

## Phosphorus-bridged calixarene phosphites: dramatic influence of a *tert*-butyl group at the upper rim of the macrocycle upon anion binding

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### Experimental

2-H-5,15-Di(*tert*-butyl)-7,13-dimethyl-2-oxodibenzo[d,g]-1,3,2-dioxaphosphocin **2**. Bis(1-hydroxy-4-methyl-6-*tert*-butylphenyl)methane **1** (0.50 g, 1.46 mmol) and ethylene chlorophosphite (0.26 ml, 2.92 mmol) in 10 ml of *p*-bromotoluene in an argon atmosphere were heated for 6 h at 150 °C. The solvent was evaporated *in vacuo*. A pure product **2** (0.57 g, 96%) was obtained by recrystallization from *n*-octane. <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ: 7.31 (d, 1H, PH, <sup>1</sup>J<sub>PH</sub> 730.9 Hz), 7.06 (s, 2H, H<sub>Ar</sub>), 7.03 (s, 2H, H<sub>Ar</sub>), 4.42 (d, 1H, Ar-CH<sub>2</sub>-Ar, <sup>2</sup>J<sub>HH</sub> 10.0 Hz), 3.65 (d, 1H, Ar-CH<sub>2</sub>-Ar, <sup>2</sup>J<sub>HH</sub> 10.0 Hz), 2.30 (s, 6H, Ar-Me), 1.40 (s, 18H, Ar-CMe<sub>3</sub>). <sup>31</sup>P NMR (CDCl<sub>3</sub>) δ: 0.5 (d, <sup>1</sup>J<sub>PH</sub> 730.9 Hz). Found (%): C, 70.93; H, 8.24; P, 7.85. Calc. for C<sub>23</sub>H<sub>31</sub>O<sub>3</sub>P (%): C, 71.48; H, 8.08; P, 8.01.

Proximal bridged *p*-*tert*-butylcalix[4]arene bis(hydrogen phosphite) **5**. Method A: *p*-*tert*-Butylcalix[4]arene **4** (1.00 g, 1.54 mmol) and ethylene chlorophosphite (0.55 ml, 6.16 mmol) in 20 ml of *p*-bromotoluene in an argon atmosphere were heated for 6 h at 150 °C. The solvent was evaporated *in vacuo*. A pure product **5** (0.90 g, 82%, mp 293 °C) was obtained by recrystallization from chloroform-tetrachloromethane. Method B: *p*-*tert*-Butylcalix[4]arene **4** (1.00 g, 1.54 mmol) and ethylene chlorophosphite (0.27 ml, 3.08 mmol) in 20 ml of *p*-bromotoluene in an argon atmosphere were heated for 6 h at 150 °C. The solvent was evaporated *in vacuo*. Yield of product **5**, 1.10 g (100%), mp 293 °C. IR (KBr, ν/cm<sup>-1</sup>): 931 [O-P(O)-O], 1193 (P-O-Ph), 1278 (P=O), 2458 (P-H). <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ: 7.26 (d, 4H, H<sub>Ar</sub>, <sup>4</sup>J<sub>HH</sub> 2.4 Hz), 7.18 (dd, 4H, H<sub>Ar</sub>, <sup>4</sup>J<sub>HH</sub> 2.4 Hz, <sup>5</sup>J<sub>PH</sub> 1.5 Hz), 5.30 (d, 1H, PH, <sup>1</sup>J<sub>PH</sub> 779.3 Hz), 4.47 (d, 2H, Ar-CH<sub>2</sub>-Ar, <sup>2</sup>J<sub>HH</sub> 14.9 Hz), 4.12 (s, 4H, Ar-CH<sub>2</sub>-Ar), 3.59 (d, 2H, Ar-CH<sub>2</sub>-Ar, <sup>2</sup>J<sub>HH</sub> 14.9 Hz), 1.32 (s, 36H, Ar-CMe<sub>3</sub>). <sup>31</sup>P NMR (CDCl<sub>3</sub>) δ: -0.4 (d, <sup>1</sup>J<sub>PH</sub> 779.3 Hz). MS (CI), *m/z*: 740.6 (M<sup>+</sup>). Found (%): C, 70.93; H, 7.81; P, 9.00. Calc. for C<sub>44</sub>H<sub>54</sub>O<sub>6</sub>P<sub>2</sub> (%): C, 71.33; H, 7.35; P, 8.36.

*Proximal bridged calix[4]arene bis(hydrogen phosphite) 7.* Calix[4]arene **6** (1.00 g, 2.35 mmol) and ethylene chlorophosphite (0.84 ml, 9.40 mmol) in 20 ml of *p*-bromotoluene in an argon atmosphere were heated for 6 h at 150 °C. The solvent was evaporated *in vacuo* and extracted by a small amount of chloroform. Undissolved powder was filtered off, washed with hot chloroform and dried to give 0.73 g (62%) of product **7** (mp 353 °C). IR (KBr,  $\nu/\text{cm}^{-1}$ ): 934 [O–P(O)–O], 1197 (P–O–Ph), 1270 (P=O), 2457 (P–H).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$ : 7.11–7.24 (m, 12H,  $\text{H}_{\text{Ar}}$ ), 5.42 (d, 1H, PH,  $^1J_{\text{PH}}$  783.8 Hz), 4.53 (d, 2H, Ar– $\text{CH}_2$ –Ar,  $^1J_{\text{HH}}$  15.4 Hz), 4.16 (s, 4H, Ar– $\text{CH}_2$ –Ar), 3.64 (d, 2H, Ar– $\text{CH}_2$ –Ar,  $^1J_{\text{HH}}$  15.4 Hz).  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$ : –0.3 (d,  $^1J_{\text{PH}}$  783.8 Hz). MS (CI),  $m/z$ : 515.9 ( $\text{M}^+$ ). Found (%): C, 65.61; H, 4.44; P, 11.58. Calc. for  $\text{C}_{28}\text{H}_{22}\text{O}_6\text{P}_2$  (%): C, 65.12; H, 4.29; P, 12.00.