

Dioximes of 1,6-heptanediones from acetylene and ketones: only three atom-economic steps

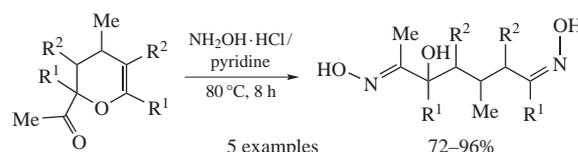
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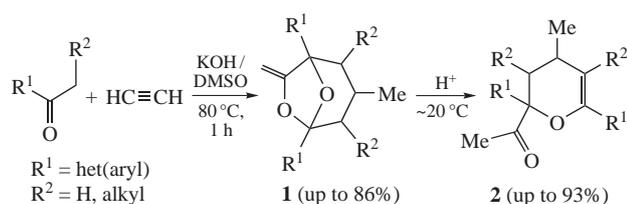
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2-Acetyl-3,4-dihydropyrans, synthesized from acetylene and ketones in two steps, react with hydroxylamine to afford 5-hydroxy-1,6-heptanedione dioximes (*E,E*-isomers) in 72–96% yields.



Over the last decade, the investigations devoted to application of acetylene in the superbasic media as the initiating and organizing molecule for one-pot assemblies of the complex molecular systems are gathering momentum.^{1–10} In the presence of strong bases, acetylene demonstrates its dual ability to act both as an electrophile (addition of nucleophilic reagents at the triple bond) and a nucleophile (addition of acetylenide ions to the electrophilic molecules).² Consequently, several molecules of acetylene and ambifunctional carbonyl compounds are often involved into the self-organization. For example, 7-methylidene-6,8-dioxabicyclo[3.2.1]octanes **1** (Scheme 1),¹¹ analogues of insect pheromones¹² and mammal hormones,¹³ are diastereoselectively assembled from two molecules of acetylene and two molecules of ketone in the KOH/DMSO superbasic system in a one preparative step.¹¹ Thus obtained bicyclooctanes **1** in the presence of trace acids at room temperature isomerize instantly and quantitatively into dihydropyrans **2**.¹⁴

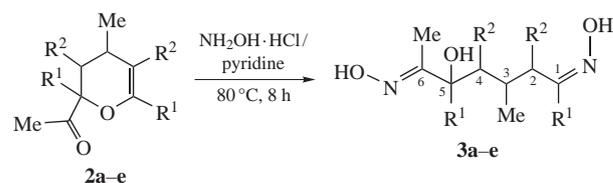


Scheme 1

Herein, we report that dihydropyrans **2a–e** upon treatment with two equivalents of hydroxylamine hydrochloride in pyridine are stereoselectively transformed (the configuration is established according to ¹³C NMR spectra)¹⁵ into dioximes of 5-hydroxy-1,6-diketones **3a–e** (Scheme 2) with the configuration of both formed oxime groups being *E*.[†]

A likely way to dioximes **3a–e** lies *via* the acid-catalyzed addition of water, released from oximation of the acetyl function,

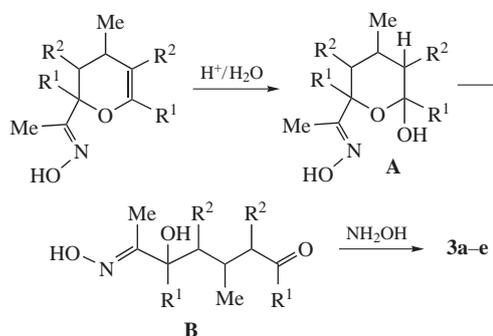
to the enol moiety of dihydropyrans **2a–e** to afford hemiacetal **A**. The latter expectedly undergoes ring-opening to give monoximes of 5-hydroxy-1,6-diketones **B**, whose further oximation leads to final dioximes **3a–e** (Scheme 3).



- a** R¹ = Ph, R² = H (94%)
b R¹ = Ph, R² = Me (72%)
c R¹ = 4-MeC₆H₄, R² = H (96%)
d R¹ = 3-MeOC₆H₄, R² = H (91%)
e R¹ = 4-PhC₆H₄, R² = H (90%)

Scheme 2

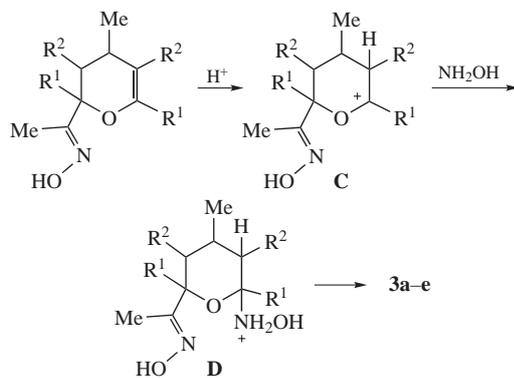
However, because of no monoximes **B** are discernible in the crude products, an alternative route to dioximes **3** seems to be also probable. Actually, the intermediate carbocation **C** may be attacked by a molecule of hydroxylamine to generate hemiacetal **D** that



