

## Self-condensation of N-substituted (4*H*-thieno[3,2-*b*]pyrrol-5-yl)methanols into bis(thienopyrrolyl)methanes

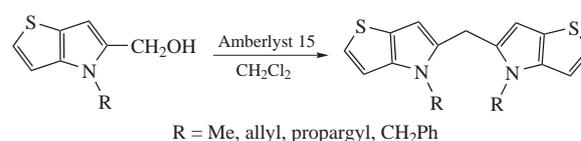
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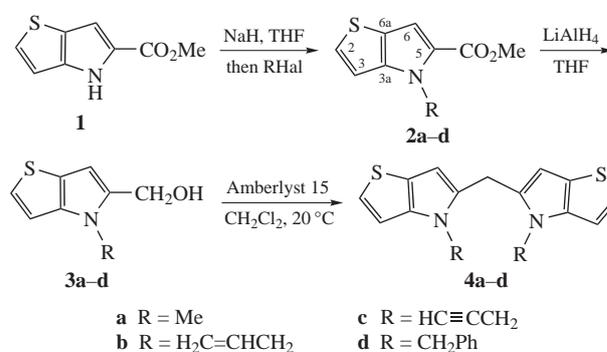
N-Substituted (4*H*-thieno[3,2-*b*]pyrrol-5-yl)methanols were obtained by alkylation of methyl 4*H*-thieno[3,2-*b*]pyrrole-5-carboxylate followed by reduction with LiAlH<sub>4</sub>. These compounds on contact with Amberlyst 15 (H-form) in CH<sub>2</sub>Cl<sub>2</sub> undergo self-condensation to produce bis(4*H*-thieno[3,2-*b*]pyrrol-5-yl)methanes.



4*H*-Thieno[3,2-*b*]pyrrole-5-carboxylic derivatives<sup>1,2</sup> are of interest in search for bioactive thienopyrrole compounds,<sup>3–8</sup> as well as base matrices in preparation of  $\pi$ -condensed systems for optoelectronics.<sup>9–12</sup> In this study, we performed N-alkylation of methyl 4*H*-thieno[3,2-*b*]pyrrole-5-carboxylate **1** with iodomethane, allyl, propargyl or benzyl bromides and obtained the corresponding N-derivatives **2a–d**, which were further reduced (LiAlH<sub>4</sub>) into the corresponding alcohols **3a–d** (Scheme 1).<sup>†</sup> We anticipated of using alcohols **3** to access more complex structures of bis(thienopyrrolyl)methane type **4** by means of self-condensation reaction. Choice of alcohols **3** for this purpose was based on known precedents of acid<sup>13(a),(b)</sup> or ion-exchange Montmorillonite resin<sup>13(c)</sup> catalyzed syntheses of pyrromethanes (dipyrrolyl-methanes) from monomeric acetoxyethylpyrroles and 5-unsubstituted pyrroles. Bis(thienopyrrolyl)methanes<sup>14</sup> are the valuable intermediates in the synthesis of porphyrins *via* pyrromethenes.<sup>15</sup> The data on self-condensation of such compounds are scarce<sup>13(a),(c)</sup> and relate to pyrroles with NH group.<sup>12,13(c),14(e),16</sup> In the case of N-substituted pyrroles, the possibilities of acid-catalyzed **3**  $\rightarrow$  **4** route are not obvious and hence they require experimental confirma-

tion. In addition, bis(thienopyrrolyl)methanes **4** seem promising in creation of cross-conjugation involving their methylene fragment.

On the way to methanes **4**, reactions of alcohol **3a** in SiO<sub>2</sub>–CH<sub>2</sub>Cl<sub>2</sub> and TsOH–benzene systems were studied at room temperature and on heating. In all cases, the reactions were sluggish and conversion of alcohol **3a** was never complete even after several hours. Meanwhile, in the Amberlyst 15 (H-form)–CH<sub>2</sub>Cl<sub>2</sub> system, alcohol **3a** was fully converted into a new product whose



