

**A new cycloadduct based on quadricyclane and perfluorocyclohexene:
synthesis, metathesis polymerization and gas-transport properties of the
obtained polymer**

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1. Materials

All starting fluorine-containing compounds were purchased from SIA "P&M-Invest" Ltd. THF was distilled for three times, first over NaOH, for the second time over sodium and for the third time over sodium with benzophenone when persisted violet color was observed. Quadricyclane was synthesized as reported previously.¹

2. Methods of characterization

NMR spectra were recorded on a Bruker Avance DRX-500 spectrometer at 500.13 MHz (¹H NMR) and 125.76 MHz (¹³C NMR), at 470.59 MHz for ¹⁹F NMR. Each sample was dissolved in a solvent (CDCl₃ or mixture C₆D₆/C₆F₆ was used due to poor solubility of **2** and hexafluorobenzene (C₆F₆) was applied also as an internal standard in ¹⁹F NMR) up to a concentration of 10%. Mass spectra were recorded on a Thermo Focus DSQ II (ionization energy 70 eV, source temperature 230°C). Differential scanning calorimetry (DSC) was performed on a Mettler TA4000 system at a heating rate 20°C/min. Thermal gravimetric measurements (TGA) were carried out using a Perkin-Elmer TGA-7 instrument.

3. Film casting

The films of the polymers studied were cast from the 2 weight % solution in C₆F₆. The solution was poured into a steal cylinder with a stretched cellophane bottom. The solvent was

allowed to evaporate slowly to yield the desired polymer films. After the films formation, cellophane was detached from the films. Before testing, the films were kept under vacuum until the constant weight is achieved. The thickness of the films was in the range of 50-80 μm .

NMR description of 1.

^1H NMR (CDCl_3 ; δ , ppm): 6.20 br.s (2H), 3.30 br.s (2H), 2.47-2.43 m (2H), 2.02 (d, $^2J=10.1$ Hz, 1H), 1.42 (d, $^2J=10.1$ Hz, 1H). ^{13}C NMR (CDCl_3 ; δ , ppm): 137.98 s, 113.38-112.53 m, 111.39-110.36 m, 109.26-108.23 m, 107.12-106.13 m, 89.08-87.99 m, 86.87-86.00 m, 44.17-44.04 m, 43.56 d ($^2J=3.6$ Hz), 40-60-40.15 m. ^{19}F NMR (CDCl_3 ; δ , ppm): -118.96-(-)119.11 m (1F), -119.55-(-)119.69 m (1F), -124.29-(-)124.49 m (1F), -124.86-(-)125.06 m (1F), -126.02-(-)126.19 m (1F), -126.00-(-)126.79 m (1F), -129.30-(-)129.51 m (1F), -129.87-(-)130.08 m (1F), -198.88 s (2F). m/z (EI): 354, M^+ (1%), 308 (4%), 91 (20%), 66 (100%).

NMR description of 2.

^1H NMR ($\text{C}_6\text{D}_6/\text{C}_6\text{F}_6$; δ , ppm): 5.93-5.34 m (2H), 3.92-3.53 m (0.5 H), 3.51-2.73 m (3.5 H), 2.47-2.10 m (1H), 1.80-1.47 m (1 H). ^{13}C NMR ($\text{C}_6\text{D}_6/\text{C}_6\text{F}_6$; δ , ppm): 132.50-131.13 m, 118.91-117.67 m, 116.75-115.56 m, 114.45-113.53 m, 112.62-111.58 m, 110.95-109.49 m, 109.01-107.39 m, 106.90-105.77 m, 95.15-87.05 m, 47.93-46.33 m, 46.14-44.48 m, 43.27-40.53 m, 36.96-36.59 m. ^{19}F NMR ($\text{C}_6\text{D}_6/\text{C}_6\text{F}_6$; δ , ppm): -120.89-(-)124.65 (2F), -125.06-(-)129.45 (4F), -129.50-(-)132.08 (2F), -181.84-(-)184.47 (2F).

