

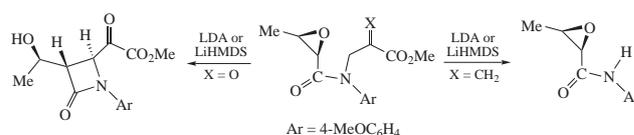
Novel azetidiones for carbapenems and fragmentation in the allylamine precursor analogue

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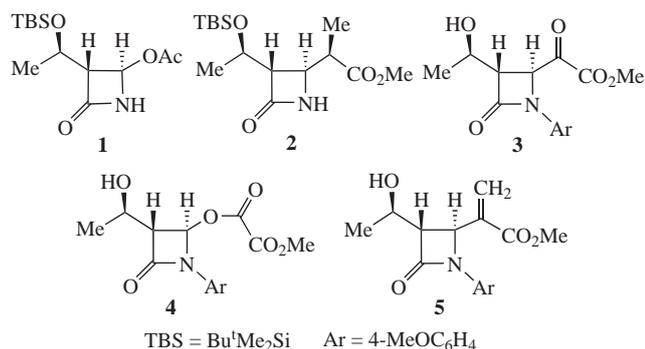
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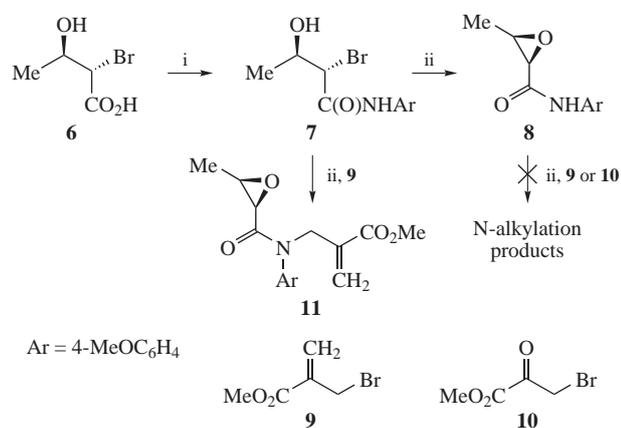
(2*R*,3*R*)-*N*-(3-Methoxy-2,3-dioxopropyl)-*N*-(4-methoxyphenyl)-2,3-epoxybutanamide on treatment with strong bases (LDA, LHMDS) gives 3-[(1*R*)-1-hydroxyethyl]azetidion-2-one derivative, while the *N*-(3-methoxy-2-methylidene-3-oxopropyl) analogue undergoes fragmentation.



Antibiotics of the carbapenem series (Meropenem, Ertapenem, Biapenem, Orapenem, Tebipenem, *etc.*) belong to much-in-demand antimicrobial agents.^{1,2} Properly functionalized key azetidione blocks, such as compounds **1** and **2**, are widely used in the synthesis of carbapenem antibiotics.³



Within the approaches to carbapenems that we are developing, we planned to obtain similar new azetidiones **3–5** from L-threonine using a version of intramolecular ‘epoxide-enolate’ cyclization.⁴ Threonine contains 3*R*-chiral center with β-oriented hydroxy group

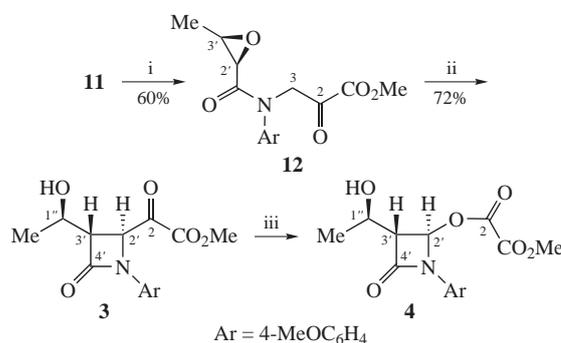


Scheme 1 Reagents and conditions: i, 4-MeOC₆H₄NH₂, DMAP, DCC, CH₂Cl₂, 0 °C; ii, 50% NaOH, Bu₄Ni (cat.), CH₂Cl₂, ~20 °C.

whose stereochemistry matches that of the side hydroxyethyl moieties in azetidiones **1** and **2** (*cf.* ref. 5).