Scheme 1

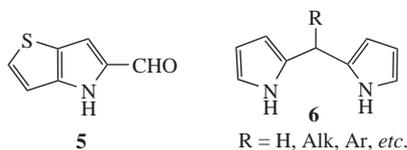
Methyl 4-methyl-4*H*-thieno[3,2-*b*]pyrrole-5-carboxylate **2a**.<sup>18</sup> A solution of pyrrole **1** (0.20 g, 1.10 mmol) in anhydrous THF (2 ml) was added to a stirred suspension of NaH (55.0 mg, 2.29 mmol) (pre-rinsed with anhydrous hexane from a 55% dispersion in mineral oil) in anhydrous THF (10 ml) under an argon atmosphere, and the mixture was stirred for 30 min. Then a solution of iodomethane (0.14 ml, 2.19 mmol) in THF (5 ml) was added dropwise, and the mixture was stirred at 40 °C until the starting pyrrole was consumed (TLC control). The mixture was cooled and quenched with saturated NH<sub>4</sub>Cl solution, the product was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3  $\times$  20 ml). The solvent was evaporated, the residue was purified by column chromatography on SiO<sub>2</sub> (light petroleum–ethyl acetate, 5:1) to give 0.17 g (79%) of compound **2a** as slightly yellow crystals, mp 62–63 °C. IR ( $\nu$ /cm<sup>-1</sup>): 1710, 1695, 1532, 1464, 1445, 1378, 1369, 1235, 1209, 1179, 1095, 1075, 962, 728. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$ : 3.86 (s, 3H, OMe), 4.06 (s, 3H, NMe), 6.95 (d, 1H, H-3, *J* 5.4 Hz), 7.18 (s, 1H, H-6), 7.34 (d, 1H, H-2, *J* 5.4 Hz). <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$ : 51.3 (NMe, OMe), 109.0 (C<sup>6</sup>), 110.0 (C<sup>3</sup>), 114.8 (C<sup>6a</sup>), 121.9 (C<sup>5</sup>), 129.13 (C<sup>2</sup>), 138.0 (C<sup>3a</sup>), 164.6 (CO<sub>2</sub>Me).

<sup>†</sup> The IR spectra were recorded from thin films on a Shimadzu IR Prestige-21 spectrometer. The <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on Bruker AM-300 (300.13 and 75.47 MHz, respectively) and Bruker Avance-500 instruments (500.13 and 125.77 MHz, respectively) using tetramethylsilane as internal standard. The mass spectra (EI, 70 eV) were obtained on Thermo Finnigan MAT 95XP and Shimadzu LCMS-2010EV mass spectrometers (sample solutions in chloroform–acetonitrile were injected by a syringe at a flow rate of 0.1 ml min<sup>-1</sup>, eluent acetonitrile–water, 95:5) in the register mode of positive ions with the ionizing potential of needle electrode 4.5 kV, temperature of capillary interface 250 °C, voltage of capillary interface 5 V, flow rate of atomizing gas (nitrogen) 1.5 dm<sup>3</sup> min<sup>-1</sup> (for chemical ionization at atmospheric pressure). The reaction course was monitored by TLC on ‘Sorbphil’ plates with visualization of compounds in the ethanol solution of anisaldehyde acidified with sulfuric acid with subsequent heating at 120–150 °C. The products were isolated by column chromatography on silica gel (30–60 g of adsorbent per 1 g of compound); freshly distilled solvents were used as eluents.

Methyl 4*H*-thieno[3,2-*b*]pyrrole-5-carboxylate **1** was obtained according to published procedure,<sup>1,2</sup> the spectral characteristics conformed to those given in literature.

spectra were in accordance with structure **4a** (see Scheme 1). In the same manner, alcohols **3b–d** underwent self-condensation into bis(thienopyrrolyl)methanes **4b–d**.<sup>‡</sup>

To our knowledge, symmetric bis(thienopyrrolyl)methanes of type **4** are not documented. The most close precedent is condensation of aldehyde **5** with C(5)-unsubstituted pyrroles leading to unsymmetrical (thienopyrrolyl)methanes used for the preparation of BODIPY complexes.<sup>16</sup> At the same time, syntheses of more simple dipyrromethanes **6** are well developed [see, e.g., refs. 14(d),17].