X-ray analysis of **1**

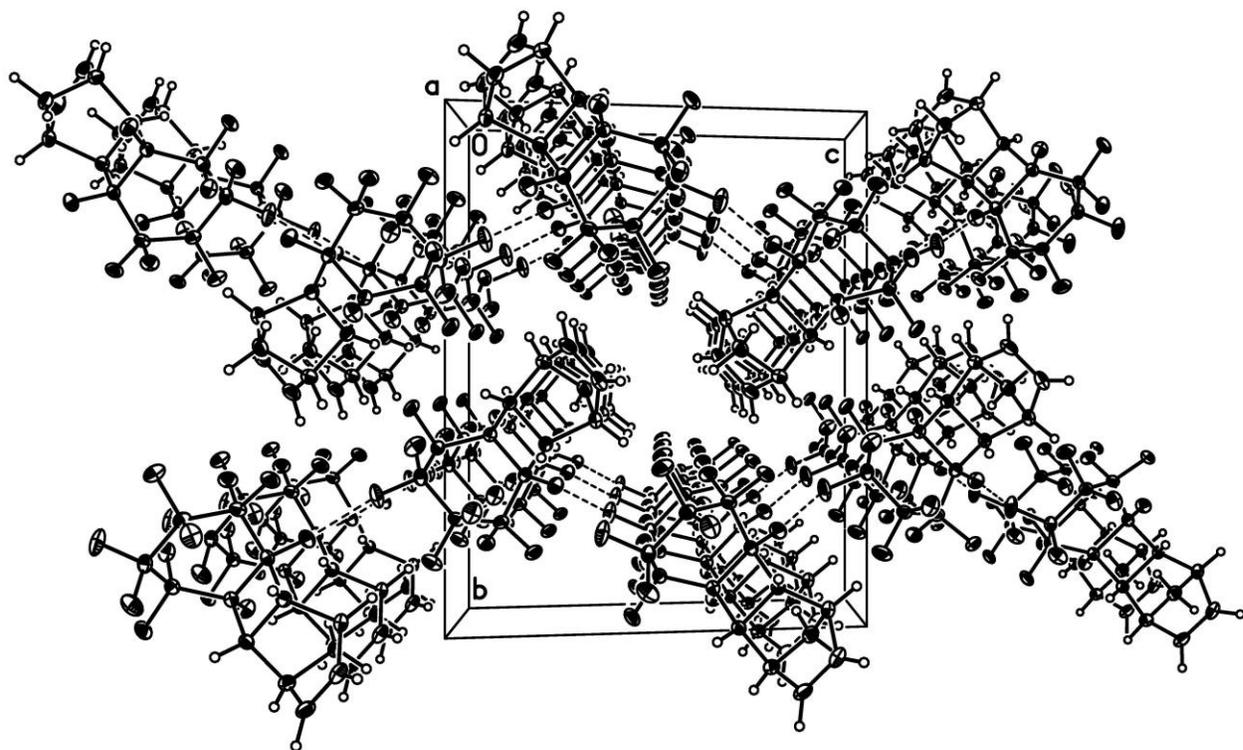


Figure S1 A fragment of the crystal packing of the compound **1** along the crystallographic axis *a*.

Table S1 Crystal data and structure refinement for **1**.

Identification code	I	
Empirical formula	C ₁₃ H ₈ F ₁₀	
Formula weight	354.19	
Temperature	120(2) K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	P 2 ₁ /n	
Unit cell dimensions	a = 6.1614(5) Å	$\alpha = 90^\circ$.
	b = 15.7377(13) Å	$\beta = 100.198(2)^\circ$.
	c = 12.8181(11) Å	$\gamma = 90^\circ$.
Volume	1223.29(18) Å ³	
Z	4	

Density (calculated)	1.923 Mg/m ³
Absorption coefficient	0.219 mm ⁻¹
F(000)	704
Crystal size	0.35 x 0.05 x 0.05 mm ³
Theta range for data collection	2.07 to 29.00°.
Index ranges	-8<=h<=8, -19<=k<=21, -17<=l<=17
Reflections collected	9998
Independent reflections	3255 [R(int) = 0.0253]
Completeness to theta = 29.00°	100.0 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.981 and 0.923
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	3255 / 0 / 208
Goodness-of-fit on F ²	1.014
Final R indices [for 2723 rfln with I>2sigma(I)]	R1 = 0.0354, wR2 = 0.0892
R indices (all data)	R1 = 0.0442, wR2 = 0.0956
Largest diff. peak and hole	0.500 and -0.221 e. Å ⁻³

