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## Synthesis and structure of new 2-aryl-substituted pyrrolidines containing phosphine oxide group

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### Experimental Section

#### General

NMR experiments were recorded on a Bruker AVANCE-600 spectrometer at 303 K equipped by 5 mm broadband probe head working at 600.1 MHz in  $^1\text{H}$  and 150.9 MHz in  $^{13}\text{C}$  experiments. The  $^{31}\text{P}$  NMR spectra were obtained on a Bruker MSL-400 NMR-Fourier spectrometer. Chemical shifts were measured relative to residual protons ( $^1\text{H}$ ) or solvent signal ( $^{13}\text{C}$ ). The pulse programs of the COSY, NOESY, HSQC and HMBC experiments were taken from Bruker software library. NMR-experiments were carried out in solutions ( $30\text{ mmol}\cdot\text{l}^{-1}$ ) at 303 K. The IR spectra of the compounds as emulsions in Nujol were recorded on a Vector 22 FTIR Spectrometer (Bruker) in the  $4000\text{--}400\text{ cm}^{-1}$  range at a resolution of  $1\text{ cm}^{-1}$ . MALDI mass spectra were run on an Ultra Flex III TOF/TOF mass spectrometer (Bruker Daltonic, Bremen, Germany) operated in the linear mode. Elemental microanalyses data were obtained on a CHN-3 analyzer and they were within  $\pm 0.4\%$  of the theoretical values for C and H. The uncorrected melting points were measured on a Boetius hot-stage apparatus. All solvents were dried according to standard protocols.

#### Synthesis

*General procedure for the synthesis of [(pyrrolidin-1-yl)methyl]phosphine oxides 3a-c.*

To the appropriate resorcinol **2a-c** (1.27 mmol) dissolved in dry benzene (7 ml), *P*-(4,4-diethoxybutylaminomethyl)-*P,P*-di-*p*-tolylphosphine oxide **1** (0.4 g, 1.28 mmol) and trifluoroacetic acid (0.2 ml, 2.5 mmol) were added. The mixture was stirred for 24 h at room temperature. The benzene layer was decanted, the oily residue was dried in vacuum, washed with diethyl ether, filtered off, and dried in vacuum to give the desired product.

**2-(2,4-Dihydroxyphenyl)-1-(di-*p*-tolylphosphorylmethyl)pyrrolidin-1-ium trifluoroacetate 3b.** Yield: 0.36 g (67%), m.p.  $130\text{--}132^\circ\text{C}$ . IR (KBr,  $\text{v}/\text{cm}^{-1}$ ): 1177, 1220 (P=O), 1602 (C=C<sub>arom</sub>), 1677 (C=O), 2588 – 2710 (NH<sup>+</sup>), 3057, 3395 (OH).  $^{31}\text{P}$  NMR (242.9 MHz),  $\delta$ : 24.9.  $^1\text{H}$  NMR (600.1 MHz, acetone-*d*<sub>6</sub>)  $\delta$ : 2.14 – 2.31 (m, 4H, CH<sub>2</sub>), 2.40 (s, 3H, CH<sub>3</sub>), 2.41 (s, 3H, CH<sub>3</sub>), 3.29 – 3.32 (m, 1H, NCH<sub>2</sub>), 3.85 – 3.88 (m, 1H, NCH<sub>2</sub>), 4.00 (dd, 1H, CH<sub>2</sub>P,  $^2J_{\text{PH}} 5.6\text{ Hz}$ ,  $^2J_{\text{HH}} 15.2\text{ Hz}$ ), 4.09 – 4.13 (m, 1H, CH<sub>2</sub>P,  $^2J_{\text{HH}} 15.1\text{ Hz}$ ,  $^2J_{\text{PH}} 7.13\text{ Hz}$ ), 4.68 (t, 1H, CH,  $^2J_{\text{HH}} 8.5\text{ Hz}$ ), 6.36 (d, 1H, ArH,  $^3J_{\text{HH}} 8.3\text{ Hz}$ ), 6.69 (s, 1H, ArH), 7.10 (d, 1H, ArH,  $^3J_{\text{HH}} 8.3\text{ Hz}$ ), 7.34 – 7.35 (m, 2H, ArH), 7.37 – 7.38 (m, 2H, ArH), 7.62 (dd, 2H, ArH,  $^3J_{\text{HH}} 11.9\text{ Hz}$ ,  $^3J_{\text{HP}} 10.9\text{ Hz}$ ), 7.72 (dd, 2H, ArH,  $^3J_{\text{HH}} 11.9\text{ Hz}$ ,  $^3J_{\text{HP}} 7.2\text{ Hz}$ ).  $^{13}\text{C}$  NMR (150.9 MHz, acetone-*d*<sub>6</sub>)  $\delta$ : 21.12 (s, C-16), 22.79 (s, C-8), 50.91 (d, C-11,  $^1J_{\text{PC}} 72.2\text{ Hz}$ ), 55.61 (s, C-10), 71.59 (d, C-7,  $^3J_{\text{PC}} 7.5\text{ Hz}$ ), 104.03 (C-5), 107.77 (C-6), 110.92 (C-1), 114.35 (q, CF<sub>3</sub>,  $^1J_{\text{CF}} 292.7\text{ Hz}$ ), 127.29 (d, C-12,  $^1J_{\text{PC}} 121.3\text{ Hz}$ ), 128.09 (d, C-12',  $^1J_{\text{PC}} 121.1\text{ Hz}$ ), 130.11 (d, C-14,  $^1J_{\text{PC}} 12.4\text{ Hz}$ ), 130.19 (d, C-14',  $^1J_{\text{PC}} 12.2\text{ Hz}$ ), 131.00 (s, C-3), 131.30, 131.37 (s, C-13), 143.85 (s, C-15), 158.02 (s, C-2),

