

Separation of magnolol from an extract of *Cortex Magnoliae Officinalis* by selective clathration

Xiao-Qing Cai,^a Kui Wu Wang^b and Zhi Min Jin^{*c}

^a College of Chemistry and Materials Engineering, Wenzhou University, Wenzhou 325035, P. R. China

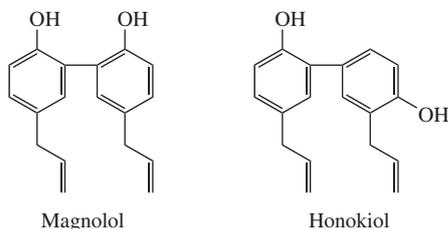
^b College of Food Science, Biotechnology and Environmental Engineering, Zhejiang Gongshang University, Hangzhou 310035, P. R. China

^c College of Pharmaceutical Sciences, Zhejiang University of Technology, Hangzhou 310032, P. R. China.
Fax: +86 571 8598 4972; e-mail: zimichem@sina.com

DOI: 10.1016/j.mencom.2011.09.014

The selective clathration of magnolol with diethylenetriamine (2:1 molar ratio) renders an amenable method for separating magnolol from an extract of *Cortex Magnoliae Officinalis* (purity, 98%; yield, 76%).

The extract of *Cortex Magnoliae Officinalis*, a well known Chinese traditional medicine,^{1,2} contains magnolol and its isomer honokiol as the main bioactive compounds. Magnolol shows various bioactivities, such as hydroxyl radical scavenging³ and antioxidative,⁴ anxiolytic,⁵ neurotrophic,⁶ antimicrobial,⁷ and antifungal⁸ effects. The selective clathration of host–guest molecules has drawn increasing attention in separating natural products, such as neoabietic acid,⁹ α/β -ionone¹⁰ and honokiol.¹¹ Here we report on separation of magnolol as a host molecule from an extract by selective clathration using diethylenetriamine (DETA) as a guest molecule.



HPLC analysis showed that the extract used consisted of magnolol (45.7%), honokiol (49.1%) and other unknown substances (5.2%). The extract (10.0 g) and DETA (1.5 g) were dissolved in 90% ethanol (25 ml) by heating to 80 °C, and yellow crystals were formed upon keeping the solution for 4 h at temperatures from –10 to 10 °C. The crystals[†] of the complex of

[†] Complex of magnolol with diethylenetriamine (2:1 molar ratio): yellow crystals, mp 87 °C. IR (KBr, ν/cm^{-1}): 3344, 2926, 2802, 2748, 2548, 2502, 1640, 1460, 1340, 1238, 1112, 996, 567. ¹H NMR (500 MHz, DMSO-*d*₆) δ : 2.59 (d, 2H, DETA, *J* 7.0 Hz), 2.69 (d, 2H, DETA, *J* 7.0 Hz), 3.28 (d, 4H, H-7, H-7', *J* 6.7 Hz), 4.98–5.08 (m, 4H, H-9, H-9'), 5.90–6.00 (m, 2H, H-8, H-8'), 6.63 (d, 2H, H-3, H-3', *J* 8.3 Hz), 6.85 (d, 2H, H-6, H-6', *J* 2.2 Hz), 6.99 (dd, 2H, H-4, H-4', *J*₁ 8.3 Hz, *J*₂ 2.2 Hz).

Crystal data: 2(C₁₈H₁₈O₂)·C₄H₁₃N₃·2H₂O, triclinic, space group $\bar{P}1$, *a* = 9.7322(11), *b* = 10.7929(13) and *c* = 18.962(2) Å, α = 85.600(2)°, β = 79.890(3)°, γ = 87.426(3)°, *V* = 1954.03(4) Å³, *Z* = 2. A Bruker APEX area-detector diffractometer was used at room temperature in measuring reflections from a crystal with a size of 0.33×0.39×0.44 mm. Among 10 159 total reflections, 5177 were unique ones. The structure was solved and refined to *R* = 0.0703 (*wR* = 0.1588) and *GOOF* = 1.155.

CCDC 765546 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif. For details, see 'Notice to Authors', *Mendeleev Commun.*, Issue 1, 2011.