Table S2 Atomic coordinates (x 10⁴) and equivalent isotropic displacement parameters (Å²x 10³) for **1**. U(eq) is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	U(eq)
C(1)	1934(2)	1907(1)	3051(1)	14(1)
C(2)	2806(2)	1191(1)	3862(1)	14(1)
C(3)	910(2)	602(1)	3341(1)	15(1)
C(4)	1503(2)	-128(1)	2619(1)	19(1)
C(5)	-748(3)	-400(1)	2028(1)	23(1)
C(6)	-1493(3)	208(1)	1332(1)	23(1)
C(7)	245(2)	905(1)	1444(1)	20(1)
C(8)	31(2)	1320(1)	2526(1)	15(1)
C(9)	2344(2)	360(1)	1729(1)	21(1)
C(10)	1283(2)	2728(1)	3549(1)	15(1)
C(11)	-87(2)	2616(1)	4441(1)	17(1)
C(12)	783(2)	1896(1)	5258(1)	17(1)
C(13)	2850(2)	1420(1)	5023(1)	15(1)

F(1)	3405(1)	2146(1)	2415(1)	22(1)
F(2)	4903(1)	916(1)	3806(1)	19(1)
F(3)	3134(1)	3163(1)	3968(1)	21(1)
F(4)	115(1)	3228(1)	2792(1)	22(1)
F(5)	-90(2)	3359(1)	4958(1)	31(1)
F(6)	-2182(1)	2441(1)	3996(1)	25(1)
F(7)	1286(2)	2232(1)	6229(1)	30(1)
F(8)	-843(1)	1326(1)	5263(1)	26(1)
F(9)	4618(1)	1919(1)	5376(1)	19(1)
F(10)	3089(1)	704(1)	5619(1)	22(1)

Table S3 Bond lengths [Å] and angles [°] for **1**.

C(1)-F(1)	1.3748(14)	C(10)-F(4)	1.3531(15)
C(1)-C(10)	1.5261(18)	C(10)-F(3)	1.3567(15)
C(1)-C(8)	1.5493(18)	C(10)-C(11)	1.5469(19)
C(1)-C(2)	1.5620(18)	C(11)-F(6)	1.3436(16)
C(2)-F(2)	1.3763(14)	C(11)-F(5)	1.3442(16)
C(2)-C(13)	1.5264(18)	C(11)-C(12)	1.5713(19)
C(2)-C(3)	1.5471(18)	C(12)-F(7)	1.3382(16)
C(3)-C(4)	1.5590(18)	C(12)-F(8)	1.3448(16)
C(3)-C(8)	1.5695(18)	C(12)-C(13)	1.5526(18)
C(3)-H(3A)	1.0000	C(13)-F(9)	1.3541(15)
C(4)-C(5)	1.520(2)	C(13)-F(10)	1.3541(15)
C(4)-C(9)	1.5392(19)	F(1)-C(1)-C(10)	105.40(10)
C(4)-H(4A)	1.0000	F(1)-C(1)-C(8)	115.68(11)
C(5)-C(6)	1.332(2)	C(10)-C(1)-C(8)	116.85(11)
C(5)-H(5A)	0.9500	F(1)-C(1)-C(2)	114.00(10)
C(6)-C(7)	1.522(2)	C(10)-C(1)-C(2)	114.68(11)
C(6)-H(6A)	0.9500	C(8)-C(1)-C(2)	90.23(9)
C(7)-C(9)	1.541(2)	F(2)-C(2)-C(13)	105.67(10)
C(7)-C(8)	1.5600(19)	F(2)-C(2)-C(3)	115.70(10)
C(7)-H(7A)	1.0000	C(13)-C(2)-C(3)	116.35(10)
C(8)-H(8A)	1.0000	F(2)-C(2)-C(1)	114.01(10)
C(9)-H(9A)	0.9900	C(13)-C(2)-C(1)	115.02(10)
C(9)-H(9B)	0.9900	C(3)-C(2)-C(1)	90.05(10)