160.02 (q, C=O,  $^2J_{CF}$  34.8 Hz), 160.47 (s, C-4).  $^{31}P$  NMR (242.9 MHz),  $\delta$ : 24.9. MS (MALDI),  $m/z$ : 422  $[M+H]^+$ . Found (%): C, 60.60; H, 5.60; F, 10.54; N, 2.58; P, 5.76. Calc. for  $C_{25}H_{28}NO_3P^*CF_3COOH$  (%): C, 60.56; H, 5.46; F, 10.64; N, 2.62; P, 5.78.

**1-(Di-*p*-tolylphosphorylmethyl)-2-(2,3,4-trihydroxyphenyl)pyrrolidin-1-ium trifluoroacetate 3c.** Yield 0.4 g (73%), m.p. 126 – 127°C. IR (KBr,  $v/cm^{-1}$ ): 1198 (P=O), 1602 (C=C), 1676 (C=O), 2754 (NH<sup>+</sup>), 3150, 3391 (OH).  $^1H$  NMR (400.1 MHz, acetone- $d_6$ )  $\delta$ : 2.1 – 2.23 (m, 4H, CH<sub>2</sub>), 2.38 (s, 3H, CH<sub>3</sub>), 2.39 (s, 3H, CH<sub>3</sub>), 3.28 – 3.35 (m, 1H, NCH<sub>2</sub>), 3.86 – 3.92 (m, 1H, NCH<sub>2</sub>), 3.96 (dd, 1H, CH<sub>2</sub>P,  $^2J_{HH}$  15.3 Hz,  $^2J_{PH}$  5.8 Hz), 4.08 (dd, 1H, CH<sub>2</sub>P,  $^2J_{HH}$  15.3 Hz,  $^2J_{PH}$  7.1 Hz), 4.67 – 4.71 (m, 1H, CHCH<sub>2</sub>), 6.36 (d, 1H, ArH,  $^3J_{HH}$  8.4 Hz), 6.96 (d, 1H, ArH,  $^3J_{HH}$  8.4 Hz), 7.30 – 7.33 (m, 2H, ArH), 7.35 – 7.38 (m, 2H, ArH), 7.53 – 7.58 (m, 2H, ArH), 7.66 – 7.71 (m, 2H, ArH).  $^{13}C$  NMR (150.9 MHz, acetone- $d_6$ )  $\delta$ : 20.68 (s, C-16), 22.33 (s, C-8), 29.26 (s, C-9), 50.15 (s, C-11,  $^1J_{PC}$  72.2 Hz), 55.54 (s, C-10), 71.11 (d, C-1,  $^3J_{PC}$  6.9 Hz), 107.31 (s, C-5), 111.07 (s, C-1), 115.42 (q, CF<sub>3</sub>,  $^1J_{CF}$  293.1 Hz), 119.17 (s, C-6), 126.12 (d, C-12,  $^1J_{PC}$  105.6 Hz), 126.82 d (12'-C,  $^1J_{PC}$  105.6 Hz), 129.71, 129.82 (d, C-14,  $^3J_{PC}$  12.5 Hz), 130.68, 130.78 (d, C-13,  $^2J_{PC}$  18.3 Hz), 133.33 (s, C-3), 132.77 (s, C-15), 143.65 (s, C-2), 146.20 (s, C-4), 160.01 (q, C=O,  $^2J_{CF}$  = 34.5 Hz).  $^{31}P$  NMR (242.9 MHz)  $\delta$ : 25.2. MS (MALDI),  $m/z$ : 438  $[M+H]^+$ . Found (%): C, 58.48; H, 5.34; F, 10.25; N, 2.51; P, 5.64. Calc. for  $C_{25}H_{28}NO_4P^*CF_3COOH$  (%): C, 58.80; H, 5.30; F, 10.33; N, 2.54; P 5.62.