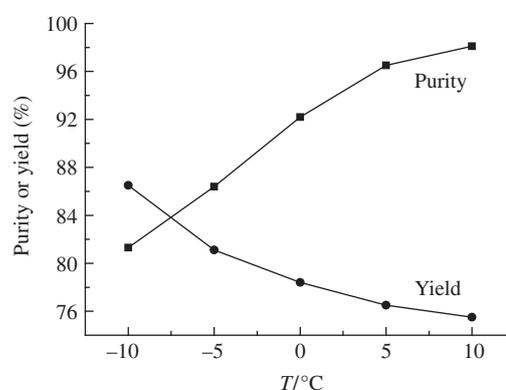


Figure 1 Effects of temperature on the yield and purity of magnolol.

magnolol, DETA and water (2:1:2) were filtered off and recrystallized from ethanol. Afterward, the crystals were gathered, then dissociated by dilute HCl, and warmed to 50 °C for two days, giving powder magnolol[‡] in a quantitative yield.

By fixing the proportion of raw materials, meanwhile changing the temperature, magnolol was obtained in 81.3–98.1% purity (HPLC) and 86.7–75.5% yield (Figure 1).

According to X-ray diffraction (Figure 2),[†] the complex contains two magnolol entities (a neutral molecule and an anion), one DETA cation (HDETA) and two water molecules; they are linked by N(3)–H...O(3), O(1)–H...O(6), N(2)–H...O(5) and O(6)–H...O(4) hydrogen bonds. The bond lengths of C(1)–O(1) [1.353(3) Å], C(19)–O(3) [1.355(3) Å] and C(28)–O(4) [1.341(3) Å] fall in the range of phenolic C–O bonds, while the length of C(10)–O(2) [1.203(3) Å] is different, indicating a phenolate C–O bond. The dihedral angles between the benzene rings of the two magnolol molecules are 35.6° and 36.7°, respectively. As the skeleton of HDETA is flexible, it may adopt different conformations resembling bended to straight lines in different extensions. In this study, HDETA demonstrated a curvable conformation, which is similar to that observed in the crystal of HDETA dichloride.¹²

[‡] Magnolol: yellowish powder, mp 103 °C. IR (KBr, ν/cm^{-1}): 3158, 1638, 1610, 1496, 1439, 1410, 1384, 1228, 1135, 992, 903, 820, 783, 731, 663, 563, 529. ¹H NMR (500 MHz, CDCl₃) δ : 3.37 (d, 4H, H-7, H-7', *J* 6.7 Hz), 5.06–5.12 (m, 4H, H-9, H-9'), 5.93–6.01 (m, 2H, H-8, H-8'), 6.95 (d, 2H, H-3, H-3', *J* 8.3 Hz), 7.08 (d, 2H, H-6, H-6', *J* 2.2 Hz), 7.14 (dd, 2H, H-4, H-4', *J*₁ 8.3 Hz, *J*₂ 2.2 Hz).

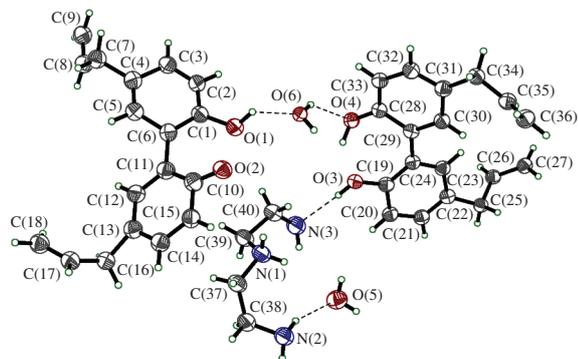


Figure 2 Molecular structure of the complex of magnolol, diethylenetriamine and water (2:1:2). Selected bond lengths (Å) and angles (°): N(1)–C(37) 1.426(4), N(2)–C(38) 1.455(4), N(1)–C(39) 1.431(4), N(3)–C(40) 1.450(4), O(1)–C(1) 1.353(3), O(2)–C(10) 1.203(3), C(4)–C(7) 1.507(4), C(6)–C(11) 1.475(4), C(7)–C(8) 1.450(4), C(8)–C(9) 1.346(4), C(16)–C(17) 1.460(4), C(17)–C(18) 1.357(4); C(8)–C(7)–C(4) 108.5(3), C(9)–C(8)–C(7) 117.6(3), C(37)–N(1)–C(39) 103.6(2); O(1)···O(6) 2.451, O(1)–H···O(6) 113.6, O(6)···O(4) 2.825(2), O(6)–H(6B)···O(4) 135.1, O(3)···N(3) 2.843, O(3)–H···N(3) 142.0, N(2)···O(5) 2.981, N(2)–H···O(5) 158.4, N(3)···O(4)($-x+1, -y, -z+1$) 2.732(3), N(3)–H···O(3)($-x+1, -y, -z+1$) 150.4.