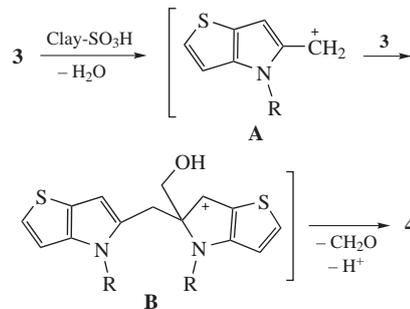


*Methyl 4-allyl-4H-thieno[3,2-b]pyrrole-5-carboxylate 2b* was prepared analogously from pyrrole **1** (0.06 g, 0.33 mmol), NaH (9.5 mg, 0.39 mmol) and allyl bromide (79.9 mg, 0.66 mmol) using Bu<sub>4</sub>NI (6 mg) as the phase transfer catalyst. The purification by column chromatography on SiO<sub>2</sub> (light petroleum–ethyl acetate, 5:1) gave 0.70 g (95%) of **2b** as an oily substance. IR ( $\nu/\text{cm}^{-1}$ ): 2948, 1703, 1699, 1532, 1464, 1441, 1395, 1303, 1256, 1216, 1175, 1103, 759, 719. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$ : 3.86 (s, 3H, OMe), 4.98 (dd, 1H, =CH<sub>2</sub>, *J* 17.1 and 1.1 Hz), 5.13 (dd, 1H, =CH<sub>2</sub>, *J* 10.4 and 1.1 Hz), 5.15 (m, 2H, NCH<sub>2</sub>), 6.01 (m, 1H, =CH, *J* 17.0, 1.5, 10.4 and 5.3 Hz), 6.92 (d, 1H, H-3, *J* 5.5 Hz), 7.21 (s, 1H, H-6), 7.33 (d, 1H, H-2, *J* 5.3 Hz). <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$ : 49.4 (CH<sub>2</sub>), 51.3 (OMe), 109.5 (C<sup>6</sup>), 110.4 (=CH<sub>2</sub>), 116.3 (C<sup>3</sup>), 122.2 (C<sup>6a</sup>), 125.9 (C<sup>5</sup>), 129.3 (C<sup>2</sup>), 133.9 (=CH), 145.1 (C<sup>3a</sup>), 161.9 (CO<sub>2</sub>Me). HRMS (ESI), *m/z*: 221.0505 [M]<sup>+</sup> (calc. for C<sub>11</sub>H<sub>11</sub>NO<sub>2</sub>S, *m/z*: 221.0510).

*Methyl 4-(prop-2-yn-1-yl)-4H-thieno[3,2-b]pyrrole-5-carboxylate 2c* was prepared in a similar manner from pyrrole **1** (0.37 g, 2.04 mmol) and 80% solution of propargyl bromide (0.45 ml, 4.10 mmol) in toluene. Bright yellow crystals, mp 83–86 °C. Yield 0.42 g (94%). IR ( $\nu/\text{cm}^{-1}$ ): 3106, 3265, 2953, 1692, 1534, 1492, 1464, 1438, 1394, 1377, 1306, 1261, 1220, 1181, 1171, 1111, 776, 732, 659. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$ : 2.32 (t, 1H,  $\equiv$ CH, *J* 2.3 Hz), 3.88 (s, 3H, OMe), 5.38 (d, 2H, NCH<sub>2</sub>, *J* 2.4 Hz), 7.09 (d, 1H, H-3, *J* 5.2 Hz), 7.22 (s, 1H, H-6), 7.38 (d, 1H, H-2, *J* 5.2 Hz). <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$ : 36.1 (CH<sub>2</sub>), 51.5 (OMe), 72.5 ( $\equiv$ CH), 78.4 ( $\equiv$ C), 110.1 (C<sup>6</sup>), 110.5 (C<sup>3</sup>), 122.6 (C<sup>6a</sup>), 125.5 (C<sup>5</sup>), 129.7 (C<sup>2</sup>), 145.3 (C<sup>3a</sup>), 162.3 (CO<sub>2</sub>Me). HRMS (ESI), *m/z*: 219.0349 [M]<sup>+</sup> (calc. for C<sub>11</sub>H<sub>9</sub>NO<sub>2</sub>S, *m/z*: 219.0354).

*Methyl 4-benzyl-4H-thieno[3,2-b]pyrrole-5-carboxylate 2d* was prepared similarly from pyrrole **1** (0.06 g, 0.33 mmol) and benzyl bromide (0.084 g, 0.66 mmol). Yellowish crystals, mp 81–83 °C. Yield 80 mg (92%). IR ( $\nu/\text{cm}^{-1}$ ): 2950, 1703, 1532, 1490, 1464, 1441, 1395, 1303, 1257, 1216, 1175, 1103, 1085, 991, 918, 781, 759, 719, 666. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$ : 3.83 (s, 3H, OMe), 5.76 (s, 2H, NCH<sub>2</sub>), 6.86 (d, 1H, H-3, *J* 5.4 Hz), 7.13 (d, 2H, Ph, *J* 7.4 Hz), 7.25 (s, 1H, H-6), 7.23–7.28 (m, 3H, Ph), 7.32 (d, 1H, H-2, *J* 5.7 Hz). <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$ : 50.4 (CH<sub>2</sub>), 51.3 (OMe), 109.8 (C<sup>6</sup>), 110.6 (C<sup>3</sup>), 122.4 (C<sup>6a</sup>), 126.1 (C<sup>5</sup>), 126.6 (C<sub>Ar</sub>), 127.3 (C<sub>Ar</sub>), 128.6 (C<sub>Ar</sub>), 129.4 (C<sup>2</sup>), 137.9 (C<sub>Ar</sub>), 145.4 (C<sup>3a</sup>), 161.9 (CO<sub>2</sub>Me). HRMS (ESI), *m/z*: 271.0662 [M]<sup>+</sup> (calc. for C<sub>15</sub>H<sub>13</sub>NO<sub>2</sub>S, *m/z*: 271.0667).