C(2)-C(3)-C(4)	117.32(11)	C(4)-C(9)-C(7)	94.35(10)
C(2)-C(3)-C(8)	90.03(9)	C(4)-C(9)-H(9A)	112.9
C(4)-C(3)-C(8)	102.99(10)	C(7)-C(9)-H(9A)	112.9
C(2)-C(3)-H(3A)	114.4	C(4)-C(9)-H(9B)	112.9
C(4)-C(3)-H(3A)	114.4	C(7)-C(9)-H(9B)	112.9
C(8)-C(3)-H(3A)	114.4	H(9A)-C(9)-H(9B)	110.3
C(5)-C(4)-C(9)	99.66(11)	F(4)-C(10)-F(3)	107.05(10)
C(5)-C(4)-C(3)	102.37(11)	F(4)-C(10)-C(1)	109.81(11)
C(9)-C(4)-C(3)	102.56(11)	F(3)-C(10)-C(1)	109.17(10)
C(5)-C(4)-H(4A)	116.6	F(4)-C(10)-C(11)	107.78(11)
C(9)-C(4)-H(4A)	116.6	F(3)-C(10)-C(11)	107.20(11)
C(3)-C(4)-H(4A)	116.6	C(1)-C(10)-C(11)	115.49(11)
C(6)-C(5)-C(4)	107.91(13)	F(6)-C(11)-F(5)	107.59(11)
C(6)-C(5)-H(5A)	126.0	F(6)-C(11)-C(10)	108.61(11)
C(4)-C(5)-H(5A)	126.0	F(5)-C(11)-C(10)	108.26(11)
C(5)-C(6)-C(7)	107.56(13)	F(6)-C(11)-C(12)	108.89(11)
C(5)-C(6)-H(6A)	126.2	F(5)-C(11)-C(12)	109.20(11)
C(7)-C(6)-H(6A)	126.2	C(10)-C(11)-C(12)	114.10(11)
C(6)-C(7)-C(9)	99.65(11)	F(7)-C(12)-F(8)	108.00(11)
C(6)-C(7)-C(8)	102.48(11)	F(7)-C(12)-C(13)	107.98(11)
C(9)-C(7)-C(8)	102.78(11)	F(8)-C(12)-C(13)	108.47(11)
C(6)-C(7)-H(7A)	116.5	F(7)-C(12)-C(11)	109.28(11)
C(9)-C(7)-H(7A)	116.5	F(8)-C(12)-C(11)	108.52(11)
C(8)-C(7)-H(7A)	116.5	C(13)-C(12)-C(11)	114.42(11)
C(1)-C(8)-C(7)	117.32(11)	F(9)-C(13)-F(10)	107.05(10)
C(1)-C(8)-C(3)	89.69(10)	F(9)-C(13)-C(2)	109.71(10)
C(7)-C(8)-C(3)	102.55(10)	F(10)-C(13)-C(2)	109.57(11)
C(1)-C(8)-H(8A)	114.6	F(9)-C(13)-C(12)	107.14(10)
C(7)-C(8)-H(8A)	114.6	F(10)-C(13)-C(12)	107.61(10)
C(3)-C(8)-H(8A)	114.6	C(2)-C(13)-C(12)	115.41(11)

Symmetry transformations used to generate equivalent atoms:

