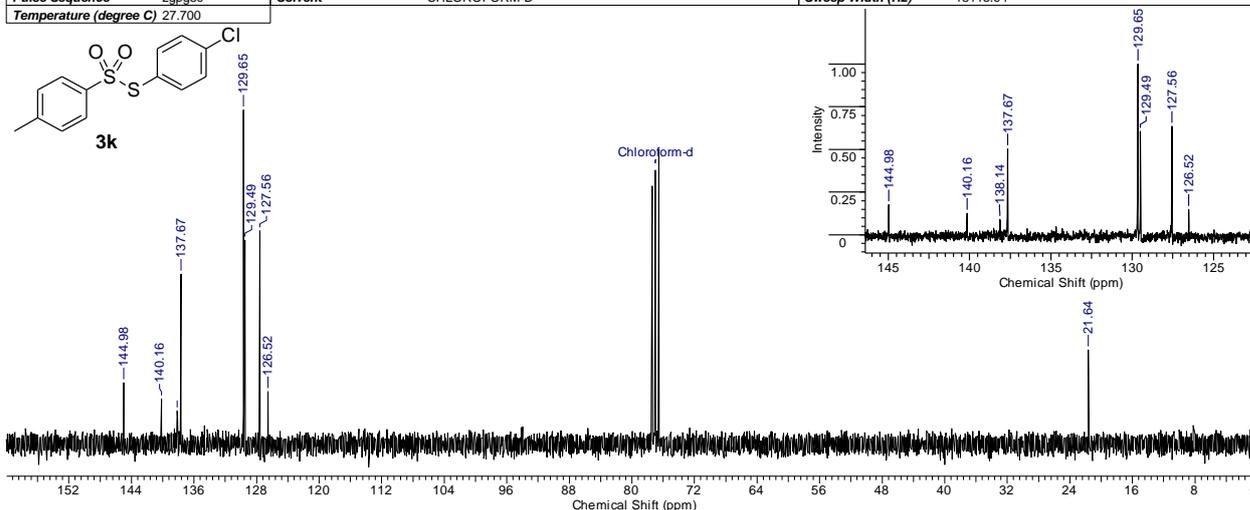
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File Name		Number of Transients	254	Original Points Count	16316	Points Count	16384
Nucleus	¹³ C	Solvent	CHLOROFORM-D		Sweep Width (Hz)	18115.94	
Pulse Sequence	zgpg30	Temperature (degree C)	25.700				



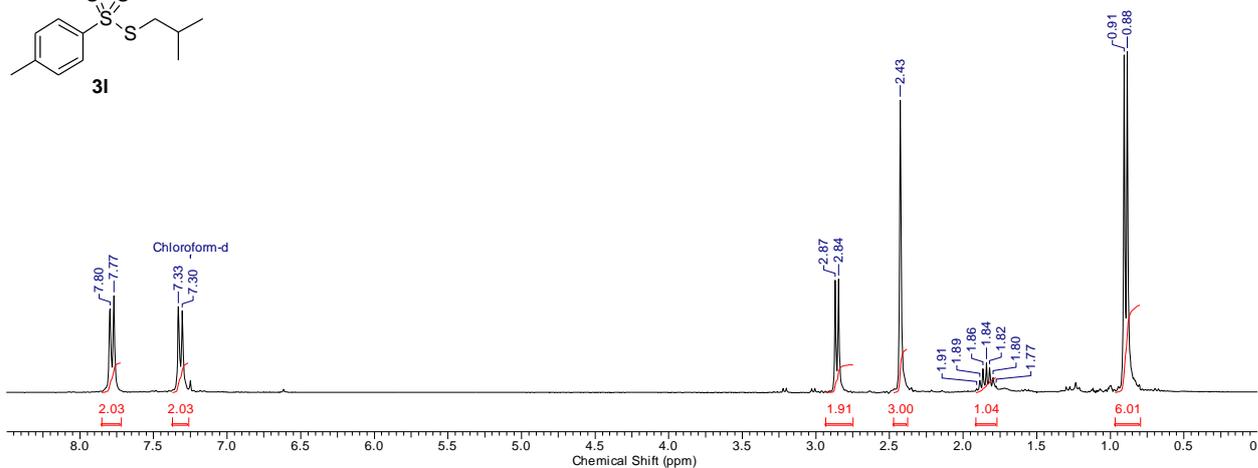
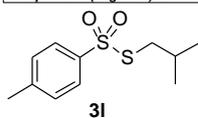
Acquisition Time (sec)	1.3518	Date	04 Jul 2018 09:14:40	Frequency (MHz)	300.13
File Name		Number of Transients	1	Original Points Count	8124
Nucleus	¹ H	Solvent	CHLOROFORM-D	Points Count	8192
Pulse Sequence	zg	Temperature (degree C)	27.700	Sweep Width (Hz)	6008.62



