

New bicyclic phosphonates of unsymmetrical structure

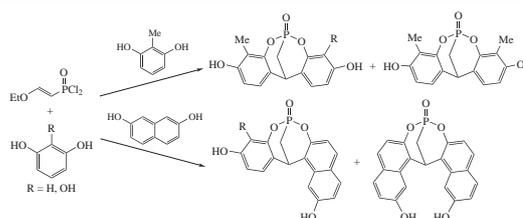
Yulia M. Sadykova,^{*a} Larisa M. Sadikova,^b Alena V. Zalaltdinova,^b
Anna G. Strelnik,^a Alexander R. Burilov^a and Mikhail A. Pudovik^a

^a A. E. Arbuзов Institute of Organic and Physical Chemistry, FRC Kazan Scientific Center of the Russian Academy of Sciences, 420088 Kazan, Russian Federation. E-mail: jsadykova@mail.ru

^b Kazan National Research Technological University, 420015 Kazan, Russian Federation

DOI: 10.1016/j.mencom.2018.11.032

New unsymmetrical bicyclic cage phosphonates have been obtained by refluxing two different phenols with (2-ethoxyvinyl)phosphonic dichloride in toluene in the presence of trifluoroacetic acid.



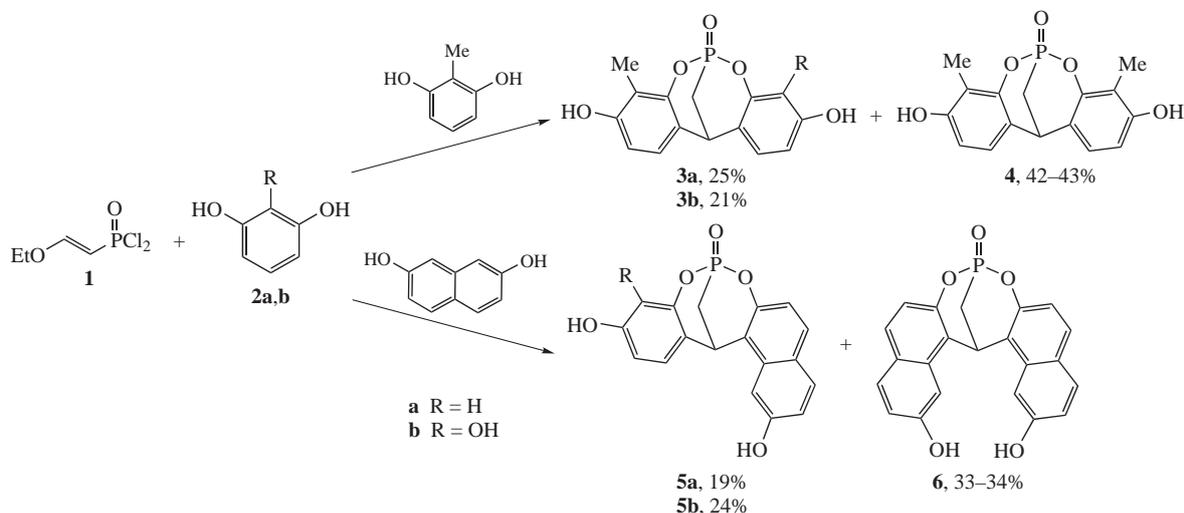
Cage compounds are of interest since the rigid framework provides strict localization of the substituents, which gives a clear knowledge of their spatial arrangement.^{1,2} Organophosphorus cage compounds with endocyclic phosphorus–carbon bonds possess unusual reactivity and physical properties and look promising from fundamental and practical viewpoints.^{3–7} Such compounds can be employed as complexing agents, ligands in metal-complex catalysis, organocatalysts,⁸ and multifunctional drugs possessing antiviral,⁹ anticancer,¹⁰ and other types of activities.^{11,12}

We have previously obtained symmetrical cage phosphonates with all endocyclic bonds by the reaction of (2-ethoxyvinyl)phosphonic dichloride **1** with resorcinol and some other active phenols.^{13–15} It seemed challenging to extend the scope of this reaction to access unsymmetrical structures. Herein, new unsymmetrical bicyclic phosphonates were prepared by multicomponent reaction of compound **1** with two different phenols. In fact, dissolution of equimolar amounts of resorcinol **2a** or pyrogallol **2b** with 2-methylresorcinol in boiled toluene in the presence of trifluoroacetic acid followed by the dropwise addition of phos-

phonic dichloride **1** led to novel unsymmetrical phosphonates **3a,b** (Scheme 1). Note that symmetrical bicyclic phosphonate **4** was also formed, while formation of symmetrical phosphonates based on resorcinol or pyrogallol was not observed. The reaction of resorcinol **2a** or pyrogallol **2b** with naphthalene-2,7-diol proceeded in the same way giving unsymmetrical phosphonates **5a,b**, along with symmetrical one **6**. All physicochemical data for compounds **4**¹³ and **6**¹⁶ coincide with our previous data.