Table S4 Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **1**. The anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2 a^{*2}U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U^{11}	U^{22}	U^{33}	U^{23}	U^{13}	U^{12}
C(1)	14(1)	14(1)	15(1)	2(1)	5(1)	0(1)
C(2)	11(1)	14(1)	16(1)	1(1)	3(1)	1(1)
C(3)	16(1)	14(1)	16(1)	1(1)	4(1)	-2(1)
C(4)	23(1)	14(1)	19(1)	-1(1)	4(1)	1(1)
C(5)	27(1)	20(1)	24(1)	-8(1)	6(1)	-6(1)
C(6)	23(1)	26(1)	20(1)	-9(1)	1(1)	-2(1)
C(7)	25(1)	20(1)	14(1)	-1(1)	3(1)	-1(1)
C(8)	15(1)	15(1)	15(1)	-1(1)	1(1)	-1(1)
C(9)	22(1)	22(1)	20(1)	-5(1)	8(1)	0(1)
C(10)	15(1)	12(1)	18(1)	2(1)	0(1)	-1(1)
C(11)	15(1)	15(1)	20(1)	-3(1)	3(1)	2(1)
C(12)	14(1)	21(1)	15(1)	-1(1)	4(1)	0(1)
C(13)	13(1)	15(1)	15(1)	2(1)	1(1)	1(1)
F(1)	24(1)	22(1)	22(1)	3(1)	12(1)	-5(1)
F(2)	12(1)	23(1)	23(1)	-3(1)	3(1)	5(1)
F(3)	20(1)	17(1)	26(1)	-1(1)	1(1)	-7(1)
F(4)	26(1)	16(1)	22(1)	5(1)	-2(1)	3(1)
F(5)	44(1)	18(1)	33(1)	-8(1)	13(1)	4(1)
F(6)	12(1)	33(1)	28(1)	5(1)	1(1)	4(1)
F(7)	29(1)	45(1)	14(1)	-8(1)	2(1)	11(1)
F(8)	17(1)	28(1)	35(1)	9(1)	10(1)	-2(1)
F(9)	13(1)	25(1)	19(1)	-3(1)	-1(1)	-2(1)
F(10)	27(1)	21(1)	18(1)	8(1)	2(1)	5(1)

Table S5 Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^{-3}$) for **1**.

	x	y	z	U(eq)
H(3A)	-137	434	3819	18
H(4A)	2481	-589	2977	23
H(5A)	-1493	-913	2134	28
H(6A)	-2861	206	854	28
H(7A)	184	1300	827	23
H(8A)	-1480	1532	2571	18
H(9A)	2598	-13	1141	25
H(9B)	3677	702	1991	25

Table S6 Torsion angles [$^\circ$] for **1**.

F(1)-C(1)-C(2)-F(2)	-0.06(15)	C(3)-C(4)-C(5)-C(6)	72.00(14)
C(10)-C(1)-C(2)-F(2)	-121.70(12)	C(4)-C(5)-C(6)-C(7)	-0.13(16)
C(8)-C(1)-C(2)-F(2)	118.37(11)	C(5)-C(6)-C(7)-C(9)	33.41(14)
F(1)-C(1)-C(2)-C(13)	122.22(12)	C(5)-C(6)-C(7)-C(8)	-72.10(14)
C(10)-C(1)-C(2)-C(13)	0.58(15)	F(1)-C(1)-C(8)-C(7)	12.95(16)
C(8)-C(1)-C(2)-C(13)	-119.35(11)	C(10)-C(1)-C(8)-C(7)	137.96(12)
F(1)-C(1)-C(2)-C(3)	-118.43(11)	C(2)-C(1)-C(8)-C(7)	-104.00(12)
C(10)-C(1)-C(2)-C(3)	119.93(11)	F(1)-C(1)-C(8)-C(3)	116.95(11)
C(8)-C(1)-C(2)-C(3)	0.01(9)	C(10)-C(1)-C(8)-C(3)	-118.04(11)
F(2)-C(2)-C(3)-C(4)	-12.20(16)	C(2)-C(1)-C(8)-C(3)	-0.01(9)
C(13)-C(2)-C(3)-C(4)	-137.14(12)	C(6)-C(7)-C(8)-C(1)	164.70(11)
C(1)-C(2)-C(3)-C(4)	104.67(12)	C(9)-C(7)-C(8)-C(1)	61.63(14)
F(2)-C(2)-C(3)-C(8)	-116.88(11)	C(6)-C(7)-C(8)-C(3)	68.46(13)
C(13)-C(2)-C(3)-C(8)	118.18(11)	C(9)-C(7)-C(8)-C(3)	-34.62(13)
C(1)-C(2)-C(3)-C(8)	-0.01(9)	C(2)-C(3)-C(8)-C(1)	0.01(9)
C(2)-C(3)-C(4)-C(5)	-165.02(11)	C(4)-C(3)-C(8)-C(1)	-118.10(10)
C(8)-C(3)-C(4)-C(5)	-68.14(12)	C(2)-C(3)-C(8)-C(7)	117.98(10)
C(2)-C(3)-C(4)-C(9)	-62.02(14)	C(4)-C(3)-C(8)-C(7)	-0.13(12)
C(8)-C(3)-C(4)-C(9)	34.86(13)	C(5)-C(4)-C(9)-C(7)	50.30(12)
C(9)-C(4)-C(5)-C(6)	-33.25(14)	C(3)-C(4)-C(9)-C(7)	-54.80(12)