*(4-Methyl-4H-thieno[3,2-b]pyrrol-5-yl)methanol 3a*. A solution of ester **2a** (70 mg, 0.36 mmol) in anhydrous THF (5 ml) was added to a stirred suspension of LiAlH<sub>4</sub> (40.0 mg, 1.05 mmol) in anhydrous THF (10 ml) under an argon atmosphere. The mixture was stirred until the starting ester was consumed (TLC control), then quenched with saturated solution of NH<sub>4</sub>Cl, THF was evaporated and the product was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3  $\times$  20 ml). The combined extracts were washed with brine, dried over MgSO<sub>4</sub>, and the solvent was evaporated. The residue was purified by column chromatography on silica gel (EtOAc–light petroleum, 1:5) to isolate 38.0 mg (63%) of white powder, mp 71–73 °C. IR ( $\nu/\text{cm}^{-1}$ ): 3527, 3230, 2727, 1530, 1377, 1366, 1337, 1295, 1241, 1135, 1078, 987, 975, 823, 763, 713, 654. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$ : 2.86 (s, 1H, OH), 3.79 (s, 3H, NMe), 4.64 (d, 2H, OCH<sub>2</sub>, *J* 5.5 Hz), 6.27 (s, 1H, H-6), 7.02 (d, 1H, H-3, *J* 5.3 Hz), 7.11 (d, 1H, H-2, *J* 5.2 Hz). <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$ : 31.3 (NMe), 56.3 (OCH<sub>2</sub>), 99.6 (C<sup>6</sup>), 110.2 (C<sup>3</sup>), 120.7 (C<sup>6a</sup>), 121.3 (C<sup>5</sup>), 122.2 (C<sup>2</sup>), 138.4 (C<sup>3a</sup>). MS (EI), *m/z* (%): 168 [M+H]<sup>+</sup> (10), 150 [M–OH]<sup>+</sup> (100).



Scheme 2

The tentative mechanism of the self-condensation is outlined in Scheme 2. Alcohol **3** on contact with highly acidic Amberlyst 15 is converted into carbocation **A** which would regioselectively attack C(2) site of the second molecule of **3**. The thus formed adduct **B** with secondary carbocation centre is supposed to

*(4-Allyl-4H-thieno[3,2-b]pyrrol-5-yl)methanol 3b* was prepared similarly from **2b** (60 mg, 0.27 mmol) and LiAlH<sub>4</sub> (12 mg, 0.32 mmol). Purification by column chromatography on SiO<sub>2</sub> (light petroleum–EtOAc, 5:1) gave 0.38 g (72%) of compound **3b** as an oil. IR ( $\nu/\text{cm}^{-1}$ ): 2948, 1703, 1699, 1532, 1464, 1441, 1395, 1303, 1256, 1216, 1175, 1103, 759, 719. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ : 1.82 (br. s, 1H, OH), 4.65 (s, 2H, OCH<sub>2</sub>), 4.75 (d, 2H, NCH<sub>2</sub>, *J* 4.8 Hz), 4.99 (d, 1H, =CH<sub>2</sub>, *J* 17.1 Hz), 5.16 (d, 1H, =CH<sub>2</sub>, *J* 10.1 Hz), 5.99 (ddd, 1H, =CH, *J* 5.1, 10.2, 5.3 Hz), 6.38 (s, 1H, H-6), 6.89 (d, 1H, H-3, *J* 5.1 Hz), 7.09 (d, 1H, H-2, *J* 5.3 Hz). <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$ : 48.0 (NCH<sub>2</sub>), 57.5 (OCH<sub>2</sub>), 101.2 (C<sup>6</sup>), 110.5 (=CH<sub>2</sub>), 116.6 (C<sup>3</sup>), 121.9 (C<sup>6a</sup>), 122.0 (C<sup>5</sup>), 123.7 (C<sup>2</sup>), 134.1 (=CH), 141.5 (C<sup>3a</sup>). Found (%): C, 62.48; H, 5.49; N, 7.46; S, 16.98. Calc. for C<sub>10</sub>H<sub>11</sub>NOS (%): C, 62.15; H, 5.74; N, 7.25; S, 16.59.