Structures of phosphonates **3a,b** and **5a,b** were elucidated from NMR (¹H, ³¹P, and ¹³C) and IR spectroscopy, mass spectrometry (MALDI) and elemental analysis. Structures of compounds **5a** and **5b** were also confirmed using 1D/2D correlation NMR experiments (for details, see Online Supplementary Materials).

In the case of compound **5a**, individual spin systems were resolved in the course of ¹H–¹H COSY experiment, while carbon atoms corresponding to these protons were found by ¹H–¹³C HSQC experiment. The fragment P–CH₂–CH is clearly resolved in ¹H spectra due to the intrinsic range of proton resonance, as well as due to the presence of coupling constants *J*_{HH} and *J*_{PH} (³*J*_{HH} 3.9 Hz, ²*J*_{PH} 16.1 Hz, and ³*J*_{HH} 3.9 Hz, ²*J*_{PH} 34.6 Hz in the case of



Scheme 1 Reagents and conditions: CF₃COOH, toluene, reflux.

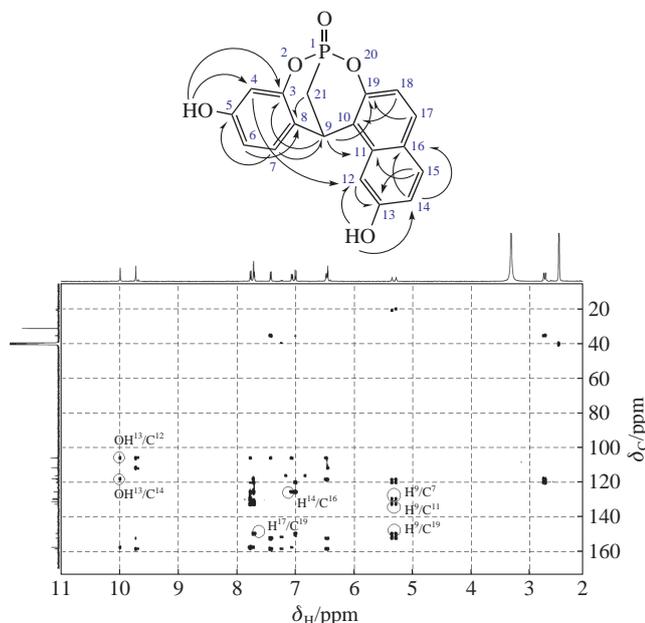
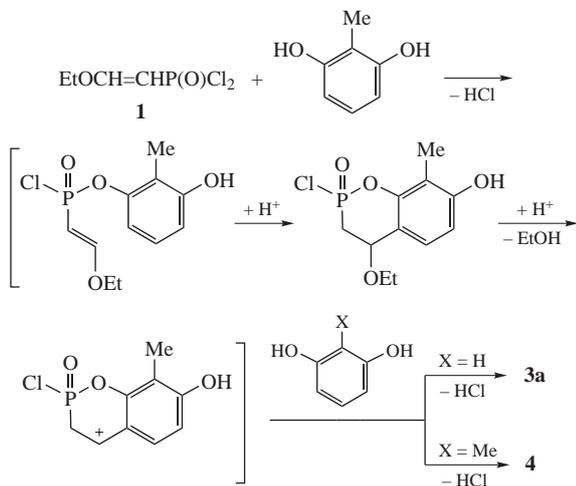


Figure 1 ^1H – ^{13}C HMBC spectrum of compound **5a** and its structure with key NMR correlations.

H^{21} and H^9 , respectively). Then, the ^1H – ^{13}C HMBC correlations along the chain from H^9 to H^3 , from H^6 to C^8 , and from OH^5 to C^4 and C^3 were established. The presence of the constant $^3J_{\text{CP}}$ 8.4 Hz allowed as to distinguish C^4H group from C^{12}H one. Analogously, the structure of naphthol fragment and its correlation with $\text{P}-\text{CH}_2-\text{CH}$ fragment were determined based on a series of ^1H – ^{13}C HMBC correlations between H^9/C^{19} , H^9/C^{11} , $\text{H}^{17}/\text{C}^{19}$, $\text{H}^{15}/\text{C}^{13}$, $\text{OH}^{13}/\text{C}^{12}$, and $\text{H}^{14}/\text{C}^{16}$ (Figure 1). The structure was additionally confirmed by the observed carbon–phosphorus coupling constants. The spatial structure was confirmed on the basis of NOE experiments; the effects H^9/H^7 and H^9/H^{12} gave an unambiguous proof of their spatial proximity.