### X-ray diffraction data

The X-ray diffraction data for the crystal of **4** were collected on Bruker Apex Smart CCD diffractometer with graphite-monochromated Mo-K $\alpha$  radiation ( $\lambda = 0.71073$  Å). The structure was solved by direct methods using the SHELXS program [S1]. Non-hydrogen atoms were first refined in the isotropic approximation and then anisotropic using the SHELXL-14 program [S2]. Hydrogen atoms were placed in the calculated positions and refined in the riding model. All calculations were performed using the WinGX [S3] and APEX2 [S4] programs. The figures were made using the OLEX2 [S5] program. CCDC 1585981 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif).

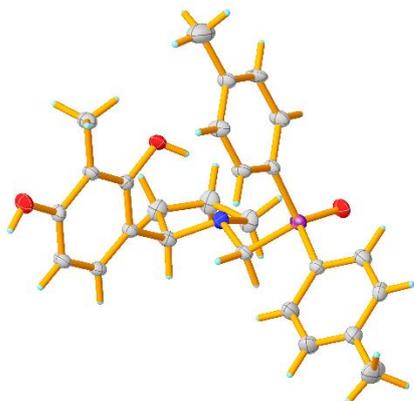
Crystallographic data for **4**: crystals of  $C_{55}H_{66}N_2O_7P_2$  ( $M = 929.03$ ) are triclinic, space group P-1, at 296 K:  $a = 10.338(4)$ ,  $b = 15.622(6)$  and  $c = 16.489(6)$  Å,  $\alpha = 80.468(5)$ ,  $\beta = 80.247(5)$ ,  $\gamma = 79.930(5)^\circ$ ,  $V = 2558.6(16)$  Å<sup>3</sup>,  $Z = 2$ ,  $d_{calc} = 1.206$  g cm<sup>-3</sup>,  $\mu(Mo) = 0.138$  mm<sup>-1</sup>,  $F(000) = 992$ . 18889 reflections were measured and 8963 independent reflections ( $R_{int} = 0.1477$ ) were used in a further refinement. The refinement converged to  $wR_2 = 0.1960$  and  $GOF = 0.803$  for all independent reflections [ $R_1 = 0.0660$  was calculated against  $F$  for 2698 observed reflections with  $I > 2\sigma(I)$ ].