Scheme 3

[†] The IR spectra were recorded on a Bruker IFS25 spectrophotometer. The NMR spectra were recorded on Bruker DPX-400 and AV-400 spectrometers (400.1 MHz for ¹H, 100.6 MHz for ¹³C and 40.5 MHz for ¹⁵N) in DMSO-*d*₆. The assignment of signals was made using COSY, NOESY, ¹H-¹³C HSQC, ¹H-¹³C HMBC, and 2D ¹H-¹⁵N HMBC experiments.

Dioximes 3 (typical procedure). A mixture of dihydropyran **2** (2 mmol) and hydroxylamine hydrochloride (4 mmol, 280 mg) in pyridine (5 ml) was stirred at 80°C for 8 h. The mixture was poured into water (20 ml), the product was collected by filtration, washed with H₂O and recrystallized from hexane.



Scheme 4

directly delivers (in the course of its decomposition) dioximes **3a–e** (Scheme 4).

1,6-Diketones and their derivatives, especially functionally substituted ones, are still hardly accessible, though synthetically and pharmaceutically prospective compounds.¹⁶ All this relates completely to dioximes of 1,6-diketones **3** bearing the additional hydroxyl function. Since the assembly of bicyclooctanes **1** (see Scheme 1) tolerates a wide range of diverse ketones,¹¹ the first syntheses of 1,6-diketone dioximes from acetylene and ketones described here can be considered as the priority application in development of general atom-economic, base- and acid-catalyzed method for the preparation of new derivatives of 1,6-dicarbonyl compounds.

(1E,6E)-5-Hydroxy-3-methyl-1,5-diphenylheptane-1,6-dione dioxime **3a**: yield 640 mg (94%), white solid, mp 160–161 °C. ¹H NMR, δ: 0.80 (d, 3H, C³Me, ³J 6.5 Hz), 1.47 (s, 3H, C⁶Me), 1.56 (m, 1H, H-3), 1.93 (m, 1H, H-4'), 2.24 (m, 2H, H-2', H-4), 2.64 (m, 1H, H-2), 5.37 (s, 1H, C⁵OH), 7.18–7.31 (m, 10H, Ph), 10.64 (s, 1H, C⁶NOH), 10.97 (s, 1H, C³NOH). ¹³C NMR, δ: 10.3 (C⁶Me), 21.3 (C³Me), 26.5 (C³), 32.4 (C²), 45.5 (C⁴), 78.0 (C⁵), 125.5, 125.8 (o-C, o'-C), 127.7, 128.1 (m-C, m'-C), 126.3, 128.2 (p-C, p'-C), 135.8 (i-C), 144.0 (i'-C), 155.9 (C¹), 159.8 (C⁶). ¹⁵N NMR, δ: –28.7 (C⁶N), –19.6 (C¹N). IR (film, ν/cm^{–1}): 3544, 3457, 3320, 3086, 3061, 2959, 2926, 2875, 1958, 1891, 1817, 1640, 1492, 1448, 1373, 1308, 1220, 1182, 1130, 1072, 999, 945, 915, 758, 699, 648, 570, 498. Found (%): C, 70.24; H, 7.05; N, 8.34. Calc. for C₂₀H₂₄N₂O₃ (%): C, 70.56; H, 7.11; N, 8.23.

(1E,6E)-5-Hydroxy-2,3,4-trimethyl-1,5-diphenylheptane-1,6-dione dioxime **3b**: yield 531 mg (72%), white solid, mp 196–197 °C. ¹H NMR, δ: 0.66 (d, 3H, C³Me, ³J 7.0 Hz), 0.89 (d, 3H, C⁴Me, ⁴J 6.9 Hz), 0.95 (d, 3H, C²Me, ²J 7.0 Hz), 1.33 (m, 1H, H-3), 1.49 (s, 3H, C⁶Me), 2.54 (m, 1H, H-2), 2.75 (m, 1H, H-4), 5.00 (br. s, 1H, C⁵OH), 7.02–7.26 (m, 10H, Ph), 10.47 (br. s, 2H, NOH). ¹³C NMR, δ: 10.0 (C⁴Me), 10.4 (C⁶Me), 11.2 (C³Me), 15.0 (C²Me), 33.7 (C³), 39.1 (C⁴), 43.9 (C²), 81.9 (C⁵), 125.8, 126.2, 127.4, 127.5, 127.8 (10C, Ph), 134.0 (i-C), 143.3 (i'-C), 158.6 (C⁶), 158.8 (C¹). IR (film, ν/cm^{–1}): 3503, 3457, 3288, 3246, 3107, 3066, 2986, 2931, 2881, 1743, 1686, 1647, 1536, 1497, 1444, 1377, 1342, 1223, 1191, 1151, 1112, 1067, 1009, 941, 895, 763, 702, 654, 580, 542. Found (%): C, 71.94; H, 7.60; N, 7.51. Calc. for C₂₂H₂₈N₂O₃ (%): C, 71.71; H, 7.66; N, 7.60.