As shown in Figure 3, the crystal structure is characterized by a layered arrangement. The hydrophilic entities including the magnolol hydroxyl, HDETA and water molecules are placed in a

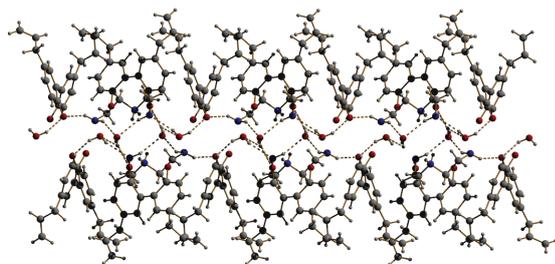


Figure 3 The layered structure of magnolol with HDETA viewed down along the *a* axis. Hydrogen bonds are illustrated by dashed lines.

layer, while the hydrophobic allyl group of magnolol is in another one. Water molecules play an important role in filling the supramolecular hole between the host and guest molecules, and in pasting two magnolol entities. As a comparison, in the crystal of a complex of honokiol with triethylenediamine, honokiol is bound with triethylenediamine, forming a spiral chain of hydrogen bonds.¹¹

It is easily predicted that both magnolol and honokiol molecules can form a supramolecular complex with DETA by N–H···O and/or O–H···N hydrogen bonds. Generally, the combination of DETA with magnolol in a crystal is preferable to that of honokiol.

References

- 1 E. Tachikawa, M. Takahashi and T. Kashimoto, *Biochem. Pharm.*, 2000, **60**, 433.
- 2 Y. Fukuyama, K. Nakade, Y. Minoshima, R. Yokoyama, H. F. Zhai and Y. Mitsumoto, *Bioorg. Med. Chem. Lett.*, 2002, **12**, 1163.
- 3 C. Y. Li, Y. Wang and M. K. Hu, *Bioorg. Med. Chem.*, 2003, **11**, 3665.
- 4 Y. H. Chen, F. Y. Lin, P. L. Liu, Y. T. Huang, J. H. Chiu, Y. C. Chang, K. M. Man, C. Y. Hong, Y. Y. Ho and M. T. Lai, *Arch. Pharm. Res.*, 2009, **32**, 221.
- 5 M. Fujita, H. Itokawa and Y. Sashida, *Chem. Pharm. Bull.*, 1972, **20**, 212.
- 6 N. Matsui, H. Nakashima, Y. Ushio, T. Tada, A. Shirono, Y. Fukuyama, K. Nakade, H. Zhai, Y. Yasui, N. Fukuishi, R. Akagi and M. Akagi, *Biol. Pharm. Bull.*, 2005, **28**, 1762.
- 7 H. Kuribara, E. Kishi, M. Kimura, S. T. Weintraub and Y. Maruyama, *Pharm. Biochem. Behavior*, 2000, **67**, 597.
- 8 Y. Fukuyama, K. Nakade, Y. Minoshima, R. Yokoyama, H. F. Zhai and Y. Mitsumoto, *Bioorg. Med. Chem. Lett.*, 2002, **12**, 1163.
- 9 Z. M. Jin, Y. J. Pan, J. G. Liu and D. J. Xu, *J. Chem. Crystallogr.*, 2000, **30**, 195.
- 10 F. Toda, K. Tanaka and T. Fujiwara, *Angew. Chem., Int. Ed. Engl.*, 1990, **29**, 662.
- 11 Z. M. Jin, W. Fu, Y. J. Pan, J. W. Zou and M. L. Hu, *J. Inclusion Phenom. Macrocyclic Chem.*, 2005, **51**, 225.
- 12 C. A. Ilioudis, K. S. B. Hancock, D. G. Georganopoulou and J. W. Steed, *New J. Chem. (Nouv. J. Chim.)*, 2000, **24**, 787.

Received: 18th March 2011; Com. 11/3700