**Table S1.** H-bond in crystal of compound **4**.

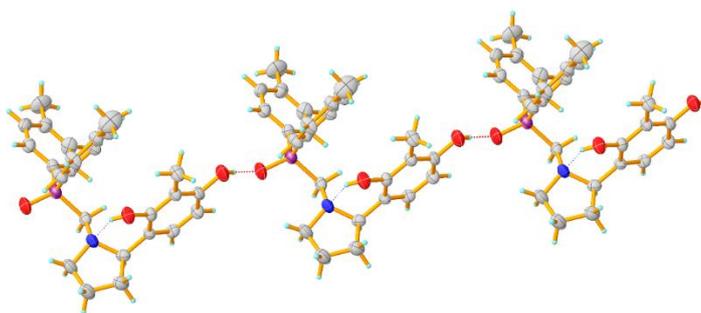
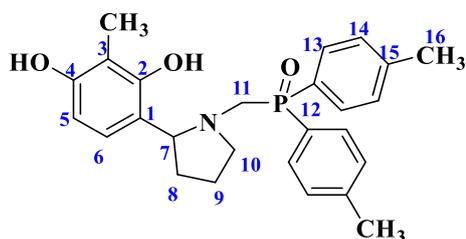
H-bond	D - H	H...A	D...A	D - H...A
O2B-H2B...N1B	0.82	2.10	2.795(6)	142
O4A-H4A...O1A	0.82	1.83	2.644(6)	174
O4B-H4B...O1B	0.82	1.83	2.644(6)	175
O6A-H6A...N1A	0.82	1.97	2.706(6)	148
C6B-H6B...O6A	0.93	2.51	3.431(7)	171
C24A-H24A...O6A	0.93	2.60	3.483(7)	159
C26A-H26B...O4A	0.96	2.30	2.761(8)	109
C26B-H26D...O2B	0.96	2.40	2.832(7)	107

**Table S2.** C-H... $\pi$  interactions in crystal of compound **4**.

Interaction	H..Cg	H-Perp	Gamma	X-H..Cg	X..Cg	X-H,Pi
C16A-H16A...Cg6	2.99	2.89	15.32	151	3.864(7)	46
C21A-H21A...Cg7	2.99	-2.89	14.96	129	3.654(7)	47
C21B-H21B...Cg3	2.93	-2.92	4.22	141	3.702(7)	55

**Figure S1.** Molecular structure of compound **4** in crystal.

Investigated crystal is the solvate with the ratio of compound **4** and acetone 2:1. There are two molecules of pyrrolidine and one molecule of acetone in independent part of a unit cell. The bond lengths, valence and torsion angles in molecules of compound **1** in a crystal are within the standard values for each type of chemical bond [S6] (Figure S1). The crystal structure of compound **4** consists of chains formed by classical O-H...O=P hydrogen bonds. Chains are connected with each other by C-H...O and CH... $\pi$  interactions (Figure S2, Tables S1 and S2).