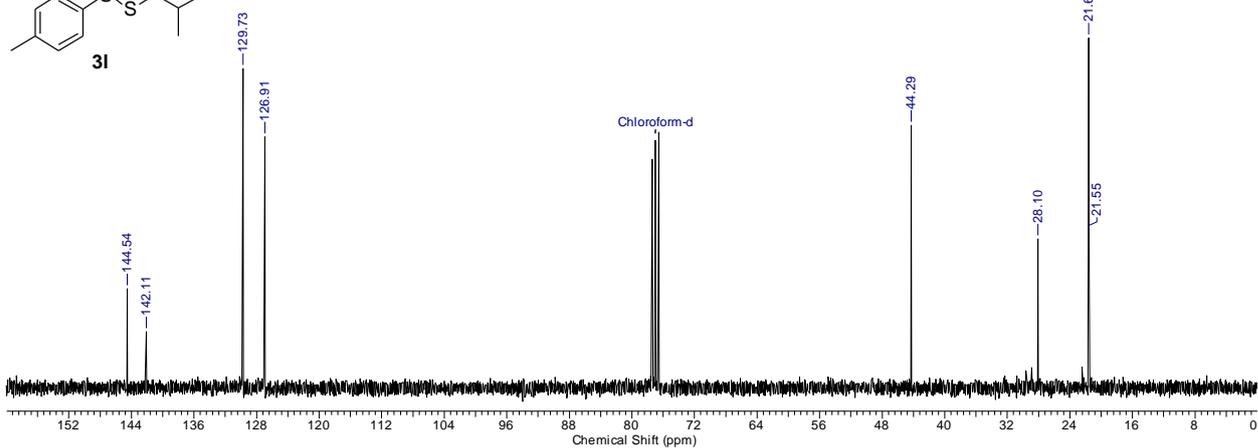
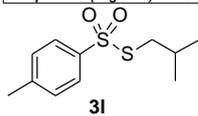
Acquisition Time (sec)	0.9006	Date	04 Jul 2018 09:16:48	Frequency (MHz)	75.48
File Name		Number of Transients	109	Original Points Count	16316
Nucleus	¹³ C	Solvent	CHLOROFORM-D	Points Count	16384
Pulse Sequence	zgpg30	Temperature (degree C)	27.700	Sweep Width (Hz)	18115.94



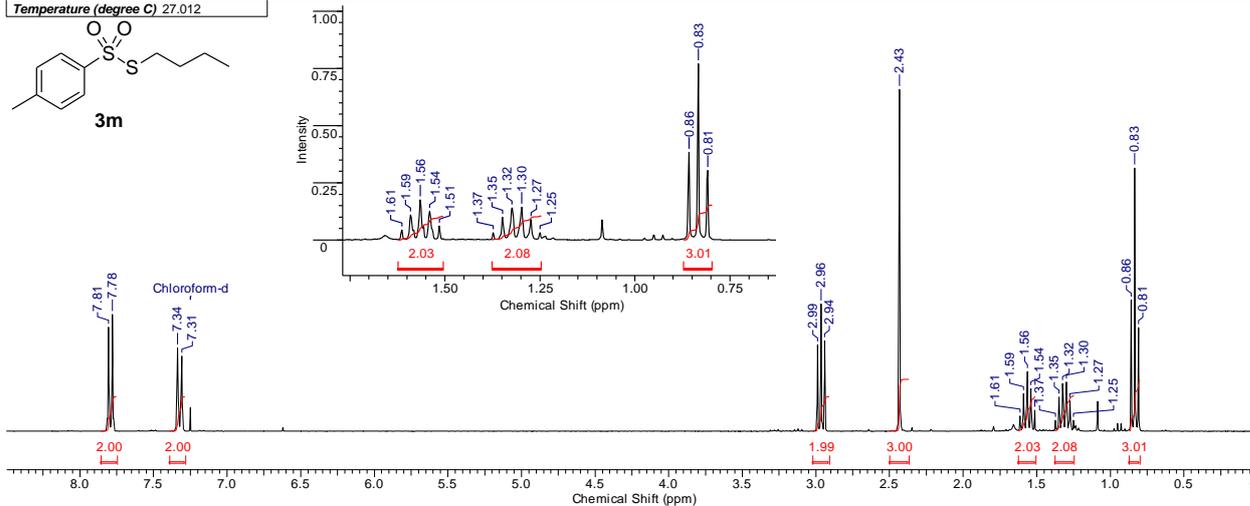
Acquisition Time (sec)	1.3518	Date	30 Jul 2018 09:04:00		Frequency (MHz)	300.13
File Name					Points Count	8192
Nucleus	1H	Number of Transients	1	Original Points Count	8124	Points Count
Pulse Sequence	zg	Solvent	CHLOROFORM-D		Sweep Width (Hz)	6009.62
Temperature (degree C)	28.400					



Acquisition Time (sec)	0.9006	Date	30 Jul 2018 09:06:08		Frequency (MHz)	75.48
File Name					Points Count	16384
Nucleus	13C	Number of Transients	294	Original Points Count	16316	Points Count
Pulse Sequence	zgpg30	Solvent	CHLOROFORM-D		Sweep Width (Hz)	18115.94
Temperature (degree C)	28.400					



Acquisition Time (sec)	2.7150	Comment		Date	04 Sep 2018 13:13:36
File Name				Frequency (MHz)	300.13
Nucleus	1H	Number of Transients	1	Original Points Count	16316
Pulse Sequence	zg	Solvent	CHLOROFORM-D	Points Count	16384
Temperature (degree C)	27.012			Sweep Width (Hz)	6009.62



Acquisition Time (sec)	0.9050	Comment		Date	04 Sep 2018 13:13:36
File Name				Frequency (MHz)	75.48
Nucleus	13C	Number of Transients	55	Original Points Count	16316
Pulse Sequence	zgpg30	Solvent	CHLOROFORM-D	Points Count	16384
Temperature (degree C)	27.055			Sweep Width (Hz)	18028.85