C(6)-C(7)-C(9)-C(4)	-50.40(12)	F(5)-C(11)-C(12)-F(8)	117.03(12)
C(8)-C(7)-C(9)-C(4)	54.86(12)	C(10)-C(11)-C(12)-F(8)	-121.69(12)
F(1)-C(1)-C(10)-F(4)	67.21(13)	F(6)-C(11)-C(12)-C(13)	121.08(12)
C(8)-C(1)-C(10)-F(4)	-62.82(14)	F(5)-C(11)-C(12)-C(13)	-121.70(12)
C(2)-C(1)-C(10)-F(4)	-166.57(10)	C(10)-C(11)-C(12)-C(13)	-0.42(16)
F(1)-C(1)-C(10)-F(3)	-49.87(13)	F(2)-C(2)-C(13)-F(9)	48.66(13)
C(8)-C(1)-C(10)-F(3)	-179.90(10)	C(3)-C(2)-C(13)-F(9)	178.56(10)
C(2)-C(1)-C(10)-F(3)	76.36(13)	C(1)-C(2)-C(13)-F(9)	-78.02(13)
F(1)-C(1)-C(10)-C(11)	-170.71(11)	F(2)-C(2)-C(13)-F(10)	-68.60(12)
C(8)-C(1)-C(10)-C(11)	59.26(15)	C(3)-C(2)-C(13)-F(10)	61.30(14)
C(2)-C(1)-C(10)-C(11)	-44.48(15)	C(1)-C(2)-C(13)-F(10)	164.73(10)
F(4)-C(10)-C(11)-F(6)	45.61(14)	F(2)-C(2)-C(13)-C(12)	169.77(10)
F(3)-C(10)-C(11)-F(6)	160.55(10)	C(3)-C(2)-C(13)-C(12)	-60.33(15)
C(1)-C(10)-C(11)-F(6)	-77.55(14)	C(1)-C(2)-C(13)-C(12)	43.10(15)
F(4)-C(10)-C(11)-F(5)	-70.93(13)	F(7)-C(12)-C(13)-F(9)	-42.37(14)
F(3)-C(10)-C(11)-F(5)	44.00(14)	F(8)-C(12)-C(13)-F(9)	-159.18(10)
C(1)-C(10)-C(11)-F(5)	165.90(11)	C(11)-C(12)-C(13)-F(9)	79.53(13)
F(4)-C(10)-C(11)-C(12)	167.27(11)	F(7)-C(12)-C(13)-F(10)	72.45(13)
F(3)-C(10)-C(11)-C(12)	-77.80(13)	F(8)-C(12)-C(13)-F(10)	-44.35(14)
C(1)-C(10)-C(11)-C(12)	44.10(16)	C(11)-C(12)-C(13)-F(10)	-165.64(11)
F(6)-C(11)-C(12)-F(7)	-117.73(12)	F(7)-C(12)-C(13)-C(2)	-164.87(11)
F(5)-C(11)-C(12)-F(7)	-0.51(15)	F(8)-C(12)-C(13)-C(2)	78.32(14)
C(10)-C(11)-C(12)-F(7)	120.77(12)	C(11)-C(12)-C(13)-C(2)	-42.96(15)
F(6)-C(11)-C(12)-F(8)	-0.19(14)		

Symmetry transformations used to generate equivalent atoms:

1. C. D. Smith, *Org. Syn.*, 1971, **51**, 133.