*{4-(Prop-2-yn-1-yl)-4H-thieno[3,2-b]pyrrol-5-yl)methanol 3c* was prepared analogously from **2c** (60 mg, 0.27 mmol) and LiAlH<sub>4</sub> (12.4 mg, 0.33 mmol). Yield 81% (42 mg). Light yellow crystals, mp 66–68 °C. IR ( $\nu/\text{cm}^{-1}$ ): 3312, 3270, 3203, 2950, 1462, 1400, 1438, 1377, 1364, 1330, 1295, 1013, 782, 721, 682, 655. <sup>1</sup>H NMR (500 MHz, acetone-*d*<sub>6</sub>)  $\delta$ : 2.79 (s, 1H, OH), 2.87 (t, 1H,  $\equiv$ CH, *J* 2.3 Hz), 4.73 (s, 2H, OCH<sub>2</sub>), 5.06 (d, 2H, CH<sub>2</sub>, *J* 2.3 Hz), 6.33 (s, 1H, H-6), 7.12 (d, 1H, H-3, *J* 5.3 Hz), 7.15 (d, 2H, H-2, *J* 5.3 Hz). <sup>13</sup>C NMR (125 MHz, acetone-*d*<sub>6</sub>)  $\delta$ : 34.4 (CH<sub>2</sub>), 56.4 (OCH<sub>2</sub>), 73.1 ( $\equiv$ CH), 78.9 ( $\equiv$ C), 100.5 (C<sup>6</sup>), 110.7 (C<sup>3</sup>), 122.1 (C<sup>6a</sup>), 122.8 (C<sup>2</sup>), 125.9 (C<sup>5</sup>), 144.1 (C<sup>3a</sup>). Found (%): C, 62.46; H, 4.49; N, 7.46; S, 16.49. Calc. for C<sub>10</sub>H<sub>9</sub>NOS (%): C, 62.80; H, 4.74; N, 7.32; S, 16.77.

*(4-Benzyl-4H-thieno[3,2-b]pyrrol-5-yl)methanol 3d* was prepared similarly from **2d** (60 mg, 0.22 mmol) and LiAlH<sub>4</sub> (10.0 mg, 0.27 mmol). Yield 34.0 mg (64%), oil. IR ( $\nu/\text{cm}^{-1}$ ): 3537, 2931, 2873, 1526, 1476, 1452, 1405, 1375, 1356, 1336, 1294, 1241, 1218, 1137, 1080, 1048, 995, 935, 826, 801, 776, 743, 592. <sup>1</sup>H NMR (500 MHz, acetone-*d*<sub>6</sub>)  $\delta$ : 2.90 (s, 1H, OH), 4.64 (s, 2H, OCH<sub>2</sub>), 5.44 (s, 2H, NCH<sub>2</sub>), 6.38 (s, 1H, H-6), 6.87 (d, 1H, H-3, *J* 5.2 Hz), 7.06 (d, 1H, H-2, *J* 5.3 Hz), 7.16 (d, 2H, Ph, *J* 7.2 Hz), 7.23–7.29 (m, 3H, Ph). <sup>13</sup>C NMR (125 MHz, acetone-*d*<sub>6</sub>)  $\delta$ : 48.7 (CH<sub>2</sub>), 56.5 (OCH<sub>2</sub>), 100.2 (C<sup>6</sup>), 110.8 (C<sup>3</sup>), 122.5 (C<sup>6a</sup>), 122.6 (C<sup>2</sup>), 126.6 (C<sub>Ar</sub>), 127.1 (C<sub>Ar</sub>), 128.5 (C<sup>5</sup>), 128.5 (C<sub>Ar</sub>), 138.6 (C<sub>Ar</sub>), 138.4 (C<sup>3a</sup>). MS (EI), *m/z* (%): 244 [MH]<sup>+</sup> (5), 226 [M–OH]<sup>+</sup> (100). Found (%): C, 68.89; H, 5.47; N, 5.44; S, 13.44. Calc. for C<sub>14</sub>H<sub>15</sub>NOS (%): C, 69.10; H, 5.39; N, 5.76; S, 13.18.