**Figure S2.** H-bonds in crystal of compound **4**.

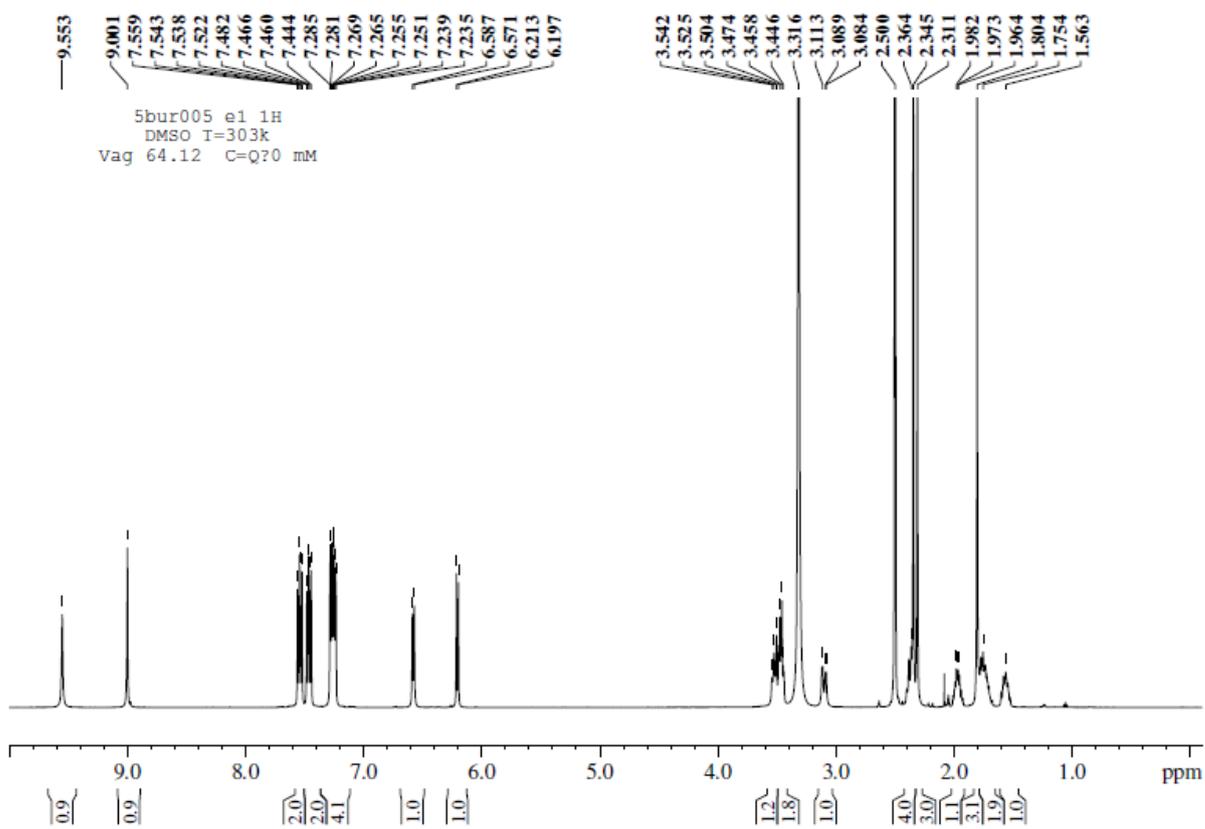


Figure S3.  $^1\text{H}$  NMR Spectrum of compound 4 (DMSO- $d_6$ ).