(1E,6E)-5-Hydroxy-3-methyl-1,5-di-p-tolylheptane-1,6-dione dioxime **3c**: yield 707 mg (96%), white solid, mp 153–155 °C. ¹H NMR, δ: 0.78 (d, 3H, C³Me, ³J 6.6 Hz), 1.45 (s, 3H, C⁶Me), 1.53 (m, 1H, H-3), 1.87 (m, 1H, H-4'), 2.21 (m, 2H, H-2', H-4), 2.25 (s, 3H, MeC₆H₄), 2.27 (s, 3H, MeC₆H₄), 2.57 (m, 1H, H-2), 5.26 (s, 1H, C⁵OH), 7.03 (m, 4H, m-H, m'-H), 7.14 (m, 4H, o-H, o'-H), 10.58 (s, 1H, C⁶NOH), 10.82 (s, 1H, C³NOH). ¹³C NMR, δ: 10.3 (C⁶Me), 20.6, 20.7 (2MeC₆H₄), 21.3 (C³Me), 26.5 (C³), 32.4 (C²), 45.4 (C⁴), 77.8 (C⁵), 125.4, 126.7 (o-C, o'-C), 128.3, 128.6 (m-C, m'-C), 132.9, 135.1 (p-C, p'-C), 137.5 (i-C), 141.0 (i'-C), 155.8 (C¹), 159.9 (C⁶). ¹⁵N NMR, δ: –28.3 (C⁶N), –20.1 (C¹N). IR (film, ν/cm^{–1}): 3544, 3297, 3090, 3059, 3030, 2955, 2923, 2870, 1910, 1641, 1619, 1511, 1455, 1411, 1369, 1307, 1237, 1184, 1104, 1041, 1020, 987, 939, 879, 814, 736, 671, 637, 585, 508, 422. Found (%): C, 71.74; H, 7.64; N, 7.58. Calc. for C₂₂H₂₈N₂O₃ (%): C, 71.71; H, 7.66; N, 7.60.

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Online Supplementary Materials

Supplementary data associated with this article can be found in the online version at doi: 10.1016/j.mencom.2018.03.010.

References

- 1 *Domino Reactions: Concepts for Efficient Organic Synthesis*, ed. L. F. Tietze, Wiley-VCH, Weinheim, 2014.
- 2 B. A. Trofimov and E. Yu. Schmidt, *Russ. Chem. Rev.*, 2014, **83**, 600.
- 3 K. S. Rodygin, A. A. Kostin and V. P. Ananikov, *Mendeleev Commun.*, 2015, **25**, 415.
- 4 M. T. Pirnot, Y.-M. Wang and S. L. Buchwald, *Angew. Chem. Int. Ed.*, 2016, **128**, 48.
- 5 K. I. Galkin and V. P. Ananikov, *Russ. Chem. Rev.*, 2016, **85**, 226.
- 6 K. S. Rodygin, G. Werner, F. A. Kucherov and V. P. Ananikov, *Chem. Asian J.*, 2016, **11**, 965.
- 7 V. P. Ananikov, K. I. Galkin, M. P. Egorov, A. M. Sakharov, S. G. Zlotin, E. A. Redina, V. I. Isaeva, L. M. Kustov, M. L. Gening and N. E. Nifantiev, *Mendeleev Commun.*, 2016, **26**, 365.
- 8 B. A. Trofimov, K. V. Belyaeva, L. V. Andriyankova, L. P. Nikitina and A. G. Mal'kina, *Mendeleev Commun.*, 2017, **27**, 109.
- 9 E. Yu. Schmidt, I. V. Tatarinova, E. V. Ivanova and B. A. Trofimov, *Mendeleev Commun.*, 2017, **27**, 283.
- 10 O. V. Petrova, L. N. Sobenina, I. A. Ushakov, A. B. Budaev, A. V. Ivanov, V. A. Samsonov, A. Ya. Tikhonov and B. A. Trofimov, *Mendeleev Commun.*, 2017, **27**, 344.
- 11 B. A. Trofimov, E. Yu. Schmidt, I. A. Ushakov, A. I. Mikhaleva, N. V. Zorina, N. I. Protsuk, E. Yu. Senotrusova, E. V. Skital'tseva, O. N. Kazheva, G. G. Alexandrov and O. A. Dyachenko, *Eur. J. Org. Chem.*, 2009, 5142.
- 12 K. Mori, in *Topics in Current Chemistry. The Chemistry of Pheromones and Other Semiochemicals I*, ed. S. Schulz, Springer, Heidelberg, 2004, vol. 239, pp. 1–50.
- 13 D. R. Greenwood, D. Comeskey, M. B. Hunt and L. E. L. Rasmussen, *Nature*, 2005, **438**, 1097.
- 14 E. Yu. Schmidt, B. A. Trofimov, N. V. Zorina, A. I. Mikhaleva, I. A. Ushakov, E. V. Skital'tseva, O. N. Kazheva, G. G. Alexandrov and O. A. Dyachenko, *Eur. J. Org. Chem.*, 2010, 6727.
- 15 A. V. Afonin, I. A. Ushakov, O. A. Tarasova, E. Yu. Schmidt, A. I. Mikhaleva and V. K. Voronov, *Russ. J. Org. Chem.*, 2000, **36**, 1777 (*Zh. Org. Khim.*, 2000, **36**, 1831).
- 16 (a) A. I. Mikhaleva, A. B. Zaitsev and B. A. Trofimov, *Russ. Chem. Rev.*, 2006, **75**, 797 (*Usp. Khim.*, 2006, **75**, 884); (b) D. K. Kölmel and E. T. Kool, *Chem. Rev.*, 2017, **117**, 10358.