<sup>‡</sup> *Bis(4-methyl-4H-thieno[3,2-b]pyrrol-5-yl)methane 4a*. Amberlyst 15 (0.15 g) was added to a stirred solution of **3a** (0.14 g, 0.84 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10 ml), the reaction mixture was stirred until the starting alcohol was consumed (5–6 h, TLC control). Amberlyst was filtered off, the CH<sub>2</sub>Cl<sub>2</sub> filtrate was concentrated, and the residue was purified by column chromatography on silica gel (EtOAc–light petroleum, 1:5) to isolate 70.0 mg (58%) of product **4a** as an oil. IR ( $\nu/\text{cm}^{-1}$ ): 3097, 3083, 2953, 2853, 1662, 1537, 1471, 1438, 1382, 1334, 1291, 1133, 1083, 845, 765, 725. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ : 3.68 (s, 6H, NMe), 4.13 (s, 2H, CH<sub>2</sub>), 6.14 (s, 2H, H-6), 6.90 (d, 2H, H-3, *J* 5.3 Hz), 7.03 (d, 2H, H-2, *J* 5.2 Hz). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$ : 26.2 (CH<sub>2</sub>), 32.2 (NMe), 100.1 (C<sup>6</sup>), 109.9 (C<sup>3</sup>), 121.6 (C<sup>6a</sup>), 121.9 (C<sup>2</sup>), 134.7 (C<sup>5</sup>), 140.1 (C<sup>3a</sup>). MS (EI), *m/z* (%): 287 [MH]<sup>+</sup> (100). Found (%): C, 63.16; H, 4.66; N, 9.49; S, 22.76. Calc. for C<sub>15</sub>H<sub>14</sub>N<sub>2</sub>S<sub>2</sub> (%): C, 62.90; H, 4.93; N, 9.78; S, 22.39.

release formaldehyde molecule and proton affording final stable bis(thienopyrrolyl)methane **4**.

In total, the herein obtained bis(4*H*-thieno[3,2-*b*]pyrrol-5-yl)methanes **4** seem promising as new scaffolds in search for new bioactive compounds and design of new cross-conjugated structures. The mechanistic aspects of the chemical transformations can also be of interest.

## References

- 1 H. Hemetsberger and D. Knittel, *Monatsh. Chem.*, 1972, **103**, 194.
- 2 V. N. Yarovenko, S. L. Semenov, I. V. Zavarzin, A. V. Ignatenko and M. M. Krayushkin, *Russ. Chem. Bull., Int. Ed.*, 2003, **52**, 451 (*Izv. Akad. Nauk, Ser. Khim.*, 2003, 431).

*Bis(4-allyl-4H-thieno[3,2-*b*]pyrrol-5-yl)methane 4b* was prepared similarly from alcohol **3b**, reaction time 8 h, oil, yield 55%. IR ( $\nu/\text{cm}^{-1}$ ): 3099, 3081, 3009, 2982, 2954, 2919, 2867, 1517, 1476, 1442, 1395, 1345, 1337, 1298, 1285, 1082, 988, 921, 798, 773, 648.  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$ : 4.07 (s, 2H,  $\text{CH}_2$ ), 4.60 (dd, 2H,  $\text{NCH}_2$ ,  $J$  1.6, 3.4 Hz), 4.61 (d, 2H,  $\text{NCH}_2$ ,  $J$  1.6, 3.4 Hz), 4.92 (dd, 2H,  $=\text{CH}_2$ ,  $J$  1.2, 17.1 Hz), 5.13 (dd, 2H,  $=\text{CH}_2$ ,  $J$  1.2, 10.2 Hz), 5.88 (m, 2H,  $=\text{CH}$ ,  $J$  5.1, 10.1 Hz), 6.15 (s, 2H, H-6'), 6.85 (d, 1H, H-3',  $J$  5.3 Hz), 6.87 (d, 1H, H-3',  $J$  5.2 Hz), 7.01 (d, 2H, H-2',  $J$  5.3 Hz).  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ )  $\delta$ : 26.0 ( $\text{CH}_2$ ), 47.9 ( $\text{NCH}_2$ ), 100.5 ( $\text{C}^6$ ), 110.3 ( $\text{C}^3$ ), 116.5 ( $=\text{CH}_2$ ), 122.2 ( $\text{C}^{6a}$ ), 122.0 ( $\text{C}^2$ ), 133.7 ( $=\text{CH}$ ), 134.5 ( $\text{C}^5$ ), 140.5 ( $\text{C}^{3a}$ ). HRMS (ESI),  $m/z$ : 338.0906 [ $\text{M}]^+$  (calc. for  $\text{C}_{19}\text{H}_{18}\text{N}_2\text{S}_2$ ,  $m/z$ : 338.0911). Found (%): C, 67.81; H, 5.47; N, 8.43; S, 18.66. Calc. for  $\text{C}_{19}\text{H}_{18}\text{N}_2\text{S}_2$  (%): C, 67.42; H, 5.36; N, 8.28; S, 18.95.

