

## **Influence of the Ni/Co ratio in bimetallic NiCo catalysts on methane conversion into synthesis gas**

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### **Synthesis**

$\text{Ce}(\text{NO}_3)_3 \cdot 6\text{H}_2\text{O}$ ,  $\text{ZrO}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ ,  $\text{Gd}(\text{NO}_3)_3 \cdot 6\text{H}_2\text{O}$ ,  $\text{Ni}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ ,  $\text{Co}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ ,  $\text{TiCl}_4$  (Acros Organics) were used as precursors. Appropriate amounts of salts were dissolved in 500 mL distilled water containing of nitric acid (pH=2) to give total concentrations of metals of 0.045 M. Then, the co-precipitation of hydroxides was carried out by addition of 2.5 M KOH solution up to pH 10 at 30°C under stirring. Ultrasonic processing (10 min, 35 kHz, 150 W, Sapphire UZV-4.0) was used during dissolution of salts in distilled water and after precipitation. The resulting precipitates were filtered, washed with distilled water-ethanol solution ( $\text{H}_2\text{O}/\text{C}_2\text{H}_5\text{OH}=9$  vol.), dried at 150°C for 12 h, and calcined in static air by heating at a rate of 4°C/min from room temperature to 500°C and kept at 500°C for 1 h in a muffle furnace.

### **Characterization**

Powder XRD data were collected at room temperature (Rigaku MiniFlex 600, Japan) with  $\text{CuK}\alpha$  radiation. Specific surface area ( $S_{\text{BET}}$ ) of the powders was measured by a conventional nitrogen adsorption method at 77K (TriStar 3000 Micromeritics). Pore-size distributions were calculated from desorption isotherm, using BJH method. TEM analyses were conducted on an Omega Leo-912AB transmission electron microscope with accelerating voltage of 100 kV. SEM analyses were conducted on a JEOL JSM-6390LA scanning electron microscope with accelerating voltage of 20 kV, equipped energy-dispersive X-ray spectrometer (EDS). Thermal analysis (TG-DSC) was carried out under air flow (20 ml/min) with Netzsch STA449F3. The samples were heated to 900°C at the rate of 20°C/min.

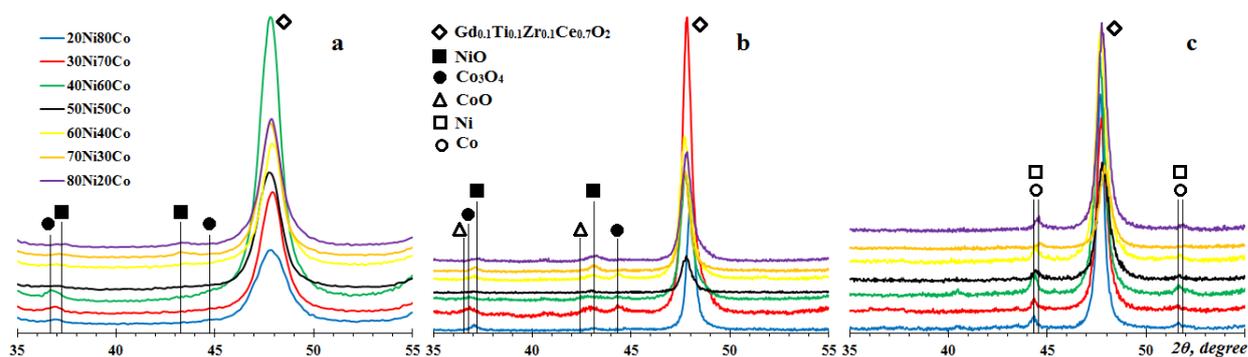
Partial oxidation and dry reforming of methane was performed using a single-pass plug-flow setup including a flow-fixed bed U-shaped quartz reactor (5 mm internal). The reactor was equipped with a pocket for a thermocouple, placed between inlet and outlet tubes. All of the tests were carried out with 0.2 g of catalyst (catalyst pellet particle size was 0.5 mm), placed at the

bottom of U-tube, at atmospheric pressure, using  $\text{CH}_4/\text{O}_2=2$  or  $\text{CH}_4/\text{CO}_2=1$  mixtures without any inert gases dilution. Catalysts were tested in the temperature range of 850-950°C with a GHSV 10 and 12  $\text{L}^1\text{g}^{-1}\text{h}^{-1}$  for POM and DRM, accordingly.