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(1E,6E)-5-Hydroxy-1,5-bis(3-methoxyphenyl)-3-methylheptane-1,6-dione dioxime **3d**: yield 729 mg (91%), white solid, mp 107–110 °C. ¹H NMR, δ: 0.81 (d, 3H, C³Me, ³J 6.5 Hz), 1.48 (s, 3H, C⁶Me), 1.63 (m, 1H, H³), 1.93 (m, 1H, H-4'), 2.23 (m, 2H, H-2', H-4), 2.57 (m, 1H, H-2), 3.69 (s, 3H, MeO), 3.72 (s, 3H, MeO), 5.37 (s, 1H, C⁵OH), 6.72–7.16 (m, 8H, Ph), 10.61 (s, 1H, C⁶NOH), 10.98 (s, 1H, C³NOH). ¹³C NMR, δ: 10.3 (C⁶Me), 21.4 (C³Me), 26.7 (C³), 32.4 (C²), 45.5 (C⁴), 54.8, 54.9 (2MeO), 77.9 (C⁵), 110.9, 111.4, 111.6, 114.1, 117.8, 118.3, 128.8, 129.1 (8C, Ph), 137.3 (i-C), 145.9 (i'-C), 155.9 (C¹), 158.9, 159.0 (m-CO, m'-CO), 159.8 (C⁶). IR (film, ν/cm^{–1}): 3327, 3081, 2950, 2840, 2083, 1928, 1646, 1593, 1483, 1455, 1434, 1375, 1312, 1293, 1243, 1175, 1138, 1044, 957, 866, 782, 710, 645, 567, 448. Found (%): C, 65.74; H, 7.09; N, 6.94. Calc. for C₂₂H₂₈N₂O₅ (%): C, 65.98; H, 7.05; N, 7.00.

(1E,6E)-1,5-Di(biphenyl-4-yl)-5-hydroxy-3-methylheptane-1,6-dione dioxime **3e**: yield 887 mg (90%), white solid, mp 216–217 °C. ¹H NMR, δ: 0.89 (d, 3H, C³Me, ³J 6.2 Hz), 1.51 (s, 4H, C⁶Me, H-3), 1.90 (m, 1H, H-4'), 2.37 (m, 2H, H-2', H-4), 2.70 (m, 1H, H-2), 5.45 (s, 1H, C⁵OH), 7.27–7.59 (m, 18H, Ph), 10.70 (s, 1H, C⁶NOH), 11.06 (s, 1H, C³NOH). ¹³C NMR, δ: 10.4 (C⁶Me), 21.6 (C³Me), 26.9 (C³), 32.7 (C²), 45.1 (C⁴), 78.3 (C⁵), 126.0, 126.2, 126.3, 126.4, 126.5, 127.3, 127.5, 128.9, 129.0, 134.7, 139.4, 139.8 (22C, Ph), 138.0 (i-C), 143.3 (i'-C), 155.6 (C¹), 159.9 (C⁶). ¹⁵N NMR, δ: –27.6 (C⁶N), –17.9 (C¹N). IR (film, ν/cm^{–1}): 3480, 3310, 3083, 3060, 3033, 2959, 2923, 2872, 1953, 1910, 1803, 1643, 1484, 1452, 1383, 1333, 1306, 1193, 1131, 1097, 1052, 1014, 940, 903, 833, 765, 734, 700, 647, 551. Found (%): C, 77.88; H, 6.65; N, 5.78. Calc. for C₃₂H₃₂N₂O₃ (%): C, 78.02; H, 6.55; N, 5.69.