*Bis(4-(prop-2-yn-1-yl)-4H-thieno[3,2-*b*]pyrrol-5-yl)methane 4c* was prepared analogously from alcohol **3c**, reaction time 8 h, oil, yield 43%. IR ( $\nu/\text{cm}^{-1}$ ): 3285, 3259, 3106, 3081, 2954, 2923, 2850, 2125, 1661, 1647, 1531, 1517, 1472, 1436, 1394, 1334, 1296, 1248, 1133, 1086, 1055, 1042, 910, 773, 715, 653.  $^1\text{H}$  NMR (300 MHz, acetone- $d_6$ )  $\delta$ : 2.33 (s, 2H,  $\equiv\text{CH}$ ), 4.32 (s, 2H,  $\text{CH}_2$ ), 4.75 (d, 4H,  $\text{NCH}_2$ ,  $J$  2.5 Hz), 6.23 (s, 2H, H-6'), 6.98 (d, 2H, H-2',  $J$  5.3 Hz), 7.06 (d, 2H, H-3',  $J$  5.3 Hz).  $^{13}\text{C}$  NMR (125 MHz, acetone- $d_6$ )  $\delta$ : 34.4 ( $\text{CH}_2$ ), 56.4 ( $\text{NCH}_2$ ), 73.1 ( $\equiv\text{CH}$ ), 78.9 ( $\equiv\text{C}$ ), 101.4 ( $\text{C}^6$ ), 110.1 ( $\text{C}^3$ ), 121.5 ( $\text{C}^{6a}$ ), 122.8 ( $\text{C}^2$ ), 133.8 ( $\text{C}^5$ ), 140.3 ( $\text{C}^{3a}$ ). HRMS (ESI),  $m/z$ : 334.0593 [ $\text{M}]^+$  (calc. for  $\text{C}_{19}\text{H}_{14}\text{N}_2\text{S}_2$ ,  $m/z$ : 334.0598). Found (%): C, 68.60; H, 4.53; N, 8.47; S, 19.56. Calc. for  $\text{C}_{19}\text{H}_{14}\text{N}_2\text{S}_2$  (%): C, 68.23; H, 4.22; N, 8.38; S, 19.17.

*Bis(4-benzyl-4H-thieno[3,2-*b*]pyrrol-5-yl)methane 4d* was prepared in a similar manner from alcohol **3d**, reaction time 9–10 h, oil, yield 62%.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$ : 3.92 (s, 2H,  $\text{CH}_2$ ), 5.13 (s, 4H,  $\text{NCH}_2$ ), 6.19 (s, 2H, H-6), 6.77 (d, 2H, H-3,  $J$  5.3 Hz), 6.99 (d, 2H, H-2,  $J$  5.2 Hz), 6.89 (m, 4H, Ph), 7.21 (m, 6H, Ph).  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ )  $\delta$ : 37.6 ( $\text{CH}_2$ ), 48.5 ( $\text{NCH}_2$ ), 99.1 ( $\text{C}^3$ ), 110.0 ( $\text{C}^6$ ), 121.0 ( $\text{C}^2$ ), 121.9 ( $\text{C}^{6a}$ ), 125.6 ( $\text{C}_{Ar}$ ), 126.0 ( $\text{C}_{Ar}$ ), 127.1 ( $\text{C}_{Ar}$ ), 128.0 ( $\text{C}_{Ar}$ ), 128.1 ( $\text{C}_{Ar}$ ), 128.5 ( $\text{C}_{Ar}$ ), 134.2 ( $\text{C}^5$ ), 137.5 ( $\text{C}_{Ar}$ ), 139.9 ( $\text{C}_{Ar}$ ), 141.5 ( $\text{C}^{3a}$ ). HRMS (ESI),  $m/z$ : 438.1219 [ $\text{M}]^+$  (calc. for  $\text{C}_{27}\text{H}_{22}\text{N}_2\text{S}_2$ ,  $m/z$ : 438.1224). Found (%): C, 73.59; H, 5.13; N, 6.54; S, 14.98. Calc. for  $\text{C}_{27}\text{H}_{22}\text{N}_2\text{S}_2$  (%): C, 73.94; H, 5.06; N, 6.39; S, 14.62.