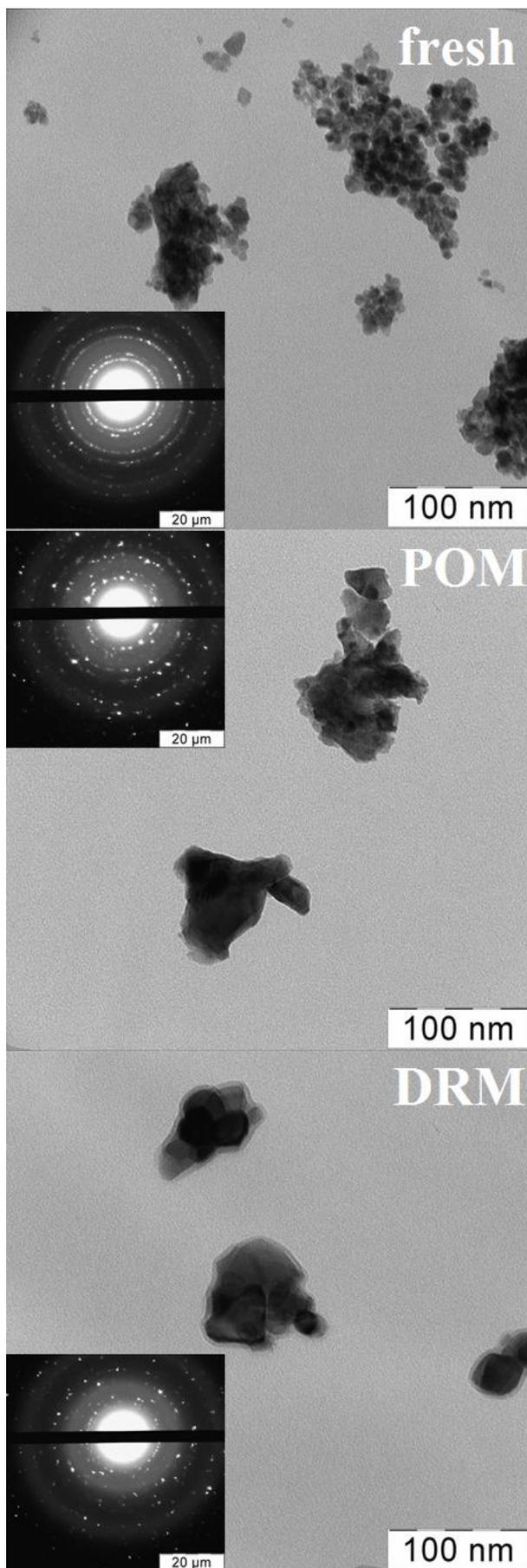
The initial gas mixture and outlet gas were analyzed using gas chromatography. Detailed catalytic test is in previous paper [1].