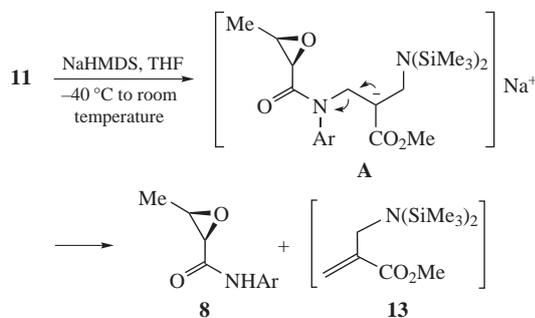
The syntheses of compounds **3–5**, we reasoned, can be started from bromo hydroxy acid **6**⁶ (Scheme 1). It was first converted into amide **7** and then into epoxide **8**. Attempts to N-alkylate epoxy amide **10** with active bromides **9**⁷ or **10**⁸ were unsuccessful. At the same time, the reaction of bromo hydroxy amide **7** with compound **9** smoothly occurred under phase transfer conditions to afford product **11**. Its ¹H NMR spectrum contained a double 1 : 1 set of rotamer signals (*cf.* ref. 9).

At the next stage, α-keto ester **12**, a precursor for intramolecular cyclization, was obtained by ozonolytic cleavage of the double bond in compound **11** (Scheme 2). The intramolecular cyclization of compound **12** was carried out in THF at –30 °C by treatment with 1.5 equiv. LiHMDS to afford azetidione **3** whose subsequent Baeyer–Villiger oxidation gave the target derivative **4**. The ¹H NMR spectrum of **3** demonstrates characteristic coupling constants of H(2) and H(3) of the azetidione ring whose small values of 2.2 Hz indicate *trans*-orientation of the substituents.⁹



Scheme 2 Reagents and conditions: i, O₃, CH₂Cl₂, –78 °C, then Me₂S; ii, LiHMDS, THF, –40 to –10 °C; iii, mCPBA, CH₂Cl₂.

The approaches to compound **5** included an attempt to generate a carbanion from **11** by treatment with bases (LDA, NaHMDS or LiHMDS). However, in all cases epoxy amide **8** was obtained as the fragmentation product (Scheme 3) in 49–60% yields. Note that a similar ‘deallylation’ of molecule **11** to give **8** was observed under Ru-catalyzed oxidation (RuCl₃–K₂OsO₄–NaIO₄).



A tentative mechanism of this fragmentation (see Scheme 3) involves a ‘conjugate addition– β -elimination’ passing through a low-temperature metallation with generation of Li(Na)-derivative **A**. Though allylamine **13**, the second product, was not isolated, TLC monitoring of the reaction mixture revealed formation of some new product along with traces of **5** (presumably, non-protonated **13**), which is lost during column chromatography on SiO₂. As one can see, the carbanion centre α -positioned to the N atom in **11** necessary to drive reaction towards **5** is not formed. This fragmentation can be formally considered as removal of the allyl-containing group by the N-deallylation protocol. In literature, Pd-catalyzed N-deallylation of imides¹⁰ and amines,¹¹ Ru-catalyzed isomerizations of N-allylamides followed by oxidation,¹² use of 2-arylallyl moieties at an N atom as readily removable protective groups¹³ were reported.

In summary, we have developed a synthesis of novel azetidinone blocks **3** and **4** and indicated an unexpected deallylation of compound **11** proceeding under the action of strong bases (NaHMDS) or oxidizing systems. The **11** \rightarrow **8** transformation can be regarded as a demonstration that N-‘methyleneacrylated’ amines can be selectively deprotected by treatment with LDA or similar bases at low temperatures.

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Online Supplementary Materials

Supplementary data associated with this article (synthetic procedures and characteristics of products) can be found in the online version at doi: 10.1016/j.mencom.2018.03.005.

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