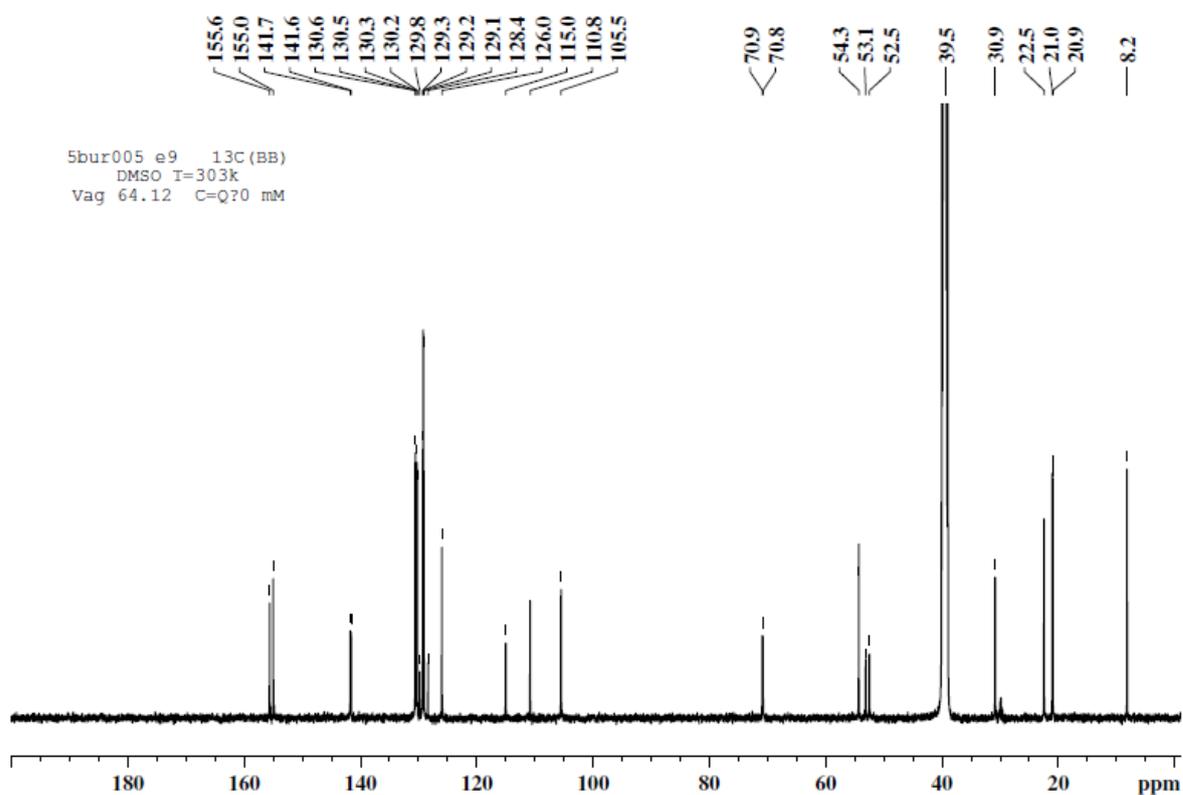
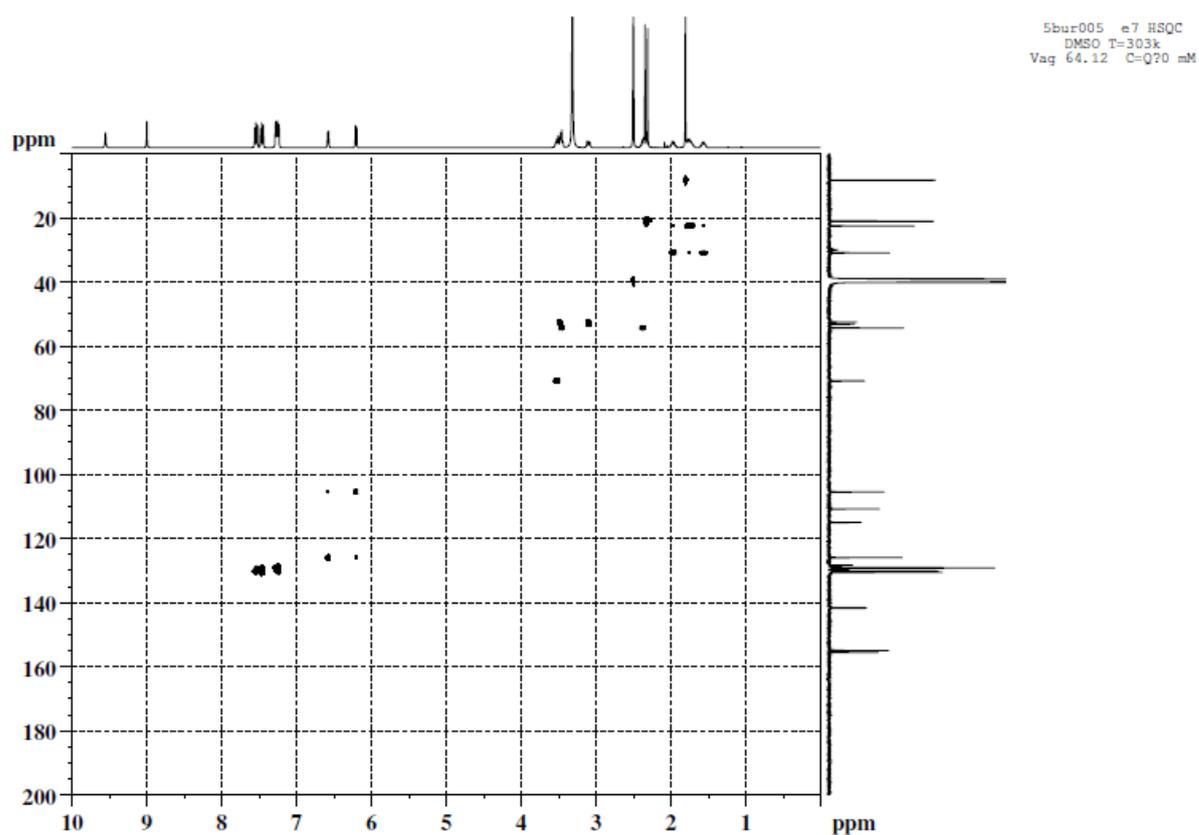
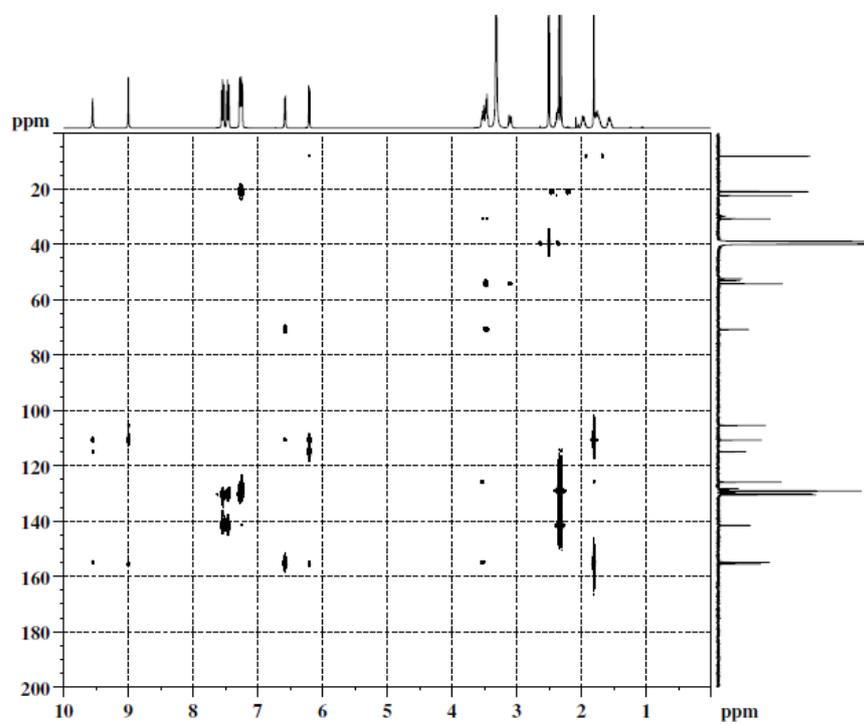