## References

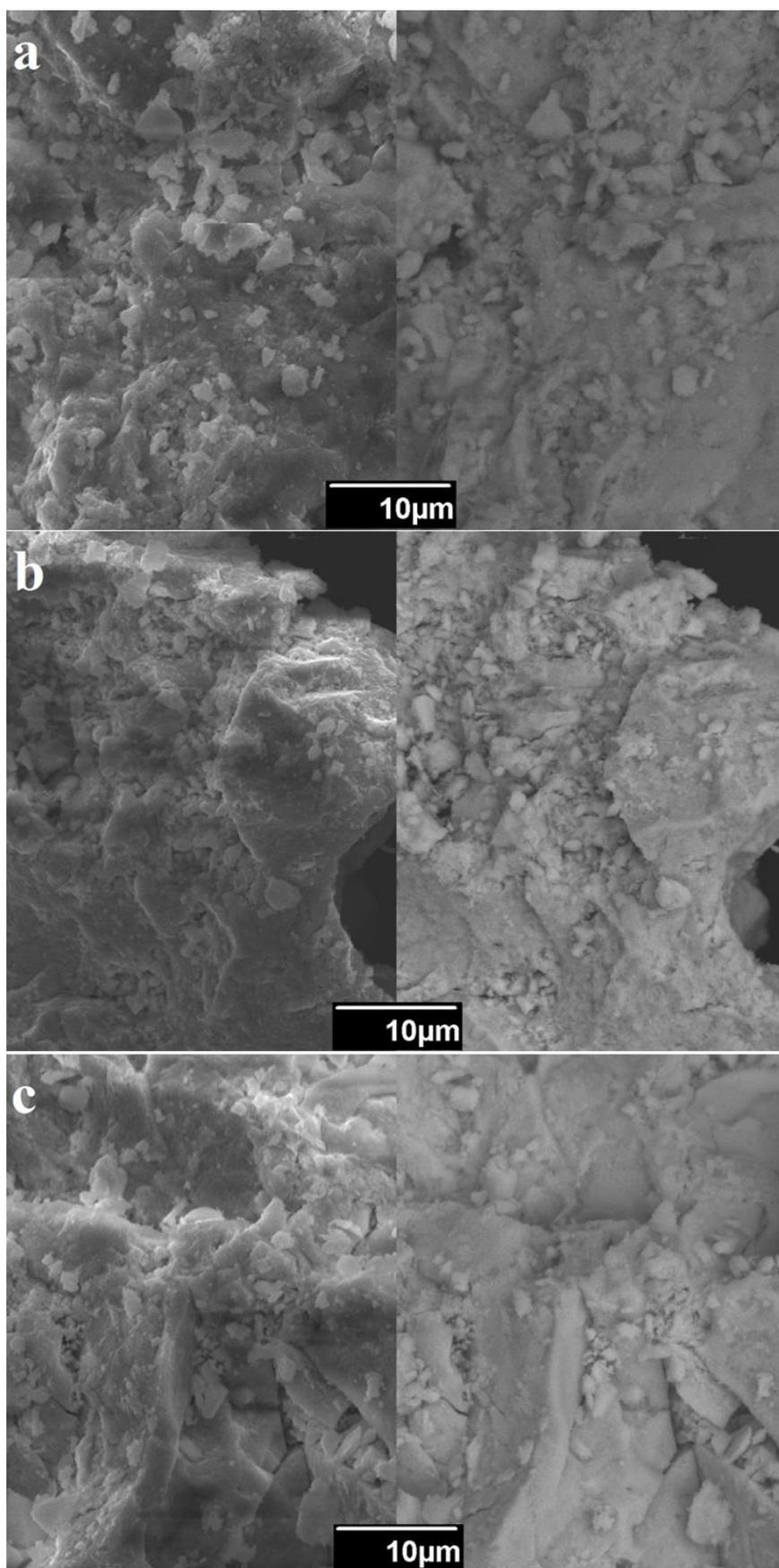
1. I.V. Zagaynov, A.S. Loktev, A.L. Arashanova, V.K. Ivanov, A.G. Dedov and I.I. Moiseev, *Chem. Eng. J.*, 2016, **290**,193.



**Figure S1** XRD patterns of fresh catalysts (a) and catalysts used after POM (b) and DRM (c).



**Figure S2** TEM pictures of fresh (a) and catalyst used after POM (b) and DRM (c) (sample 80Ni/20Co).



**Figure S3** SEM pictures of fresh (a) and catalyst used after POM (b) and DRM (c) (sample 80Ni/20Co) in SE (left) and BSE (right) mods.