Scheme 2

A plausible mechanism for formation of unsymmetrical bicyclic phosphonates is outlined on the example of phosphonic dichloride **1**, 2-methylresorcinol and resorcinol as the reactants (Scheme 2). Initially, phosphorylation of one hydroxyl group of 2-methylresorcinol occurs with the formation of phosphonochloridate. The further cascade reactions, intermolecular cyclization, elimination of EtOH and subsequent attack by the second molecule of 2-methylresorcinol or resorcinol at carbocation centre, lead to the final bicyclic phosphonates (see Scheme 2).

In summary, the first representatives of unsymmetrical bicyclic phosphonates were synthesized by the one-pot reaction of (2-ethoxyvinyl)phosphonic dichloride with two different phenols.

This work was supported by the Russian Foundation for Basic Research (grant no. 17-03-00254).

Online Supplementary Materials

Supplementary data associated with this article can be found in the online version at doi: 10.1016/j.mencom.2018.11.032.

References

- I. R. Knyazeva, A. R. Burilov, M. A. Pudovik and W. D. Habicher, *Russ. Chem. Rev.*, 2013, **82**, 150.
- P. Finocchiaro, S. Failla and G. Consiglio, *Russ. Chem. Bull., Int. Ed.*, 2005, **54**, 1355 (*Izv. Akad. Nauk, Ser. Khim.*, 2005, 1313).
- A. M. Caminade and J. P. Majoral, *Chem. Rev.*, 1994, **94**, 1183.
- G. A. Consiglio, S. Failla, P. Finocchiaro and V. Siracusa, *Phosphorus Sulfur Silicon Relat. Elem.*, 1998, **134**, 413.
- I. S. Antipin, E. Kh. Kazakova, W. D. Habicher and A. I. Kononov, *Russ. Chem. Rev.*, 1998, **67**, 995 (*Usp. Khim.*, 1998, **67**, 995).
- I. Bauer and W. D. Habicher, *Phosphorus Sulfur Silicon Relat. Elem.*, 1997, **130**, 89.
- V. F. Mironov, N. R. Khasiyatullina and D. B. Krivolapov, *Tetrahedron Lett.*, 2015, **56**, 7132.
- T. Hatakeyama, S. Hashimoto and M. Nakamura, *Org. Lett.*, 2011, **13**, 2130.
- Yu. N. Klimochkin, V. A. Shiryaev and M. V. Leonova, *Russ. Chem. Bull., Int. Ed.*, 2015, **64**, 1473 (*Izv. Akad. Nauk, Ser. Khim.*, 2015, 1473).
- G. Süß-Fink, *Dalton Trans.*, 2010, **39**, 1673.
- D. A. Tatarinov, D. M. Kuznetsov, A. D. Voloshina, A. P. Lyubina, A. S. Strobyskina, F. K. Mukhitova, F. M. Polyancev and V. F. Mironov, *Tetrahedron*, 2016, **72**, 8493.
- N. R. Khasiyatullina, A. M. Vazykhova, V. F. Mironov, D. B. Krivolapov, Yu. K. Voronina, A. D. Voloshina, N. V. Kulik and A. S. Strobyskina, *Mendeleev Commun.*, 2017, **27**, 134.
- Yu. M. Sadykova, N. V. Dalmatova, Yu. K. Voronina, A. R. Burilov, M. A. Pudovik and O. G. Sinyashin, *Heteroat. Chem.*, 2012, **23**, 340.
- Yu. M. Sadykova, N. V. Dalmatova, A. R. Burilov and M. A. Pudovik, *Russ. Chem. Bull., Int. Ed.*, 2012, **61**, 2009 (*Izv. Akad. Nauk, Ser. Khim.*, 2012, 1991).
- Yu. M. Sadykova, N. V. Dalmatova, M. F. Nagimova, Yu. K. Voronina, A. R. Burilov, M. A. Pudovik and O. G. Sinyashin, *Heteroat. Chem.*, 2014, **25**, 55.
- Yu. M. Sadykova, L. M. Sadikova, A. R. Burilov and M. A. Pudovik, *Russ. J. Gen. Chem.*, 2017, **87**, 1913 (*Zh. Obshch. Khim.*, 2017, **87**, 1429).

Received: 14th May 2018; Com. 18/5576