- 3 K.-C. Ching, Y.-W. Kam, A. Merits, L. F. P. Ng and C. L. L. Chai, *J. Med. Chem.*, 2015, **58**, 9196.
- 4 K.-C. Ching, T. N. Q. Tran, S. N. Amrun, Y.-W. Kam, L. F. P. Ng and C. L. L. Chai, *J. Med. Chem.*, 2017, **60**, 3165.
- 5 L. Sartori, C. Mercurio, F. Amigoni, A. Cappa, G. Fagá, R. Fattori, E. Legnagli, G. Ciossani, A. Mattevi, G. Meroni, L. Moretti, V. Cecatiello, S. Pasqualato, A. Romussi, F. Thaler, P. Trifiró, M. Villa, S. Vultaggio, O. A. Botrugno, P. Dessanti, S. Minucci, E. Zagarrí, D. Caretoni, L. Iuzzolino, M. Varasi and P. Vianello, *J. Med. Chem.*, 2017, **60**, 1673.
- 6 D. Ashok, P. S. Goud, D. Shrivani, J. Rajaiiah, M. Sarasija and K. Bhaskar, *Indian J. Heterocycl. Chem.*, 2010, **20**, 81.
- 7 A. P. Ilyin, I. G. Dmitrieva, V. A. Kustova, A. V. Manaev and A. V. Ivachtchenko, *J. Comb. Chem.*, 2007, **9**, 96.
- 8 J. A. Sindac, B. D. Yestrepky, S. J. Barraza, K. L. Bolduc, P. K. Blakely, R. F. Keep, D. N. Irani, D. J. Miller and S. D. Larsen, *J. Med. Chem.*, 2012, **55**, 3535.
- 9 E. V. Verbitskiy, P. A. Slepukhin, M. S. Valova, E. M. Cheprakova, A. V. Schepochkin, G. L. Rusinov and V. N. Charushin, *Eur. J. Org. Chem.*, 2014, 8133.
- 10 M. M. Krayushkin, V. N. Yarovenko, S. L. Semenov, I. V. Zavarzin, A. V. Ignatenko, A. Yu. Martynkin and B. M. Uzhinov, *Org. Lett.*, 2002, **4**, 3879.
- 11 C. Jones, D. Boudinet, Y. Xia, M. Denti, A. Das, A. Facchetti and T. G. Driver, *Chem. Eur. J.*, 2014, 5938.
- 12 X.-D. Jiang, H. Zhang, Y. Zhang and W. Zhao, *Tetrahedron*, 2012, **68**, 9795.
- 13 (a) D. H. Burns, Y. H. Li, D. C. Shi and T. M. Caldwell, *J. Org. Chem.*, 2002, **67**, 4536; (b) S. E. Bari, J. Iturraspe and B. Frydman, *Tetrahedron*, 1995, **51**, 2255; (c) A. H. Jackson, R. K. Pandey, K. R. Nagaraja Rao and E. Roberts, *Tetrahedron Lett.*, 1985, **26**, 793.
- 14 (a) C.-H. Lee and J. S. Lindsey, *Tetrahedron*, 1994, **50**, 11427; (b) K. Singh, S. Sharma and A. Sharma, *Synth. Commun.*, 2011, **41**, 3491; (c) T. K. Chan, S. K. Jana, M. Rajeswara Rao, M. S. Shaikh and M. Ravikanth, *Inorg. Chim. Acta*, 2012, **383**, 257; (d) T. E. Wood and A. Thomson, *Chem. Rev.*, 2007, **107**, 1831; (e) E. A. Mikhailitsyna, V. S. Tyurin, S. E. Nefedov, S. A. Syrbu, A. S. Semeikin, O. I. Koifman and I. P. Beletskaya, *Eur. J. Inorg. Chem.*, 2012, 5979.
- 15 (a) T. D. Lash, *Chem. Eur. J.*, 1996, **2**, 1197; (b) Z. Fang and B. Liu, *Tetrahedron Lett.*, 2008, **49**, 2311; (c) L. T. Nguen, M. O. Senge and K. M. Smith, *J. Org. Chem.*, 1996, **61**, 998.
- 16 L. Wu and K. Burgess, *Chem. Commun.*, 2008, 4933.
- 17 (a) J. S. Lindsey, *Acc. Chem. Res.*, 2010, **43**, 300; (b) A. Sharma and S. Obrai, *Der Chemica Sinica*, 2015, **6**, 57; (c) W. Zhao and E. M. Carreira, *Chem. Eur. J.*, 2006, **12**, 7254; (d) N. A. M. Pereira and T. M. V. D. Pinhole Melo, *Org. Prep. Proced. Int.*, 2014, **46**, 183.
- 18 S. Lotz, M. Landman, H. Goerls, C. Crause, H. Nienaber and A. Olivier, *Z. Naturforsch., B: Chem. Sci.*, 2007, **62**, 419.

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