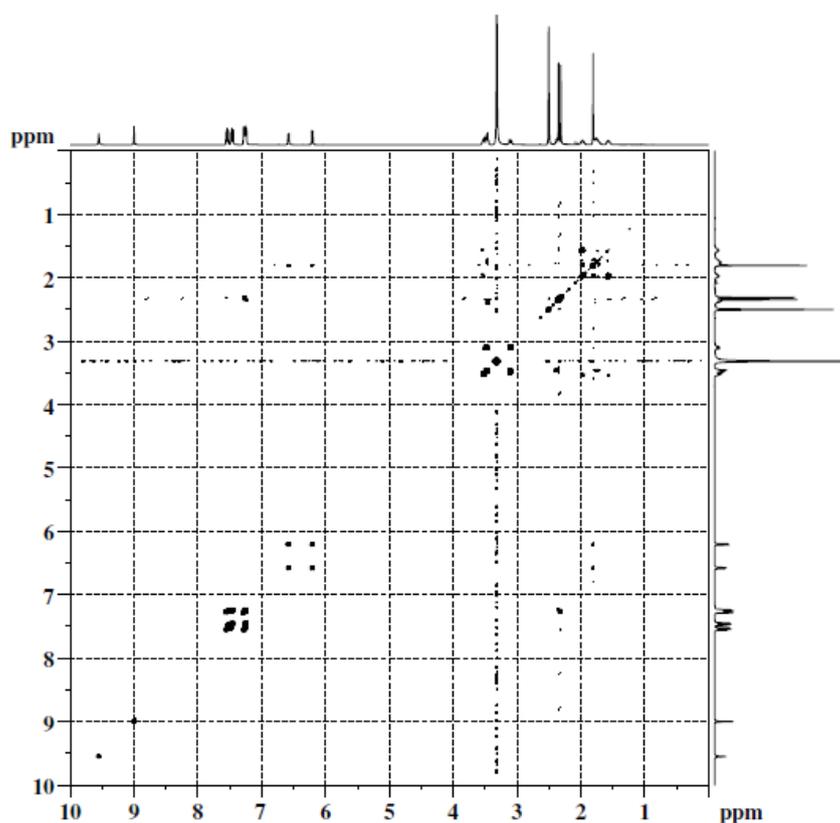
Figure S4.  $^{13}\text{C}$  spectrum of compound 4 (DMSO- $d_6$ )



**Figure S5.** HSQC-spectrum of compound 4.



**Figure S6.** HMBS spectrum of compound 4



**Figure S7.** COSY spectrum of compound **4**.

## References

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